

## Short communication

 $\text{Ce}^{4+}$ -modified cordierite ceramicsZ.M. Shi<sup>a,\*</sup>, X. Bai<sup>b</sup>, X.F. Wang<sup>a</sup><sup>a</sup>*School of Materials Science and Engineering, Inner Mongolia University of Technology, 010062 Hohhot, China*<sup>b</sup>*School of Chemical Engineering, Inner Mongolia University of Technology, 010062 Hohhot, China*

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**Abstract**

A dense, low thermal expansion cordierite ceramic was prepared by the co-precipitation method and  $\text{Ce}^{4+}$  addition.  $\text{Ce}^{4+}$  addition can inhibit  $\mu$ -cordierite crystallization and improve  $\alpha$ -cordierite crystallization. Adding 3.2 wt.%  $\text{Ce}^{4+}$  enables density and mechanical strength to reach maximum values with a minor increase in thermal expansion coefficient.

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**1. Introduction**

Cordierite ceramic is a good candidate for the honeycomb structure of catalyst converters and diesel particulate trappers due to its low thermal expansion coefficient [1]. In order to increase the catalytic and trapping efficiency, it is necessary to decrease the thickness of walls of the honeycomb and to increase cell density. Because the durability of the honeycomb depends on its strength and thermal stability [2], the ceramic should possess enough high strength and a low thermal expansion coefficient to avoid wall failing.

Stoichiometric cordierite ceramic fabricated by oxide-powder sintering possesses very low density and strength [3,4]. Therefore, many processes have been adopted to improve its mechanical and thermal expansion properties. One of the processes is adding a low-melting flux such as  $\text{K}_2\text{O}$ ,  $\text{Na}_2\text{O}$ ,  $\text{Li}_2\text{O}$ ,  $\text{CaO}$ ,  $\text{B}_2\text{O}_3$ ,  $\text{NaF}$  and  $\text{LiF}$  [3,5–9]. The flux is helpful for obtaining dense cordierite ceramic and improving cordierite transformation, but it will deteriorate the thermal expansion coefficient once a glass phase occurs. The sol-gel method is another route to prepare high-performance cordierite ceramics [10–12]. Suzuki [13] observed that the densification of cordierite ceramics

prepared by this process appeared at 800–900 °C without any sintering aid. Sintering at 1300 °C for 12 h provided a dense  $\alpha$ -cordierite ceramic with a flexural strength of 120 MPa. Addition of rare earths to cordierite ceramics can decrease sintering temperature and porosity, facilitate cordierite transformation and result in less increase of the thermal expansion coefficient [14–17]. Okamura [18] found that a dense cordierite glass-ceramic was obtained by adding  $\text{La}_2\text{O}_3$  at sintering temperatures between 1355 and 1420 °C. Moreover, cordierite ceramics can be reinforced by graphite,  $\text{SiC}$  and stainless-steel fibers or mullite and zirconia secondary phases [19–23].

An additive, for cordierite ceramic to be effective, not only should increase the mechanical properties but also limit the thermal expansion coefficient. The present work aims at preparing high density and low thermal expansion coefficient cordierite ceramic by the co-precipitation method and  $\text{Ce}^{4+}$  addition.

**2. Experimental procedures**

Tetraethyl silicate (TEOS), aluminum isopropoxide ( $\text{Al}(\text{OC}_3\text{H}_7)_3$ ), magnesium isopropoxide ( $\text{Mg}(\text{OC}_2\text{H}_5)_2$ ) and cerium nitrate ( $\text{Ce}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ ) were used as precursory materials. The sheer hydrolyzed alkoxides and nitrate were mixed into solutions with stoichiometric

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cordierite composition and different amounts of  $\text{Ce}^{4+}$ , and strongly stirred to form sols. NaOH solution was added into the sols to precipitate amorphous powders. Such powders were cleaned with deionized water, dried at 100 °C, calcined at 450 °C, ball-milled for 3 h, mechanically pressed at 50 MPa, and ultimately sintered at different temperatures in air atmosphere.

