

# Atomic structure and chemistry of dense nano-precipitates in $\text{MgB}_2$ ceramic

Xueyan Song\*

*Department of Mechanical and Aerospace Engineering, West Virginia University, Morgantown, WV 26506, USA*

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## Abstract

Ceramic  $\text{MgB}_2$  superconductor has great potential for applications in the devices. Doping  $\text{MgB}_2$  with carbon or SiC has been successful in improving critical current density  $J_c$ , the upper critical field  $H_{c2}$  and flux-pinning. However, the presence of nano-precipitates, and their microstructure and chemistry, upon alloying/doping  $\text{MgB}_2$  grains are largely unexplored. In this paper, the atomic structure and chemistry of 5–10 nm dense nano-precipitates were discovered in purified  $\text{MgB}_2$  pellets by removing  $\text{B}_2\text{O}_3$ . The nano-precipitates are platelet-shaped and fully coherent with the  $\text{MgB}_2$  grains and having the longer axis of the plate perpendicular to the  $c$ -axis of  $\text{MgB}_2$  grains. Systematic conventional and analytical Transmission Electron Microscopy (TEM) was carried out to study the atomic structure and chemistry of those nano-precipitates. The practical experimental approach of optimizing the composition and synthesis heat treatment of  $\text{MgB}_2$  to induce dense uniformly distributed nano-precipitates is also discussed.

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## 1. Introduction

$\text{MgB}_2$  superconductor, with superconducting temperature  $T_c$  of 39 K, has great potential for application in the electric power transportation and devices. Because the grain boundaries (GBs) do not block current flow [1,2] as they do in the high temperature superconductors (HTS), the technological barriers to  $\text{MgB}_2$  conductor fabrication are very much less than that for HTS [3,4]. But in the  $\text{MgB}_2$  material, the critical current density ( $J_c$ ) drops rapidly with increasing magnetic field strength [5]. One of the practical challenges of the development of  $\text{MgB}_2$  superconducting wires is to improve the  $J_c$ . The magnitude and field dependence of the critical current are related to the presence of structural defects, which can ‘pin’ the quantized magnetic vortices that permeate the material. Currently,  $J_c$  of  $\text{MgB}_2$  is severely affected by the materials microstructure issues such as the high porosity, bad GB-connectivity and the lack of the crystal defects that could act as flux-pinning centers within the  $\text{MgB}_2$  grains. Accordingly, one critical challenge for development of high

performance  $\text{MgB}_2$  is to understand and to optimize the flux pinning in  $\text{MgB}_2$ .

Various research have been attempted to improve the flux pinning for  $\text{MgB}_2$ , such as irradiation [6], chemical doping [7] and reducing the grain size [8,9].  $\text{MgB}_2$  is clearly able to incorporate a variety of nanoscale pins, such as  $\text{MgO}$  [10], SiC-doping [11], and  $\text{Y}_2\text{O}_3$ -doping [12], with significant improvement of bulk pinning. However, the practical difficulty of incorporating nanoparticles in the  $\text{MgB}_2$  is that adding those materials will increase the normal state resistivity, and those added materials often end up sitting in the GBs of  $\text{MgB}_2$  and reduce the active cross-section of carrying the super-current [13–15]. Their influence of blocking the current can easily exceed the benefits of having effective pinning centers in high fields.

By contrast, inducing the dense uniformly distributed nano-precipitates into the  $\text{MgB}_2$  grains, without having the accumulation of the precipitates at the  $\text{MgB}_2$  GBs would be an ideal case for improving the flux pinning properties. Precipitates in a superconductor can pin vortices [16–19] by reduction or suppression of the condensation energy within their volume, and more indirectly by creating defects such as stacking faults, dislocations in the superconducting matrix.

\*Tel.: +1 304 293 3269; fax: +1 304 293 6689.

