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Effect of TiO₂ ratio on BaO–Nd₂O₃–TiO₂ microwave ceramics

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Abstract

The dielectric properties of BaO–Nd₂O₃–TiO₂ ceramics with various TiO₂ contents at microwave frequencies were investigated, by varing the range of composition from 1:1:4 to 1:1:5 (BaO:Nd₂O₃:TiO₂). Sintered ceramics were studied by X-ray diffractometer (XRD), scanning electron microscopy (SEM), and dielectric data measured at microwave frequencies. The XRD results revealed that the matrix phase was BaNd₂Ti₅O₁₄, and that BaTi₄O₉ and Ba₂Ti₉O₂₀ secondary phases were present. The dielectric constant saturates at 80–83 with various TiO₂ content. $Q \times f$ values of 5700–11,560 (at 4 GHz) can be obtained by doping with various the ratio of TiO₂, sintering at 1300 °C for 2 h. The temperature coefficient of resonant frequency ranges from 62 to 151 ppm/°C.

Keywords: D. TiO2; Dielectric resonator; Microwave dielectric properties

1. Introduction

Recently, dielectric resonators with high dielectric constant (ε_r) in the range 25–100, low loss ($\tan \delta < 5 \times 10^{-4}$, where $\tan \delta$ is dielectric loss), and small frequency temperature coefficient (τ_f) within ± 20 ppm/°C; have been widely used in microwave integrated circuits and microwave filters [1]. Several materials have been developed as microwave dielectrics, such as Ba₂Ti₉O₂₀ [2], (Zn,Sn)TiO₄ [3], Ba(Mg_{1/3}Ta_{2/3})O₃ [4], Ba(Zn_{1/3}Ta_{2/3})O₃ [5] and BaO–Ln₂O₃–TiO₂ (Ln = La, Nd, Sm) [6–8]. Among these dielectrics with BaO:Nd₂O₃:TiO₂ of around 1:1:4 and a high dielectric constant (ε_r = 70–90) have been commonly applied in mobile telephone systems. They can be used to miniaturize resonance circuits without degrading the Q value.

Numerous approaches existed for modifying the characteristics of microwave dielectrics including (1) doping with additives of Bi₂O₃, Bi₄Ti₃O₁₂ [9], PbO [10] and rare-

earth oxide [6], and (2) varying the composition. The dielectric characteristics of BaO–Nd₂O₃–TiO₂ ceramics strongly depend on their crystal structure, stoichiometry, grain size, additives, and phase composition [11]. Optimizing dielectric properties and ensuring reproducibility are important. This study reports the effect of the proportion of TiO₂ in BaO–Nd₂O₃–TiO₂ ceramics upon their microwave dielectric properties and microstructures.

2. Experiment procedures

2.1. Sample preparation

Samples were prepared from reagent-grade BaCO₃, TiO₂, and Nd₂O₃. The resonator was prepared as BaO–Nd₂O₃–nTiO₂ with n = 4.0, 4.3, 4.7, and 5.0. Initially, appropriate mixtures of the powders were ground finely by ball mill in a plastic jar using deionized water and zirconia balls. After it had been dried, the powder was calcined from 1000–1200 °C for 2 h. Subsequently, an organic binder was added to produce granulation. Cylindrical specimens were pressed

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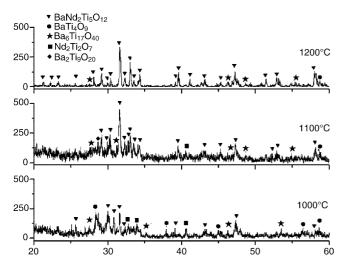


Fig. 1. XRD pattern of BaO-Nd₂O₃-4.7TiO₂ ceramic powder calcined at various temperatures for 2 h.

under a pressure of 1000 kg/cm² and sintered in air at 1300 °C for 2 h.

