

Dielectric properties and microstructures of CaSiO_3 ceramics with B_2O_3 addition

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Abstract

The effects of B_2O_3 additives on the sintering behavior, microstructure and dielectric properties of CaSiO_3 ceramics have been investigated. The B_2O_3 addition resulted in the emergence of $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass phase, which was advantageous to lower the synthesis temperature of CaSiO_3 crystal phase, and could effectively lower the densification temperature of CaSiO_3 ceramic to as low as 1100 °C. The 6 wt% B_2O_3 -doped CaSiO_3 ceramic sintered at 1100 °C possessed good dielectric properties: $\epsilon_r = 6.84$ and $\tan \delta = 6.9 \times 10^{-4}$ (1 MHz).

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1. Introduction

Low temperature co-fired ceramic (LTCC) technology has been playing an important role in modern wireless communication systems. In order to process ceramic with electrode material, such as 30 Pd/70 Ag (mp 1167 °C) and silver (mp 961 °C), it is required to sinter the dielectrics at temperatures lower than the melting temperature of the co-fired electrode material. Research on the development of co-fired dielectrics with different dielectric constants ϵ_r has been going for years to fill the need for various RF ranges [1–6]. However, low ϵ_r material is still the most popularly used one, such as DuPont 951 (ϵ_r 7.8, $Q \times f$ 700 GHz at 3 GHz, τ_f 8 ppm/°C), since it provides fast transmission in communication systems [7,8].

CaSiO_3 ceramic was proved as a low-permittivity dielectric ceramic system and possessing good dielectric properties: $\epsilon_r = 5$, $\tan \delta = (1.0\text{--}3.0) \times 10^{-4}$ [9]. However, the sintering temperature range of pure CaSiO_3 was very narrow and it was difficult to obtain dense CaSiO_3 ceramics materials by the traditional solid-state preparation process [10]. The synthesis and microwave dielectric properties of CaSiO_3 nanopowder prepared by sol–gel method have been investigated, and it was

also difficult to obtain dense CaSiO_3 ceramics by sintering the CaSiO_3 nanopowders at 1280–1320 °C [11].

In order to improve the sintering characteristic, the study on the synthesis of $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ (CBS) glass-ceramics and decreasing sintering temperature with aids have been investigated. Zhu et al. found that the optimal sintering temperature for $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass-ceramic is 820 °C for 15 min, and the glass-ceramic possesses excellent dielectric properties: $\epsilon_r = 6.5$, $\tan \delta < 2 \times 10^{-3}$ at 10 MHz with the major phases are CaSiO_3 , CaB_2O_4 and SiO_2 [12]. Chang and co-workers demonstrated that the dielectric properties of the CBS system which comprised of CaSiO_3 , $\text{Ca}_6\text{Si}_4\text{O}_3$ and CaB_2O_4 are: $\epsilon_r = 5.1$, $\tan \delta = 0.0013$ (1 GHz) [13,14]. Cai et al. reported that the excessive SiO_2 and clay can reduce the sintering temperature (1170–1230 °C) of CaSiO_3 , and the dielectric properties are: $\epsilon_r = 7.0\text{--}9.0$, $\tan \delta = (1.0\text{--}5.0) \times 10^{-4}$ [9].

In this work, B_2O_3 was used as the sintering aid to lower the sintering temperature of CaSiO_3 ceramic. The resultant dielectric properties were analyzed based upon the densification, the X-ray diffraction patterns and the microstructures of the ceramics.

2. Experimental procedure

Specimen powders were prepared by a conventional solid-state method. High-purity oxide powders (>99.5%): CaCO_3 ,

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SiO₂ and B₂O₃ were used as raw materials. The powders were weighed according to the composition of CaSiO₃, and ground in ethanol for 24 h in a balling mill with agate balls. Prepared powders were dried and calcined at 1100–1200 °C for 2 h in air. The powders calcined at 1200 °C were mixed as desired composition of CaSiO₃ with 0.5–8 wt% B₂O₃ additions and remilled for 24 h. The fine powder together with the organic binder (5 wt% PVA) was pressed into pellets with dimensions of 15 mm in diameter and 5 mm in thickness. These pellets were sintered at temperatures of 1000–1200 °C for 2 h in air with the heating rate of 5 °C/min.

