

# Synthesis and characterization of $\text{La}_2\text{O}_3/\text{BaO}$ -doped $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$ microwave ceramics

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## Abstract

$(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$  microwave ceramics using  $\text{La}_2\text{O}_3$  and BaO as sintering aids were prepared by the conventional mixed oxide route and investigated by XRD, SEM, EDS and dielectric measurements. The sintering temperature could be effectively lowered by using  $\text{La}_2\text{O}_3$  and BaO additives.  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST ceramics sintered at  $1350^\circ\text{C}/4\text{ h}$  showed very good dielectric properties:  $\epsilon_r = 41$ ,  $\tau_f = -3.79\text{ ppm}/^\circ\text{C}$ , and  $Q = 9800$  at  $1.8\text{ GHz}$ . In  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST ceramics, a major ZST solid solution is formed with three minor phases: (i)  $(\text{Ti}_{0.8}\text{Zr}_{0.1}\text{Sn}_{0.1})\text{O}_{2\text{ss}}$ , a  $\text{TiO}_2$ -based solid solution, (ii) orthorhombic  $\text{La}_{2/3}\text{TiO}_3$  containing 3.4 at.% Ba, (iii) orthorhombic  $\text{Ba}_2\text{TiO}_4$  containing 2.2 at.% Zr and 2.8 at.% Sn. © 2002 Published by Elsevier Science Ltd and Techna S.r.l.

**Keywords:** B. Microstructure; C. Dielectric properties;  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$ ; Phase

## 1. Introduction

Zirconium tin titanate ceramics have attracted noticeable interest as microwave dielectric materials in communication networks. Among them,  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$  (ZST) has proved to be one of the most popular dielectric resonator materials, because of a high dielectric constant  $\epsilon_r$ , a high quality factor  $Q$ , and a low temperature coefficient of resonant frequency  $\tau_f$  [1]. ZST ceramics are prepared either by attrition milling [2] or by chemical processing methods [3]. Being difficult to fully densify ZST ceramics at a low temperature, a combination of two or more oxides are used as sintering aids, i.e.  $\text{La}_2\text{O}_3$ , ZnO, NiO,  $\text{Fe}_2\text{O}_3$  and  $\text{Co}_2\text{O}_3$ , etc. [4–9].

Duk-Jun Kim and his co-workers [10] reported that all the alkaline earth metals were effective for densification of ZST ceramics, especially BaO. BaO additions did not affect the values of dielectric constant  $\epsilon_r$  and temperature coefficient of resonant frequency  $\tau_f$ , and can improve the quality factor  $Q$ . Although  $\text{La}_2\text{O}_3$ –BaO– $\text{TiO}_2$  has the ability to form eutectic liquids at

temperature of  $1350^\circ\text{C}$  or below [11], it has not yet been used as sintering aid in the ZST system. This paper reports an investigation on phase, microstructure and dielectric properties of ZST microwave ceramics doped with  $\text{La}_2\text{O}_3$  and BaO.

## 2. Experimental procedures

The powders were prepared via the conventional mixed oxide route. The raw materials were high purity ( $>99.0\%$ )  $\text{ZrO}_2$ ,  $\text{TiO}_2$ ,  $\text{SnO}_2$ ,  $\text{La}_2\text{O}_3$ , and  $\text{BaCO}_3$ . To study the effects of different additives upon phase, microstructure and dielectric properties of ZST ceramics, four compositions were chosen:

- (a) undoped  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$
- (b)  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4 + 1.0\text{ wt.}\% \text{ La}_2\text{O}_3$
- (c)  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4 + 2.0\text{ wt.}\% \text{ BaO}$
- (d)  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4 + 1.0\text{ wt.}\% \text{ La}_2\text{O}_3 + 2.0\text{ wt.}\% \text{ BaO}$

The above compositions were milled with  $\text{ZrO}_2$  balls for 24 h in deionized water, dried, calcined at  $1150^\circ\text{C}$  for 2 h, remilled with an organic dispersant and binder,

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molded into disks 10 mm in diameter and  $\sim 2$  mm thick under a pressure of 100 MPa, then sintered in air at 1550 and 1350 °C for 4 h for undoped ZST and doped ZST, respectively. The heating rate was 300 °C/h. The cooling rate was 600 °C/h down to 800 °C, then furnace cooling to room temperature.