The phase analysis was conducted by XRD (D/max-2400X, Rigaku, with Cu K $\alpha$  radiation, 40 kV, 120 mA and scanning speed of 4°/min). Backscattered electron images of surface-polished samples covered with carbon by ion-sputtering were observed by SEM (CSM950, Opton). The apparent porosity of samples was measured by the water displacement technique. The three-point flexural strength of 3 mm  $\times$  4 mm  $\times$  40 mm samples was measured by using a testing machine (AGS-10KNG, Shimadzu) with 30 mm span and loading rate of 0.5 mm/min. The thermal expansion coefficient was measured with a TMA92, Setaram apparatus at a heating rate of 10 °C/min and in Ar protective atmosphere.

### 3. Results and discussions

#### 3.1. Crystallization

Fig. 1 shows the XRD patterns of ceramics sintered at different temperatures, and Fig. 2, the morphology of  $\text{CeO}_2$  particles in the ceramics with different  $\text{Ce}^{4+}$  additions.  $\text{CeO}_2$  easily precipitates from the  $\text{Ce}^{4+}$ -modified amorphous at low temperatures, and the content of  $\text{CeO}_2$  particles increase with the increase of the  $\text{Ce}^{4+}$  addition. When sintered at 1200 °C for 3 h, the  $\text{Ce}^{4+}$ -modified ceramics are composed of  $\alpha$ -cordierite and  $\text{CeO}_2$ , while the  $\text{Ce}^{4+}$ -free ceramic is composed of larger amounts of  $\mu$ -cordierite and small amounts of  $\alpha$ -cordierite phases. This proves  $\text{Ce}^{4+}$  addition to promote the transformation of  $\mu$ -cordierite to  $\alpha$ -cordierite.

This is related to  $\text{Ce}^{4+}$  densifying the microstructure of glass or amorphous phase [24] due to its high valence (+4) and large coordination number [25], so the  $\text{Ce}^{4+}$ -modified amorphous gets more stable than the  $\text{Ce}^{4+}$ -free amorphous towards  $\mu$ -cordierite crystallization. Moreover, because of the insignificant solubility of  $\text{Ce}^{4+}$  in crystalline cordierite, and no reaction between  $\text{Ce}^{4+}$  and cordierite occurring [26],  $\text{Ce}^{4+}$  concentration in the amorphous decreases with the precipitation of  $\text{CeO}_2$  particles, resulting in a decreasing stability and so an incomplete crystallization of  $\mu$ -cordierite.

Since the transformation of  $\alpha$ -cordierite in the  $\text{Ce}^{4+}$ -modified ceramics is quite completed when sintered at 1200 °C, we selected this sintering temperature to examine the effect of  $\text{Ce}^{4+}$  addition on the microstructure and properties for all of the samples.

#### 3.2. Densification and mechanical properties

Fig. 3 shows the variation of linear shrinkage with sintering temperature. Remarkable shrinkage appears at

