



CERAMICS INTERNATIONAL

www.elsevier.com/locate/ceramint

Ceramics International 37 (2011) 813-817

A comparative study of MoSi₂ coatings manufactured by atmospheric and vacuum plasma spray processes

Xiaoai Fei a,b, Yaran Niu a,b, Heng Ji a,b, Liping Huang a,b, Xuebin Zheng a,b,*

^a Key Laboratory of Inorganic Coating Materials, Chinese Academy of Sciences, Shanghai 200050, China
^b Shanghai Institute of Ceramics, Chinese Academy of Sciences, Shanghai 200050, China
Received 22 January 2010; received in revised form 12 June 2010; accepted 8 October 2010
Available online 17 November 2010

Abstract

In this work, MoSi₂ coatings were manufactured by atmospheric plasma spraying (APS) and vacuum plasma spraying (VPS) technologies, respectively. Phase composition and microstructure of the coatings were characterized by X-ray diffraction and scanning electron microscopy. Microhardness, void and oxygen content of the coatings were also determined. Oxidation behavior of the coatings at high temperature was examined. The results showed that the surface of VPS-MoSi₂ coating was dense and homogeneous. However, there were many microcracks formed on the surface of APS-MoSi₂ coating. The VPS-MoSi₂ coating also had lower void and oxygen contents, higher Vickers hardness compared with those of APS-MoSi₂ coating. Besides, oxidation resistance of the VPS-MoSi₂ coating was better than that of APS-MoSi₂ at 1500 °C. Published by Elsevier Ltd and Techna Group S.r.l.

Keywords: Plasma spray; MoSi2 coatings; Microstructure; Oxidation behavior

1. Introduction

MoSi₂ has been extensively studied during the last three decades for its rather low density, high electrical conductivity and enhanced oxidation resistance at high temperatures, even in harsh environments [1]. It has been widely used as heating elements operating at high temperatures, such as gas turbine engines, missile nozzles, diesel engine grow plugs, among other applications [2]. One of its main applications is oxidation resistant coatings [3]. The protection against rapid oxidation is accomplished by the formation of a thin, coherent and adherent SiO₂ layer on the surface of MoSi₂, thereby retarding oxygen diffusion to the substrate [4].

Many techniques have been suggested to deposit MoSi₂ coatings, such as chemical vapor deposition [5], pack siliconizing [6], vacuum sintering [7], solid surface reaction [8], pulsed laser ablation and deposition (PLAD) [9], among the main considered processes. Among the various coating

E-mail addresses: xiaoaif@yahoo.cn (X. Fei), xbzheng@mail.sic.ac.cn (X. Zheng).

manufacturing methods, plasma spraying is considered to be a versatile technology. Plasma spray technology enables the processing of a broad variety of coating materials, including refractory materials. It can be a suitable way for protective coatings manufacturing, because of its ability to cover large surfaces with thick coatings at high deposition rate and relatively low operating cost than most of the other coating deposition techniques [10].

Both atmospheric (APS) and vacuum plasma spraying (VPS) technologies use the energy of a high velocity Ar-H₂ or Ar-He hot plasma jets to melt and accelerate feedstock particles onto a pre-treated substrate surface to form coatings [11]. However, there are some differences between them. In case of APS technology, the process is carried out in air environment. Therefore, the feedstock particles would interact with air engulfed in the plasma flow, which may limit the choice of the spray material since reaction products such as oxides or secondary phases could be built into the coatings. The APS applications are normally related to wear and corrosion protection, which are often based on oxide and carbide ceramic materials. But APS technology has some issues with metal and oxygen sensitive coating materials processing such as MoSi₂ and B₄C. [12]. For VPS system, the coating deposition process takes place in inert atmosphere in a closed chamber with

^{*} Corresponding author at: Key Laboratory of Inorganic Coating Materials, Chinese Academy of Sciences, Shanghai 200050, China. Tel.: +86 21 52411050; fax: +86 21 52413903.

