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# Influence of $V_2O_5$ substitutions to $Bi_2(Zn_{1/3}Nb_{2/3})_2O_7$ pyrochlore on sintering temperature and dielectric properties

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#### Abstract

The dielectric properties of the monoclinic zirconolite-like structure compound,  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3}\_x\text{V}_x)_2\text{O}_7$  ( $0 \le x \le 0.1$ ), were investigated. We found a small vanadium substitution (x) to  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3})_2\text{O}_7$  ceramics lowered sintering temperature from 950 to 850 °C. Low-temperature sintered  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3}\_x\text{V}_x)_2\text{O}_7$  maintained high dielectric constant ( $\varepsilon_r$ ),  $\sim$ 80 and Qxf value,  $\sim$ 3000 at 6 GHz. By increasing x, second phase, bismuth vanadate ( $\text{Bi}_4\text{V}_2\text{O}_{11}$ ) was detected. The formation of second phase is accompanied by a significant decrease in dielectric loss. It was found that  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3})_2\text{O}_7$  had two phases, one was for the stoichiometric composition and the other was for Bi-deficient composition. The chemical compatibility of silver electrode and  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3}\_x\text{V}_x)_2\text{O}_7$  (x = 0.005) has also been investigated using SEM micrograph and EDS line scan of the interface between a silver electrode and the ceramic body, for low-temperature cofired ceramics (LTCC) applications.

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#### 1. Introduction

Multilayer microwave components have been investigated to miniaturize passive devices for volume efficiency and low-temperature cofired ceramics (LTCC) are required for manufacturing multilayer components. In the case of LTCC, low-melting-point glasses are frequently added as sintering aids and the resulting ceramic-glass dielectrics can be cofired with low-resistivity conductor layers, such as silver. By the way,  $V_2O_5$  has been also used as a sintering aid for the materials, such as BiNbO<sub>4</sub>, SrBi<sub>2</sub>Nb<sub>2</sub>O<sub>7</sub>, and Pb(Zr,Ti)O<sub>3</sub>, etc. [1–3] because  $V_2O_5$  has low melting temperature of  $\sim 690\,^{\circ}$ C.

Recently, bismuth-based dielectric ceramics have been studied for its relatively low sintering temperature (less than  $1000\,^{\circ}$ C) and high dielectric constant [4]. Bi<sub>2</sub>(Zn<sub>1/3</sub>Nb<sub>2/3</sub>)<sub>2</sub>O<sub>7</sub> ceramics with a dielectric constant in the range of 80–210, dielectric losses ( $\tan \delta$ ) as low as  $1 \times 10^{-4}$  at frequency of 1 MHz, were reported [5,6]. But

not much work has been done on the sintering behaviors and related dielectric properties as LTCC materials.

In our study, as a way of lowering sintering temperature, Nb ion was substituted with V ion in Bi<sub>2</sub>( $Zn_{1/3}Nb_{2/3}_{-x}V_x$ )<sub>2</sub> O<sub>7</sub> sample with  $0 \le x \le 0.1$ . The sintering behavior, phase, and microwave dielectric properties of low-temperature sintered Bi<sub>2</sub>( $Zn_{1/3}Nb_{2/3}_{-x}V_x$ )<sub>2</sub>O<sub>7</sub> were examined.

#### 2. Experimental procedure

The powders were prepared by conventional mixed oxide method. Bi<sub>2</sub>O<sub>3</sub>, ZnO, Nb<sub>2</sub>O<sub>5</sub> and V<sub>2</sub>O<sub>5</sub> (High Purity Chemical Lab., Japan) powders with 99.9% purity were weighed and mixed for 24 h with stabilized zirconia media and ethanol. The mixed powders were calcined at 800–900 °C for 2 h and then ball milled for 24 h. The milled powders were pressed into disks 8 mm in diameter and 2–4 mm thick under a pressure of 1000 kg/cm<sup>2</sup>. Pellets were sintered at 750–1000 °C for 2 h in air with a heating rate of 5 °C/min. The bulk density of the sintered specimens was determined by the Archimedes method. Shrinkage of the specimens

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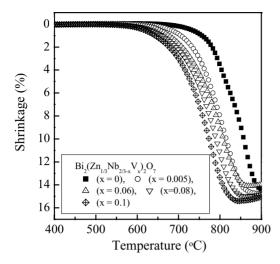


Fig. 1. Shrinkage curves of  $Bi_2(Zn_{1/3}Nb_{2/3_{-x}}V_x)_2O_7$  samples as a function of sintering temperature.