The bulk densities of the sintered bodies were measured by the Archimedes method. X-ray diffraction (XRD) analysis (D/Max-IIIB, Rigaku, Japan) was used to identify the crystalline phases of the sintered samples. For microstructural analysis samples were studied by scanning microscopy (SEM) (JSM-6301F, Jeol, Japan) in the as-sintered condition and polished, thermally etched condition. The phase composition was investigated by EDS (Link ISIS-300, UK).

Dielectric properties of sintered samples were measured at 1.8 GHz with a HP 4291B RF Impedance/Material analyzer. Silver paste was used for the electrodes. The temperature coefficient of dielectric constant  $\tau_\epsilon$  was determined at 1 MHz with a precision LCR meter (HP 4284A) equipped with a thermostat in the temperature 20–85 °C. The temperature coefficient of resonant frequency  $\tau_f$  was calculated as follows,

$$\tau_f = -1/2\tau_\epsilon - \alpha$$

where  $\tau_\epsilon$  is the temperature coefficient of dielectric constant and  $\alpha$  is the linear thermal expansion coefficient of the material [8].

### 3. Results and discussion

Fig. 1 shows the X-ray diffraction patterns of the sintered specimens. A major phase indexed in terms of an

orthorhombic unit cell  $\text{ZrTiO}_4$  (JCPDS 34-415) could be observed. In undoped ZST and in  $\text{La}_2\text{O}_3$ -doped ZST, there were no secondary phases, whereas  $\text{Ba}_2\text{TiO}_4$  was detected in the BaO-doped specimen;  $\text{TiO}_2$ ,  $\text{La}_{2/3}\text{TiO}_3$ ,  $\text{Ba}_2\text{TiO}_4$  and an unknown phase were detected in the specimen doped with  $\text{La}_2\text{O}_3/\text{BaO}$ . These results are due to the differences of the ionic sizes between the substitution ions and the tetravalent ions in  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$ . The  $\text{La}^{3+}$  and  $\text{Ba}^{2+}$  could not be substituted for the tetravalent element of  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$  being the ionic radii of  $\text{La}^{3+}$  (1.061 Å) and  $\text{Ba}^{2+}$  (1.35 Å) much larger than for  $\text{Zr}^{4+}$  (0.72 Å),  $\text{Sn}^{4+}$  (0.69 Å) and  $\text{Ti}^{4+}$  (0.605 Å) ions. Thus, with additions of  $\text{La}_2\text{O}_3$  and/or BaO,  $\text{La}_2\text{O}_3$  and BaO react with  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$  and second phases are formed. Although  $\text{La}_2\text{O}_3$  is well known to form a boundary phase, a secondary phase was not observed with addition of only 1 wt.%  $\text{La}_2\text{O}_3$ , due to the fact that detection of a minor phase by XRD is extremely difficult.

SEM micrographs of the as sintered specimens surfaces are shown in Fig. 2. Major differences between the microstructures are apparent. As shown in Fig. 2(a), the undoped ZST ceramics are dense at 1550 °C. The grains are rounded and uniform in shape. With  $\text{La}_2\text{O}_3$  and/or BaO additions, a much easier densification of ZST ceramics at a lower temperature occurs, as shown in Fig. 2(b)–(d), grain sizes are larger than for the undoped sample and abnormal grain growth occurs.  $\text{La}_2\text{O}_3/\text{BaO}$  doped ZST consists of the largest grains and is characterized by round and bar shaped grains, the fraction of the bar grains being much lower than for round shaped grains. Different grain shape indicates that some second phases possibly formed. A similar abnormal grain growth was observed by Kudesia et al. [4] on addition of  $\text{La}_2\text{O}_3/\text{ZnO}$ , and by Huang et al. [12] on addition of  $\text{CuO}/\text{ZnO}$ .

