

# Effect of $\text{Al}_2\text{O}_3$ on the microstructure and mechanical properties of $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$ in situ composites synthesized by reactive hot pressing

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## Abstract

$\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  in situ composites with different  $\text{Al}_2\text{O}_3$  contents were successfully synthesized from the powder mixture of Ti, TiC, Al and  $\text{TiO}_2$  by a reactive hot-pressing method at 1350 °C. The effect of  $\text{Al}_2\text{O}_3$  on the microstructure and mechanical properties of the composites was investigated in detail. The results indicate that the as-fabricated products mainly consist of  $\text{Ti}_3\text{AlC}_2$ ,  $\text{Al}_2\text{O}_3$  and a small amount of TiC. With increasing the  $\text{Al}_2\text{O}_3$  content, the flexural strength of  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites increase gradually, the fracture toughness reaches the peak value of 8.21  $\text{MPa m}^{1/2}$  as the  $\text{Al}_2\text{O}_3$  content increasing to 9 wt%, the hardness attains the maximum value of 10.16 GPa for 12 wt%  $\text{Al}_2\text{O}_3$ . The strengthening mechanism of the composites was also discussed.

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## 1. Introduction

$\text{Ti}_3\text{AlC}_2$  is a layered machinable ceramic representative of  $\text{M}_{n+1}\text{AX}_n$  phases (also called MAX), with  $n = 1, 2$  or 3, where M is an early transition metal, A is an A-group element and X is C or N [1–4]. It combines both the merits of metals and ceramics. Like metals, it is good electrical and thermal conductive, relatively low hardness, and plastic at the high temperatures, and is easy to be machined by both electrical discharge method and conventional cutting tools. Like ceramics, it possesses high melting point, high modulus, low density, high thermal stability and good antioxidation [5,6]. In addition,  $\text{Ti}_3\text{AlC}_2$  also has nice self-lubricating property. However, the potential application of  $\text{Ti}_3\text{AlC}_2$  as a high temperature structural material is limited due to its low hardness and unsatisfactory strength [7,8].

Incorporation of a second phase is an effective way to overcome these weaknesses of MAX carbides [8–14]. Considering the high hardness and modulus, excellent chemical

stability and approximate thermal expansion coefficient,  $\text{Al}_2\text{O}_3$  is chosen as reinforcement to increase the mechanical properties of MAX, such as  $\text{Ti}_3\text{SiC}_2$  [9,10] and  $\text{Ti}_3\text{AlC}_2$ . Wu et al. [11] reported that the addition of  $\text{Al}_2\text{O}_3$  into  $\text{Ti}_3\text{AlC}_2$  matrix was shown to enhance its friction behavior and wear resistance. Chen et al. [12,13] successfully fabricated  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  directly from the powder mixture of Ti, C, Al and  $\text{Al}_2\text{O}_3$  by in situ hot pressing/solid–liquid reaction process at 1500 °C. In comparison to monolithic  $\text{Ti}_3\text{AlC}_2$ , the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites are substantially improved in the hardness, toughness, compressive strength, and flexural strength. However, the large size and inhomogeneous distribution of the  $\text{Al}_2\text{O}_3$  particles hindered further increase of its mechanical properties. Recently, Yeh et al. [14] successfully prepared the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  and  $\text{Ti}_2\text{AlC}/\text{Al}_2\text{O}_3$  composites through in situ synthesis in the mode of self-propagating high-temperature synthesis (SHS) in Ti–Al–C– $\text{TiO}_2$  systems, but work on the strengthening  $\text{Ti}_2\text{AlC}$  and  $\text{Ti}_3\text{AlC}_2$  by incorporation of second phase  $\text{Al}_2\text{O}_3$  is not reported. However, to our knowledge, the method of in situ reactive hot pressing has not been applied into the synthesis of the  $\text{Ti}_3\text{AlC}_2$  matrix composites. The reactive hot pressing represents an in situ processing technique for the fabrication of composites, which takes the advantage of low energy requirement, contaminant-free interface.

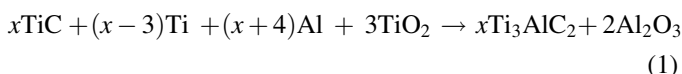
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So the objective of the present work is to fabricate  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites in the Ti–TiC–Al– $\text{TiO}_2$  system using the in situ reactive hot pressing technique. The effect of the content of in situ formed  $\text{Al}_2\text{O}_3$  on the phase composition, microstructure, density and room-temperature mechanical properties including hardness, flexural strength, and fracture toughness of the composites were investigated in detail. And the strengthening mechanism of the composites was also discussed.