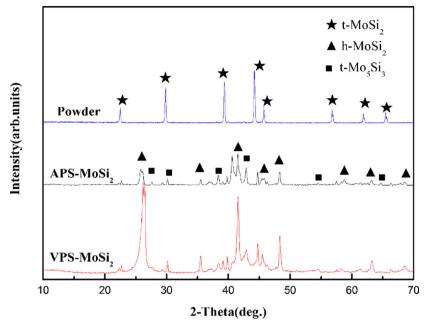


Fig. 1. XRD patterns of feedstock powders and as-sprayed APS-MoSi₂ and VPS-MoSi₂ coatings.

reduced pressure. The chamber is firstly pumped to pressures lower than 10 Pa and then filled with inert argon gas at about 5000–40,000 Pa. In order to tailor the working chamber pressure, efficient pump systems have to be implemented to remove steadily injected plasma gases. The limited reaction between melted particles and environment gases allows avoiding the formation of subproducts and the resulting coatings exhibit lower oxygen content as well as lower void content due to higher particles momentum upon impact [13,14].

Although several plasma sprayed silicide coatings were manufactured in previous studies [3,15], very few reports were focused on comparison of oxidation resistance of atmospheric and vacuum plasma sprayed MoSi $_2$ coatings. In this work, the microstructure and oxidation resistance behavior between APS-MoSi $_2$ and VPS-MoSi $_2$ coatings were investigated. Phase composition and microstructure of coatings were characterized by X-ray diffraction (XRD) and scanning electron microscopy (SEM). Some other characteristics, including microhardness, void content and oxygen content of coatings were determined. The oxidation behavior of the two types of MoSi $_2$ coatings was assessed at 1500 °C.

2. Experimental

2.1. Materials and preparation

Commercial $MoSi_2$ (Zhengzhou Chida Tungsten & Molybdenum Products Co., Ltd., Zhengzhou, China) feedstock powders with the medium size d_{50} of 17.6 μ m were chosen as feedstock. The powders are made of tetragonal $MoSi_2$ (Fig. 1). The coatings deposition was carried out by a plasma spray system (A-2000, Sulzer Metco, Wohlen, Switzerland) equipped with a F4-MB torch for APS and a F4-VB torch for VPS processes, respectively. Argon and hydrogen were used as the plasma forming gases. The optimized spray operating

parameters are summarized in Table 1. The powders were preheated at 60 °C for 20 min. C/C composite material with a specific mass of 1.8 g/cm³ and thermal conductivity of 20 W/m K was used as a substrate. It was grit-blasted to the waviness Wt of 1.92 μm , average roughness Ra of 2.05 μm and maximum peak-to-valley height Rt of 7.62 μm , and then cleaned with ethanol prior to the spraying process. The powders were also simultaneously sprayed onto a polished carbon steel substrate (average roughness Ra of 1.93 μm). Caused by the smooth surface and the different thermal coefficients of expansion the coatings were peeled off easily from the substrate.

2.2. Coating composition and microstructure characterization protocols

Crystalline phases of feedstock powders and coatings were analyzed by X-ray diffraction (XRD, RAX-10, Rigaku, Japan) operating with Cu K α (λ = 1.5406 nm) radiation. The scan

Table 1 Atmospheric and vacuum plasma spray operating parameters of $MoSi_2$ coatings.

Parameters	APS	VPS
Power (kW)	35	35
Gas Ar (slpm)	42	42
Gas H ₂ (slpm)	10	10
Spray distance (mm)	150	300
Carrier gas Ar (slpm)	4	4
Powder feed rate (g/min)	14	14
Surrounding atmosphere pressure (Pa)	$\sim \! 10^{5}$	10^{4}
Anode nozzle diameter at exit (mm)	6	6
Plasma mass enthalpy (J/kg)	8×10^{6}	8×10^{6}
Torch scanning velocity (m/s)	1.2	1.2
Scanning step (mm/pass)	4	4
Substrate average temperature during deposition (°C)	<350	<350