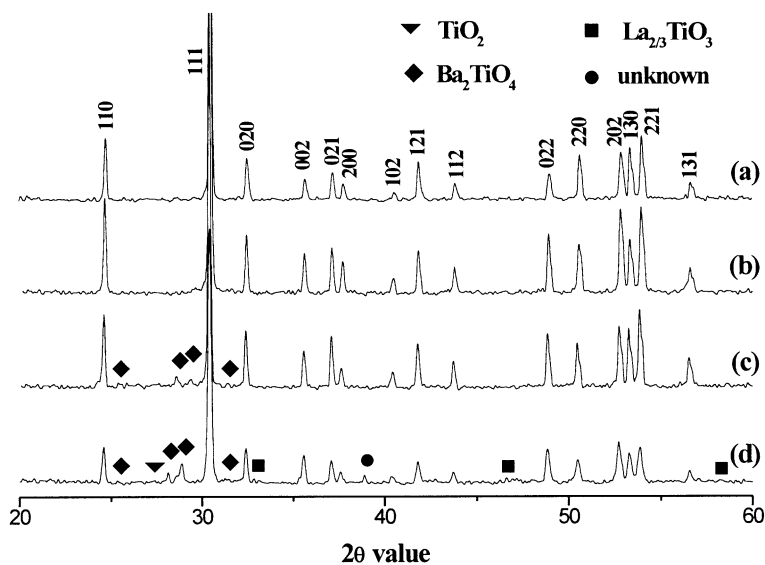


Fig. 1. X-ray diffraction patterns of  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$  ceramics. (a) 1550 °C, undoped; (b) 1350 °C, with 1 wt.%  $\text{La}_2\text{O}_3$ ; (c) 1350 °C, with 2 wt.% BaO; (d) 1350 °C, with 1 wt.%  $\text{La}_2\text{O}_3$  + 2 wt.% BaO.

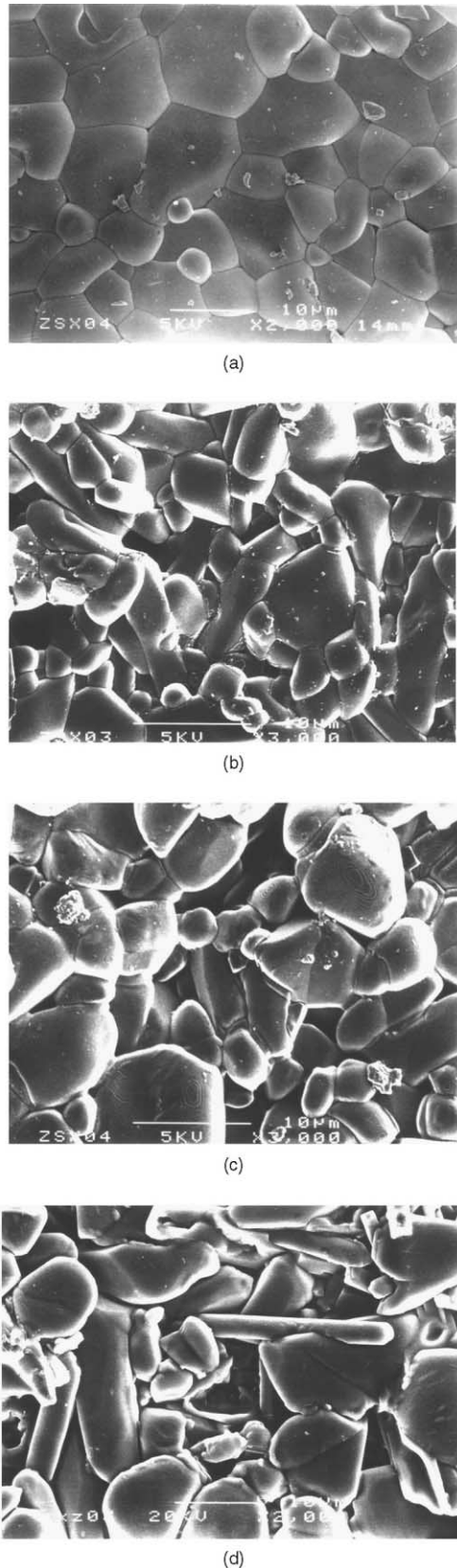


Fig. 2. SEM micrographs of the surfaces of as sintered ( $\text{Zr}_{0.8}\text{Sn}_{0.2}\text{TiO}_4$ ) ceramics. (a) 1550 °C, undoped; (b) 1350 °C, with 1 wt.%  $\text{La}_2\text{O}_3$ ; (c) 1350 °C, with 2 wt.% BaO; (d) 1350 °C, with 1 wt.%  $\text{La}_2\text{O}_3$  + 2 wt.% BaO.

Such discontinuous-type grain growth is frequently encountered in titanate systems that contain a transient liquid phase [13].

In order to verify the presence of minor phases, SEM and EDS analyses were carried out on polished and etched surfaces. In the undoped ZST and  $\text{La}_2\text{O}_3$ -doped ZST, no second phases were observed, whereas minor second phases were found in BaO-doped ZST and  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST. In BaO-doped ZST, a 'Ti-Ba' rich phase was present, whereas a phase rich only in 'Ti' was not detected. In  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST, three minor second phases were found out, i.e. a 'Ti' rich phase, a 'Ti-La' rich phase and a 'Ti-Ba' rich phase.