2.2. Characterization

The calcined and sintered samples were characterized by X-ray powder diffractiometer (XRD, Rigaku D/Max-II). The microstructure of the sintered specimens was observed by scanning electron microscopy (SEM, JEOL JSM-6330F). The bulk density of the sintered pellets was measured by the Arichimedes method. The dielectric constant (ε_r) and the quality factor (Q) at microwave frequencies were measured using the Hakki-Coleman dielectric resonator method [12], as modified and improved by Courtney [13] and Kobayashi [14]. Two dielectric resonators were prepared under the same sintering conditions. They had the same diameter, but one was twice as high as another sample. The temperature coefficient of resonant frequency (τ_f) at microwave frequencies was measured at temperature from 25 to 80 °C, and was defined as follows $\tau_f = \Delta f_o / f_o \Delta T$ (ppm/°C), where Δf_0 is the shift in the center frequency introduced by the temperature change ΔT . The above-mentioned microwave properties were calculated by the resonant frequency of the TE_{011} and TE_{012} resonant modes.

3. Results and discussion

Fig. 1 presents the X-ray diffraction patterns of BaO–Nd₂O₃–4.7TiO₂ calcined powder at different temperatures. It reveals that the ceramics powders calcined at 1000 °C for 2 h contains BaTi₄O₉, Ba₂Ti₉O₂₀, Nd₂Ti₂O₇, Ba₆Ti₁₇O₄₀ and Nd₂Ti₂O₇ phases. As the temperature of calcinations was increased, the BaNd₂Ti₅O₁₄ gradually became major phase. Calcination conditions are well known to be able to affect greatly the densification behavior. However, the densification process primarily affects characteristics of the

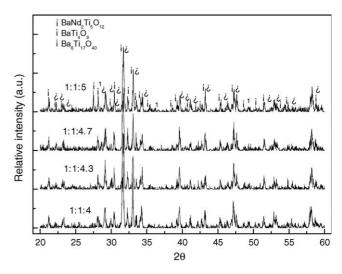


Fig. 2. XRD pattern of BaO–Nd $_2$ O $_3$ –TiO $_2$ ceramics with various TiO $_2$ content, sintered at 1300 $^{\circ}$ C for 2 h.

particles of the calcined powder. A lower calcination temperature will yield in the presence of more unreacted reagents, whereas a high calcination temperature will result in grain growth during calcination. Both calcination conditions suppress densification. Accordingly, these results indicate that a calcination temperature of $1100 \,^{\circ}\text{C}$ is appropriate for BaO–Nd₂O₃–nTiO₂ ceramics (n = 3–5).

BaO-Nd₂O₃-TiO₂ ceramics with various TiO₂, which phase change during heat treatment was examined by X-ray diffraction analyses. Fig. 2 presents the X-ray diffraction patterns of the BaO-Nd₂O₃-TiO₂ ceramics with compositions from 1:1:4 to 1:1:5 (BaO:Nd₂O₃:TiO₂) after sintering at 1300 °C for 2 h. The patterns are similar for all the compositions and indicate that BaNd₂Ti₅O₁₄ is the dominant phase. The patterns in all compositions include additional peaks that correspond to BaTi₄O₉ and Ba₂Ti₉O₂₀. These impurity phases were present probably due to the solid state reaction generated an inhomogeneous microstructure in

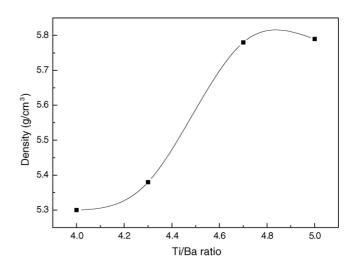


Fig. 3. Densities of BaO–Nd₂O₃–TiO₂ ceramics with various TiO₂ content, sintered at 1300 $^{\circ}$ C for 2 h.