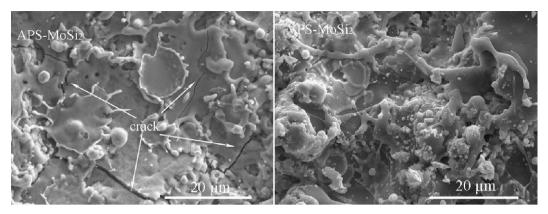


Fig. 2. Surface morphologies of as-sprayed APS-MoSi₂ (left) and VPS-MoSi₂ (right) coatings.

ranges from 10° to 70° , with step size of 0.02° and counting time of 1 s at each step. The surface and cross-section morphologies of coatings were investigated by field emission scanning electron microscopy (FESEM, JSM-6700F, JEOL, Japan). The composition of coatings was estimated by energy dispersive spectrometer (EDS, INCA ENERGY, UK). Vickers hardness measurements were conducted on polished cross-section of the coatings using a microhardness instrument (HX-100, Shanghai Second Optical Instrument Factory, China) under loading weight of 1.96 N and dwell time of 15 s. The depicted values represent an average of twenty point measurements randomly located along the crosssection. The void content of as-sprayed coatings was measured on polished cross-sections (resolution: 270 PPI) by image analysis (IMAGE, National Institute of Health, Springfield, MD, USA). Ten measurements were carried out for each sample. Surface roughness was measured by HOMMEL WERKE (T8000-C, Wave, Germany). The oxygen content in the coatings was detected by nitrogen/oxygen analyzer (TC600, Leco, USA).

2.3. Oxidation behavior characterization

Freestanding coatings were cut into small buttons of 7 mm in diameter and 1 mm thick for the oxidation test. The surfaces were mechanically polished with surface average roughness Ra of 0.38 μ m using corundum sand of F60 grade at a blasting angle nearly 90° for about 20 s and blasting distance 120 mm, and then cleaned ultrasonically. The specimens were placed in

an alumina crucible in a box type furnace and were oxidized at $1500\,^{\circ}$ C in ambient air at atmospheric pressure up to $50\,\text{h}$. The samples were furnace-heated and cooled with the rate of $5\,^{\circ}$ C/min. The mass change of each sample was measured with an electronic balance (sensitivity: $10^{-4}\,\text{g}$). Weight changes/surface areas of the samples were then calculated.

3. Results and discussion

3.1. Phase composition and microstructure of as-sprayed coatings

The XRD patterns (Fig. 1) show that both the as-sprayed APS-MoSi₂ and VPS-MoSi₂ coatings are composed of MoSi₂ tetragonal and hexagonal phases and Mo₅Si₃ tetragonal phase. Phase compositions of both coatings were changed compared with the feedstock MoSi₂ powders. A fraction of the tetragonal MoSi₂ was converted to hexagonal phase due to high cooling rate of the deposited molten particles (about 10⁶ K/s). The high temperature hexagonal MoSi₂ phase was kept as main phase in the coatings [16]. The existence of Mo₅Si₃ phase is very likely due to oxidation of the MoSi₂ according to Eq. (1).

$$5\text{MoSi}_{2(s)} + 7\text{O}_{2(g)} \rightarrow \text{Mo}_5\text{Si}_{3(s)} + 7\text{SiO}_{2(s)}$$
 (1)

It is worth noticing that the ratio of the intensity of Mo₅Si₃/MoSi₂ in APS-MoSi₂ coating was higher than that in VPS-

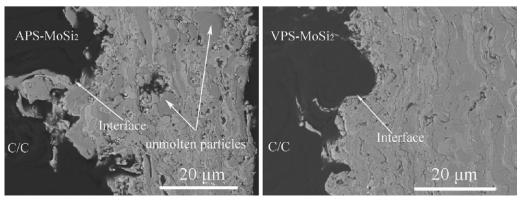


Fig. 3. Cross-section morphologies of as-sprayed APS-MoSi₂ (left) and VPS-MoSi₂ (right) coatings.

