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Addition of Al–Ti–B master alloys to improve the performances of alumina matrix ceramic materials

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Abstract

In the present work, a new technique to improve the performances of alumina matrix ceramic materials is presented, in which Al, Al $_3$ Ti and TiB $_2$ are incorporated into alumina matrix ceramic materials in the form of Al–Ti–B master alloys. Composites of Al $_2$ O $_3$ /TiB $_2$ /AlN/TiN are fabricated by the technology of transient liquid phase sintering, during which new phases such as AlN and TiN are produced by the chemical reactions taking place among Al, Ti and N $_2$ (the protective atmosphere). The densification rate of the composites as a function of Al–Ti–B volume content is discussed. The fundamental properties of the composites such as hardness, fracture toughness and bending strength are examined. The relations of volume content of Al–Ti–B master alloys and mechanical properties of alumina matrix ceramic materials are analyzed. The effects of fracture mechanism on mechanical properties of the composites are researched together with the refining performances of Al–Ti–B master alloys. © 2006 Elsevier Ltd and Techna Group S.r.l. All rights reserved.

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1. Introduction

With several special properties like high hardness, good chemical inertness, high wear resistance and low cost, alumina is attracting considerable interest for advanced engineering applications such as refractory materials, grinding media, cutting tools, and high temperature bearings. However, the brittleness of monolithic Al₂O₃ limits its application in many engineering fields. It has been a key subject for many years that how to improve the bending strength and fracture toughness of monolithic Al₂O₃ [1–8]. Researchers have attempted to improve the mechanical properties of monolithic Al₂O₃ by incorporating intermetallic compounds such as Fe₃Al [9], Fe–Al [10] and AlTiC [11] into alumina matrix, and many interesting results have been obtained. The low price of Al–Ti–B master alloys permits its widespread applications to the refining technology of Al and its alloys [12,13], however, there

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are few articles reporting application of Al-Ti-B master alloys on refining alumina matrix ceramic materials.

Cai et al. [14] considered that AlN, which was formed by the reaction of Al and N₂ during the sintering process, could make significantly improvements in mechanical properties of Al₂O₃/ TiC composites. As additives, TiN [15,16] and TiB₂ [17,18] also play important roles in toughening alumina matrix ceramic materials. Being composed of Al, Al₃Ti and TiB₂ phases [19], Al–Ti–B master alloys are introduced in alumina matrix ceramic materials with different volume contents, and N₂ is used as the protective atmosphere in the hot-pressing sintering process. The specific objective of the study is to determine whether Al in Al–Ti–B and Ti, being separated from Al₃Ti, can react with N₂ to produce AlN and TiN. Newly formed AlN and TiN phases can cooperate with TiB₂ to improve the performance of alumina matrix materials.

2. Experimental procedure

Commercial Al_2O_3 powder of high purity (99.99%) and small grain size (0.5–1 μ m) was used as the starting materials. Al–Ti–B master alloys, which are developed by the department

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Table 1 Compositions of hot pressed composites

Specimens	Composition
AT_0	100 vol.% Al ₂ O ₃ + 0 vol.% Al–Ti–B
AT_5	95 vol.% Al ₂ O ₃ + 5 vol.% Al–Ti–B
AT_{10}	90 vol.% Al ₂ O ₃ + 10 vol.% Al-Ti-B
AT_{15}	85 vol.% Al ₂ O ₃ + 15 vol.% Al–Ti–B
AT_{20}	80 vol.% Al ₂ O ₃ + 20 vol.% Al-Ti-B
AT_{25}	75 vol.% $Al_2O_3 + 25$ vol.% $Al-Ti-B$

of materials science and engineering in Shandong University, were used as additives. The alloy used in present study has the composition Al–5% Ti–1% B. The volume content of Al–Ti–B master alloys ranged from 5 to 25 vol.% as listed in Table 1. (The suffix in AT $_0$, AT $_5$, AT $_{10}$, AT $_{15}$, AT $_{20}$ and AT $_{25}$ represents the volume content of Al–Ti–B. For example, AT $_5$ means the volume content of Al–Ti–B is 5 vol.%.)

