FISFVIFR

Contents lists available at ScienceDirect

## **Cement & Concrete Composites**

journal homepage: www.elsevier.com/locate/cemconcomp



# Characterizing the nano and micro structure of concrete to improve its durability

P.J.M. Monteiro a,\*, A.P. Kirchheim , S. Chae , P. Fischer , A.A. MacDowell , E. Schaible , H.R. Wenk

- <sup>a</sup> Department of Civil and Environmental Engineering, University of California at Berkeley, USA
- <sup>b</sup> Center for X-ray Optics, Lawrence Berkeley National Laboratory, Berkeley, CA, USA
- <sup>c</sup> Advanced Light Source, Lawrence Berkeley National Laboratory, USA
- <sup>d</sup> Earth and Planetary Science, University of California at Berkeley, USA

#### ARTICLE INFO

Article history:
Received 13 September 2008
Received in revised form 15 December 2008
Accepted 17 December 2008
Available online 25 December 2008

Keywords: Concrete Durability Microscopy Neutron Tomography X-ray

#### ABSTRACT

New and advanced methodologies have been developed to characterize the nano and microstructure of cement paste and concrete exposed to aggressive environments. High resolution full-field soft X-ray imaging in the water window is providing new insight on the nano scale of the cement hydration process, which leads to a nano-optimization of cement-based systems. Hard X-ray microtomography images of ice inside cement paste and cracking caused by the alkali-silica reaction (ASR) enables three-dimensional structural identification. The potential of neutron diffraction to determine reactive aggregates by measuring their residual strains and preferred orientation is studied. Results of experiments using these tools are shown on this paper.

© 2009 Elsevier Ltd. All rights reserved.

#### 1. Introduction

Classical studies of the microstructure of cement paste and concrete have been largely done using electron microscopy. Transmission electron microscopy has been used to investigate the complex nanostructure of calcium silicate hydrates, and scanning electron microscopy (both in secondary and backscattered mode) is frequently used to characterize the phase distributions and assess damage of the material exposed to aggressive environments. Both these techniques normally require high vacuum which can modify or even damage the fragile hydration products. The existence of high vacuum also prevents the in situ imaging of the hydration reactions. X-ray based imaging techniques, which are available at large scale synchrotron radiation facilities, have the potential to revolutionize the current understanding of hydration mechanisms of cements. Neutron scattering has been used to study the hydration of cement minerals and this paper shows how neutron diffraction can be employed to measure the degree of deformation and reactivity in granitic rocks, this being critical information in the study of alkali silica reaction.

Recent research using the full-field soft transmission X-ray microscope XM-1 located at beamline 6.1.2 at the Advanced Light Source (ALS) in Berkeley, California, and operated by the Center for

X-ray Optics at Lawrence Berkeley National Laboratory since 1994, is described below. Originally designed for the observation of biological cells in their natural, wet state at atmospheric pressure, this microscope has wide application in nano-sciences, such as nanomagnetism and nano-materials sciences. Because this technique allows high resolution soft X-ray imaging of cement in aqueous solution and of the early formation of the hydration products, since 1998 numerous studies on cementitious materials have been carried out (see Refs. [1–11]).

Characterizing the three-dimensional network of microcracks using non-destructive techniques has proved challenging. The traditional method of first impregnating the concrete and then cutting and polishing the sample may introduce artifacts. In addition, all observations are limited to whatever can be determined from the surface of the sample; any three-dimensional analysis is impossible.

Microtomography has produced insightful information on the topology of cracks in the cement paste matrix and how they are deflected by the dispersed hard inclusions. This study reports on current research using the Hard X-ray Microtomography (XMT) located at beamline 8.3.2 at the ALS, which analyzes images obtained of cement paste exposed to freezing conditions and of three-dimensional cracks forming in mortar, as a result of the alkali-aggregate reaction.

Neutron scattering can provide unique information on potentially reactive aggregates for concrete. In many rocks, minerals exhibit a preferred orientation pattern, and this is expressed in

<sup>\*</sup> Corresponding author. Address: Department of Civil Engineering, University of California, 721 Davis Hall, Berkeley, CA 94720, USA. Tel.:+1 510 643 82 51. E-mail address: monteiro@ce.berkeley.edu (P.J.M. Monteiro).

anisotropy of physical properties. It is recognized that the deformation state, grain size, and development of foliation in granitic rocks used as aggregate in concrete influence the alkali–silica reaction. By performing an analysis of preferred orientation in conjunction with expansion tests, Monteiro et al. [12] showed that there is a quantitative relationship between the degree of deformation and reactivity in granitic rocks. In this paper, we summarize our ongoing microstructural characterization of reactive rocks by neutron diffraction and transmission electron microscopy (TEM), observing dislocations and determining dislocation densities of quartz grains in granitic rocks of different deformation states.