## 2. Experimental

Ti (50  $\mu\text{m}$ , 99.3% purity), TiC (75  $\mu\text{m}$ , 99.05% purity), Al (75  $\mu\text{m}$ , 99.5% purity), and  $\text{TiO}_2$  (25  $\mu\text{m}$ , 99.5% purity) powders were used as the starting materials. For the preparation of  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  in situ composites, the powder mixtures containing Al,  $\text{TiO}_2$ , Ti, and TiC were formulated with stoichiometry given according to the following reaction. Considering the vaporization of Al at high temperatures, the content of TiC was reduced to 90 wt% and the content of Al was increased to 10 wt% in the starting powders. The detailed compositions of the samples were listed in Table 1.



First, the blend of powders was ball milled in alcohol for 1 h and then dried at 50 °C for several hours. Then the milled powders were compacted uniaxially under 10 MPa in a graphite mold and coated with BN inside. The compacted samples were hot press sintered in vacuum (less than  $6.0 \times 10^{-2}$  Pa) at a rate of 5 °C/min to 1350 °C and held for 2 h under a pressure of 16 MPa. Finally, the samples were cooled down to room temperature in the furnace. The surfaces of samples were ground to remove graphite layer. The phase analysis of the sintered bulks was identified by X-ray diffraction (XRD) using a diffractometer with Cu K $\alpha$  radiation (D/max-2200PCX). The microstructures of the samples were investigated by scanning electron microscope (SEM, JEOL JSM-6700F) and energy dispersive spectroscopy (EDS).

The measurements of all the mechanical properties were performed at room temperature by averaging 6 individual measurements. The sample densities were measured by Archimedes' method. Hardness measurements were performed on the HXD-1000 tester (Shanghai Second Optical Ltd., China) with a diamond indenter under a 1000 N load for 15 s. The three-point bending flexural strength tests were conducted by

Table 1  
The compositions of the samples (wt%).

Specimens No.	Ti	TiC	Al	$\text{TiO}_2$	Theoretical targeted $\text{Al}_2\text{O}_3$ content
YL0 <sup>a</sup>	28.49	54.87	15.68	1.17	0
YL5	24.50	52.67	17.39	5.85	5
YL9	20.79	50.61	18.99	10.22	9
YL12	17.95	48.99	20.23	13.61	12
YL15	14.62	47.19	21.64	17.47	15

<sup>a</sup> YLXX refers to the sample number, and XX is the targeted weight percent of  $\text{Al}_2\text{O}_3$  phase.

the PC-1036PC (Perfect Instrument Co. Ltd., Taiwan, China) material tester with a span of 20 mm at a cross-head speed of 0.5 mm/min with the specimen dimensions of 4 mm  $\times$  4 mm  $\times$  30 mm. The fracture toughness  $K_{\text{IC}}$  was measured on the PC-1036PC testing machine by using the single edge pre-cracked beam (SEPB) method with a notch depth of 0.4W (W is the width of specimen), at a cross-head speed of 0.05 mm/min, with a loading span of 20 mm. The flexural strength and fracture toughness were calculated with the following equations, respectively.

$$\sigma = \frac{3PL}{2bh^2} \quad (2)$$

$$K_{\text{IC}} = Y \times \frac{3PLa^{1/2}}{2BW^2} \quad (3)$$

## 3. Results and discussion

The X-ray diffraction patterns of monolithic  $\text{Ti}_3\text{AlC}_2$  and  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites hot pressed at 1350 °C for 2 h are shown in Fig. 1. It can be clearly seen that the monolithic  $\text{Ti}_3\text{AlC}_2$  from the sample YL0 of Ti:TiC:Al = 1.2:1.8:1.1 (in mole ratio) is composed of  $\text{Ti}_3\text{AlC}_2$  and small amount of TiC (Fig. 1(a)). The previous work [15] also revealed that the TiC often existed in the final products as an intermediate phase for synthesis of  $\text{Ti}_2\text{AlC}$  and  $\text{Ti}_3\text{AlC}_2$  due to the short reaction time and large amounts of reaction heat released from the thermal explosion reaction between the Ti and C. When small amount of 5.85 wt%  $\text{TiO}_2$  was incorporated in the initial materials (Fig. 1(b)), a small amount of  $\text{Al}_2\text{O}_3$  phase was also found in addition to the  $\text{Ti}_3\text{AlC}_2$  phase. It is worth noting that that the as-sintered sample is composed of dominant  $\text{Ti}_3\text{AlC}_2$  and a small amount of  $\text{Al}_2\text{O}_3$  and TiC, and the ratio of  $\text{Al}_2\text{O}_3$  content to  $\text{Ti}_3\text{AlC}_2$  content increases with increasing  $\text{TiO}_2$  amount in the initial materials (Fig. 1(c)–(e)). Compared with the previous works [4,14], the synthesis temperature of 1350 °C in this work is much lower than that of 1400 °C. The presence of the above-mentioned phases in the composite samples confirms the