E-mail address: [xueyan.song@mail.wvu.edu](mailto:xueyan.song@mail.wvu.edu)

For effective pinning centers, the size and the separation of the foreign particles are critical [20]. For example, pinning centers in conventional methods of fabrication of low  $T_c$  superconductors are developed by modifying the thermomechanical processing. Finely dispersed nano-sized non-superconducting Ti phase precipitated during heat treatment is strong flux pinners in NbTi. For effective pinning the typical thickness of the  $\alpha$ -Ti phase must be 1–4 nm with a separation of 5–20 nm [21,22]. For HTS  $\text{YBa}_2\text{Cu}_3\text{O}_{7-x}$  (YBCO) coated conductors, both microstructural and electromagnetic evidences were found for an enhanced vortex pinning in Sm substituted coated conductors which produces 17 vol% of mixed  $(\text{Y},\text{Sm})_2\text{O}_3$  nano-precipitates that are  $\sim 10$  nm in size and separated by  $\sim 15$  nm. Such precipitate arrays can be a powerful way to enhance the critical current density  $J_c$  and high-field performance of YBCO coated conductors [23]. By similar approach, for having the strong flux pinning in  $\text{MgB}_2$ , the intragranular nano-precipitates should be dense and with the size comparable to the  $\text{MgB}_2$  superconducting coherence length of 5–12 nm [4,20]. However, for the  $\text{MgB}_2$ , the production of dense uniformly distributed nano-precipitates remains challenging in  $\text{MgB}_2$  synthesized in any form, such as bulk, thin film and wires. The presence of nano-precipitates, and their microstructure and chemistry, upon alloying/doping  $\text{MgB}_2$  grains are largely unexplored.

In this paper, the dense and evenly distributed nano-precipitates are found in the bulk scale  $\text{MgB}_2$  ceramic samples. The atomic structure and chemistry of the precipitates were analyzed using advanced TEM. The practical experimental approach of optimizing the composition and synthesis heat treatment of  $\text{MgB}_2$  to induce dense uniformly distributed nano-precipitates, and the effect of such precipitates on the electromagnetic properties and application of  $\text{MgB}_2$  is discussed.

## 2. Experimental details

As-received amorphous B powder (Alfa-Aesar 99%) was pressed into 8–10 mm diameter pellets by cold isostatic pressing, and then heat treated at 950 °C for 48 h in flowing 4%  $\text{H}_2$  in Ar in order to remove  $\text{B}_2\text{O}_3$ . Such pellets were reacted at 950 °C for 24 h with Mg vapor to form  $\text{MgB}_2$ . The synthesis and processing procedure of the  $\text{MgB}_2$  pellets were detailed in a paper [24]. In this paper, the detailed TEM analysis is presented on the morphology, distribution, atomic structure and chemistry of the nano-precipitates found in those purified  $\text{MgB}_2$  samples.

Samples for TEM were prepared by mechanical polishing and ion milling. Diffraction contrast and high-resolution TEM (HRTEM) investigations were performed using a microscope (Philips CM200) operated at 200 kV with point resolution of 0.19 nm. High-angle annular dark-field (HAADF) Z-contrast imaging in a scanning TEM (STEM) was performed using a 200 kV (JEOL 2010F) microscope focused to a  $\sim 0.14$  nm diameter electron probe with the inner cut-off angle of the HAADF detector  $> 52$  mrad. The nanoscale chemistry was analyzed using a

0.2 nm probe and on-axis electron energy loss spectroscopy (EELS) with an energy dispersion of 0.2 eV per channel and an energy resolution of 1.0 eV in a JEOL 2010F STEM.