The raw materials were blended with one other according to certain proportions and ball milled for 60 h in an alcohol medium to obtain a homogeneous mixture. Then the slurry was dried in vacuum and screened. Hot-pressing was used to sinter the powder mixture in a graphite die under nitrogen protection. The specimens were heated up to $1600~^{\circ}\text{C}$ ($30~^{\circ}\text{C}$ min $^{-1}$) under a pressure of 30 MPa for 35 min.

The sintered specimens were machined with a grinding machine. Standard test pieces (3 mm \times 4 mm \times 36 mm) were obtained through rough grinding, finish grinding with diamond wheels, and polishing. Three-point-bending mode was used to measure the bending strength with a span of 20 mm and at a crosshead speed of 0.5 mm/min. The bending strength was calculated by the following formula [20]:

$$\sigma_{\rm f} = \frac{3PL}{2bh^2} \tag{1}$$

where σ_f is bending strength (MPa) and P is load (N) under which the samples break, b and h are width and height (mm), respectively, and L is span (mm).

Vickers hardness was measured on polished surface with a load of 9.8 N for 5 s with a micro-hardness tester (MH-6). Fracture toughness measurement was performed using indentation method with a hardness tester (Hv-120), and results were obtained by the formula proposed by Cook and Lawn [21].

The X-ray diffraction (XRD) analysis was undertaken to identify the crystal phases present before and after sintering. Microstructural observations of fracture surfaces were done by scanning electron microscopy (HITACHI S-570).

3. Results and discussion

3.1. X-ray diffraction phase analysis

The X-ray diffraction analysis of AT₁₅ powders before sintering and AT₁₅ specimen after sintering at 1600 °C, under 30 MPa for 35 min, are shown in Figs. 1 and 2, respectively. It is clear from Fig. 1 that there exist Al₂O₃, TiB₂, Al₃Ti and Al phases in the ball-milled powders, which indicates that Al-Ti-

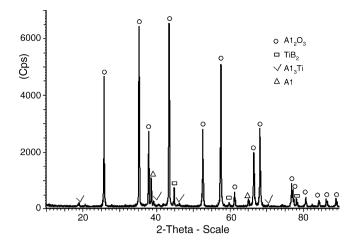


Fig. 1. XRD pattern of AT₁₅ powder before sintering

B master alloys are composed of TiB₂, Al₃Ti and Al phases [19]. Further, there exist in AT₁₅ specimen the Al₂O₃, TiB₂, TiN and AlN phases. Newly formed TiN and AlN may be produced in the following ways: when the temperature goes up to 1000 °C [22], labile Al₃Ti will dissolve to deliver Ti and Al particles:

$$Al_3Ti \rightarrow 3Al + Ti$$
 (2)

The sintering temperature is $1600\,^{\circ}\text{C}$, which is higher than the melting point of Al ($660\,^{\circ}\text{C}$). Hence, previously mentioned Al in Al–Ti–B master alloys and the delivered Al may be in the molten state, some may escape during the sintering process, the other reacting with N_2 (the protective atmosphere) to form the new phase AlN:

$$2Al + N_2 \rightarrow 2AlN \tag{3}$$

TiN is from the reaction taking place between delivered Ti and N_2 :

$$2Ti + N_2 \rightarrow 2TiN \tag{4}$$

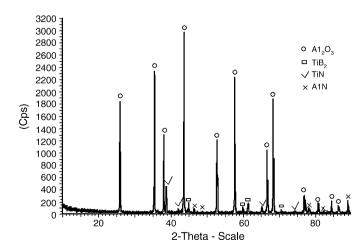


Fig. 2. XRD pattern of AT₁₅ specimen after sintering.

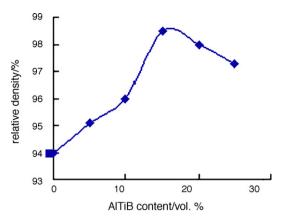


Fig. 3. Relation of Al-Ti-B content and relative density.