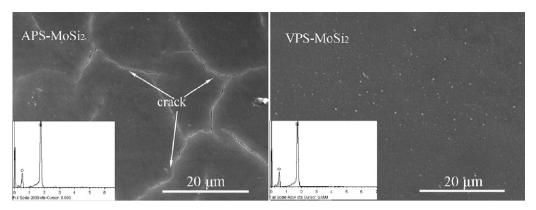


Fig. 4. Surface morphologies of APS-MoSi₂ (left) and VPS-MoSi₂ (right) coatings heated at 1500 °C for 50 h.

 $MoSi_2$ coating, indicating that more $MoSi_2$ were oxidized during APS process. Due to the residue of O_2 in the large chamber and in plasma gases, small amounts of Mo_5Si_3 were also detected in VPS- $MoSi_2$ coatings.

Surface morphologies of as-sprayed APS-MoSi $_2$ and VPS-MoSi $_2$ coatings are depicted in Fig. 2. Well-flattened lamellar along with some spherical features were observed on the surfaces of both of the coatings, which exhibit typical morphologies of plasma sprayed coatings. However, the surface of APS-MoSi $_2$ coating contains some microcracks, which are supposed to be caused by the internal residual stresses during the spraying process [14]. Besides, the coefficients of thermal expansion (CTE) of the tetragonal Mo $_5$ Si $_3$ is highly anisotropic, and thermal expansion anisotropy is undesirable since it gives rise to microcrack [17]. In comparison, no obvious microcracks were found for VPS-MoSi $_2$ coating. The VPS system limited the oxidation of MoSi $_2$ and little amount Mo $_5$ Si $_3$ was embedded in the coatings.

Polished cross-section morphologies of the as-sprayed coatings (Fig. 3) allow observing coating structures and their interface with substrates. Structure of the APS-MoSi₂ coating appeared non-compact, and the particles seemed distributed unevenly with numerous unmolten particles, whereas the VPS-MoSi₂ coating exhibited a dense and homogeneous structure, and the lamellar morphology was much clearly revealed. It was worth noticing that neither interfacial delamination nor void was found at substrate/coating interfaces for both considered manufacturing routes.

Some coating characteristics are displayed in Table 2. Void content in APS-MoSi $_2$ coating was $12 \pm 2\%$, while it was only $5 \pm 1\%$ in VPS-MoSi $_2$ coating as determined by image analysis. The low void content plays very likely a role in decreasing the void network connectivity, therefore increasing the oxidation resistance of coatings. Oxygen contents of APS-MoSi $_2$ and VPS-MoSi $_2$ coating were 0.86 wt.% and 0.25 wt.%, respectively. The oxygen content of VPS-MoSi $_2$ coating was significantly lower than that of APS-MoSi $_2$ coating owing to the intrinsic characteristics of VPS technology. Vickers hardness of VPS-MoSi $_2$ coating was 4.06 GPa, while the one of APS-MoSi $_2$ coating was 4.06 GPa. The higher microhardness of VPS-MoSi $_2$ coating can be explained by its lower void content,

higher inter-splat cohesion and fewer defects because of less oxidation [18].

3.2. Oxidation resistance of coatings

Both the freestanding APS-MoSi₂ and VPS-MoSi₂ coatings were oxidized at 1500 °C to evaluate their oxidation resistance behavior. After oxidized for 50 h, mass gains of the VPS-MoSi₂ and APS-MoSi₂ coatings were increased by 1.46 mg/cm² and 1.68 mg/cm², respectively. The VPS-MoSi₂ coatings exhibited a lower mass gain in comparison with the APS-MoSi₂ coating. Fig. 4 illustrates the surface morphologies of APS-MoSi₂ and VPS-MoSi₂ coatings oxidized at 1500 °C for 50 h. It can be seen that both the coatings had smooth surfaces, which was proved to be composed of SiO₂ by EDS. However, some microcracks appeared on the surface of oxidized APS-MoSi₂ coating (about 10 µm to be compared to about 20 µm). Corresponding polished cross-section pictures (Fig. 5) revealed that the oxide layer of VPS-MoSi₂ coating was thinner than that of APS-MoSi₂ coating, which indicated it had a better oxidation resistance [19]. At the same time, it was found that for APS-MoSi₂ coating, the formed microcracks penetrated into, or even through the SiO₂ film, as illustrated in Fig. 5 (left).