during heating was measured using a horizontal-loading dilatometer with Al<sub>2</sub>O<sub>3</sub> rams and boats (Model DII, 420C, Netzsch Instruments, Germany). The formation of second phase was investigated using powder X-ray diffractometer (XRD: Model M18XHF, Mac Science Instruments, Japan). Polished and thermally etched surfaces of sintered specimens were examined using scanning electron microscopy (SEM: Model XL20, Philips, The Netherlands). EDS line scan was used to characterize the diffusion of silver and the interface reactions. The microwave dielectric properties of sintered samples were measured at x-band frequencies (8–10 GHz) using a network analyzer (model HP8720C, Hewlett Packard, Palo Alto, CA). To determine the dielectric properties in the frequency range from 1 kHz to 10 MHz, the capacitance and  $\tan \delta$  were measured using an impedance/grain-phase analyzer (Model HP 4194A, Hewlett Packard, USA).

## 3. Results and discussion

# 3.1. Sintering behavior of $Bi_2(Zn_{1/3}Nb_{2/3-x}V_x)_2O_7$

Fig. 1 shows shrinkage curves of the green compact of pure  $Bi_2(Zn_{1/3}Nb_{2/3})_2O_7$  (BZN) and  $Bi_2(Zn_{1/3}Nb_{2/3}_xV_x)_2$   $O_7$  (BZNV). As can be seen in the figure, a rapid shrinkage occurred near 700 °C in the case of BZNV specimens. The onset of shrinkage occurred at lower temperatures as x increased. Shrinkage was initiated at approximately 800 °C for pure BZN specimen. This demonstrates that V substitution lowered the sintering temperature about 100 °C.

Generally,  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3})_2\text{O}_7$  is known to be sintered above 950 °C [4], but the bulk density of  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3_{-x}}\text{V}_x)_2\text{O}_7$  with x=0.005 sintered at 850 °C for 2 h reaches ~97% of the theoretical density of  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3})_2\text{O}_7$ , as shown in Fig. 2. Therefore, these results reveal that sintering of  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3})_2\text{O}_7$  below 900 °C is possible by  $\text{V}_2\text{O}_5$ 

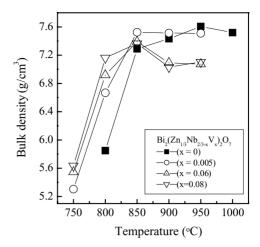


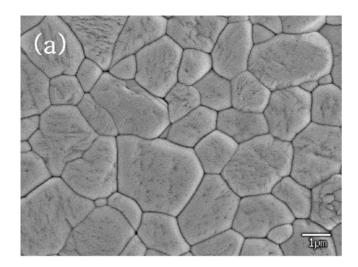
Fig. 2. Bulk density of  ${\rm Bi_2}({\rm Zn_{1/3}Nb_{2/3}}_x{\rm V}_x)_2{\rm O_7}$  samples as a function of sintering temperature.

substitution, which indicates that cofiring with silver inner electrode can be done.

Fig. 3 shows scanning electron micrographs of BZN sintered at 950 °C and BZNV (x=0.005) sintered at 850 °C for 2 h. The dense microstructure of BZNV (x=0.005) was observed. No second phase or liquid phase was observed, and sintered Bi<sub>2</sub>(Zn<sub>1/3</sub>Nb<sub>2/3</sub>)<sub>2</sub>O<sub>7</sub> samples containing V<sub>2</sub>O<sub>5</sub> had a smaller grain size (0.4–1  $\mu$ m) compared with pure Bi<sub>2</sub>(Zn<sub>1/3</sub>Nb<sub>2/3</sub>)<sub>2</sub>O<sub>7</sub> sample sintered at 950 °C for 2 h.