#### 2. Soft X-ray microscopy

Soft X-ray microscopy uses advances in nanotechnology to provide high quality X-ray optical components. Fresnel zone plates, i.e., circular gratings with a radially increasing line density, are used to focus the X-rays, which constitute the basis for performing X-ray microscopy because with a refractive index close to unity, conventional lenses cannot be applied. State-of-the-art Fresnel zone plate optics have now achieved a 15 nm-spatial resolution [13,14]. The optical design for the soft X-ray microscope used here is an X-ray analogue to a conventional light microscope, whereby Fresnel zone plates are used for the condenser and objective lenses (see Fig. 1).

Because the wavelength is dependent on its focal length (together with a pinhole close to the specimen as a monochromator with a spectral resolution  $E/\Delta E \sim 500$ ), X-rays emitted from a bending magnet source propagate to a condenser zone plate, which provides a partially incoherent hollow-cone illumination of the sample. The current condenser zone plate has an outer diameter of about 9 mm, about 50,000 rings, and an outermost zone width  $\Delta r$  of about 40 nm. The X-rays penetrating the specimen are projected onto an X-ray sensitive CCD by a micro zone plate, which is positioned at its focal length downstream of the sample; this creates a full-field image. The backside-illuminated, thinned CCD has  $2048 \times 2048$  pixels, with a physical pixel size of 13.5  $\mu m$ . Because of the high flux of synchrotron radiation, typical exposure times per image are only a few seconds, covering a field of view of about 10 µm diameter at a magnification of about 2500. The micro zone plate, i.e., the high resolution objective lens, has a diameter of about 30–60 μm, typically about 500 rings, and an outermost zone width  $\Delta r$  down to 15 nm, a critical factor in determining the spatial resolution possible using soft X-ray microscopy.

Fig. 2 shows the main steps necessary in preparing the samples to be subjected to soft X-ray microscopy. The samples must be highly diluted for transmission of the soft X-rays. In order to retard the dissolution of the cement minerals in the highly diluted samples, and to provide calcium and sulfate ions for the reactions, a solution saturated with Ca(OH)2 and CaSO4·2H2O was used as the liquid media (pH = 12.3). The solution was prepared using freshly boiled de-ionized water inside a glove bag filled with nitrogen gas to avoid carbonation. Teflon and polyethylene flasks were used to prepare the solution to avoid alkali-silica reaction. The solid particles were mixed in the solution for 60 s inside a test tube, and approximately 2 ml of the mixture was centrifuged for 15 s. A small droplet – around  $2 \mu l$  – was taken from the supernatant with a micropipette and squeezed between two silicon nitride windows in a metallic sample holder, which was then placed in the microscope for examination.

#### 2.1. Images and results

Shrinkage of concrete on drying frequently leads to cracking. Such premature deterioration affects durability, integrity and long-term service of concrete structure. Recognized as a serious problem in concrete design and construction practice, especially in regard to pavements, floors, and relatively thin structural members, one solution is to use expansive cements (e.g., Type K cement, which contains  $C_4A_3\bar{S}$  as the principal source of the reactive aluminate) or expansive admixtures [15]. Expansive cements and admixtures, unlike Portland cement, expand during the early hydration period after setting. The expansion is a consequence of the formation of ettringite (calcium sulfoaluminate hydrate) in considerable quantities during the first few days of curing. Tests conducted by past researches containing expansive admixtures in concrete have typically focused on stress development, critical cracking condition of concrete, and creep. This paper aim to show how this expansive admixtures works along and in combination with portland cement during hydration. Two separate research projects using soft X-ray microscopy to study this phenomenon are reported therein and preliminary results presented below.

In the first series of tests, the hydration of the  $C_4A_3\bar{S}$  in a  $Ca(OH)_2 + CaSO_4 \cdot 2H_2O$  saturated solution was analyzed. The solution/solid materials ratio was 50 ml/g. The morphology of this compound in absence and in presence of 10% of  $Ca(OH)_2$  is presented over the time. Fig. 3a and b shows a sequence of large ribbons arranged in a fan-like configuration or cross shape for a

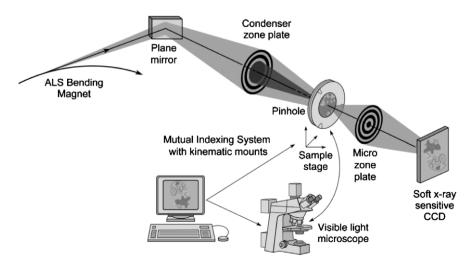


Fig. 1. Schematic layout of the soft X-ray microscopy beamline XM-1 at the ALS using X-rays emitted from a bending magnet. The micro zone plate projects a full-field image onto a CCD camera that is sensitive to soft X-rays.



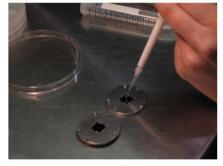




(a) preparation of the solutions

(b) silicon nitride windows

(c) aspect of the supernatant solution after centrifugation



(d) placement of the cement mixture into the windows



(e) sample holder in the soft x-ray microscope

(f) data acquisition

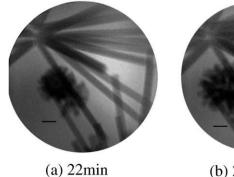
Fig. 2. Sequence of sample preparation.

saturated Ca(OH)<sub>2</sub> solution, while in the presence of 10% Ca(OH)<sub>2</sub> the same ribbons are formed, but at a smaller size; this is assumed to be a retardation of the  $C_4A_3\bar{S}$  hydration. The same retardation was observed by Mehta [16].