In previous studies, the oxidation resistance of atmospheric plasma sprayed WSi₂ (very similar of MoSi₂) was unsatisfactory owing to their high void contents [15]. They suggested one possible means of creating a denser coating would be vacuum plasma spraying. Apart from the denser coatings obtainable with the VPS process, this homogeneous coating structure may have a positive effect on the formation of a continuous dense SiO₂ layer. And Tiwari et al. [3] reported also that the oxidation resistance of the vacuum plasma sprayed MoSi₂ matrix was very good under cyclic and isothermal conditions. Oxygen is reported to have a very low diffusion coefficient in SiO₂ films. The dense glassy SiO₂ can retard transportation of oxygen,

Table 2 Some characteristics of the as-sprayed coatings.

Coatings	Void content (%)	O content (wt.%)	Microhardness (GPa)
APS-MoSi ₂	12 ± 2	0.86	4.06
VPS-MoSi ₂	5 ± 1	0.25	6.85

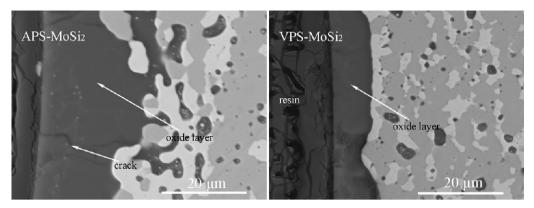


Fig. 5. Cross-section morphologies of APS-MoSi₂ (left) and VPS-MoSi₂ (right) coatings heated at 1500 °C for 50 h.

therefore provide a good oxidation resistance [6]. However, microcracks in the SiO₂ might not be self-sealed completely because that the fluidity of SiO₂ glass is poor below 1700 K [20]. Oxygen can hence diffuse through the formed microcracks and directly react with attack internal substrate. In our present work, the cracks on the surface of APS-MoSi₂ coating accelerated oxygen transport to substrate diffusion and aggravated the oxidation process, leading therefore to a comparatively worse oxidation resistant property compared with VPS-MoSi₂ coating [21,22].

4. Conclusion

MoSi $_2$ coatings were manufactured by atmospheric and vacuum plasma spray processes, respectively. The as-sprayed VPS-MoSi $_2$ coating exhibited a dense and homogeneous microstructure. However, there were many microcracks formed on the surface of APS-MoSi $_2$ coating. Besides, the VPS-MoSi $_2$ had lower void and oxygen contents, higher Vickers hardness in comparison with those of APS-MoSi $_2$ coating. Glassy SiO $_2$ films formed on both APS-MoSi $_2$ and VPS-MoSi $_2$ coatings after oxidation test. Due to the absence of microcracks in the SiO $_2$ film, the oxidation resistant property of VPS-MoSi $_2$ coating was better than that of APS-MoSi $_2$ coating at 1500 °C.

References

- A.L. Dumont, J.P. Bonnet, T. Chartier, MoSi₂/Al₂O₃ FGM: elaboration by tape casting and SHS, J. Eur. Ceram. 21 (2001) 2353–2360.
- [2] E.K. Nyutu, M.A. Kmetz, S.L. Suib, Formation of MoSi₂–SiO₂ coatings on molybdenum substrates by CVD/MOCVD, Surf. Coat. Technol. 200 (2006) 3980–3986.
- [3] R. Tiwari, H. Herman, S. Sampath, Vacuum plasma spraying of MoSi₂ and its composite, Mater. Sci. Eng. 155A (1992) 95–100.
- [4] Z.D. Liu, S.X. Hou, D.Y. Liu, L.P. Zhao, B. Li, J.J. Liu, An experimental study on synthesizing submicron MoSi₂-based coatings using electrothermal explosion ultra-high speed spraying method, Surf. Coat. Technol. 202 (2008) 2917–2921.
- [5] G.A. West, K.W. Beeson, Chemical vapor deposition of molybdenum silicide, J. Electrochem. Soc. 135 (7) (1988) 1752–1757.
- [6] J.K. Yoon, K.H. Lee, G.H. Kim, J.K. Lee, J.M. Doh, K.J. Hong, Growth kinetics of MoSi₂ coating formed by a pack siliconizing process, J. Electrochem. Soc. 151B (2004) 309–318.