The crystalline phases of calcined powders and sintered ceramics were identified by X-ray diffraction pattern (XRD, ARL XTRA, Cu K_α). The microstructure observation of the sintered ceramics was performed by means of scanning electron microscopy (SEM, JSM-5601). The bulk densities of the sintered pellets were measured by the Archimedes method. The dielectric constant (ϵ_r) and dielectric loss ($\tan \delta$) at room temperature were determined from capacitance measurements by an LCR meter (HEWLETT PACKARD 4278A) at 1 MHz, and the electric voltage value was 1 V. The dielectric constant was calculated from a parallel-plate capacitor equation, e.g., $\epsilon_r = Cd/\epsilon_0 A$, where C was the capacitance of the specimens, d and A were, respectively, the thickness and the area of the electrode, and ϵ_0 was the dielectric permittivity of vacuum (8.854×10^{-12} F/m). Silver paste was used for the electrodes.

3. Results and discussion

Fig. 1 shows the X-ray diffraction patterns of CaCO₃–SiO₂ powders calcined at different temperatures. For the sample calcined at 1100 °C, the SiO₂ phase was predominant with a great deal of CaSiO₃ phase. As the calcination temperature was raised to 1200 °C, the CaSiO₃ phase became the main phase accompanied with a little SiO₂ phase.

The X-ray diffraction patterns of CaSiO₃ ceramics sintered at 1100 °C with different B₂O₃ additions are shown in Fig. 2. With 0.5 wt% B₂O₃ addition, the single crystal phase of CaSiO₃

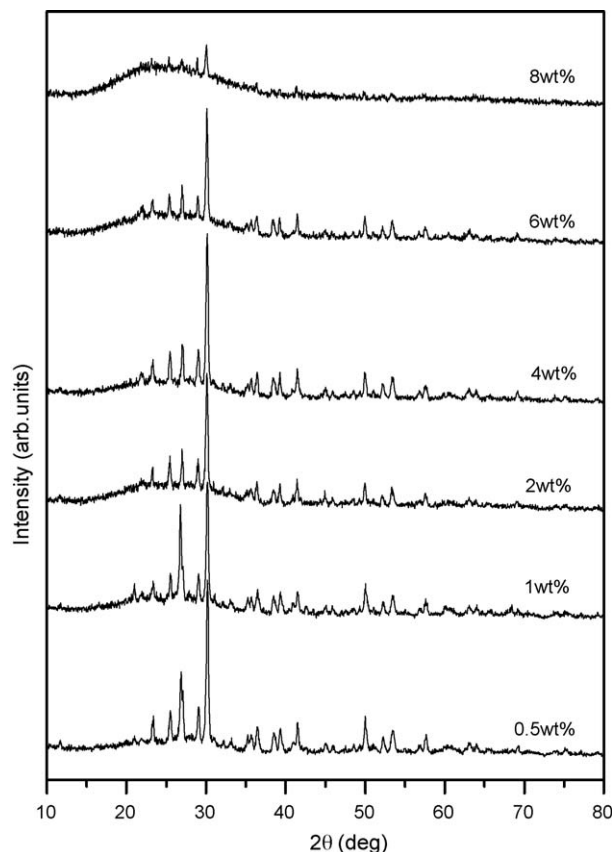


Fig. 2. X-ray diffraction patterns of CaSiO₃ ceramics sintered at 1100 °C with different B₂O₃ additions.

was obtained, and no second phase was observed, which indicated that the addition of B₂O₃ could lower the synthesis temperature of CaSiO₃ crystal phase. Moreover, it is clear that the diffraction peaks of CaSiO₃ were weakened and the amorphous phase was increased with the B₂O₃ content. The results showed that the B₂O₃ addition resulted in the emergence of CaO–B₂O₃–SiO₂ glass phase, which was advantageous to lower the synthesis temperature of CaSiO₃ crystal phase.