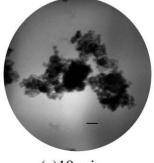
In the second series of tests, one type of expansive admixtures based on calcium sulfoaluminate was blended with Type I–II cement to analyze how expansive admixture works alone and in combination with Portland cement. The expansive admixture is used to reduce shrinkage and the associated cracking, it is a suitable product for multiple applications, including concrete pavement. Fig. 4 shows soft X-ray images of Type I–II cement (Fig. 4a) and pure expansive agent (Fig. 4b) after 35 min in solution.

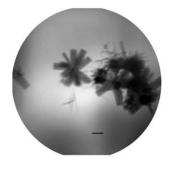
Fig. 4c and d indicates how the expansive admixture modifies the microstructure. The solution/solid materials ratio was 10 ml/g.

The reaction between calcium, sulfate, aluminate, and hydroxyl ions within a few minutes of cement hydration resulted in needle-shaped crystals of a calcium sulfoaluminate hydrate (ettringite) appearing before 21 min (see Fig. 4c). When cement is blended with 6% of expansive admixture and hydrated, basically ettringite (needles) and calcium hydroxide (platelets) were formed close to the surface of the cement particles, which agree with Nagataki and Gomi [17]. Differently those which can be seen in Fig. 4a, the platelets are formed close to the surface of the particle, this is important in order to achieve the desired amount of expansion.





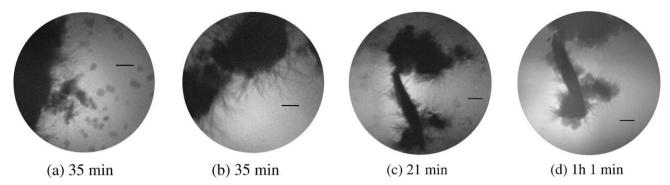




(c)19 min

(d) 2 h 47 min

**Fig. 3.** In situ soft X-ray images of hydrating  $C_4A_3\bar{S}$  particles in a saturated calcium hydroxide–calcium sulfate solution, s/c = 50 ml/g. Figs. (a) and (b): hydration in  $Ca(OH)_2 + CaSO_4 \cdot 2H_2O$  saturated solution. Figs. (c) and (d): hydration in solutions with the presence of 10%  $Ca(OH)_2$ ; hydration time is indicated. Scale bar corresponds to 1  $\mu$ m. (a) 22 min (b) 2 h 45 min (c)19 min (d) 2 h 47 min.



**Fig. 4.** (a) *In situ* soft X-ray images of around 35 min hydration of Type I–II cement, (b) commercial expansive admixture hydrated for 35 min, (c) and (d) blend of Type I–II cement and 6% of expansive admixture over the time in a saturated calcium hydroxide–calcium sulfate solution. Hydration time is indicated. Scale bar corresponds to 1 μm. (a) First freeze, (b) first thaw (c) second freeze.

### 3. X-ray microtomography

In synchrotron XMT, a wide parallel X-ray beam passes through the sample, and the transmitted fraction is detected as an image by a charge couple device (CCD) area detector. The sample is mounted on a stage with four degrees of freedom: x, y, and z transition for positioning and centering of the sample, and rotation for data acquisition. The rotary stage allows for many X-ray transmission images of the sample to be recorded for a series of orientations. According to the Nyquist theorem, the minimum number of radiographs needed for a full reconstruction within 180° of rotation is  $N\pi/2$ , where N is width of the sample in pixels. In practice, little difference is observed for image numbers  $>N\pi/4$ .

At the ALS superbend beamline 8.3.2, the transmitted beam is detected by a scintillator that converts X-rays to visible light, which is imaged by magnifying lenses onto a CCD. The available optics allow for spatial resolutions between 0.44  $\mu m/pixel$  and 11.5  $\mu m/pixel$ . The monochromator is tunable to X-ray energies between 5 keV and 35 keV by means of a double multilayer mirror arrangement. The monochromator optics can also be removed from the beam, allowing a white light mode of operation that is broadly polychromatic in the 30–70 keV range by means of metal filters.

Tomography imaging can be done either in absorption or phase contrast mode. In absorption mode, each pixel of the projection image represents the intensity of the transmitted beam that has not been absorbed by the sample. According to Beer's law:

$$I(x) = I_0 e^{-\mu x} \tag{1}$$

where I(x) is the final intensity of the beam after the beam passes through sample thickness x,  $I_o$  is the initial intensity of the beam, and  $\mu$  is the material's linear attenuation coefficient, which can be expressed as:

$$\mu = \sum f_i \mu_i \rho; \tag{2}$$

where i represents each atomic element,  $f_i$  is the atomic weight fraction of the element in the material,  $\mu_i$  is the mass attenuation coefficient at the beam energy, and  $\rho$  is the density of the material.