Fig. 3 shows SEM micrographs and EDS maps of a polished and etched surface of  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST, whereas Table 1 collects average elementary composition. The sample exhibits four distinct phases: the major phase ZST; a 'Ti' rich phase with Zr (10.8 at.%) and Sn (10.7 at.%) that we notice ' $(\text{Ti}_{0.8}\text{Zr}_{0.1}\text{Sn}_{0.1}\text{O}_{2.55})$ '; a 'Ti-La' rich phase close to the composition ' $\text{La}_{2/3}\text{TiO}_3$ ' with a small amount of Ba (3.4 at.%), a similar phase has been observed by D. Houivet et al. [6]; a 'Ti-Ba' phase close to the composition ' $\text{Ba}_2\text{TiO}_4$ ' with a small amount of Zr (2.2 at.%) and Sn (2.8 at.%). In the main ZST phase, La and Ba were not detected. Small amounts of La and Ba are difficult to quantify because of the overlap of Ti and La, Ba energy peaks. About 0.2 at.% of La and 1.6 at.% of Ba are quantified in opposition to the 0.6 and 2.7 at.% expected, respectively. So, La and Ba prefer to gather with Ti in minor phases at the grain boundaries.

Dielectric properties of the almost fully densified ZST ceramics containing different additives are listed in Table 2. Except for the comparatively high  $\tau_f$  value of the  $\text{La}_2\text{O}_3/\text{BaO}$ -doped specimen, the additives do not affect substantially the values of  $\epsilon_r$  and  $\tau_f$ . However,  $Q$  is very sensitive to the additives. Namely,  $\text{La}_2\text{O}_3$ - and/or BaO-doped specimens exhibit  $Q$  values higher than undoped specimen. The changes in the values of  $\epsilon_r$ ,  $Q$  and  $\tau_f$  can be explained by the effects of secondary phases. The high  $Q$  value of  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST ceramics are caused by the high densification and the existence of second phases  $\text{TiO}_2$  ( $\epsilon_r \sim 104$ ,  $Q \sim 14,000$  at 3 GHz) [14] and  $\text{La}_{2/3}\text{TiO}_3$  ( $\epsilon_r \sim 64.6$ ,  $Q \sim 4770$  at 3.3 GHz) [6,15].

Table 1  
EDS concentration values (at.%) for  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST

	Ti	Zr	Sn	Ba	La
Bulk	48.4	32.7	17.1	1.6	0.2
Surface	48.3	31.6	18.4	1.1	0.6
Plot a	47.1	34.7	18.2	—	—
Plot b	78.5	10.8	10.7	—	—
Plot c	58.0	—	—	3.4	38.6
Plot d	31.6	2.2	2.8	63.4	—

