

# The comparison of mechanical behavior of $\text{MgO-MgAl}_2\text{O}_4$ with $\text{MgO-ZrO}_2$ and $\text{MgO-MgAl}_2\text{O}_4\text{-ZrSiO}_4$ composite refractories

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## Abstract

Mechanical properties of different compositions obtained from the additions of 5, 10, 20 and 30 wt.% zircon ( $\text{ZrSiO}_4$ ) into the  $\text{MgO}$ –spinel composite refractories and  $\text{ZrO}_2$  into  $\text{MgO}$  have been examined, the variations that occurred have been determined, and the parameters affecting those factors have been investigated with the reasons. The density, strength, Young's modulus, fracture toughness, fracture surface energy and work of fracture were measured and evaluated. Microstructural variations and fracture surfaces have been examined and the formation of new phases has been identified depending on the additive type and quantity. The relationships between mechanical properties and structural variations for different compositions have been examined. In  $\text{MgO}$ –spinel materials, strength, Young's modulus and fracture toughness values decrease up to 20% spinel addition and stay almost constant for further loads.  $\text{ZrO}_2$  addition displays same trend but not as effective as spinel. Besides, since  $\text{ZrO}_2$  is stable in cubic form, it does not show any toughening mechanism. Forsterite formation is the most important factor for 2-fold improvement in the mechanical properties of  $\text{MgO}$ –spinel–zircon refractories. The more the zircon addition, the more the mechanical properties improve. The generation of natural bonding between matrix particles with forsterite formation, on the other hand, causes the fracture path to turn to transgranular fracture with an increase in fracture surface energy and a decrease in work of fracture, among which the latter is considered as an indicator of thermal shock resistance of the materials being high.

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## 1. Introduction

Magnesia–chromite ( $\text{MgO-Cr}_2\text{O}_3$ ) bricks have been used as standard lining for the hot kiln zones since 1940. Their properties, e.g. thermal conductivity, thermal shock resistance and thermal expansion, mainly depend on the refractory grade chrome ores [1]. However, because of the risk of contamination of ground water by  $\text{Cr}^{VI}$  ions leaching from waste refractory materials, and because  $\text{Cr}^{VI}$  has in particular been associated with allergic skin ulceration and carcinomas, the replacement of the chrome in these refractories has become a challenge [2–4]. For this reason, attention has been given to the use of magnesium aluminate spinel,  $\text{MgAl}_2\text{O}_4$ , as an alternative to chrome ore [5–7]. While magnesite is often used alone, magnesium aluminate spinel is now

widely incorporated in magnesite-based refractories [8,9]. A further major advantage of the magnesite–spinel refractories is that they have improved thermal shock resistance, and it is claimed that two to three times longer service life can be obtained, compared to a conventional magnesite–chrome brick [10,11].

The optimum spinel content, reported so far, is variable depending on the properties of spinel. Cooper and Hudson [12] found 40 wt.% spinel refractories exhibited the best combination of properties with superior resistance against thermal shock while Aksel and Riley [13] reported 20 wt.% was sufficient.

While service life of refractories is prolonged especially in connection with the increase in work of fracture energy by addition of spinel particles to  $\text{MgO}$ , a reduction occurs in other mechanical properties [5,6].

This study has been conducted for further improving the mechanical performances of materials and further prolonging service life by deriving high thermomechanical properties as a result of the increase in resistance to fracture by addition of zircon ( $\text{ZrSiO}_4$ ) in different proportions to  $\text{MgO}$ –spinel

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composite refractories containing various quantities of spinel ( $\text{MgAl}_2\text{O}_4$ ) and  $\text{ZrO}_2$  to  $\text{MgO}$ . The relationships between mechanisms affording improvement in the mechanical behaviors of refractories and microstructural changes and parameters affecting these have been analyzed in detail.

## 2. Experimental

Recipes were prepared by adding  $\text{ZrO}_2$  in proportion of 5%, 10%, 20% and 30% to  $\text{MgO}$  (M– $\text{ZrO}_2$ ) and zircon in the same proportions to compositions derived by adding 5%, 10%, 20% and 30% spinel by weight to  $\text{MgO}$  (M–S–zircon). Chemical analyses of raw materials are given in Table 1. Batches prepared using fine, medium and large sized  $\text{MgO}$  (0–1 mm), spinel (0–1 mm) and zircon ( $\sim 13 \mu\text{m}$ ) were shaped into samples of  $\sim 8 \text{ mm} \times 8 \text{ mm} \times 60 \text{ mm}$  applying  $\sim 100 \text{ MPa}$  pressure. The samples were sintered for 2 h in kiln (Nabertherm HT16/18) at  $1600^\circ\text{C}$  at  $5^\circ\text{C}/\text{min}$  heating and cooling rates. After the surfaces of samples sintered were polished for  $\sim 2 \text{ min}$  using 800 and 1200 grade SiC paper, mechanical tests were performed after dehydration at  $\sim 110^\circ\text{C}$  in the drying oven.