### 3.2. Phase analysis

Fig. 4 shows the XRD patterns of the specimens of  $Bi_2(Zn_{1/3}Nb_{2/3_{-x}}V_x)_2O_7$  (x = 0.0  $\sim$  0.08) sintered at  $850\,^{\circ}\text{C}$  for 2 h. The structure of  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3})_2\text{O}_7$ belongs to a monoclinic zirconolite-like structure [C2/c](No. 15) space group, a = 13.1037(9) Å, b = 7.6735(3) Å, $c = 12.1584(6) \text{ Å}, \beta = 101.318(5)^{\circ}$ , reported by Levin et al. [7]. The range of solid-solution formation for  $Bi_2(Zn_{1/3}Nb_{2/3}, V_x)_2O_7$  was very narrow ( $x \approx 0.005$ ). Second phase of Bi<sub>4</sub>Vi<sub>2</sub>O<sub>11</sub> and BiNbO<sub>4</sub> were found at the specimens of  $Bi_2(Zn_{1/3}Nb_{2/3}V_x)_2O_7$  with  $x \ge 0.01$ . It is of interest that the splitting of the fundamental reflections near 27, 34, 37, 47, and 49° was abruptly decreased with increasing x (Fig. 5). One of the changes found with increasing x is the formation of Bi-based second phases, which indirectly indicates that the matrix phase of Bi<sub>2</sub>( $Zn_{1/3}Nb_{2/3}$ ,  $V_x$ )<sub>2</sub>O<sub>7</sub> was bismuth deficient state. In order to confirm this hypothesis, Bi-deficient phase of  $Bi_{2-\delta}(Zn_{1/3}Nb_{2/3})_2O_{7-(3/2\delta)}$  ( $\delta = 0.1$ ) was prepared and its XRD data is shown in Fig. 5. As can be seen in the figure, Bi<sub>1.9</sub>(Zn<sub>1/3</sub>Nb<sub>2/3</sub>)<sub>2</sub>O<sub>7</sub> exhibited almost same fundamental reflections with  $Bi_2(Zn_{1/3}Nb_{2/3_{-x}}V_x)_2O_7$  (x = 0.08). Therefore, it can be said that Bi<sub>2</sub>(Zn<sub>1/3</sub>Nb<sub>2/3</sub>)<sub>2</sub>O<sub>7</sub> exhibit two phases determined by the Bi stoichiometry.



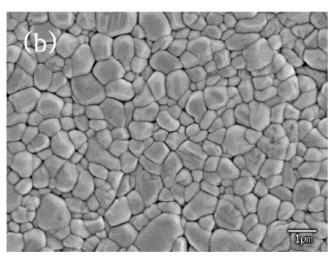


Fig. 3. SEM micrograph of the Bi $_2$ (Zn $_{1/3}$ Nb $_{2/3}$   $_x$ V $_x$ ) $_2$ O $_7$  specimen with (a) x=0.0 sintered at 950  $^{\circ}$ C for 2 h, (b) x=0.005 sintered at 850  $^{\circ}$ C for 2 h.

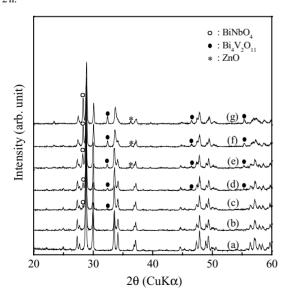


Fig. 4. XRD patterns of  $\mathrm{Bi_2(Zn_{1/3}Nb_{2/3_-x}V_x)_2O_7}$  sintered at 850 °C for 2 h: (a) x=0.0, (b) x=0.005, (c) x=0.01, (d) x=0.02, (e) x=0.04, (f) x=0.06, (g) x=0.08.

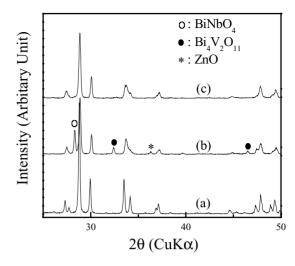


Fig. 5. XRD patterns of  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3_{-x}}\text{V}_x)_2\text{O}_7$  for: (a) x=0.0 sintered at 950 °C for 2 h, (b) x=0.08 sintered at 850 °C for 2 h, (c)  $\text{Bi}_{1.9}(\text{Zn}_{1/3}\text{Nb}_{2/3})_2\text{O}_{6.85}$  sintered at 850 °C for 2 h.

# 3.3. Microwave properties of low-fired $Bi_2(Zn_{1/3}Nb_{2/3-x}V_x)_2O_7$

The dielectric properties of  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3_{-x}}V_x)_2\text{O}_7$  are summarized in Tables 1 and 2.  $\text{Bi}_2(\text{Zn}_{1/3}\text{Nb}_{2/3_{-x}}V_x)_2\text{O}_7$  (x=0.001) sintered at 850 °C had a dielectric constant ( $\varepsilon_r$ ) of 78 and a Qxf value of 3800 measured at x-band. By the way, the dielectric loss was increased at x=0.01 and exhibited large increase at the specimens with  $x\geq0.02$ . Because of large dielectric loss, dielectric properties of the specimens with  $x\geq0.02$  were measured at 1 MHz. This increase of dielectric loss was related to the formation of