## 3. Results and discussion

### 3.1. Morphology and crystallographic features of precipitates in the $\text{MgB}_2$ grains

The electron diffraction contrast images taken from the  $\text{MgB}_2$  grains in Fig. 1 shows that the  $\text{MgB}_2$  grains are with size of microns. There are pores with the size 100–200 nm. Some of the pores are faceted. There are spherical MgO particles with the size of 40 nm or bigger within the  $\text{MgB}_2$  grains. Pores and MgO particles are commonly observed in the  $\text{MgB}_2$  film, wire and bulk samples. The unique feature of this sample is found when the electron beam is imaging along the [100] direction of  $\text{MgB}_2$  grains. Diffraction contrast images under higher magnification in Fig. 2 shows the uniform distribution of dense nano-precipitates. The precipitates are platelet-shaped with the size of 5–15 nm in longer axis and 2–5 nm in the shorter axis. The nano-precipitates are with well-defined crystallography relationship with the  $\text{MgB}_2$  matrix and the longer axis of the nano-precipitates is lying on the basal plane of the hexagonal  $\text{MgB}_2$  lattice. HRTEM images taken from the [100] zone axis of  $\text{MgB}_2$  grains in Fig. 3 confirm that the platelet precipitates all lie with their long axis perpendicular to the [001] direction of  $\text{MgB}_2$  grains. The HRTEM images also show the precipitates have the same lattice fringes as the  $\text{MgB}_2$  matrix except for having the periodic contrast modulating along the [001] direction of the  $\text{MgB}_2$ . The modulation period of nano-precipitates is twice of the  $\text{MgB}_2$  lattice constant along the  $c$ -axis. In consistent with the image contrast, the corresponding Fourier transformation from the area containing nano-precipitates (Fig. 3a)

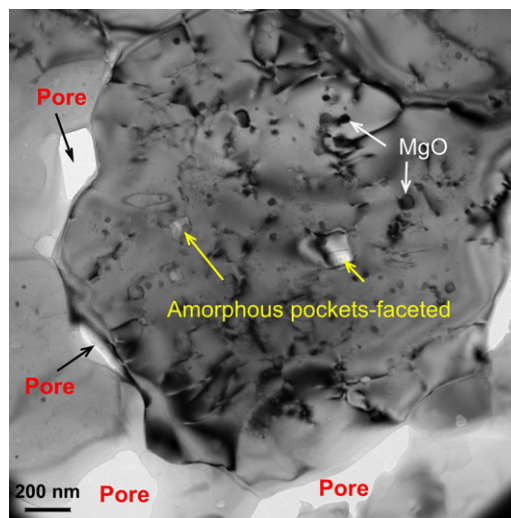


Fig. 1. TEM diffraction contrast images revealing the morphology of large  $\text{MgB}_2$  grains, amorphous pockets and some spherical shaped MgO particles.

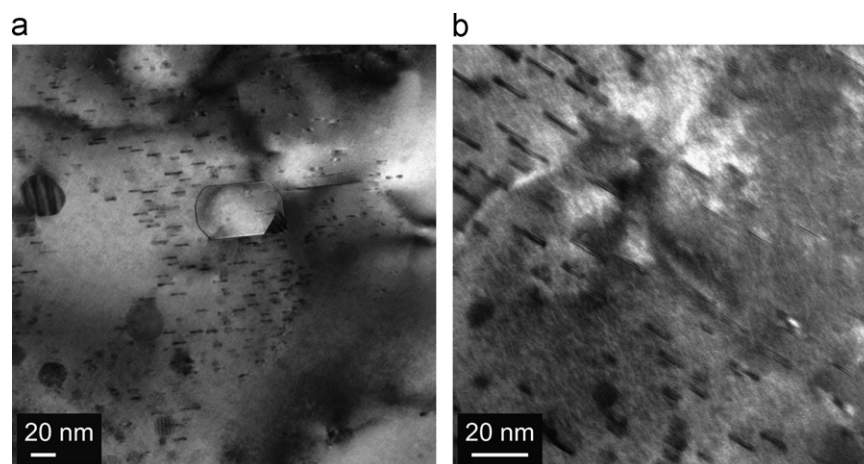


Fig. 2. TEM diffraction contrast images showing the morphology and distribution of the nano-precipitates. Images (a) and (b) were taken from different regions.