The attenuated value for each pixel,  $P(\varphi, \mathbf{x}')$  is then determined by:

$$P(\varphi, \mathbf{x}') = -\ln\left[I(\varphi, \mathbf{x}')/I_{o}(\varphi, \mathbf{x}')\right],\tag{3}$$

where for each angle of rotation for the sample,  $\varphi$ , x' indicates the axis parallel to the X-ray direction and perpendicular to the sample rotation;  $I_o(\varphi,x')$  is the initial intensity of the beam, and  $I(\varphi,x')$  is the intensity of the transmitted beam.

The attenuation can be expressed as an integral of the linear attenuation coefficient over the sample:

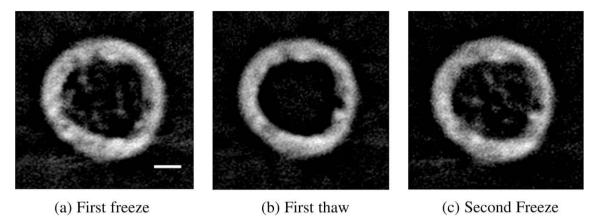
$$P(\varphi, x', z) = \int \int_{-\infty}^{\infty} \mu(x, y, z) \delta(x \cos \varphi + y \sin \varphi - x') dx dy, \tag{4}$$

where the  $\delta$  function defines the path of the X-rays through the sample, and  $\mu$  is the material's linear attenuation coefficient. The function  $P(\varphi,x',z)$  is known as the Radon transform or sinogram. Reconstruction involves inverting this equation to yield  $\mu(x,y,z)$  at each voxel location [18]. Ultimately, the reconstruction procedure combines the two-dimensional projection images to yield a three-dimensional data set of the absorbance of the sample at each voxel. The processed data contains a series of cross-sectional slices (tomograms) at set height z of the sample.

The absorption mode scanning is usually applied using monochromatic light. If polychromatic beam is used, Beer's law no longer applies because the high-energy X-rays penetrate farther than those of low energy. This causes the "beam hardening effect," making the resulting image less quantitative. Mathematical approximations are necessary to adjust for the beam hardening effect, and a rigorous reconstruction can only be made if it is assumed that all components of the sample have the same energy dependence of absorption.

The phase contrast mode relies on the variation of the real part of the index of refraction to provide image contrast. Materials differing in their refractive index result in differing phase shifts to the transmitted X-rays. The effect of the differing phase shifts is only seen at the edges between materials of different indices. The interaction of X-rays with differing phase shifts result in intensity cancellation or enhancement at the detector plane. Edges are seen as oscillating dark and bright lines and generally have much higher contrast than the corresponding absorption images. At X-ray wavelengths, the index of refraction is about 10–100 times larger than the absorption index. Thus the phase contrast mode of operation may, in principle, be more sensitive than the absorption mode. Phase contrast tomography is harder to interpret quantitatively, but progress is being made in this new area [19]. The phase contrast mode is useful for imaging light materials for which the absorption contrast is very small, such as for low density materials or for materials with very similar absorption properties but where small changes in the refraction index can be observed.

With the radiograph size limit of 4000 pixels by 2500 pixels, and the source beam size being less than 4 cm wide and usually several millimeters high, the choice of magnification becomes a tradeoff between the desired magnification and the scale of the sample to be tested. The sample must fit completely within the width of the produced radiograph, otherwise the tomographic reconstruction becomes more approximate. Although smaller samples may allow for greater magnification and finer dimensions of the smallest observable detail, the data might not be appropriate given that concrete is generally used in the construction of large



**Fig. 5.** Reconstructed tomograms of the sample in the midsection of one of the larger air voids. The scale bar is 50 μm. The ice crystals (a) in dendritic form; (b) disappear upon thawing; and (c) reappear in changed morphology during the second freeze.

structures. Using larger samples, however, results in lower magnification and coarser dimensions of the smallest observable particle size

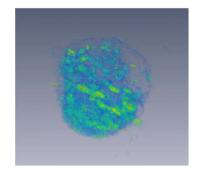
As an example of the application of microtomography to study the durability of concrete, reported herein are preliminary conclusions of the analyses of images obtained in the study of *in situ* ice formation in entrained air voids in hydrated cement paste. When cement paste is exposed to freezing temperatures it experiences deformation caused by: (a) volumetric change as liquid water solidifies, (b) cryo-suction driving liquid water towards the frozen pores, (c) drainage of liquid water from the freezing sites, (d) thermo-mechanical response of the porous matrix, the liquid and solid water. The imaging of ice inside the cement paste can be done by low-temperature scanning electron microscopy and directional cooling microscopy but these techniques do not provide the three-dimensional information which is easily obtained by microtomography.