The mechanical tests, performed using a load cell of 2 kN moving at a velocity of 0.5 mm/min, were applied to minimum 5–6 samples in order to take average values. Bend strength [14] was measured in three point loading with support roller span ( $L$ ) of 50 mm. Young's modulus ( $E$ ) was calculated on the basis of the initial, steepest straight-line portion of the load–deflection curve, with correction for machine stiffness [15], using the standard relationship:

$$E_m = \frac{L^3}{4WD^3} m \quad (1)$$

where  $m$  is the initial slope of the curve,  $W$  is the bar width, and  $D$  is the bar thickness.  $K_{1C}$  values were calculated by using notches (by single edge notched beam (SENB) method) at a depth of  $\sim 25\%$  of thickness of the material using a diamond disk  $700 \mu\text{m}$  thick on specimens using standard equation [16–19]:

$$K_{1C} = \frac{3PL\sqrt{C}}{2WD^2} \quad (2)$$

where  $Y$  is given by:

$$Y = A_0 + A_1\left(\frac{C}{D}\right) + A_2\left(\frac{C}{D}\right)^2 + A_3\left(\frac{C}{D}\right)^3 + A_4\left(\frac{C}{D}\right)^4 \quad (3)$$

For  $L/D \sim 8$  [18]:

$$A_0 = +1.96, \quad A_1 = -2.75, \quad A_2 = +13.66, \quad A_3 = -23.98, \quad A_4 = +25.22$$

Table 1  
Chemical analysis of raw materials (in wt.%).

	$\text{SiO}_2$	$\text{Al}_2\text{O}_3$	$\text{MgO}$	$\text{CaO}$	$\text{ZrO}_2$	$\text{Fe}_2\text{O}_3$	$\text{HfO}_2$	Total
MgO (fine)	1.16	0.35	95.61	1.33	–	0.93	–	99.38
MgO (medium–large)	1.00	0.36	95.78	1.25	–	0.93	–	99.32
Spinel (medium–large)	0.44	64.05	34.13	0.54	–	0.53	–	99.69
Zircon	34.23	1.70	–	0.08	62.30	0.15	–	98.46
$\text{ZrO}_2$	0.16	–	–	–	97.87	0.05	1.94	100.02

Notch depth was measured using an Olympus BX60M brand optical microscope. SEM studies were carried out using Zeiss Evo 50 device to analyze the microstructures and fracture surfaces of materials. Using photographs taken on surfaces of specimens, which were polished and thermally etched for 10 min at  $1450^\circ\text{C}$ , the average  $\text{MgO}$  grain size was calculated using standard equation employing the linear intersection technique [20]. Bulk density and apparent porosity of bars were measured using the standard water immersion method [21]. X-ray diffraction measurements were carried out using a Rigaku RINT 2000 series set and a  $\text{Cu K}\alpha$  X-ray radiation. A scanning speed of  $2^\circ/\text{min}$  was used for the  $2\theta$   $10$ – $70^\circ$ .

## 3. Results and discussion

Based on phase analysis results,  $\text{MgO}$  and cubic- $\text{ZrO}_2$  were determined in M– $\text{ZrO}_2$  system where forsterite ( $2\text{MgO} \cdot \text{SiO}_2$ ) and cubic- $\text{ZrO}_2$  phases were determined in the M–S–Zircon system beside  $\text{MgO}$  and  $\text{MgAl}_2\text{O}_4$  phases. Forsterite phase, having a high melting temperature, was formed after reaction of  $\text{MgO}$  with  $\text{SiO}_2$  which was released after decomposition of zircon ( $\text{ZrSiO}_4$ ) as  $\text{SiO}_2$  and  $\text{ZrO}_2$ . It was observed that sintering was more effective with the formation of the new phase. The average density values measured for  $\text{MgO}$  and M–spinel refractories were  $\sim 2.8 \text{ g}/\text{cm}^3$  going up to  $\sim 3.1 \text{ g}/\text{cm}^3$  with increasing addition of zircon. Porosity, on the other hand, was  $\sim 22\%$  in  $\text{MgO}$  and M–spinel materials, displaying a reduction trend down to  $\sim 14\%$  as the quantity of added zircon was increased. Addition of spinel to  $\text{MgO}$  had a very little effect on density and porosity, since theoretical density of both components was identical ( $d_{\text{MgO}}: 3.583$  and  $d_{\text{spinel}}: 3.578$ ) and usage of spinel did not have a promoting effect on sintering. Inclusion of zircon to M–spinel and  $\text{ZrO}_2$  to  $\text{MgO}$ , however, increased density and decreased porosity as a result of pinning effect of  $\text{ZrO}_2$  and forsterite formation in the case of M–S–zircon (Fig. 1). Mean grain size of the compositions  $\text{MgO}$ , M–5S, M–10S and M–5S–20zircon were measured as  $67 \mu\text{m}$ ,  $40 \mu\text{m}$ ,  $38 \mu\text{m}$  and  $27 \mu\text{m}$ , respectively.