Based on thermodynamic analysis of the reactions, we can get the following rankings [23]:

$$\begin{split} \Delta G_{T2\,(1273\,\mathrm{K})}^{\theta} &= -182549.9635 < 0, \qquad \Delta G_{T3\,(1873\,\mathrm{K})}^{\theta} \\ &= -218483.1739 < 0, \qquad \Delta G_{T4\,(1873\,\mathrm{K})}^{\theta} \\ &= -324689.4844 < 0 \end{split}$$

where $\Delta G_{T2\,(1273\,\mathrm{K})}^{\theta}$, $\Delta G_{T3\,(1873\,\mathrm{K})}^{\theta}$ and $\Delta G_{T4\,(1873\,\mathrm{K})}^{\theta}$ represent the Gibbs free energy of the reactions (2)–(4), respectively, and the temperature are 1000, 1600, and 1600 °C, respectively. The XRD analysis shows that Al₂O₃, TiB₂, TiN and AlN phases actually exist in sintered specimen of AT₁₅ as previously mentioned.

3.2. Densification of the composites

Fig. 3 shows the relative density curves for AT_0 , AT_5 , AT_{10} , AT_{15} , AT_{20} and AT_{25} sintered at $1600\,^{\circ}$ C, under 30 MPa for 35 min. It is clear from the figure that the relative density of AT_0 is only 94%, and that of the $Al_2O_3/TiB_2/AlN/TiN$ composites increases sharply (up to 98.5%) as the addition amount of Al-Ti-B master alloys is raised from 0 to 15 vol.%. This trend reverses with further addition of Al-Ti-B master alloys, and the relative density of AT_{25} specimen decreases to 97.3%. This is mainly because the amount of TiB_2 , which is difficult to be densified [24], is increasing. Obviously, the densification rate of

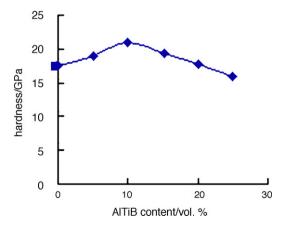


Fig. 4. Relation of Al-Ti-B content and hardness.

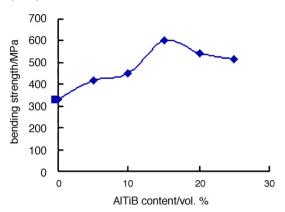


Fig. 5. Relation of Al-Ti-B content and bending strength.

the composites increases with the volume content of Al–Ti–B master alloys increasing from 0 to 15%. The highest relative density value of AT_{15} is enhanced by 4.8% compared to that of monolithic Al_2O_3 .

3.3. Mechanical properties

The relations of Al-Ti-B volume content and hardness. bending strength, fracture toughness of hot-pressed alumina matrix ceramic materials are shown in Figs. 4–6, respectively. The hardness and bending strength of monolithic Al₂O₃ is respectively 17.64 GPa and 331.5 MPa, and the fracture toughness is 3.52 MPa m^{1/2}. Kim et al. [25] have observed that the mechanical properties such as hardness, fracture toughness and strength of the composites increase as the improvements in the relative density of sintered specimens are obtained. As seen from Figs. 4–6, the variation in hardness, fracture toughness and strength of the composites with Al-Ti-B volume content is analogous to that of relative density curve. The hardness of the composites increases as Al–Ti–B content is raised from 0 to 10 vol.% and reaches its maximum value of 21 GPa, then decreases from 10 to 25 vol.%. As a general rule, the hardness of the composites should decrease with the increasing amount of low-hardness Al-Ti-B, whereas the curve in Fig. 4 is a parabola with a maximum value. There are two factors influencing the hardness of composites, namely a

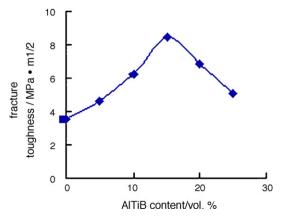


Fig. 6. Relation of Al-Ti-B content and fracture toughness.