Table 1 Microwave properties of Bi<sub>2</sub>(Zn<sub>1/3</sub>Nb<sub>2/3\_x</sub>V<sub>x</sub>)<sub>2</sub>O<sub>7</sub> for  $x = 0.0 \sim 0.1$ 

Samples	Sintering temperature (°C/2 h)	Qxf (GHz)	$\mathcal{E}_{\mathbf{r}}$	Bulk density (g/cm <sup>3</sup> )
BZN	950	2981	76.23	7.6059
BZNV (0.001)	850	3799	78.55	7.5479
BZNV (0.003)	850	3486	78.62	7.5371
BZNV (0.005)	850	3143	78.01	7.5238
BZNV (0.01)	850	2200	78.37	7.5121

Table 2 Dielectric properties of Bi<sub>2</sub>(Zn<sub>1/3</sub>Nb<sub>2/3\_x</sub>V<sub>x</sub>)<sub>2</sub>O<sub>7</sub> for  $x=0.02\sim0.08$  and Bi<sub>4</sub>V<sub>2</sub>O<sub>11</sub> at 1 MHz

. 2					
Samples	Sintering temperature (°C/2 h)	tan δ (1 MHz)	$\mathcal{E}_{\mathrm{r}}$	Bulk density (g/cm <sup>3</sup> )	
BZNV (0.02)	850	0.00024	72.723	7.5141	
BZNV (0.04)	850	0.00154	72.526	7.3938	
BZNV (0.06)	850	0.00185	72.337	7.3910	
BZNV (0.08)	850	0.00276	67.817	7.3648	
$Bi_4V_2O_{11}\\$	750	0.089	87.329	7.2143	
BZNV (0.06) BZNV (0.08)	850 850	0.00185 0.00276	67.817	7.3910 7.3648	

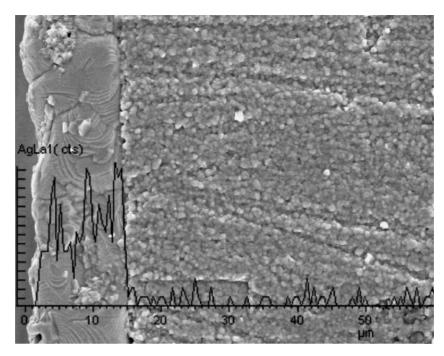


Fig. 6. SEM micrograph and EDS line scan of the interface between a silver electrode and  $Bi_2(Zn_{1/3}Nb_{2/3_{-x}}V_x)_2O_7$  (x = 0.005) cofired at 850 °C for 2 h.

second phase  $Bi_4V_2O_{11}$  since  $Bi_4V_2O_{11}$  ceramic exhibited large dielectric loss;  $\tan \delta$  value of  $\sim 0.089$ .

Fig. 6 shows an SEM micrograph and EDS line scan results of the interface between the silver electrode and Bi<sub>2</sub>  $(Zn_{1/3}Nb_{2/3_{-x}}V_x)_2O_7$  (x=0.005) cofired at 850 °C for 2 h. The silver profile decreases sharply at the interface, which indicates that the reaction of Bi<sub>2</sub> $(Zn_{1/3}Nb_{2/3_{-x}}V_x)_2O_7$  with silver electrode did not occur.

Consequently,  $Bi_2(Zn_{1/3}Nb_{2/3})_2O_7$  with small amount of  $V_2O_5$  substitution can be selected as a suitable candidate for low-temperature cofired ceramics (LTCC) because of its high dielectric constant and compatibility with silver electrode.

### 4. Conclusions

Influence of  $V_2O_5$  substitution to  $Bi_2(Zn_{1/3}Nb_{2/3})_2O_7$  on sintering and dielectric properties were investigated. Substitution of vanadium ion induced two structural changes. One is the formation of Bi-based second phase  $Bi_4V_2O_{11}$  and  $BiNbO_4$ . The other is the change of  $Bi_2(Zn_{1/3}Nb_{2/3})_2O_7$  phase from stoichiometric form to Bi-deficient form. The  $Bi_2(Zn_{1/3}Nb_{2/3}\__xV_x)_2O_7$  (x=0.001) sintered at  $850\,^{\circ}$ C for 2h exhibited good microwave dielectric properties:

Qxf = 3800 at 6 GHz,  $\varepsilon_r = 78.6$  and this composition showed compatibility with silver inner electrode.

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