Addition of up to 20% of the spinel to  $\text{MgO}$  decreased both strength and Young's modulus significantly (75%), as well as toughness. Further additions caused no further loss (Figs. 2–4). During cooling from production temperatures, the large difference in thermal expansion coefficient between  $\text{MgO}$  ( $13 \times 10^{-6}^\circ\text{C}^{-1}$ ) and spinel ( $7.6 \times 10^{-6}^\circ\text{C}^{-1}$ ) generates very large hoop tensile stress around reinforced particles, causing extensive microcracking [22] (Fig. 5). At a certain temperature, microcracks can only occur when the dispersed particle size is larger than a critical value.  $\text{MgO}$ –spinel composites with

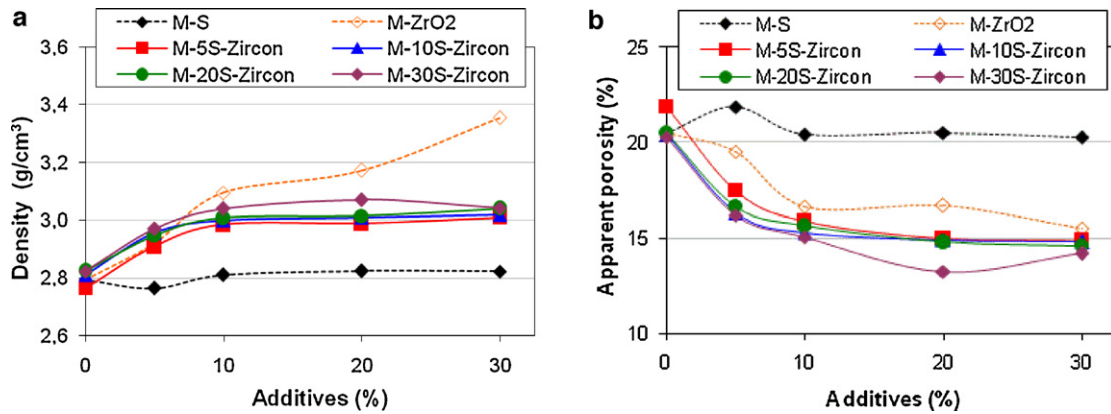


Fig. 1. (a) Density and (b) apparent porosity values as a function of additives.

$\alpha_{\text{spinel}} < \alpha_{\text{MgO}}$  show low strengths because the radial cracks produced readily link together [23,24]. Thermal expansion mismatch between MgO and spinel therefore can lead to a reduction in strength because of an increase in crack length and

a decrease in Young's modulus [25–30]. The microcracks decrease the overall strength and stiffness [31,32]. For more than 20% spinel addition, owing to the interlinking of microcracks, no more decrease occurs. Because the mean

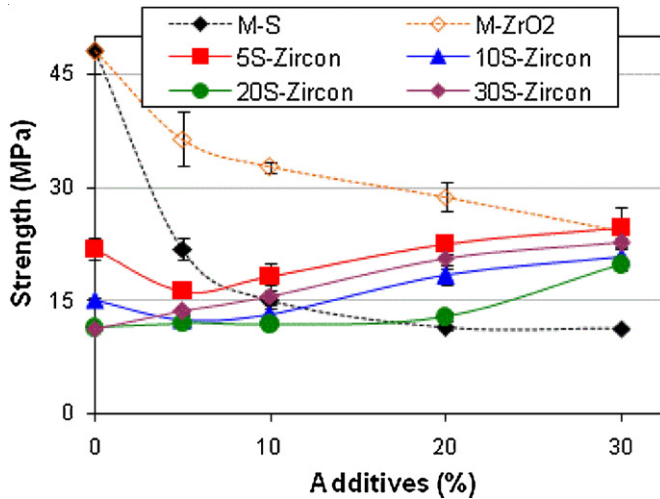
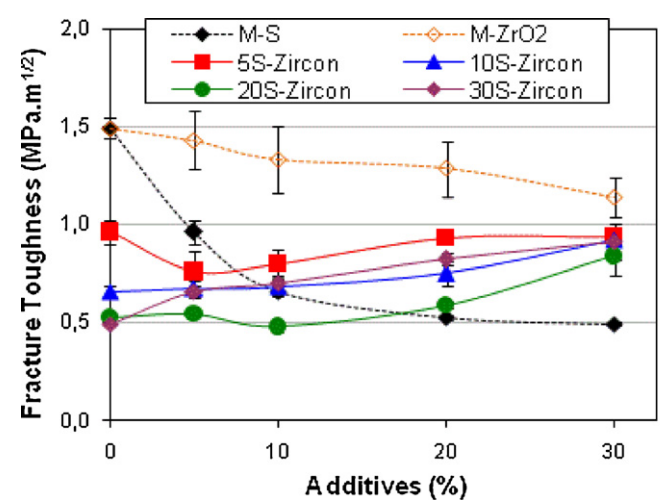
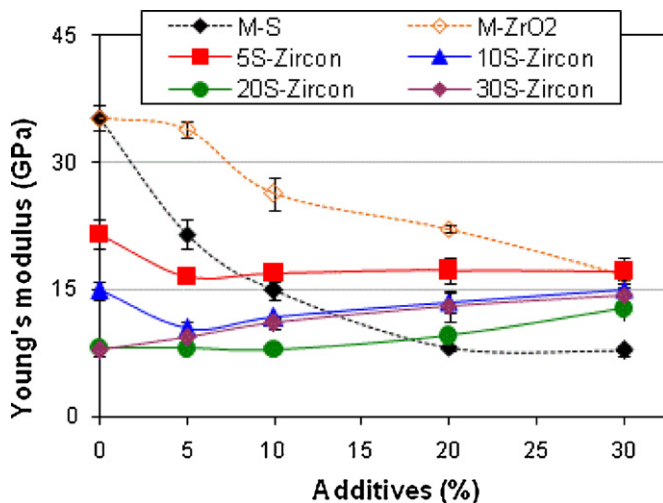
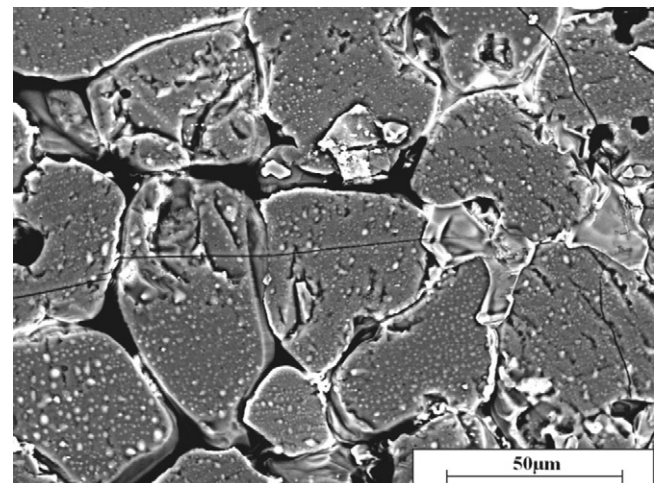
Fig. 2. Strength as a function of additives (M: MgO, S: MgAl<sub>2</sub>O<sub>3</sub>, and Zircon: ZrSiO<sub>4</sub>).Fig. 4. Fracture toughness as a function of additives (M: MgO, S: MgAl<sub>2</sub>O<sub>3</sub>, and Zircon: ZrSiO<sub>4</sub>).Fig. 3. Young's modulus as a function of additives (M: MgO, S: MgAl<sub>2</sub>O<sub>3</sub>, and Zircon: ZrSiO<sub>4</sub>).