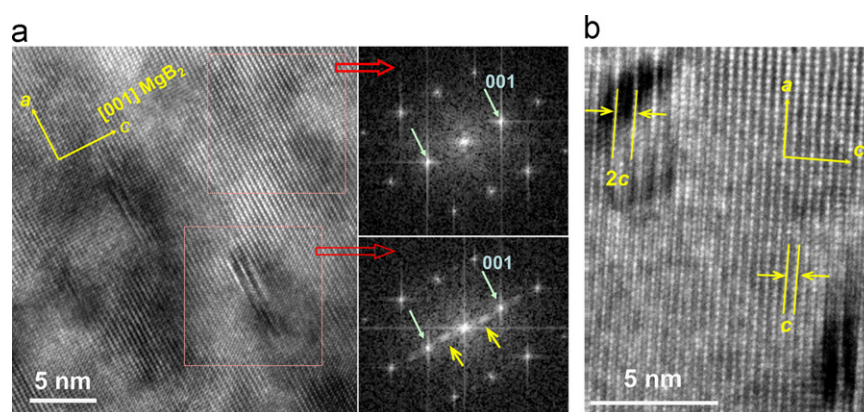


Fig. 3. (a) HREM images and corresponding Fourier transformation of the nano-precipitates and (b) high resolution images showing the contrast modulation in the precipitates.

shows the extra diffraction spots along the  $[001]$  direction with a vector equal to  $\frac{1}{2}[001]$ . Because of the nano-precipitates are having the structure modulation along the  $[001]$  direction of  $\text{MgB}_2$  lattice, under TEM, the contrast of nano-precipitates is pronounced only as the incident electron direction is close to the  $[100]$  direction of  $\text{MgB}_2$ . Under TEM, over several tens of  $\text{MgB}_2$  grains were studied to exam the commonality of the existence of such precipitates. It was found that as long as the  $[100]$  direction  $\text{MgB}_2$  grains are tilted to be parallel to the electron beam, the dense nano-precipitates are observed. Accordingly, it is believed that the existence of nano-precipitates in the  $\text{MgB}_2$  grain is common structural feature of the entirely purified  $\text{MgB}_2$  polycrystalline samples [24].

### 3.2. Atomic structure and chemistry of precipitates

The atomic structure of the precipitates was imaged using STEM equipped with both the bright field and HAADF detector. The bright field images and the HAADF Z-contrast images are taken simultaneously. The bright field STEM

image in Fig. 4(a and b) clearly shows the precipitates. By contrast, the corresponding HAADF Z-contrast images in Fig. 4(c) did not show any contrast from the precipitates. In the HAADF Z-contrast images taking along the  $[100]$  directions of  $\text{MgB}_2$ , the bright columns are corresponding to the Mg columns. By contrast, the B atoms are not visible due to the low scattering amplitude of B atoms. Since the HAADF images did not show the contrast difference between the  $\text{MgB}_2$  bulk and nano-precipitates, it clearly indicated that nano-precipitates are having the same Mg atom arrangements as that for  $\text{MgB}_2$  bulk, and the structure modulation in the  $c$ -direction of the precipitates probably originates from the structure and chemistry changes in the Boron lattices, which are not shown in the Z-contrast images at all.

To reveal the chemistry of the nano-precipitates, the EELS was taken from the precipitates and the  $\text{MgB}_2$  grains. Only Mg, B and O were identified in the precipitates, and only Mg and B were found in the  $\text{MgB}_2$  grains. No other elements were present in the sample. Boron K-edge EELS spectra taken from the nano-precipitates