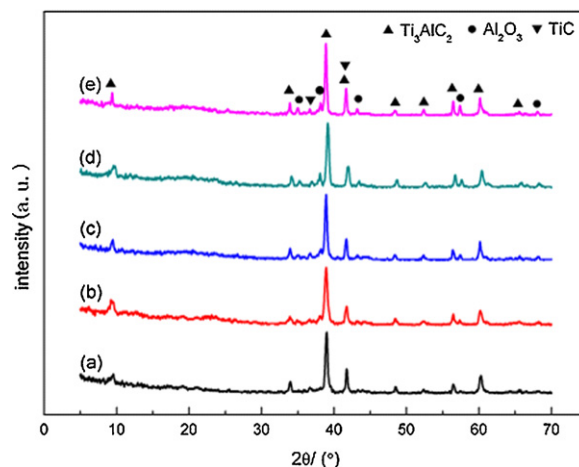


Fig. 1. X-ray diffraction patterns of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites with various  $\text{Al}_2\text{O}_3$  contents. (a) 0 wt%; (b) 5 wt%; (c) 9 wt%; (d) 12 wt%; and (e) 15 wt%.

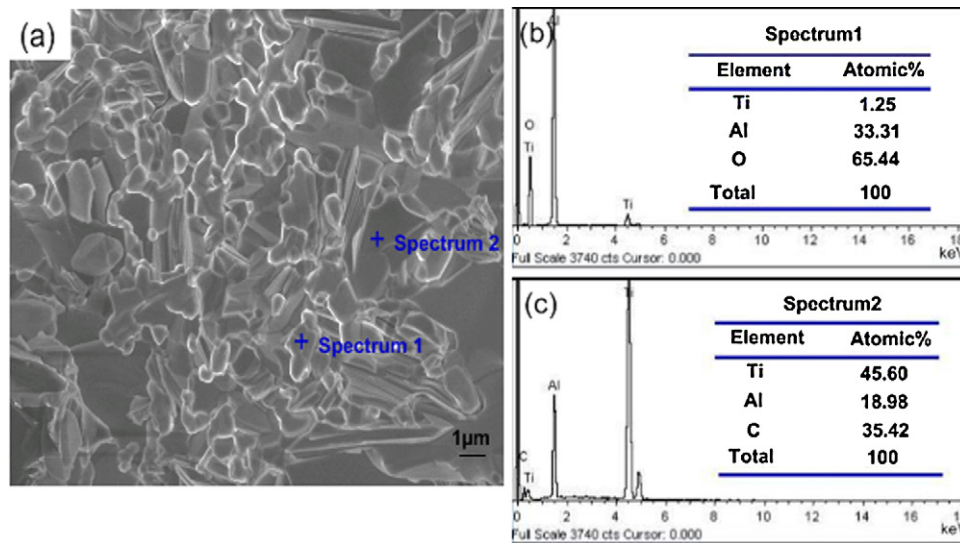


Fig. 2. (a) SEM micrographs and (b) and (c) EDS analysis results of the fracture surface of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  with 9 wt%  $\text{Al}_2\text{O}_3$  composites sintered at 1350 °C.

feasibility of the overall reaction expressed as Eq. (1). The in situ formation of  $\text{Al}_2\text{O}_3$  along with  $\text{Ti}_3\text{AlC}_2$  is resulted from the thermite reaction of Al and  $\text{TiO}_2$ . It should be noted that the thermite reaction between Al and  $\text{TiO}_2$  is a highly exothermic reaction, which releases large amounts of reaction heat. The increasing temperature facilitates faster and more complete diffusion of the reactants during reactive hot pressing, leading to a more uniform distribution of the second phase,  $\text{Al}_2\text{O}_3$ , and lower synthesis temperature of 1350 °C. The presence of TiC in the final product has often been observed in the formation of  $\text{Ti}_3\text{AlC}_2$  and  $\text{Ti}_2\text{AlC}$  as an intermediate phase. So the reaction (1) could be divided into following reactions:

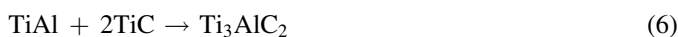


Fig. 2 displays the SEM micrograph of the fracture surface and the EDS analysis of  $\text{Ti}_3\text{AlC}_2/9$  wt%  $\text{Al}_2\text{O}_3$  composite. It is obvious that the material consists of the lamellar matrix phase and the particle-like second phase. The size of particles is about 1 μm, the platelike matrix grains are closely packed into a laminated configuration, which is characteristic of the layered ternary carbides,  $\text{Ti}_3\text{AlC}_2$ . Combined with the EDS analysis, the chemical composition indicates that the particle phase is enriched with Al, O and a small amount of Ti, and the atomic compositions of Al and O are 33.41% and 65.44% in mole ratio, respectively, which is close to stoichiometric  $\text{Al}_2\text{O}_3$ . From Fig. 1(c), for the lamellar matrix phase the quantitative atomic ratio of Ti:Al:C is 45.60:19.89:35.42, which is quite consistent with the predicted one of 3:1:2 for  $\text{Ti}_3\text{AlC}_2$ . Therefore, the composite is mainly composed of the matrix  $\text{Ti}_3\text{AlC}_2$  and the second reinforcement phase,  $\text{Al}_2\text{O}_3$ . The latter one disperses at the grain boundaries of the matrix phase.

Fig. 3 displays the SEM micrographs of the in situ reactive hot pressed  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites with different  $\text{Al}_2\text{O}_3$  contents. When the  $\text{Al}_2\text{O}_3$  content is 5 wt%, the matrix  $\text{Ti}_3\text{AlC}_2$

are coarse with a grain size of 10–15 μm (Fig. 2(a)). When increasing the  $\text{Al}_2\text{O}_3$  content to 9 wt%, the size of the matrix grain decreases significantly to 5–10 μm, and  $\text{Al}_2\text{O}_3$  grains disperses uniformly in the matrix (Fig. 2(b)). With further increasing the  $\text{Al}_2\text{O}_3$  content to 15 wt%, surface pores and local agglomeration of  $\text{Al}_2\text{O}_3$  grains tend to occur at the grain boundaries (Fig. 2(c)). It seems that too many  $\text{Al}_2\text{O}_3$  grains would hinder the densification process of the composites.

Fig. 4 shows that the density and hardness of  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites with different  $\text{Al}_2\text{O}_3$  contents. With increasing  $\text{Al}_2\text{O}_3$  contents, the density of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites gradually decreases, due to the fact that the density of  $\text{Al}_2\text{O}_3$  is smaller than that of  $\text{Ti}_3\text{AlC}_2$ . It is worth noting that the measured densities of all synthesized  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites are 99–99.7% of the theoretical density, which indicates that the as fabricated samples are nearly full dense. From Fig. 4, it can also be seen that with increasing  $\text{Al}_2\text{O}_3$  content, the Vickers hardness of  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  gradually increases. It is reported that the hardness of monolithic  $\text{Ti}_3\text{AlC}_2$  is about 3.5–4.7 GPa [4]. In this study, the maximum value of hardness reaches 10.16 GPa as the  $\text{Al}_2\text{O}_3$  content increases to 12 wt%, which is almost 3 times of that of the monolithic  $\text{Ti}_3\text{AlC}_2$ . This partially derives from the higher hardness of the reinforcing  $\text{Al}_2\text{O}_3$  phase, and partially from the refined grain size of the composites resulted from the second phase of  $\text{Al}_2\text{O}_3$ . However, when the  $\text{Al}_2\text{O}_3$  content is high enough (15 wt%), the hardness decreases, which is attributed to the loose microstructure caused by too many  $\text{Al}_2\text{O}_3$  content, which is consistent with the above SEM results.