The density of B₂O₃-doped CaSiO₃ ceramics at different sintering temperatures is shown in Fig. 3. With increasing sintering temperature, the apparent density was found to increase to a maximum value, and higher B₂O₃ content shifts the obtainable maximum density to a lower temperature. The bulk densities of CaSiO₃ ceramics sintered at 1100 °C with 0.5 wt%, 1 wt%, 2 wt%, 4 wt%, 6 wt% and 8 wt% B₂O₃ additions were 2.22 g cm⁻³, 2.51 g cm⁻³, 2.69 g cm⁻³, 2.78 g cm⁻³, 2.85 g cm⁻³ and 2.82 g cm⁻³, respectively. The optimal sintering temperatures of CaSiO₃ ceramics with 0.5 wt%, 1–2 wt%, 4 wt%, 6 wt% and 8 wt% B₂O₃ additions were 1200–1250 °C, 1150–1200 °C, 1150 °C, 1100 °C and 1100–1150 °C, separately. The decrease of sintering temperature with the increase of B₂O₃ addition was due to the augment of CaO–B₂O₃–SiO₂ glass phase, which would transform into liquid phase and promote the sintering process. As it is obvious, the CaSiO₃ ceramic sintered at 1100 °C with 6 wt% B₂O₃ addition reached above 98% of the theoretical density (the

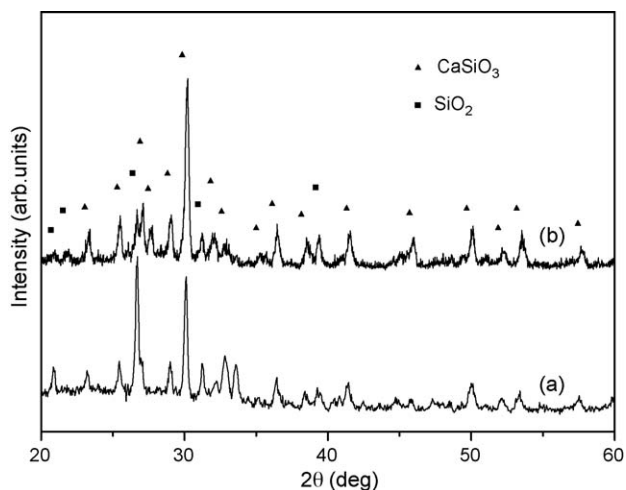


Fig. 1. X-ray diffraction patterns of CaCO₃–SiO₂ powders calcined at (a) 1100 °C and (b) 1200 °C.

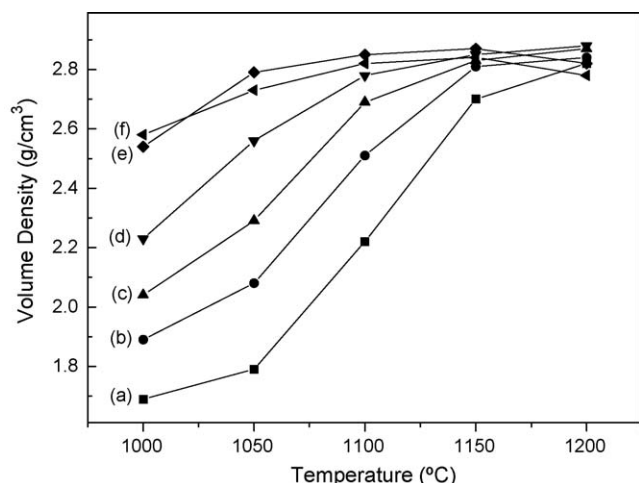


Fig. 3. Dependence of volume density on sintering temperature for CaSiO_3 ceramics with (a) 0.5 wt%, (b) 1 wt%, (c) 2 wt%, (d) 4 wt%, (e) 6 wt% and (f) 8 wt% B_2O_3 additions.

theoretical density of CaSiO_3 ceramic is 2.91 g cm^{-3}), which could be co-fired with 30 Pd/70 Ag electrode material.

The microstructures of CaSiO_3 ceramics sintered at 1100°C with different B_2O_3 additions are presented in Fig. 4. From picture

(a), it can be seen that the specimen was not dense and the grain did not grow with 0.5 wt% B_2O_3 addition. Picture (b), picture (c) and picture (d) show that the grain size increased with the increase of B_2O_3 addition due to the liquid phase effect resulted from the addition of B_2O_3 , and there were many grains which grow up to $5 \mu\text{m}$ with 4 wt% B_2O_3 added in. However, the CaSiO_3 ceramics had many pores with 1–4 wt% B_2O_3 addition, which indicated that it is difficult to lower down the sintering temperature of CaSiO_3 ceramics to 1100°C by adding 1–4 wt% B_2O_3 . This was probably due to an insufficient amount of liquid phase ($\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass phase). When the B_2O_3 content increased to 6 wt%, the grains of CaSiO_3 ceramic were in close contact and there was little porosity, which was consistent with the result of bulk densities. However, a low grain growth rate was observed by adding 8 wt% B_2O_3 , which was due to the decreasing of CaSiO_3 phase and the augment of $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass phase.