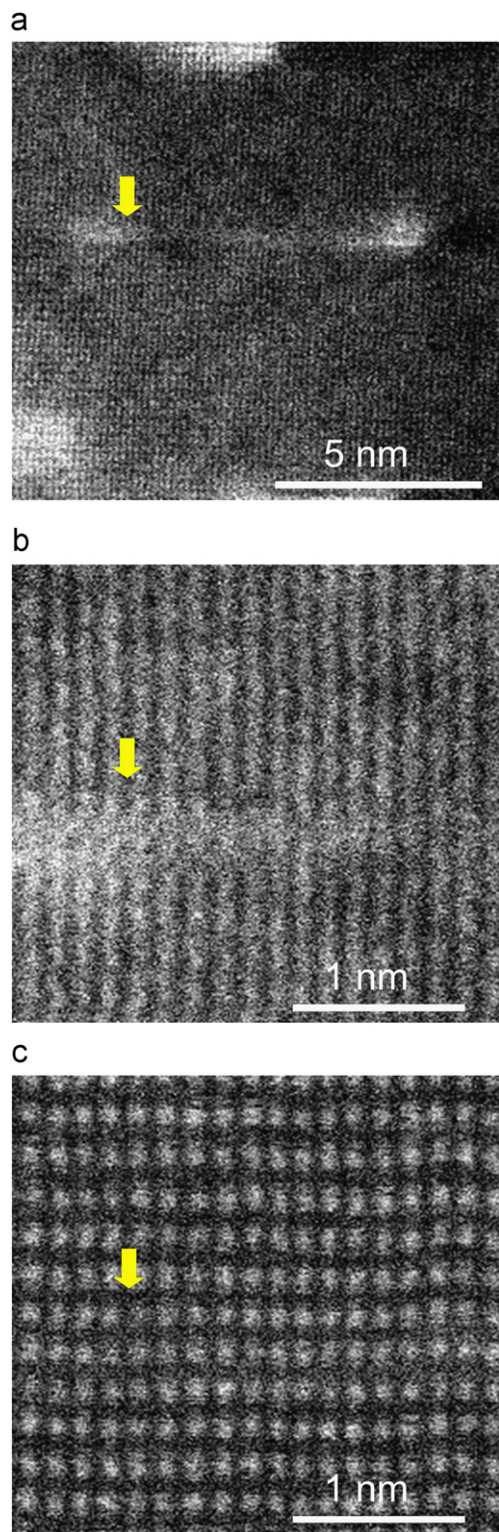


Fig. 4. STEM images of the precipitates, indicated by the arrow: (a) low magnification bright-field images; (b) high magnification of BF-STEM images and (c) simultaneous HAADF image corresponding to (b).

and bulk are shown in Fig. 5. The EELS is processed for background subtraction and corrected for multiple-scattering contributions. It can be seen the near edge fine structure of EELS from the precipitates are different from

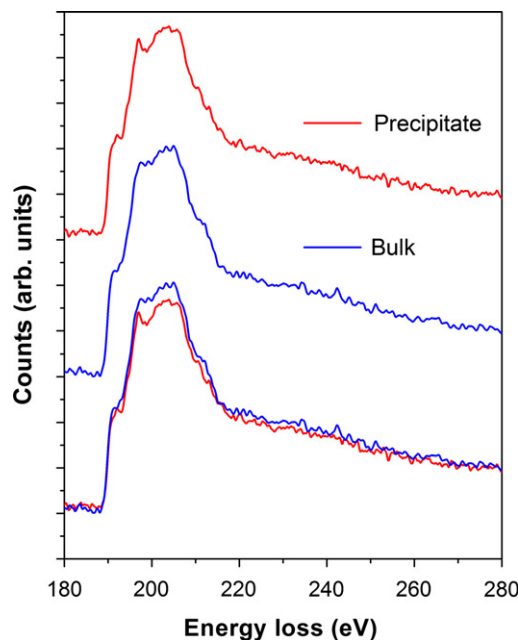


Fig. 5. EELS of boron K-edge of precipitates and bulk  $\text{MgB}_2$ .

that of the bulk and the pre-peak intensity of the boron K-edge spectra of nano-precipitates is slightly higher than that of the  $\text{MgB}_2$ . The EELS spectra from oxygen K-edge of the nano-precipitates, shown in Fig. 6, also indicate the significant amount of oxygen in the nano-precipitates. By contrast, the intensity of oxygen K-edge EELS from  $\text{MgB}_2$  matrix is very low. The above results show that the composition of the nano-precipitates is  $\text{Mg}(\text{BO})_x$  phases. Because the HRTEM images and FFT results show that the nano-precipitates lattices are having the structure/chemistry modulation with period twice of the  $\text{MgB}_2$  lattice constant along the  $c$ -axis, in combination with the  $Z$ -contrast and EELS results, it is reasonable to assume that, the nano-precipitates are having the oxygen ordering every other B lattices along the  $c$ -axis of  $\text{MgB}_2$  grains. Nevertheless, because  $Z$ -contrast imaging was not able to resolve the boron and oxygen atoms, it is unclear about the exact atomic positions that oxygen incorporate into the boron lattice.