To prevent frost damage in concrete, a well-distributed system air voids is introduced in the porous cement paste. These air voids may also act as an efficient cryo-pump because ice crystals, initially formed in air voids at atmospheric pressure, make the liquid water diffuse towards the air voids, rather than towards the larger already frozen pores, resulting in a beneficial lowering of the inpore crystal pressure. There is a strong technological incentive to further characterize the formation and distribution of ice crystals inside and around the air voids so a new generation of air-entraining admixtures can be developed. For this preliminary experiment, XMT scans were performed on air-entrained, hydrated cement paste cast in 0.25 mm outer diameter glass capillary tube while subjecting the sample to several cycles of evaporated liquid nitrogen gas cooling and ambient thawing. The samples were frozen using an Oxford 600-series Cryostream which generates a jet of cold nitrogen gas with a coaxial outer sheath of dry nitrogen for frost prevention. The scan settings were optimized for both absorption and phase contrast imaging. The absorption contrast provided imaging of the dense hydrated cement matrix, while the phase contrast provided imaging of the ice-air phases that would coexist in the partially filled air void. A 13 KeV monochromatic X-ray energy provided sufficient penetration through the hydrated cement paste, while an approximately 30 cm sample-to-detector distance enhanced the phase contrast imaging and made visible the iceair interface within the air voids. The magnification was set at  $5.4 \times$  with the resulting pixel size of 1.54 µm/pixel.

Fig. 5 shows a cross section through one of the cement air voids. The donut shape is the cement material constrained within the glass tube. Fig. 5a shows the sample when frozen, with dendritic ice growth obvious within the void. The sample was then allowed

to warm up to room temperature (Fig. 5b). The ice formation disappears and then reappears with a different structure as the sample was frozen once again (see Fig. 5c). Fig. 6 shows a three-dimensional reconstruction of the air/ice interface in a single air void; it is roughly that of a spherical shell. Upon thawing, the interface vanished, confirming that the image of the ice was successful.

Microtomography is also useful for performing time-scale experiments. Here, a modified ASTM C-1260 test of aggregate reactivity was conducted. In this experiment, three 1 in.  $\times$  1 in.  $\times$  12 in. mortar bars were cast following the specifications detailed in ASTM C-1260, using known inert aggregate, while 50% of the aggregates by weight were substituted with partially strained quartz sand. The goal was to supplement the expansion results of the samples obtained by the ASTM C-1260 test with visual representation of damage in progress. Considering the scale and the density of the sample, the samples were scanned under a white beam filtered with aluminum plates and under low magnification. The resulting spatial resolution was 11.55 um/pixel. The scans were obtained under phase contrast mode in an attempt to enhance crack widths smaller than spatial resolution. Because the 1 in.  $\times$  1 in. cross section of the mortar bars was wider than the beam width, a narrower sample (0.75 in.  $\times$  0.75 in.  $\times$  9 in.) was cast. The large bars and the smaller bar were subject to the same test conditions prescribed by ASTM C-1260. The larger bars were designed to provide strain data while the smaller bar would be used for imaging purposes. The smaller bar underwent a slightly amplified but comparable rate of expansion and was deemed to be a suitable representation of the standard-sized mortar bar expansion. Fig. 7 shows internal sections of the same sample after



**Fig. 6.** A 3D reconstruction of ice formation in a single air void. The cement matrix and the glass capillary have been removed via post reconstruction image processing. The air void is approximately 150  $\mu$ m in diameter. (a) 7 Days (b) 14 days.

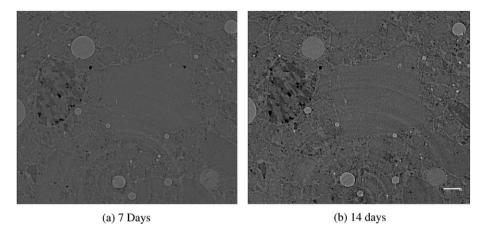


Fig. 7. Tomogram of reduced sized mortar bars at (a) 7 days and (b) 14 days of ASTM C-1260 testing: new microcrack developments (arrows) are visible. The ring-like structure overlaid on the images is an artifact of the tomographic reconstruction process. The scale bar is 500 μm.

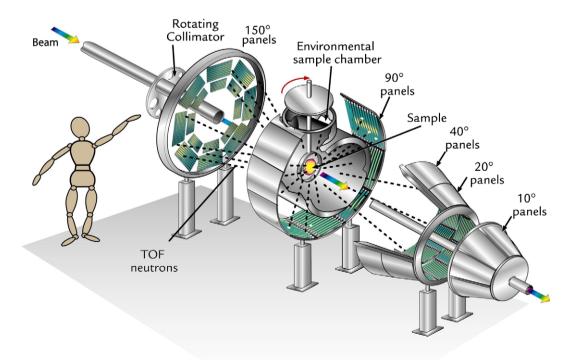
7 and 14 days. Close examination of the older sample shows the development of microcracks.

Because most existing high resolution characterization techniques require drying of the sample or examination in vacuum, *in situ* studies of corrosion are challenging to perform. Both corrosion reactions and corrosion products are very sensitive to environmental conditions. Microtomography can be an ideal tool to image the evolution of the corrosion products and cracking of the matrix.