- [7] R. Chow, D. Nichols, Properties of MoSi₂ films deposited from a composite target, Thin Solid Films 118 (1984) 139–147.
- [8] S.C. Deevi, N.N. Thadhani, Reaction synthesis of high temperature silicides, Mater. Sci. Eng. 193 (1995) 604–611.
- [9] S. Madhukar, S. Aggarwal, A.M. Dhote, R. Ramesh, S.B. Samavedam, S. Choopun, R.P. Sharma, Pulsed laser-ablation deposition of thin films of molybdenum silicide and its properties as a conducting barrier for ferroelectric random-access memory technology, J. Mater. Res. 14 (1999) 940–947.
- [10] X.C. Zhang, B.S. Xu, F.Z. Xuan, H.D. Wang, Y.X. Wua, Microstructural and porosity variations in the plasma-sprayed Ni-alloy coatings prepared at different spraying powers, J. Alloys Compd. 473 (2008) 145–151.
- [11] O. Knotek, in: R.F. Bunshah (Ed.), Handbook of Hard Coatings: Deposition Technologies, Properties and Applications, Noyes Pub./William Andrew Publishing, LLC, Park Ridge, New Jersey, USA/Norwich, New York, USA, 2001.
- [12] M. Bhusari, in: Proceedings of the International Thermal Spray Conference, Singapore, May 28–30, (2001), pp. 1289–1297.
- [13] P.S. Sidky, M.G. Hocking, Review of inorganic coatings and coating processes for reducing wear and corrosion, Br. Corros. J. 34 (1999) 171– 183
- [14] H. Singh, B.S. Sidhu, D. Puri, S. Prakash, Use of plasma spray technology for deposition of high temperature oxidation/corrosion resistant coatings – a review, Mater. Corros. 58 (2007) 92–102.
- [15] O. Knotek, R. Elsing, H.-R. Heintz, Corrosion and oxidation resistance of plasma-sprayed WSi₂ coatings, Surf. Coat. Technol. 30 (1) (1987) 107– 114
- [16] G. Reisel, B. Wielage, S. Steinhäuser, I. Morgenthal, R. Scholl, High temperature oxidation behavior of HVOF-sprayed unreinforced and reinforced molybdenum disilicide powders, Surf. Coat. Technol. 146 (2001) 19–26.
- [17] J.H. Schneibel, C.J. Rawn, E.A. Payzant, C.L. Fu, Controlling the thermal expansion anisotropy of Mo₅Si₃ and Ti₅Si₃ silicides, Intermetallics 12 (2004) 845–850.
- [18] C. Huang, X. Zhou, C. Ding, Investigation of the thermo-mechanical properties of a plasma-sprayed nanostructured zirconia coating, J. Eur. Ceram. Soc. 23 (9) (2003) 1449–1455.
- [19] S. Lohfeld, M. Schütze, A. Böhm, V. Güther, R. Rix, R. Scholl, Oxidation behavior of particle reinforced MoSi₂ composites at temperatures up to 1700 °C, Mater. Corros. 56 (2005) 93–96.
- [20] J. Zhao, Q.G. Guo, J.L. Shi, G.T. Zhai, L. Liu, SiC/Si-MoSi₂ oxidation protective coatings for carbon materials, Surf. Coat. Technol. 201 (2006) 1861–1865.
- [21] V.I. Zmii, A.P. Patokin, V.L. Khrebtov, B.M. Shirokov, Molybdenum-based oxidation resistant oxidation MoSi₂–Al₂O₃ and WSi₂–Al₂O₃ coatings, Powder Metall. Met. Ceram. 47 (2008) 11–12.
- [22] K. Natesan, S.C. Deevi, Oxidation behavior of molybdenum silicides and their composites, Intermetallics 8 (2000) 1147–1158.