Fig. 5. SEM micrograph showing the microstructure of a M-20spinel composite.



grain size of  $\text{ZrO}_2$  is much smaller than spinel,  $\text{ZrO}_2$  addition was not as effective as spinel even though the difference in coefficient of thermal expansion is higher than M–spinel. An overall increase was observed in strength with zircon addition as well as in Young's modulus. Among all the compositions, 30% zircon addition gave the highest strength values. Strength of M–30S–30zircon had 2-fold improvement compared to that of M–30spinel.

As  $\geq 10\%$  zircon was added to refractories containing various proportion of spinel, the fracture toughness values of materials basically rose with increasing amount of zircon and they were higher than the fracture toughness of non-additive

M–spinel materials. The fracture toughness of all compositions containing 30% zircon were close each other, displaying  $\sim 2$ -fold improvement compared to non-additive M–spinel (Fig. 4).

It is clear from Fig. 5 that the loss of strength in the composites is not the result of the influence of spinel on magnesia grain size. The observed changes in composite strength with increasing spinel loading are consistent with the generation of microcracks. Direct evidence of microcracking is seen in polished faces, typified by Figs. 5 and 7. These microcracks, it has been assumed, are the result of the large thermal expansion mismatch between the MgO matrix grains and the spinel particles. In fact in very many cases, the