The dielectric properties (dielectric constant ϵ_r and dielectric loss $\tan \delta$) of B_2O_3 -doped CaSiO_3 ceramics sintered at different temperatures ($1050\text{--}1200^\circ\text{C}$) are illustrated in Figs. 5 and 6. The relationships between dielectric properties and sintering temperature followed similar trend to those between density and sintering temperature, due to a fact that a higher density signifying a lower porosity. The dielectric constant (ϵ_r) increased

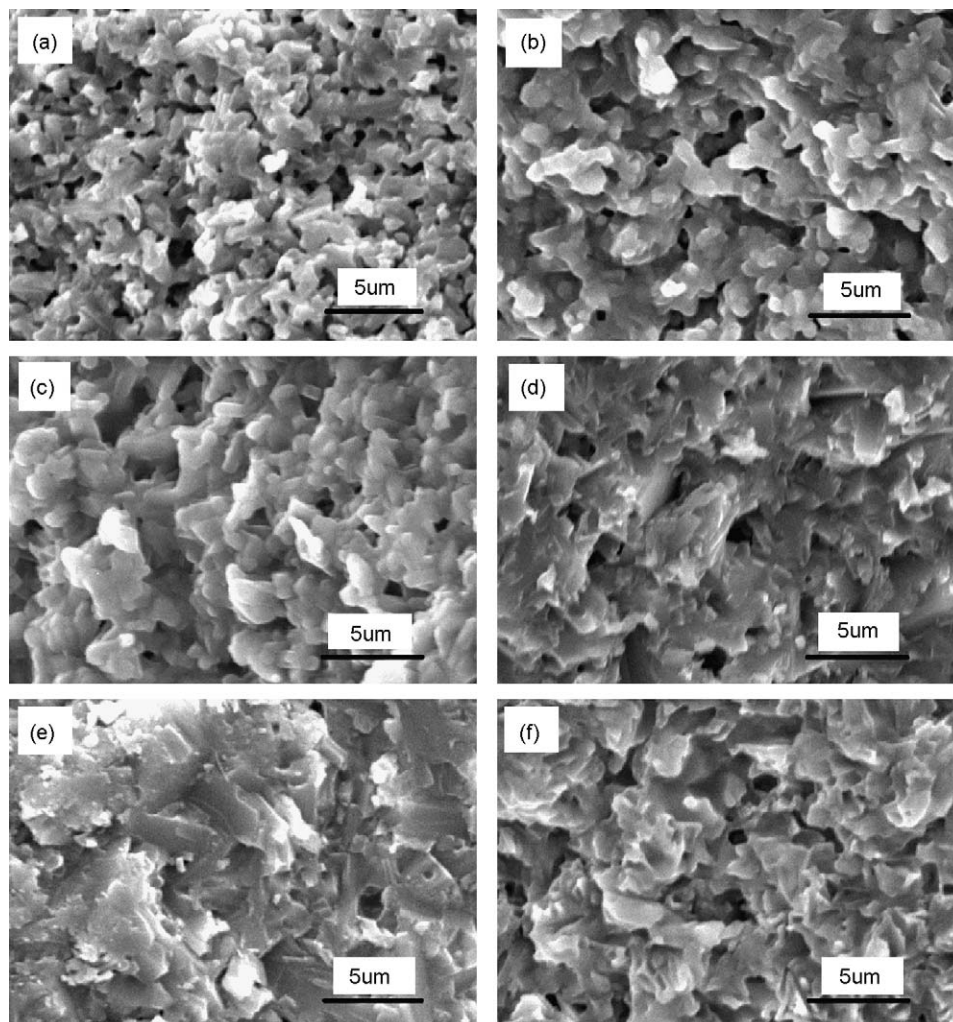


Fig. 4. SEM photographs of CaSiO_3 ceramics sintered at 1100°C with (a) 0.5 wt%, (b) 1 wt%, (c) 2 wt%, (d) 4 wt%, (e) 6 wt% and (f) 8 wt% B_2O_3 additions.