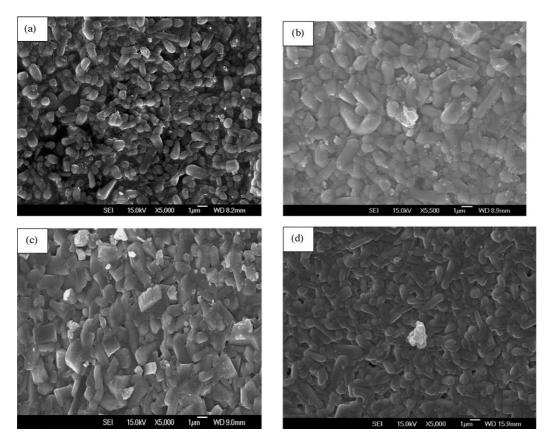


Fig. 4. Microstructures of BaO-Nd₂O₃-TiO₂ ceramics with various TiO₂ content for (a) 1:1:4, (b) 1:1:4.3, (c) 1:1:4.7, and (d) 1:1:5 samples, sintered at 1300 $^{\circ}$ C for 2 h.

BaO-Nd₂O₃-TiO₂ ceramics. Barium titanates compounds may have formed because the performed BaTiO₃ reacted with excess TiO₂. Fig. 3 plots the density of BaO-Nd₂O₃-TiO₂ ceramics against TiO₂ content, sintered at 1300 °C for 2 h. The density of BaO-Nd₂O₃-TiO₂ ceramics increases with TiO₂ content. The above results imply that the density of BaO-Nd₂O₃-TiO₂ ceramics is related to their sintering temperature and the proportion of TiO₂. Fig. 4 displays SEM micrographs of BaO-Nd₂O₃-TiO₂ ceramics with different amount of TiO₂. The microstructures of these ceramics exhibit a uniform grain size of approximately 2-3 µm, with BaNd₂Ti₅O₁₄ as the major phase. The presence of second phases such as BaTi₄O₉ and Ba₂Ti₉O₂₀ is very common in these types of compounds [15]. The porosity of BaO-Nd₂O₃-TiO₂ ceramics declines as TiO₂ content increases. This behavior is associated with the replacement mechanism of Ba²⁺ + 2Nd³⁺ \leftrightarrow 2Ti⁴⁺ [16]. These results are consistent with the findings concerning density.

Various factors influence the microwave properties of dielectric ceramics materials, including the content of individual crystalline, secondary phases, and the degree of densification. Fig. 5 presents the dielectric constant of BaO–Nd₂O₃–TiO₂ ceramics with variousTiO₂ contents, sintered at 1300 °C for 2 h. The dielectric constant increases gradually with TiO₂ content. The relationship between dielectric constant and proportion of TiO₂ follow the same trend as that between density and proportion of TiO₂

content. Fig. 6 plots $Q \times f$ and τ_f of BaO–Nd₂O₃–TiO₂ ceramics as a function proportion of TiO₂, following sintering at 1300 °C for 2 h. As the proportion of TiO₂ increase, the $Q \times f$ value initial declines, and then, reaching a minimum (5700 GHz) at the ratio of 1:1:4.7, whereas BaO–Nd₂O₃–TiO₂ ceramics in the ratio of 1:1:5 exhibit excellent microwave dielectric properties, which the $Q \times f$ value of 11,560 (at 4 GHz). This behavior may be due to the fact that BaO–Nd₂O₃–TiO₂ ceramics in the ratio of 1:1:5

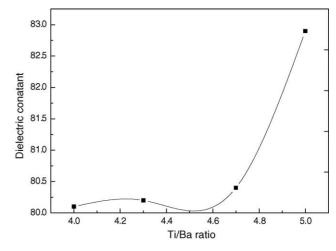


Fig. 5. Dielectric constant of BaO–Nd₂O₃–TiO₂ ceramics with various TiO₂ content, sintered at 1300 $^{\circ}C$ for 2 h.