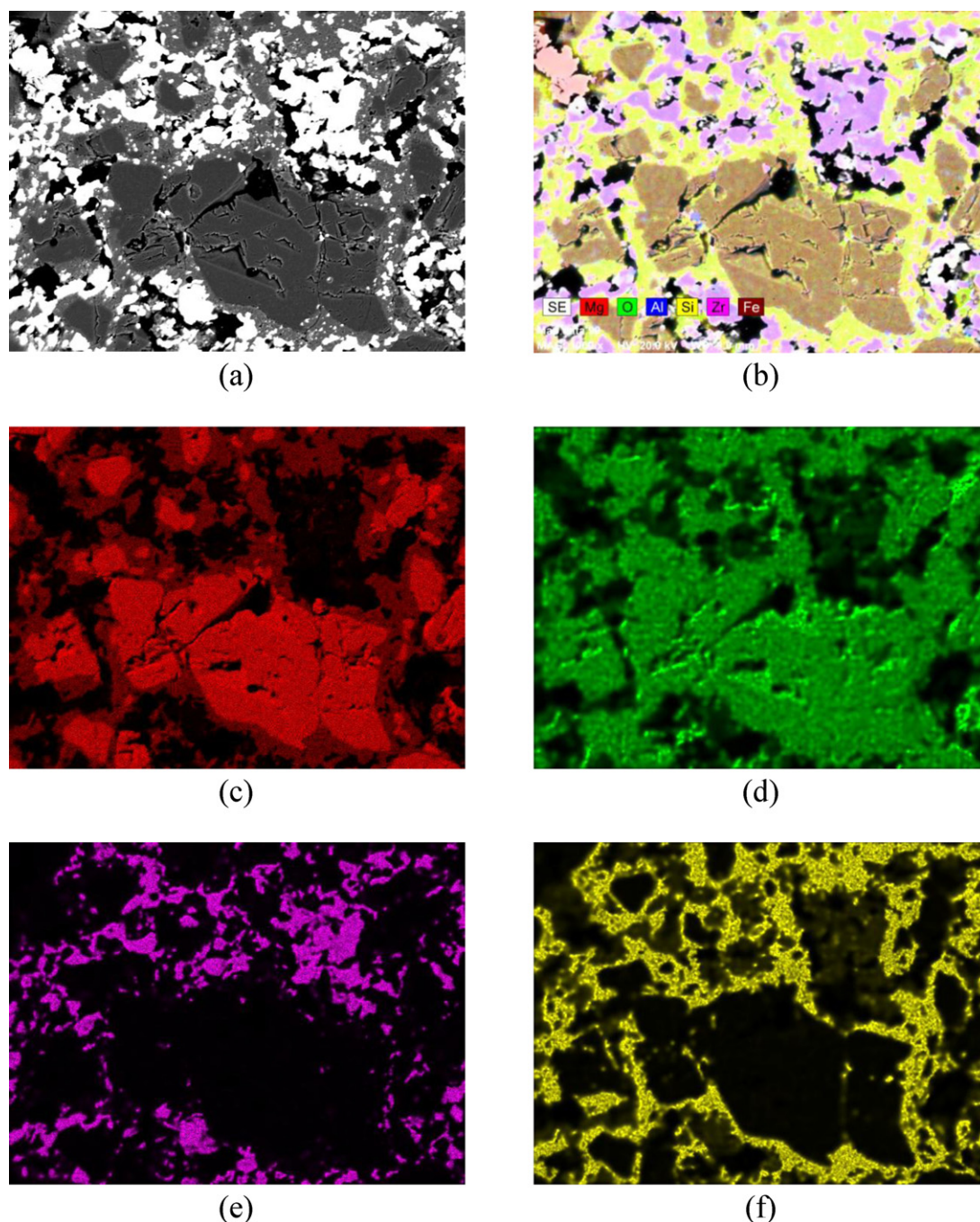


Fig. 6. Electron mapping images of the sample M–20S–20zircon: (a) and (b) general image, (c) Mg, (d) O, (e) Zr, and (f) Si.

intergranular cracking might be more accurately described as grain boundary separation. The microcracking becomes more extensive, the higher the spinel content. Much of the cracking appears to be intergranular, but with the more extensively cracked materials, cracks clearly also extend through MgO grains, following the natural cleavage planes. The decrease in strength of the spinel composites is, therefore, more likely to be related to a decrease of fracture toughness with decreasing  $E$  and fracture energy [13,22].

Integration of zircon to MgO–spinel composite refractories has two particular effects: (i) the difference in coefficient of thermal expansion that is effective for compositions including low rate of spinel (i.e. M–5spinel or M–10spinel) causes more microcrack generation when interlinking has not started yet [33,34], and (ii) pinning effect of  $ZrO_2$  inhibits grain growth. Amount of reinforcement has a significant effect on strength both for MgO–spinel and for MgO–spinel–zircon but not as much as  $SiO_2$  which forms forsterite ( $Mg_2SiO_4$ ) and causes a natural bond between matrix MgO grains (Fig. 6). Even though forsterite brick, without any addition, does not have enough refractivity or corrosion resistance, secondary forsterite formation in basic bricks improves densification and slag leakage resistance through joining matrix grains [35,36].

Since  $ZrO_2$  stays in the cubic form even after cooling, advance in the fracture toughness with zircon addition cannot be associated with transformation toughening of  $ZrO_2$ , but the forsterite formation that aids sintering.

The microstructural image of M–5S–20zircon composition is given in Fig. 7. The white colored  $ZrO_2$  grains released as a result of the decomposition of zircon mostly lie in the MgO boundaries. It is observed that the microcracks formed in the structure are connected to each other and progress for a short distance and stop when they reach cubic- $ZrO_2$  grains or when they reach to the pores.

Fig. 8 shows that for spinel composites, there was no such a fall either in strength or in Young's modulus like for the compositions prepared with more than 10% spinel and zircon. On the other hand, zircon had a little effect on the Young's

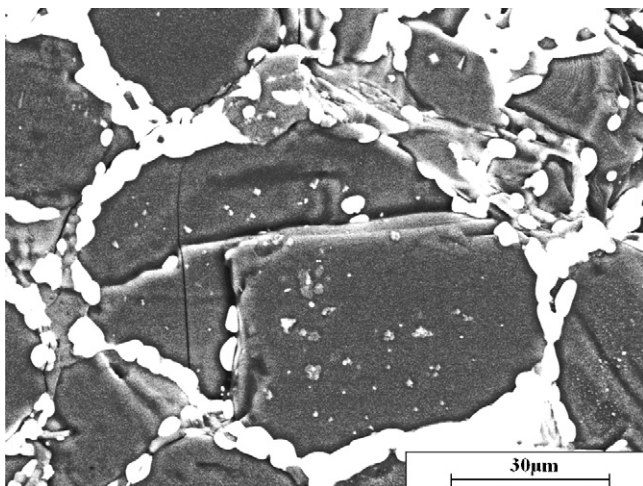


Fig. 7. SEM micrograph showing the microstructure of a M–5S–20zircon composite.