Fig. 5 shows the flexural strength and fracture toughness of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites with different  $\text{Al}_2\text{O}_3$  contents. It can be clearly seen that the flexural strength increases from 497 to 720 MPa as increasing  $\text{Al}_2\text{O}_3$  content from 0 to 15 wt%. However, the  $K_{IC}$  increases with increasing the  $\text{Al}_2\text{O}_3$  content, and reach a maximum value of 8.2 MPa m<sup>1/2</sup> at 9 wt%  $\text{Al}_2\text{O}_3$  content, and then decreases dramatically. Combined with the SEM analysis results (Fig. 3), the fracture toughness of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites begins to

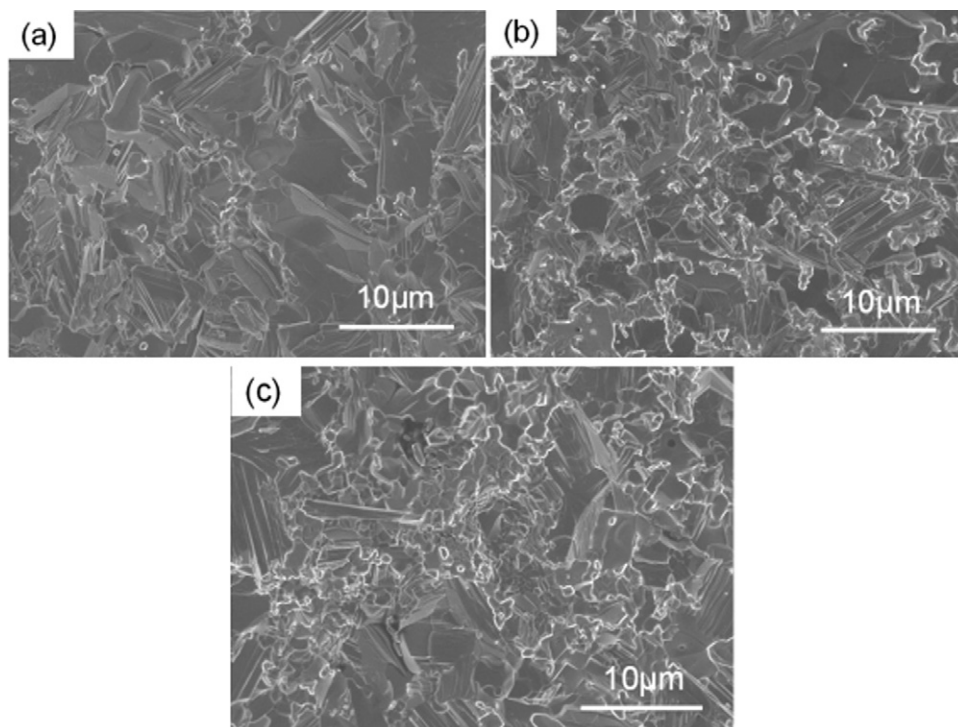


Fig. 3. SEM microstructures of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites with different  $\text{Al}_2\text{O}_3$  contents. (a) 5 wt%; (b) 9 wt%; and (c) 15 wt%.

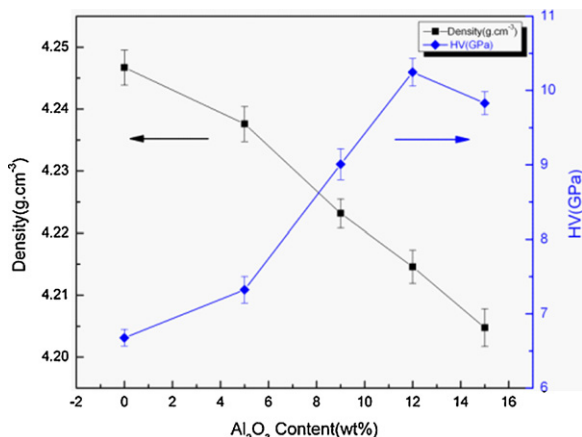


Fig. 4. Density and hardness of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites with different  $\text{Al}_2\text{O}_3$  contents.

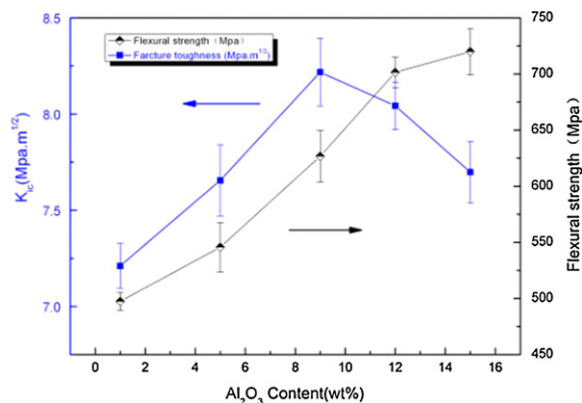


Fig. 5. Flexural strength and fracture toughness of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites with different  $\text{Al}_2\text{O}_3$  contents.