hardness effect due to the addition of Al-Ti-B master alloys and a densification effect of the composites. The hardness of the composites increases for the reason that the densification effect is stronger than the hardness effect, otherwise the hardness of the composites will decrease. In the first stage (0 to 10 vol.% addition), the densification effect is dominant and the hardness is enhanced, while the hardness is reduced as the hardness effect is in turn becoming dominant (10-25 vol.% addition). The effects of Al-Ti-B master alloys on the bending strength and fracture toughness of the composites are similar to each other as shown in Figs. 5 and 6. They increase with the increasing amount of Al-Ti-B master alloys before 15 vol.%, and then reach their maximum values. For 15 vol.%, the bending strength and toughness are 600 MPa, 8.44 MPa m^{1/2}, respectively, and decrease after 15 vol.%. This trend of bending strength and fracture toughness correlates with the microstructures of the composites, which will be discussed in the following section.

It is obvious that the fabricated $Al_2O_3/TiB_2/AlN/TiN$ composites, sintered at $1600\,^{\circ}C$, under $30\,MPa$ for $35\,min$, exhibit significant improvements in mechanical properties than monolithic Al_2O_3 . Composite with the addition of $15\,vol.\%$ Al–Ti–B master alloys shows better comprehensive performances, the hardness, bending strength and the fracture toughness of the composite reach $19.3\,GPa$, $600\,MPa$ and $8.44\,MPa\,m^{1/2}$, respectively, which are enhanced by $9.41,\,81$ and 139.8%, respectively, with respect to monolithic Al_2O_3 sintered under the same conditions.

3.4. Analysis of microstructures

SEM photomicrograph of fracture surface of monolithic Al_2O_3 sintered at 1600 °C is shown in Fig. 7, those of AT_5 , AT_{15} and AT_{25} are shown in Figs. 8–10, respectively. There are

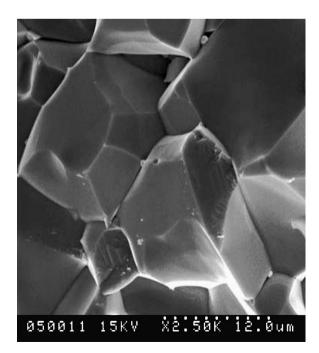


Fig. 7. SEM photomicrograph of monolithic Al₂O₃.

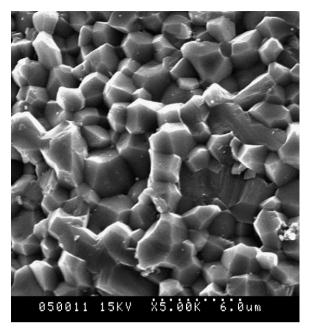


Fig. 8. SEM photomicrograph of AT₅ specimen.

significant microstructural differences among the composites. The fracture mode of monolithic Al₂O₃ is mainly intergranular failure, and the mean grains size is about 20 μm. The addition of Al–Ti–B master alloys makes the microstructures of the composites finer and more homogeneous. The fracture mode keeps invariant with incorporation of 5 vol.% Al–Ti–B master alloys and then turns to the combination of transgranular failure and intergranular failure as the addition amount of Al–Ti–B master alloys reaches 15 vol.%, as is shown in Figs. 8 and 9. Further addition of Al–Ti–B master alloys (25 vol.%) makes the fracture mode mainly intergranular failure, which is accompanied with little transgranular failure (Fig. 10). It can be

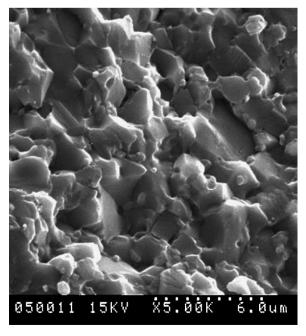


Fig. 9. SEM photomicrograph of AT₁₅ specimen.