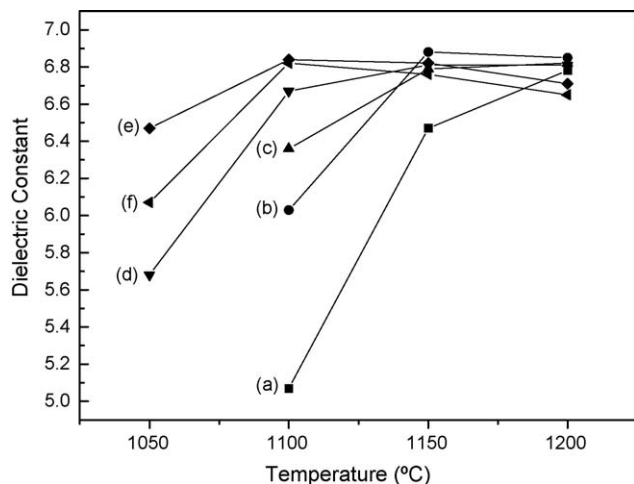


Fig. 5. Dependence of dielectric constant on sintering temperature for CaSiO_3 ceramics with (a) 0.5 wt%, (b) 1 wt%, (c) 2 wt%, (d) 4 wt%, (e) 6 wt% and (f) 8 wt% B_2O_3 additions.

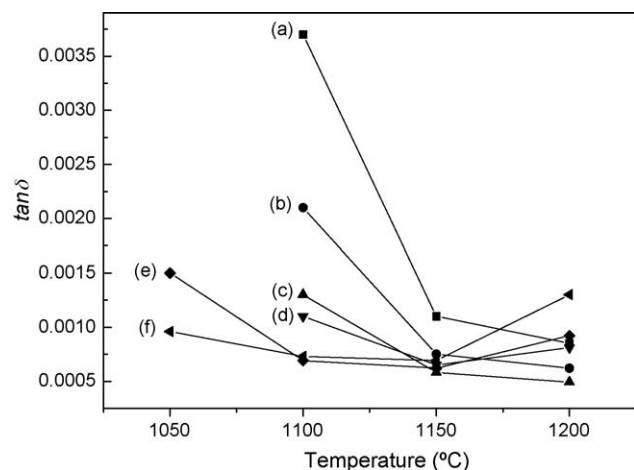


Fig. 6. Dependence of dielectric loss on sintering temperature for CaSiO_3 ceramics with (a) 0.5 wt%, (b) 1 wt%, (c) 2 wt%, (d) 4 wt%, (e) 6 wt% and (f) 8 wt% B_2O_3 additions.

and the dielectric loss ($\tan \delta$) decreased with the increasing sintering temperature. Sintered at 1100 °C, the CaSiO_3 ceramic with 6 wt% B_2O_3 addition possessed good dielectric properties: $\epsilon_r = 6.84$ and $\tan \delta = 6.9 \times 10^{-4}$ (1 MHz), which was a better result than the dielectric properties of $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass-ceramic [12–14]. The dielectric properties of CaSiO_3 ceramic with 8 wt% B_2O_3 addition were: $\epsilon_r = 6.76$ and $\tan \delta = 7.3 \times 10^{-4}$ (1 MHz). It suggested that the dielectric constant decreased and the dielectric loss increased with the increasing of B_2O_3 addition, although it was advantageous to lower the sintering temperature by adding sufficient B_2O_3 . These results also indicated that the dielectric constant of $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass was lower than which of CaSiO_3 ceramic and the dielectric loss of $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass was higher than which of CaSiO_3 ceramic.

4. Conclusion

The dielectric properties and microstructures of B_2O_3 -doped CaSiO_3 ceramics were investigated. B_2O_3 was not detected in

the ceramics but residual as a $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass phase. The emergence of $\text{CaO-B}_2\text{O}_3\text{-SiO}_2$ glass phase, acting as a sintering aid, can effectively lower the synthesis temperature of CaSiO_3 crystal phase and the sintering temperature of CaSiO_3 ceramic. Sintered at 1100 °C, the 6 wt% B_2O_3 -doped CaSiO_3 ceramic possesses good dielectric properties: $\epsilon_r = 6.84$ and $\tan \delta = 6.9 \times 10^{-4}$ (1 MHz), which can be co-fired with 30 Pd/70 Ag electrode material.

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