### 4. Neutron diffraction and texture analysis of reactive rocks

The previous discussion has mainly focused on the application of synchrotron X-rays in cement research. Other opportunities are provided by neutron scattering (e.g. Wenk, [20]). Compared to X-rays neutrons have low absorption for most elements and are thus ideal to investigate bulk samples with penetration of centimeters to tens of centimeters. It is easy to perform *in situ* exper-

iments at low/high temperature and pressure and record changes during chemical reactions and thus to study kinetics. For cement the interaction of neutrons with hydrogen is of particular importance: contrary to X-rays which scatter on electrons and thus hydrogen with one electron is largely invisible, neutrons scatter on the nucleus and scattering on hydrogen is of similar magnitude as for heavy elements. For hydrogen (H with one proton) there is a large interaction and energy exchange with thermal neutrons resulting in strong inelastic scattering. For deuterium (D with a proton and a neutron) there is little inelastic interaction and compounds with D are preferred for crystallographic studies. Finally, for neutrons, scattering on the small nucleus the scattering factor does not decrease appreciably with scattering angle  $2\theta$ and thus high intensities are observed to small d-spacings, resulting in good resolution for structure refinements. The application of neutron scattering to cement has been reviewed by Livingston et al. [21].



**Fig. 8.** HIPPO time-of-fight neutron diffractometer at LANSCE with a large sample chamber (person is for scale) and 32 detector panels (green) arranged on four banks. Each detector views differently oriented crystals (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.).

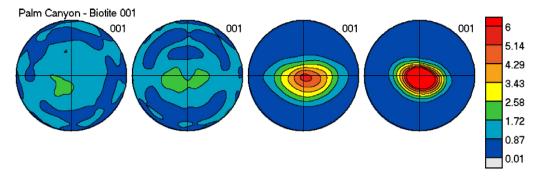
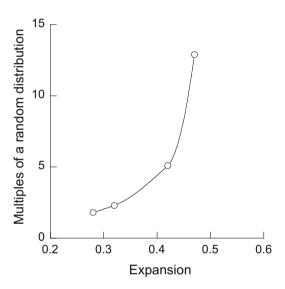


Fig. 9. Pole figures of (001) biotite for selected samples of deformed granite calculated from the orientation distribution based on neutron diffraction data, in order of increasing deformation. The pole figures are projected onto the foliation plane. Linear scale (in m.r.d.), equal area projection. Deformation increases from left to right.

Neutrons have been critical to determine the detailed crystal structures of ettringite [22] and portlandite [23] and were used for quantitative phase analysis of clinkers (e.g. Peterson et al. [24], Pritula [25]). Small angle scattering lends itself to quantify porosity in concrete (e.g. Allen et al. [26], Jennings et al. [27]) as well as to investigate the still enigmatic structure of C-S-H see Refs. [28–31] and kinetic studies of hydration [32]. High resolution neutron strain diffractometers have been used to determine internal stresses imposed during freezing of concrete [33] and recently there has been interest in determining the nature of water bonds with inelastic neutron scattering [34], which is likely to become an important application with new high resolution instruments such as the Fourier Difference Spectrometer (FDS) at LANSCE or similar instruments in ISIS [35]. Finally neutrons are very complementary to X-rays in tomography [36]. X-rays are excellent to determine high atomic number contrast such as steel reinforcements in concrete, or pore space. With neutrons it is possible to obtain 3D images of water migration and reactions of hydrous phases.

We have been making use of the time-of-flight neutron diffractometer HIPPO (High Pressure Preferred Orientation [37] for a rather unique concrete application. This instrument has a large sample chamber for ancillary equipment such as pressure cells and furnaces and many detectors for fast data acquisition (Fig. 8). It has been used to investigate structure changes in portlandite with pressure and temperature [38]. We used the capability that each detector records differently oriented crystals in the sample and crystal orientation patterns are related to the deformation history of a rock.

We have observed that the deformation state of a granitic rock has a profound impact on the long-term stability of concrete if used as aggregate due to enhanced susceptibility to the alkali-silica reaction [39,40]. An investigation of the microstructure of granitic rocks revealed that, as these rocks become progressively deformed from granite to mylonite and phyllonite they develop a microstructure that makes them increasingly unsuitable for use as aggregates. ASTM C 1260 expansion tests were performed and indicate expansion of 0.25 in granite up to 0.55 in phyllonite after 30 days. To obtain a bulk measure of the deformation state we have determined preferred orientation of mica with HIPPO. Mica platelets align during ductile deformation and this "directionality" is most commonly represented in pole figures. A pole figure is constructed by placing a polycrystalline sample inside a sphere and intersecting a crystal direction (e.g., the pole to the (001) platelet plane of mica) with the surface of the sphere. Pole densities can then be contoured and expressed as a smooth distribution, usually normalized such that contours express pole densities in multiples of a uniform distribution. These pole figures (Fig. 9) document that preferred orientation increases from maximum pole densities of 1.4 m.r.d. in weakly deformed granite to 12 m.r.d. in phyllonite.



**Fig. 10.** A plot of the average 30-day expansion value of each rock type from ASTM C 1260 tests versus the (001) pole figure maximum for biotite in multiples of a random distribution.