### 3.3. Formation of nano-precipitates and approach of introducing different defects into $\text{MgB}_2$

The integrated TEM, HREM,  $Z$ -contrast STEM imaging and EELS carried out on the precipitates have indicated the composition of nano-precipitates as  $\text{Mg}(\text{BO})_x$ . In the meanwhile, it is believed that  $\text{MgB}_2$  grains are free or with very tiny amount of oxygen, which is evidenced by the EELS spectrum and by the high  $T_c$  of 38.5 K, measured for the sample [24]. In the meanwhile, the oxygen contamination in the starting materials is regarded as the source of having dense nano-precipitates. In the present study, those  $\text{Mg}(\text{BO})_x$  precipitate sizes are comparable with the  $\text{DyB}_4$  precipitates [25] that acts as the effective pinning centers for enhancing

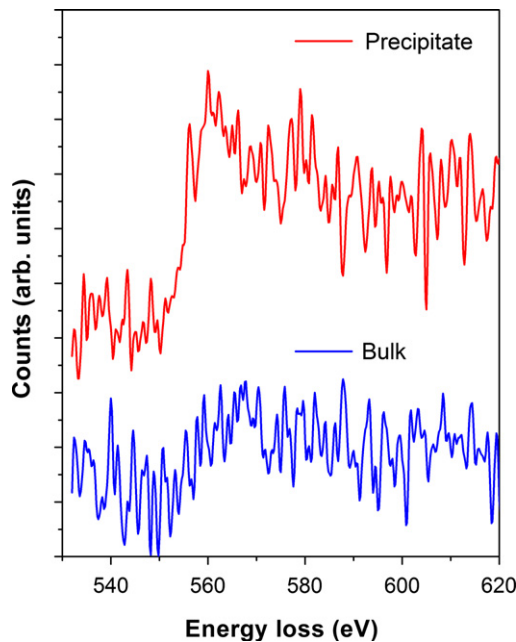


Fig. 6. EELS of oxygen K-edge of precipitates and bulk  $\text{MgB}_2$ .

the critical current density. For the  $\text{MgB}_2$ , the coherence length values along the  $ab$ -plane range between 3.7 and 12.8 nm and along the  $c$ -axis between 1.6 and 5.0 nm. Dense nano-precipitates with the size of 5–15 nm in longer axis and 2–5 nm in the shorter axis, comparable with the superconducting coherence length 5–12 nm for  $\text{MgB}_2$ , could act as very effective flux-pinning centers under high magnetic field.

The presence of such dense precipitates in the bulk  $\text{MgB}_2$  samples are very intriguing. Significant effort has also been put to investigate the effect of oxygenation on the superconducting properties of  $\text{MgB}_2$  and it is found that  $\text{MgB}_2$  are remarkably stable against damage from oxygen [26]. The author in this paper had also investigated many  $\text{MgB}_2$  samples using TEM before [10,12,27–30]. Those samples are usually with  $\text{MgO}$  particles and definitely no plate-shaped nano-precipitates. Interestingly, in the present study, TEM results indicated that the existence of nano-precipitates in the  $\text{MgB}_2$  grain is common structural feature of the entire polycrystalline  $\text{MgB}_2$  samples that we previously synthesized. It must be pointed out that, for the present study, the nano-precipitates in the samples prepared in the attempt of removing the  $\text{B}_2\text{O}_3$  impurities in the starting precursor materials. Those starting precursor materials of boron powders were initially processed by “heat treated at 950 °C for 48 h in flowing 4%  $\text{H}_2$  in air in order to remove  $\text{B}_2\text{O}_3$ ”. X-ray diffraction and microstructure analysis show that removing  $\text{B}_2\text{O}_3$  results in a decrease in  $\text{MgO}$  from 5.3 to 1.5 mol% in the final reacted  $\text{MgB}_2$  [24]. It is from those samples that made from starting boron powders subjected to  $\text{B}_2\text{O}_3$  purification; such dense nano-precipitates were found. Accordingly, it is thus believed there is a “fine range” of oxygen concentration in the starting precursor materials to render the formation

of the nano-precipitates without forming the large dispersed  $\text{MgO}$  particles.