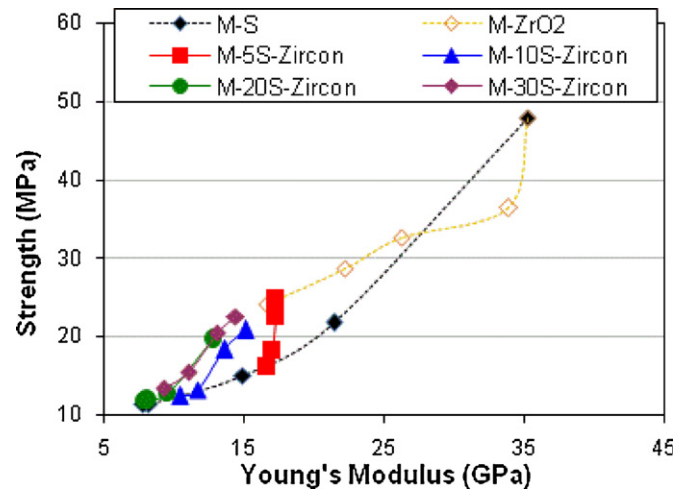


Fig. 8. Strength as function of Young's modulus (M: MgO, S:  $MgAl_2O_3$ , and Zircon:  $ZrSiO_4$ ).

modulus of the refractory materials with 5% spinel whereas strength declined dramatically, which shows these materials are more sensitive to additives than the others.

Fracture surface energy values obtained from modulus and SENB toughness values using the standard relationship:

$$\gamma_i = \frac{K_{IC}^2}{2E} \quad (4)$$

are given in Fig. 9.

Surface energy decreased by  $\sim 50\%$  with the additions of spinel up to 10% and values remained almost constant for further additions whereas more than 10%  $ZrO_2$  did not affect the fracture surface energy after 10% increase. As the microcracks, generated during cooling, extended with spinel addition, transgranular fracture faces of pure MgO showed a marked tendency towards intergranular fracture, the tendency being more marked the higher the spinel loading, as illustrated in Fig. 10.

In the case of MgO–spinel–zircon, fracture surface energy of the refractory materials increased with zircon incorporation,

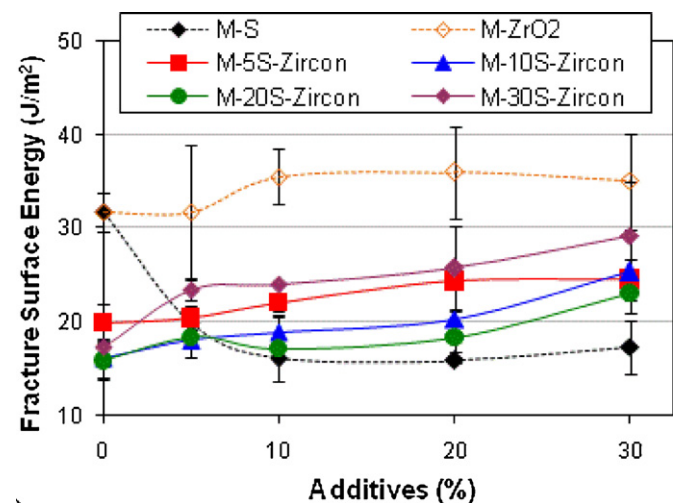


Fig. 9. Fracture surface energy as a function of additives (M: MgO, S:  $MgAl_2O_3$ , and Zircon:  $ZrSiO_4$ ).