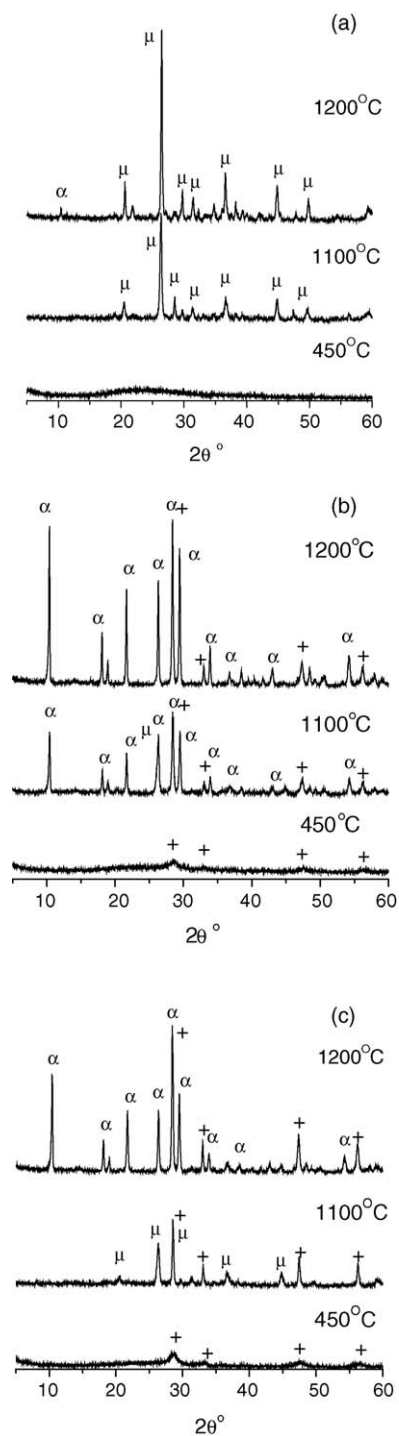


Fig. 1. XRD patterns of ceramics ( $\alpha$ :  $\alpha$ -cordierite;  $\mu$ :  $\mu$ -cordierite; +:  $\text{CeO}_2$ ; soaking time: 3 h). (a)  $\text{Ce}^{4+}$ -free. (b) 3.2 wt.% of  $\text{Ce}^{4+}$ . (c) 4.8 wt.% of  $\text{Ce}^{4+}$ .

temperatures less than 900 °C below cordierite crystallization. The largest shrinkage occurred at 3.2 wt.%  $\text{Ce}^{4+}$  addition. Fig. 4 shows porosity and flexural strength versus  $\text{Ce}^{4+}$  addition. The lowest porosity and the highest flexural strength also appeared at 3.2 wt.%  $\text{Ce}^{4+}$ .

The  $\text{Ce}^{4+}$  addition may decrease the softening temperature of the amorphous [17], so the density may increase with

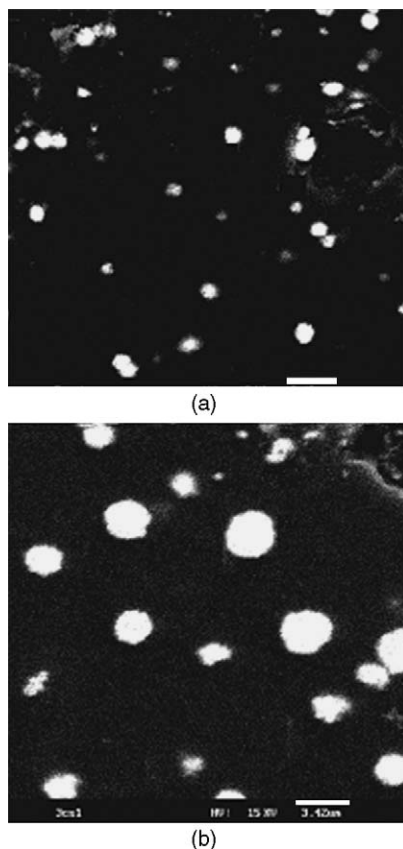


Fig. 2. Backscattered electron images of Ce<sup>4+</sup>-modified cordierite ceramics. (a) 3.2 wt% of Ce<sup>4+</sup>. (b) 4.8 wt% of Ce<sup>4+</sup>.

Ce<sup>4+</sup> additions. On the other hand, the CeO<sub>2</sub> particles inhibit the densification, the effect becoming stronger with the increase of both particle content and size. This simultaneous action results in the highest density and strength at 3.2 wt.% of Ce<sup>4+</sup>.