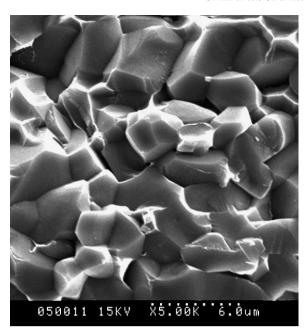


Fig. 10. SEM photomicrograph of AT₂₅ specimen.

seen from Fig. 9, the bondings of grains become stronger when the fracture mode turns to the combination of transgranular failure and intergranular failure. In other words, a fracture mode changing from intergranular failure to the combination of transgranular failure and intergranular failure is observed and may bring on a significant increase in bending strength and fracture toughness. The main cause is that the grain boundaries of the composites are strengthened owing to the addition of Al—Ti–B master alloys.

The mean grains size of monolithic Al_2O_3 is about 20 μ m as previous mentioned, while that of AT₅, AT₁₅ and AT₂₅ are 3, 2 and 5 µm, respectively. The grains size of monolithic Al₂O₃ is larger than that of the composites toughened by Al-Ti-B master alloys under identical conditions. It indicates that the grains of Al₂O₃ in composites may be restrained from abnormal growth owing to the addition of Al-Ti-B master alloys. The reason is as follows, Al-Ti-B master alloys are composed of TiB₂, Al₃Ti and Al phases, the melting point of Al is 660 °C, so the liquid-phase sintering begins as the temperature goes above 700 °C. The reaction between Al and N₂ first occurs on the surface of the disc compacts. However, this development of the reaction is slow owing to high chemical bond strength of Nitrogen molecule and it is difficult for N₂ to infiltrate through the compacted AlN by reason of the very low solubility of N₂ in Al and strong compactness of AlN. So it takes long time to complete the transformation of Al to AlN [14]. TiAl₃ can be a very effective nucleant for aluminum [13], but this phase dissolves quickly under the sintering temperature and then Ti is provided by the dissolution of TiAl₃. It has been reported that Ti plays a significant role in refining pure Al [26]. Furthermore, TiB₂ particles are dispersed and being virtually insoluble in molten aluminum, act as heterogeneous nucleation sites. The behavior is merely a result of particle setting and agglomeration [12]. The existing of Ti changes the interface between liquid Al and solid alumina, and its reaction with N₂ or O (O is provided by alumina, the compounds of Ti and O are too small to be detected by XRD analysis.) improves the wettability of Al and alumina [27,28]. When the temperature is 800–1000 °C, dissolution of alumina in Al occurs, and upon cooling alumina crystals heteroepitaxially nucleate on the pre-existing alumina grains, leading to grain refinement. At the end of the sintering process, expect those may have escaped by the way of flowing out or vaporizing, Al is almost entirely transformed into AlN, which has a grains size on the order of nanometers. These nanometer grains of AlN and TiN, which are formed by in situ synthesis, distribute evenly over alumina matrix ceramic materials and contribute to the refining effect [14].

4. Conclusion

Al₂O₃/TiB₂/AlN/TiN composites are fabricated by hotpressing technology. Transient liquid phase sintering comes into being, during which new phases such as AlN and TiN are produced by the chemical reactions taking place among Al, Ti and N₂. The relative density of AT₁₅ specimen reaches its maximum value of 98.5%. The composite with the addition of 15 vol.% Al-Ti-B master alloys shows better comprehensive performances, and the hardness, bending strength and the fracture toughness of the composite are 19.3 GPa, 600 MPa and 8.44 MPa m^{1/2}, respectively. The fracture mode of monolithic Al₂O₃ is mainly intergranular failure. The addition of Al-Ti-B master alloys makes the microstructures of the composites finer and more homogeneous. The fracture mode turns to the combination of transgranular failure and intergranular failure as the addition amount of Al-Ti-B master alloys is 15 vol.%, which may bring on the increase in bending strength and fracture toughness. One of the main causes is that the grain boundaries of the composites are strengthened, and the other is that the grains of Al₂O₃ in composites may be restrained from abnormal growing owing to the addition of Al-Ti-B master alloys.

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