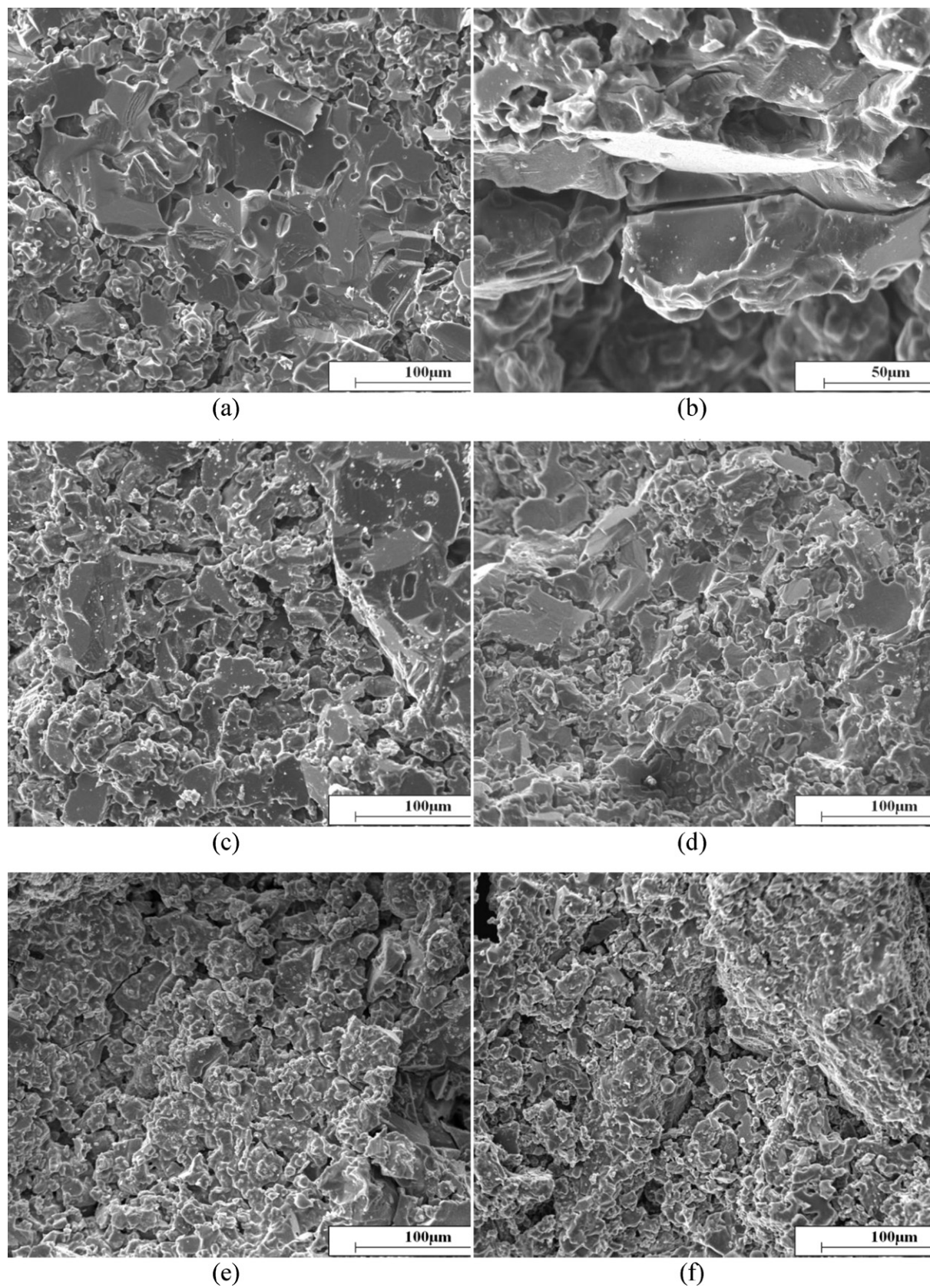


Fig. 10. SEM micrograph of a fracture surface of (a) and (b) pure MgO, (c) M-5spinel, (d) M-10spinel, (e) M-20spinel and (f) M-30spinel.

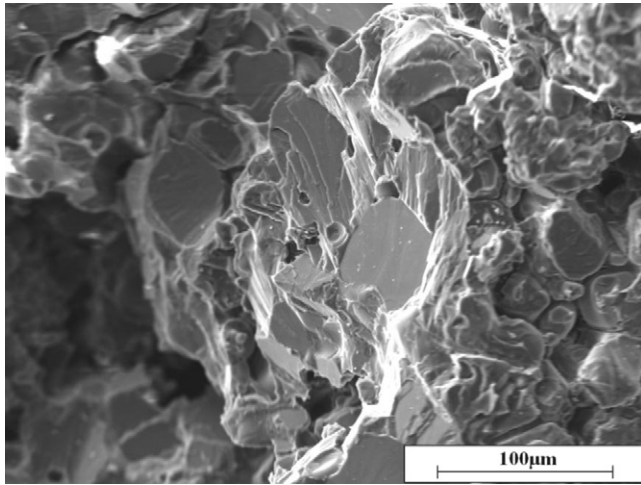


Fig. 11. SEM micrograph of a fracture surface of M-5S-20zircon.

M-30S-30zircon being ~2-fold higher than M-30spinel. Forsterite formation not only caused a natural bonding between matrix particles but also diminished weak zones within the entire material which are basically the source of initiation of fracture propagation. As a result, fracture path turned to transgranular along with defective zones decrease (Fig. 11).

Values for the work of fracture ( $\gamma_{\text{WOF}}$ ) were calculated from load–deflection curves obtained from notched bars deformed in three-point bend, by measuring the area ( $U$ ) under the load–deflection curve.  $\gamma_{\text{WOF}}$  is given by the following equation [37,38]:

$$\gamma_{\text{WOF}} = \frac{U}{2W(D-c)} \quad (5)$$

There was a general increase in  $\gamma_{\text{WOF}}$  with increasing spinel content (Fig. 12). Fracture surfaces of pure MgO showed a large proportion of transgranular cracks, with some intergranular cracks (Fig. 10a–d). At low rate additions of spinel, transgranular cracks were still present with some intergranular cracks, in the fracture surfaces of each spinel composite. However, mostly intergranular fracture occurred at higher additions of spinel (20–30%) (Fig. 10e and f). It therefore appears that higher values