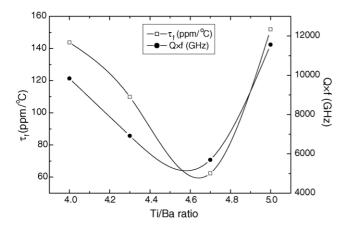


Fig. 6. $Q \times f$ and τ_f of BaO–Nd₂O₃–TiO₂ ceramics with various TiO₂ content, sintered at 1300 °C for 2 h.

exhibit more densification. As indicated in Fig. 4(d), the grains were uniformly and the porosities reduced. Moreover, the $\tau_{\rm f}$ value decreases as the proportion of TiO₂, after reaching a minimum (62 ppm/°C) at the ratio 1:1:4.7, $\tau_{\rm f}$ increases slightly. The relationship between $Q \times f$ value and the proportion of TiO₂ follows the same trend as that between $\tau_{\rm f}$ value and the proportion of TiO₂.

4. Conclusions

In this study, the microwave dielectric properties of BaO– $\mathrm{Nd_2O_3}$ – $\mathrm{TiO_2}$ ceramics with various proportion of $\mathrm{TiO_2}$ were investigated by the conventional solid-state ceramic route. The composition content of $\mathrm{TiO_2}$ was controlled to yield excellent microwave dielectric properties. BaO– $\mathrm{Nd_2O_3}$ – $\mathrm{TiO_2}$ ceramics with an optimum ratio of 1:1:5 exhibits excellent microwave dielectric properties. Their dielectric

constant is 82.9; $Q \times f$ value is 11,560 (at 4 GHz), and $\tau_{\rm f}$ value is 151.2 ppm/°C. These characteristics are appropriate for many microwave devices such as filters, oscillators, patch antenna, and others.

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References

- [1] J.S. Sun, C.C. Wei, L. Wu, J. Mater. Sci. 27 (1992) 5818.
- [2] J.K. Pourde, D.F. Linn, H.N. O'Bryan, J. Thomson, J. Am. Ceram. Soc. 58 (1975) 418.
- [3] G. Wolfram, H.E. Gobel, Mater. Res. Bull. 16 (1981) 1455.
- [4] S. Nomura, K. Toyama, K. Kaweta, Jpn. J. Appl. Phys. 21 (1982) 264.
- [5] S. Kawashina, M. Nishada, I. Ueda, H. Ouchi, J. Am. Ceram. Soc. 66 (1983) 421.
- [6] P. Laffez, G. Desgardin, B. Raveau, J. Mater. Sci. 27 (1992) 5229.
- [7] D. Kolar, Z. Stadler, S. Gaberseck, D. Suvorov, Ferroelectrics 27 (1980) 269.
- [8] J.M. Durand, J.P. Boilot, J. Mater. Sci. Lett. 6 (1987) 134.
- [9] Y.J. Wu, X.M. Chen, J. Eur. Ceram. Soc. 19 (1999) 1123.
- [10] K. Wakino, K. Minai, H. Tamura, J. Am. Ceram. Soc. 67 (1984) 278.
- [11] S. Skapin, D. Kolar, D. Suvorov, Z. Samardzija, J. Mater. Res. 13 (5) (1998) 1327.
- [12] B.W. Hakki, P.D. Coleman, IEEE Trans. Microwave Theory Tech. 8 (1960) 402.
- [13] W.E. Courtney, IEEE Trans. Microwave Theory Tech. 18 (1970) 476.
- [14] Y. Kobayashi, M. Katoh, IEEE Trans. Microwave Theory Tech. 33 (1985) 586
- [15] H. Sreemoolanadhan, M.T. Sebastian, P. Mohanan, Br. Ceram. Trans. 95 (2) (1996) 79.
- [16] K.M. Cruickshank, X. Jing, G. Wood, E.E. Lachowski, A.R. West, J. Am. Ceram. Soc. 79 (1996) 1605.