Biotite preferred orientation (and thus overall deformation) correlates perfectly with the observed expansion (Fig. 10).

#### 5. Conclusions

A well-documented understanding underlying cement hydration has remained elusive given the complexity of the reaction. The recent application of advanced methodologies has been developed to characterize the nano and microstructure of cement paste and concrete exposed to aggressive environments. High resolution full-field soft X-ray imaging in the water window is providing new insight into the cement hydration process on the nano scale. This technique is also appropriate to characterize the new generation of nanohybrid CSH which can increase the durability of concrete.

Preliminary results of studying freezing of cement paste and alkali-aggregate reaction with microtomography indicates that the technique has the following advantages: (a) rapid data acquisition, (b) high resolution; (c) non-destructive examination; (d) characterization of samples at ambient and low temperatures. The use of phase contrast was a powerful technique to image ice inside cement paste and to allow the use of larger samples for the ASR study.

Texture analysis of quartz grains showed no simple relationship between quartz preferred orientation and the susceptibility to the alkali–silica reaction. However, texture analysis of biotite, a common accessory mineral in granites, shows a linear relationship with reactivity. These results suggest that preferred orientation of biotite may be used to estimate the susceptibility of rocks containing both biotite and quartz to the alkali–silica reaction.

#### Acknowledgments

Paulo Monteiro and Rudy Wenk acknowledge the financial support from the National Science Foundation Grant 062464 and from KAUST. Ana Paula Kirchheim acknowledge the financial support of CAPES (Fundação Coordenação de Aperfeiçoamento de Pessoal de Nível Superior – Ministério da Educação – Brasil). The soft X-ray microscope and the Advanced Light Source are supported by the Director, Office of Science, Office of Basic Energy Sciences, of the U.S. Department of Energy under Contract No. DE-AC02-05CH11231. The authors would like to thank Cruz Carlos and Roula Rbeiz for their assistance in the ASTM-1260 preparation and testing.

#### References

- Kurtis KE, Monteiro PJM, Brown JT, Meyer-Ilse W. Imaging of ASR gel by soft Xray microscopy. Cem Concr Res 1998;28:411–21.
- [2] Kurtis KE, Monteiro PJM, Meyer-Ilse W. Soft X-ray spectromicroscopy for in situ study of corrosion. J Corros Sci 2000;42(8):1327–36.
- [3] Gartner E, Kurtis KE, Monteiro PJM. Proposed mechanism of C-S-H growth tested by X-ray microscopy. Cem Concr Res 2000;30(5):817-22.
- [4] Lamour VHR, Monteiro PJM, Scrivener KL, Fryda H. Microscopic studies of the early hydration of calcium aluminate cements. In: International conference on calcium aluminate cement (CAC), July, 2001, Edinburgh, UK Proceedings. p. 169–80.
- [5] Kurtis KE, Collins CL, Monteiro PJM. The surface chemistry of the alkali-silica reaction: a critical evaluation and X-ray microscopy. Concr Sci Eng 2002;4:1–11.
- [6] Kurtis KE, Monteiro PJM. Chemical additives to control expansion of alkalisilica reaction gel: proposed mechanisms of control. J Mater Sci 2003;38(9):2027–36.
- [7] Juenger MCG, Lamour V, Monteiro PJM, Gartner EM, Denbeaux GP. Direct observation of cement hydration by X-ray microscopy. J Mater Sci Lett 2003;22(19):1335–7.
- [8] Juenger MCG, Monteiro PJM, Gartner EM, Denbeaux GP. A soft X-ray microscope investigation into the effects of calcium chloride on tricalcium silicate hydration. Cem Concr Res 2005;35(1):19–25.
- [9] Silva DA, Monteiro PJM. Hydration evolution of C3S EVA composites analyzed by soft X-ray microscopy. Cem Concr Res 2005;35(2):351–7.
   [10] Silva DA, Monteiro PJM. The influence of polymers on the hydration of
- [10] Silva DA, Monteiro PJM. The influence of polymers on the hydration of portland cement phases analyzed by soft X-ray transmission microscopy. Cem Concr Res 2006;36(8):1501–7.
- [11] Silva DA, Monteiro PJM. Early formation of ettringite in tricalcium aluminatecalcium hydroxide-gypsum dispersions. J Ceram Soc 2007;90(2):614-7.
- [12] Monteiro PJM, Shomglin K, Wenk HR, Hasparyk NP. Effect of aggregate deformation on alkali-silica reaction. ACI Mater J 2001;98(2):179–83 [Farmington Hills: Michigan].
- [13] Chao W, Harteneck BH, Liddle JA, Anderson EH, Attwood DT. Soft X-ray microscopy at a spatial resolution better than 15 nm. Nature 2005;435:1210-3.
- [14] Kim DH, Fischer P, Chao W, Anderson E, Im MY, Shin SC, et al. Magnetic soft X-ray microscopy at 15 nm resolution probing nanoscale local magnetic hysteresis (invited). Virt J Nanoscale Sci Technol 2006;13(17) [May 1].