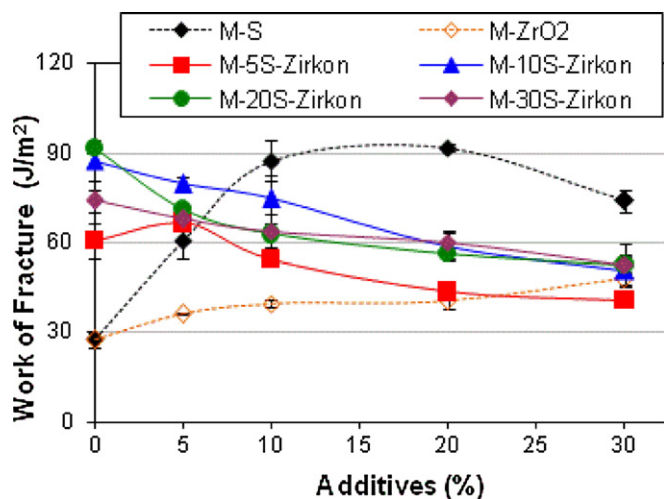


Fig. 12. Work of fracture as a function of additives (M: MgO, S: MgAl<sub>2</sub>O<sub>3</sub>, and Zircon: ZrSiO<sub>4</sub>).

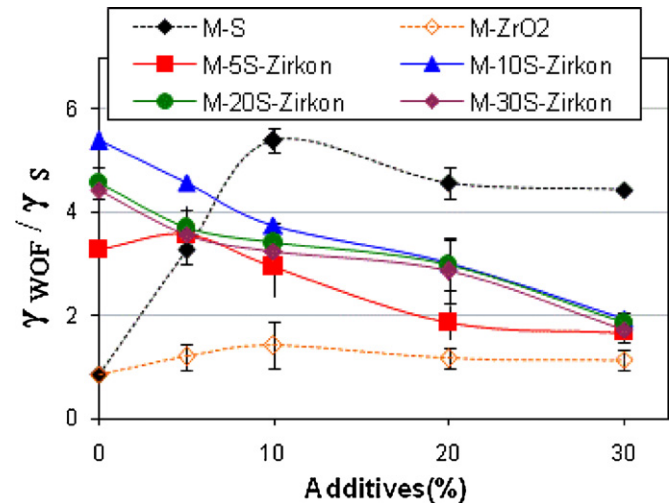


Fig. 13.  $\gamma_{\text{WOF}}/\gamma_i$  ratio as a function of additives (M: MgO, S: MgAl<sub>2</sub>O<sub>3</sub>, and Zircon: ZrSiO<sub>4</sub>).

of  $\gamma_{\text{WOF}}$  are associated with the occurrence of more intergranular fracture with increasing spinel additions. The fracture of the magnesia–spinel composites is either semi-stable or stable, but never catastrophic like MgO. It may be concluded that crack propagation is a much greater energy consuming process than crack initiation in these materials. For many industrial applications, the initiation of fracture is less important than  $\gamma_{\text{WOF}}$  and the degree of damage (e.g. further loss of mechanical integrity by strength and material loss through large scale fracture behind the hot face) [39]. Large values of the  $\gamma_{\text{WOF}}/\gamma_i$  ratio were obtained in these composites (Fig. 13). This is a basic requirement for refractory materials to show good thermal shock damage resistance [40]. Fig. 13 illustrates that the M-10spinel composite showed the highest  $\gamma_{\text{WOF}}/\gamma_i$  ratio, by a factor of ~6, as compared to pure MgO. For refractory materials prepared with MgO–ZrO<sub>2</sub>,  $\gamma_{\text{WOF}}$  values had an increasing trend but not as much as in MgO–spinel owing to the fact that ZrO<sub>2</sub> had much smaller grain size which kept fracture in transgranular form. In the case of M–S–zircon,  $\gamma_{\text{WOF}}$  generally decreased with increasing amount of zircon. Forsterite formation, at the grain boundaries, strengthens the microstructure and causes the fracture type to change into transgranular form (Fig. 11). None of the compositions, containing zircon, has as high  $\gamma_{\text{WOF}}$  as M–spinel. On the other hand, even these materials could have high thermal shock resistance due the fact that they have higher resistance to fracture initiation.

#### 4. Conclusions

Cubic-ZrO<sub>2</sub> and forsterite were detected in MgO–spinel–zircon refractories accompanied with the matrix periclase and spinel. Mechanical properties of MgO–spinel refractories deteriorated with the increasing amount of spinel up to 20% and higher loads caused no more damage, for which MgO–ZrO<sub>2</sub> refractories showed the same trend but not as effective as MgO–spinel. With the addition of zircon, both strength and Young's modulus as well as fracture toughness and fracture surface energy increased for almost 2 times higher than MgO–spinel composite refractories with the aid of forsterite formation which formed a natural bonding

between matrix grains. Forsterite formation not only increased the mechanical properties but also caused the fracture path to change into transgranular form which is an indicator of resistance to fracture initiation, but caused  $\gamma_{\text{WOF}}$  to be decreased.

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