### 3.3. Thermal expansion coefficient

The thermal expansion coefficient of the ceramics sintered at 1200 °C for 3 h increases with Ce<sup>4+</sup> content and temperature (Fig. 5). It can be found that the Ce<sup>4+</sup>-modified cordierite ceramics no longer present the intrinsic

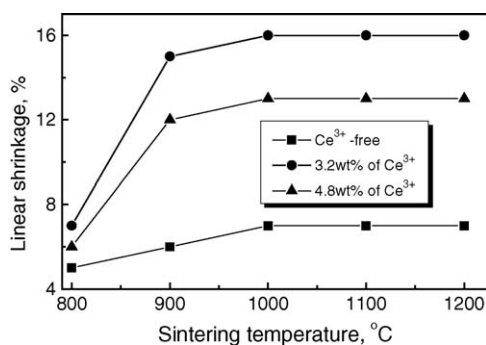


Fig. 3. Linear shrinkage of ceramics sintered at different temperatures for 2 h.

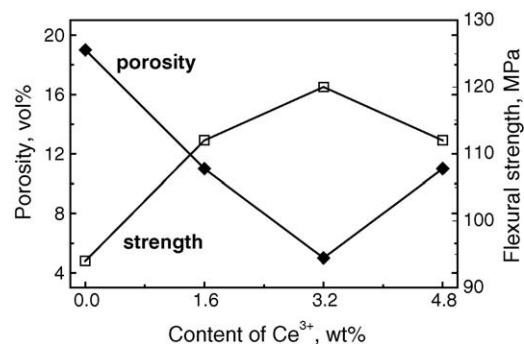


Fig. 4. Variations of porosity and flexural strength with Ce<sup>4+</sup> content.

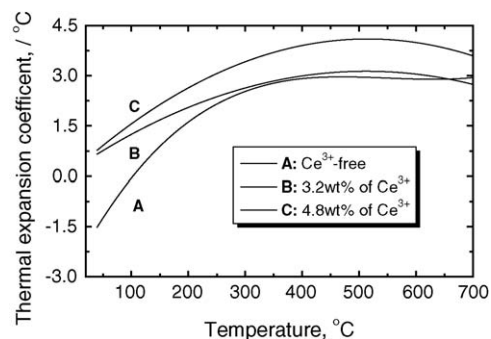


Fig. 5. Variation of thermal expansion coefficient with Ce<sup>4+</sup> content and temperature.

character of negative thermal expansion at low temperatures. The averages of thermal expansion coefficients are  $2.22 \times 10^{-6}/^{\circ}\text{C}$ ,  $2.48 \times 10^{-6}/^{\circ}\text{C}$  and  $3.21 \times 10^{-6}/^{\circ}\text{C}$  in 20–700 °C for the ceramics with 0, 3.2 and 4.8 wt.% Ce<sup>4+</sup> additions, respectively.

Because of the existence of large amounts of  $\mu$ -cordierite, the thermal expansion coefficient of the Ce<sup>4+</sup>-free ceramic is higher than for stoichiometric  $\alpha$ -cordierite crystal ( $1.5 \times 10^{-6}/^{\circ}\text{C}$ ). Moreover, because the CeO<sub>2</sub> particles increase the thermal expansion coefficient, the Ce<sup>4+</sup>-modified ceramic possesses a higher thermal expansion coefficient than the Ce<sup>4+</sup>-free.

## 4. Conclusions

The addition of Ce<sup>4+</sup> effectively depresses the formation of  $\mu$ -cordierite and stimulates the transformation of  $\alpha$ -cordierite. The Ce<sup>4+</sup>-modified cordierite ceramic, when sintered at 1200 °C for 3 h, is only composed of  $\alpha$ -cordierite and CeO<sub>2</sub> particles.

The density and strength of the cordierite ceramics reach the maximum values at 3.2 wt.% Ce<sup>4+</sup> addition, and the thermal expansion coefficient increases with Ce<sup>4+</sup> content and temperature.

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