- [15] Mehta PK, Monteiro PJM. Microstructure, properties and materials. 3rd ed. McGraw-Hill; 2006. 659 pp.
- [16] Mehta PK. Effect of lime on hydration of pastes containing gypsum and calcium aluminates or calcium sulfoaluminate. J Am Ceram Soc 1973;56(6).
- [17] Nagataki S, Gomi H. Expansive admixtures (mainly ettringite). Cem Concr Compos 1998;20(2-3):163-70.
- [18] Kak AC, Slaney M. Principles of computerized tomographic imaging, classics in applied mathematics, vol. 33. Philadelphia (PA): Society for Industrial and Applied Mathematics; 2001.
- [19] Groso A, Abela R, Stampanoni M. Implementation of a fast method for high resolution phase contrast tomography. Opt Expr 2006;14:8103–10.
- [20] Wenk HR, editor. Neutron scattering in earth sciences. Review in mineralogy and geochemistry, vol. 63. Mineral Soc Amer; 2006. 471pp.
- [21] Livingston RA, Neumann DA, Allen AJ, Fitzgerald SA, Berliner R. Application of neutron scattering to Portland cement. Neutron News 2000;11.
- [22] Hartman MR, Berliner R. Investigation of the structure of ettringite by time-of-flight neutron powder diffraction. Cem Concr Res 2006;36:364-70.
- [23] Leinenweber K, Partin DE, Schuelke U, O'Keeffe M, Von Dreele RT. He structure of high pressure Ca(OD)<sub>2</sub>II from powder neutron diffraction: relationship to the ZrO<sub>2</sub> and Eul<sub>2</sub> structures. J Solid State Chem 1997;132:267–73.
- [24] Peterson V, Hunter B, Ray A, Aldrige LP. Rietveld refinement of neutron, synchrotron and combined powder diffraction data of cement clinker. Appl Phys A 2004;74:s1409–11.
- [25] Pritula O. X-ray and neutron Rietveld quantitative phase analysis of industrial Portland cement clinkers. Powder Diffr 2004;19(3).
- [26] Allen AJ, Windsor CG, Rainey V, Pearson D, Double DD, Alford NMcN. A smallangle neutron-scattering study of cement porosities. J Phys D 1982;15:1817–33.
- [27] Jennings HM, Thomas JJ, Gevrenov JS, Constantinides G, Ulm FJ. A multitechnique investigation of the nanoporosity of cement paste. Cem Concr Res 2007:37:329–36.
- [28] Allen AJ, Thomas JJ, Jennings HM. Composition and density of nanoscale calcium-silicate-hydrate in cement. Nature Mater 2007;6. April.
- [29] Allen AJ, Thomas JJ. Analysis of cement paste and C-S-H gel by small-angle neutron scattering. Cem Concr Res 2007;37:319–24.
- [30] Thomas JJ, Jennings HM, Allen AJ. The surface area of cement paste as measured by neutron scattering – evidence for two C–S–H morphologies. Cem Concr Res 1998;28(6):897–905.
- [31] Thomas JJ, FitzGerald SA, Neumann DA, Livingston RA. The state of water in hydrating tricalcium silicate and Portland cement pastes as measured by quasi-elastic neutron scattering. J Am Ceram Soc 2001;84:1811-6.
- [32] Fitzgerald SA, Thomas JJ, Neumann DA, Livingston RA. A neutron scattering study of the role of diffusion in cement hydration. Cem Concr Res 2002;32:409–13.
- [33] Schulson EM, Swainson IP, Holden TM. Internal stress within hardened cement paste induced through thermal mismatch: calcium hydroxide versus calcium silicate hydrate. Cem Concr Res 2001;31:1785–91.
- [34] Padureanu I, Aranghel D, Rotarescu G, Dragolici F, Turcanu C, Kozlov ZA, et al. An inelastic neutron scattering study of the water dynamics in cement paste. Rom I Phys 2005:50:561–74.
- [35] Parker SF. Inelastic neutron scattering spectroscopy. In: Handbook of vibrational spectroscopy. Wiley; 2006.
- [36] Lehmann EH, Frei G, Kuehne G, Boillat P. The micro-setup for neutron imaging: a major step forward to improve the spatial resolution. Nucl Instr Methods A 2007;576:389–96.
- [37] Wenk HR, Lutterotti L, Vogel S. Texture analysis with the new HIPPO TOF diffractometer. Nucl Instr Methods A 2003;515:575–88.
- [38] Xu H, Zhao Y, Zhang J, Hickmott DD, Daemen LL. In situ neutron diffraction study of deuterated portlandite Ca(OD)<sub>2</sub> at high pressure and temperature. Phys Chem Mineral 2007;34:223–32.
- [39] Monteiro PJM, Shomglin K, Wenk HR, Hasparyk NP. Effect of aggregate deformation on alkali-silica reaction. Am Concr Inst Mater J 2001;98:179–83.
- [40] Wenk HR, Monteiro P, Shomglin K. Relationship between aggregate microstructure and concrete expansion. A case study of deformed granitic rocks from the Santa Rosa Mylonite Zone. J Mater Sci 2007;43:1278–85.