decrease when the  $\text{Al}_2\text{O}_3$  content is exceeded to 9 wt%, due to the agglomeration of the  $\text{Al}_2\text{O}_3$  phase. So the optimum  $\text{Al}_2\text{O}_3$  concentration is 9 wt%, at which the flexural strength and the fracture toughness are increased by 25.95% and 13.86%, respectively. The typical properties of the composite with 9 wt%  $\text{Al}_2\text{O}_3$  are summarized in Table 2. For comparison, the results of the monolithic  $\text{Ti}_3\text{AlC}_2$  fabricated by the solid-liquid reaction synthesis and simultaneous in situ hot pressing process in Ref. [4],  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composite fabricated from the powder mixture of Ti, C, Al and  $\text{Al}_2\text{O}_3$  by in situ hot pressing process in Ref. [12] and  $\text{Ti}_3\text{AlC}_2/\text{TiC}-\text{Al}_2\text{O}_3$  composites obtained by in situ synthesis in the mode of thermal explosion from  $3\text{TiO}_2-5\text{Al}-2\text{C}$  system in Ref. [16] are also included. It can be seen that flexural strength and fracture toughness of the present sample are much higher than those of the monolithic  $\text{Ti}_3\text{AlC}_2$  and  $\text{Ti}_3\text{AlC}_2/\text{TiC}-\text{Al}_2\text{O}_3$  reported in Refs. [4] and [16], respectively. The fracture toughness is similar to that of the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composite reported in Ref. [12], and the hardness and flexural strength are much higher, which are resulted from the finer microstructure with uniform distribution. Compared with Ref. [16], the hardness of the present samples are much lower than  $\text{Ti}_3\text{AlC}_2/\text{TiC}-\text{Al}_2\text{O}_3$  due to the higher content of  $\text{Al}_2\text{O}_3$  in the latter.

The modification of the strength and fracture toughness is mainly attributed to the in situ formed  $\text{Al}_2\text{O}_3$ . The previous work [10,12] revealed that the addition of  $\text{Al}_2\text{O}_3$  would improve the mechanical properties due to its high hardness, 18 GPa, and high Young's modulus, 386 GPa. The fine grain size controlled by the presence of dispersed fine  $\text{Al}_2\text{O}_3$  grains and sintering at relatively low temperature of 1350 °C is another important reason to strengthen the  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites.



Table 2

Summary of typical properties of the as-prepared  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composite (with  $\sim 9$  wt%  $\text{Al}_2\text{O}_3$ ), together with monolithic  $\text{Ti}_3\text{AlC}_2$  [4],  $\text{Ti}_3\text{AlC}_2/\text{TiC}-\text{Al}_2\text{O}_3$  [16] and  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composite [12] for comparison.

Properties	Flexural strength (MPa)	Fracture toughness ( $\text{MPa m}^{1/2}$ )	Density ( $\text{g cm}^{-3}$ )	Hardness (GPa)	References
$\text{Ti}_3\text{AlC}_2$	340	7.2	4.24	3.5–4.7	[4]
$\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$	500	8.8	–	5.5	[12]
$\text{Ti}_3\text{AlC}_2/\text{TiC}-\text{Al}_2\text{O}_3$	$466 \pm 39$	$5.8 \pm 0.3$	–	$13.3 \pm 1.1$	[16]
$\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$	$626 \pm 30$	$8.2 \pm 0.2$	4.224	$8.9 \pm 0.2$	In this work

#### 4. Conclusions

The  $\text{Ti}_3\text{AlC}_2/\text{Al}_2\text{O}_3$  composites were synthesized by the in situ hot pressing method under a pressure of 16 MPa at 1350 °C for 2 h based on the Ti–TiC–Al–TiO<sub>2</sub> system. The phases of the as-sintered samples are mainly made of small sized  $\text{Ti}_3\text{AlC}_2$  and  $\text{Al}_2\text{O}_3$ . The introduction of  $\text{Al}_2\text{O}_3$  improves the hardness, fracture toughness and flexural strength of the composites. When the  $\text{Al}_2\text{O}_3$  content is 9 wt%, the hardness, flexural strength and fracture toughness of the in situ composites increase by 62.18%, 25.95% and 13.86%, respectively.

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