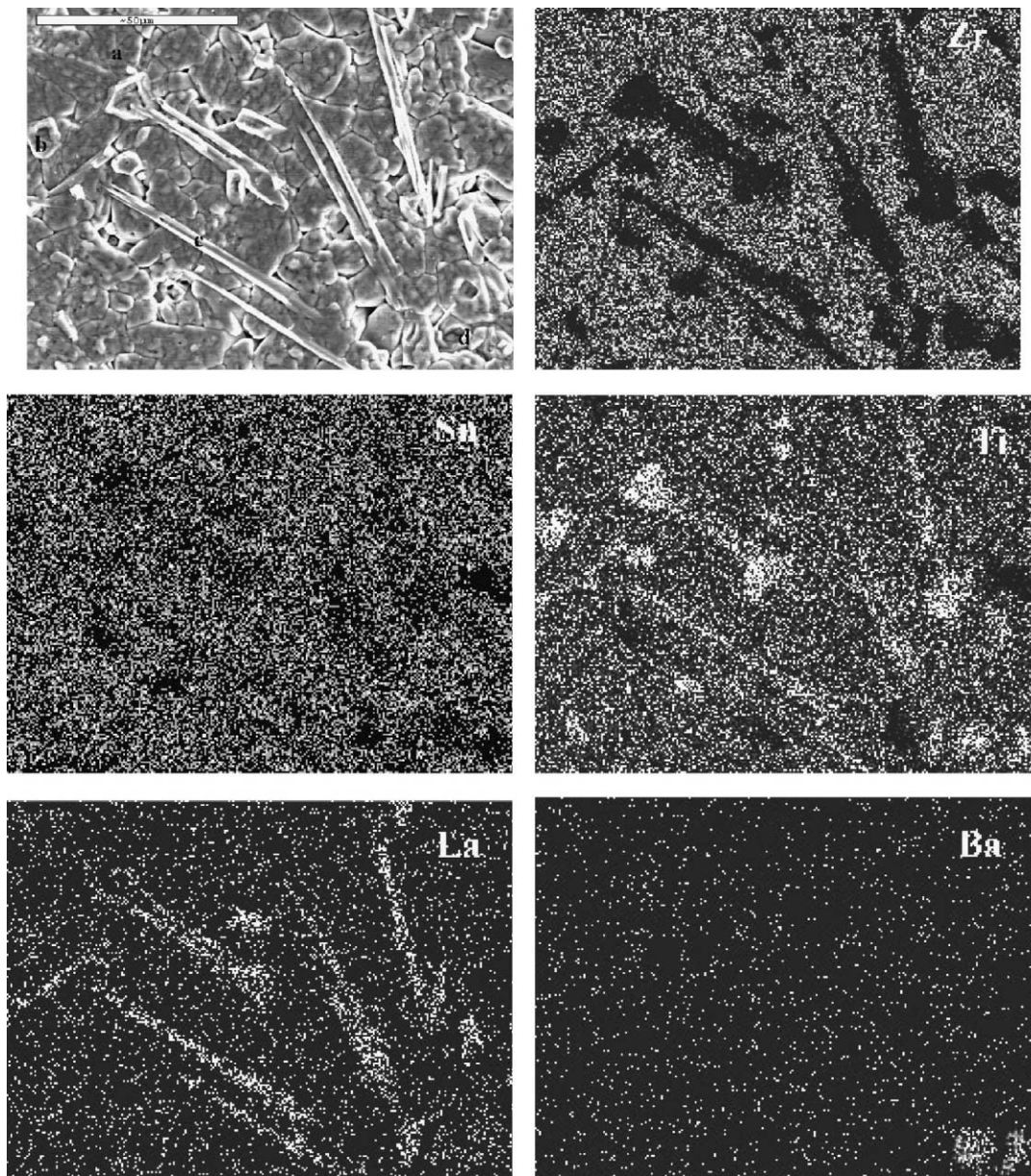


Fig. 3. SEM micrographs and EDS maps of a polished and etched surface of  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST.

Table 2  
Dielectric properties of ZST ceramics with different additives

Sample	Additives (wt.%)	Sintering temperature ( $^{\circ}\text{C}$ )	Density ( $\text{g}/\text{cm}^3$ )	$\epsilon_r$	$Q$	$\tau_f(\text{ppm}/^{\circ}\text{C})$
(a)	Undoped	1550	4.968	37	6900	0
(b)	1 wt.% $\text{La}_2\text{O}_3$ ,	1350	4.776	38	8100	2.61
(c)	2 wt.% BaO	1350	5.000	39	8600	−0.37
(d)	1 wt.% $\text{La}_2\text{O}_3$ , 2 wt.% BaO	1350	5.101	41	9800	−3.79

#### 4. Conclusions

The effects of  $\text{La}_2\text{O}_3$  and/or BaO additions on the phase, microstructure and dielectric properties of  $(\text{Zr}_{0.8}\text{Sn}_{0.2})\text{TiO}_4$  microwave ceramics were investigated.

$\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST can be sintered to high density at  $1350^{\circ}\text{C}$ , and exhibit very good dielectric properties:  $\epsilon_r=41$ ,  $Q=9800$ ,  $\tau_f=-3.79$  ppm/ $^{\circ}\text{C}$  at 1.8 GHz. According to SEM and XRD analysis, in  $\text{La}_2\text{O}_3/\text{BaO}$ -doped ZST, a major ZST solid solution is formed

accompanied by three minor phases: (i)  $(\text{Ti}_{0.8}\text{Zr}_{0.1}\text{Sn}_{0.1})\text{O}_{2\text{ss}}$ , a  $\text{TiO}_2$ -based solid solution, (ii) orthorhombic  $\text{La}_{2/3}\text{TiO}_3$  phase containing 3.4 at.% Ba, (iii) orthorhombic  $\text{Ba}_2\text{TiO}_4$  phase containing 2.2 at.% Zr and 2.8 at.% Sn.

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