Unlike the dispersed large  $\text{MgO}$  particles in the  $\text{MgB}_2$  matrix [24], the nano-precipitates are having well-defined orientation relationship with  $\text{MgB}_2$  matrix, and having their longer axis of platelet perpendicular to the  $c$ -axis of  $\text{MgB}_2$  matrix, and oxygen order every other unit cell in the  $c$ -direction of  $\text{MgB}_2$  lattice. In other words, during the  $\text{MgB}_2$  reactions, the oxygen goes into the  $\text{MgB}_2$  lattice as impurity by forming oxygen point defects in the B site and leaving the Mg site un-disturbed. Those oxygen impurity atoms accumulate at the scale of several nanometers and order every other unit cell in the  $c$ -direction of  $\text{MgB}_2$  lattice. Thus, heat-treatment conditions that promote the oxygen atoms going into the  $\text{MgB}_2$  lattice and subsequently ordering is the key. As a fact, this is fully consistent with the conclusion by Kovac et al. [31] in stating that “the level of  $\text{MgO}$  should be optimized as a compromise between inter-grain connectivity and introducing pinning centers”. Furthermore, the present study provides the microstructure evidence of existence of dense nano-precipitates  $\text{Mg}(\text{BO})_x$  and their atomic structure.

By contrast to the precipitates, the EELS results have demonstrated that  $\text{MgB}_2$  grain matrix as shown in Fig. 2 is free or with very tiny amount of oxygen. EELS did not show any trace of carbon contamination in the  $\text{MgB}_2$  grains. The clean  $\text{MgB}_2$  grain matrix provides possibility of having additional defects in the  $\text{MgB}_2$ . The synergetic combination of “Nano-precipitates in the  $\text{MgB}_2$  grain” and “Point defects such as carbon impurity in the  $\text{MgB}_2$  lattice” will be very possible and interesting. Recently, synergetic combination of different types of defects to optimize the pinning landscape of HTS YBCO, has been proved to be very successful and produces thick films with remarkable critical current [32]. For  $\text{MgB}_2$ , the pinning provided by intragranular nano-precipitates, C substitution in  $\text{MgB}_2$  lattice may take place at the same time at superconducting temperatures, resulting in a synergetic combination of optimal defects from all sources. A full understanding of the way to control the formation of the above mentioned different defects in  $\text{MgB}_2$  will guide future technological development of  $\text{MgB}_2$  for commercial exploitation.

#### 4. Summary

Dense platelet shaped nano-precipitates with the size of 5–10 nm were found in the  $\text{MgB}_2$  bulk samples. The nano-precipitates are fully coherent with the  $\text{MgB}_2$  grains and having the longer axis of the plate perpendicular to the  $c$ -axis of  $\text{MgB}_2$  grains. Under TEM, the precipitates show the periodic contrast modulating along the  $[001]$  direction of the  $\text{MgB}_2$ . The modulation period of nano-precipitates is twice of the  $\text{MgB}_2$  lattice constant along the  $c$ -axis. Using the combination of atomic resolution bright field, dark field STEM imaging, and EELS, the composition of the precipitates were identified as  $\text{Mg}(\text{BO})_x$ . The oxygen is

assumed to substitute B and form ordering every other B lattices along the *c*-axis of MgB<sub>2</sub> grains. The precipitates are common feature of the samples investigated. It is from those samples that made from starting boron powders subjected to B<sub>2</sub>O<sub>3</sub> purification; such dense nano-precipitates were found. Accordingly, it is thus believed there is a “fine range” of oxygen concentration in the starting precursor materials to render the formation of the nano-precipitates without forming the large dispersed MgO particles. The nano-precipitates with the size of 5–15 nm longer axis and 2–5 nm in the shorter axis could be very effective pinning centers. With the presence of nano-precipitates, the MgB<sub>2</sub> grains are free of oxygen. The clean MgB<sub>2</sub> lattice provides possibility of having point defects to further enhance the flux pinning properties. The synergetic combination of “nano-precipitates in the MgB<sub>2</sub> grain” and “point defects such as carbon impurity in the MgB<sub>2</sub> lattice” is suggested for increasing the critical current density of MgB<sub>2</sub>.

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