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KrF excimer laser irradiated nanoporous TiO₂ layers for dye-sensitized solar cells: Influence of laser power density

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Abstract

This paper reports the performance of dye-sensitized solar cells (DSSCs) with nanoporous TiO_2 photoanodes irradiated by KrF excimer laser beams of various power densities. The laser induces surface remelting and solidification to create textures on the TiO_2 layers. After laser irradiation, TiO_2 also undergoes a phase transition from anatase to rutile; the amount of the transformed phase increases with the laser power density. For dye-anchored TiO_2 layers, light absorption increases first and then decreases as the laser power density increases. The assembled cell efficiency, strongly correlated with the photocurrent density, also increases first and then decreases as the laser power density rises. This indicates that laser treatment creates surface textures on TiO_2 that improve light trapping in the dye-anchored TiO_2 photoanodes, thereby increasing the photocurrent level and cell efficiency. For high laser irradiation power density, the ablation of the TiO_2 layer becomes significant, leading to a decrease in cell efficiency. The surface remelting and solidification process reduces the density of the surface recombination centers on TiO_2 , resulting in an increase of open-circuit voltage. © 2012 Elsevier Ltd and Techna Group S.r.l. All rights reserved.

Keywords: Excimer laser; Titania; Dye-sensitized solar cells; Texture

1. Introduction

Dye-sensitized solar cells (DSSCs) have become a promising technology for photovoltaics since the great breakthrough achieved in 1991 [1]. In DSSCs, photoelectron generation and transport are separated in two distinct materials. The electrons are excited in dyes, and then transported through the nanoporous photoanodes to the external load. The nanoporous photoanode is thus important in both optical and electrical aspects. Light passes through the photoanodes and is absorbed by the dyes. During the passage of light, light trapping/scattering can greatly influence cell performance. Photoanodes are also transport routes for the electrons. The morphology and electrical properties of the photoanodes strongly affect the electron flows. Therefore, the design and fabrication of

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the photoanodes can significantly influence cell efficiency [2–17].

There have been several papers reporting on the application of laser technology on TiO_2 photoanodes for DSSCs. Pan et al. performed spray deposition of TiO_2 nanoparticles using a KrF excimer laser [6]. Kim et al. prepared mesoporous TiO_2 photo-electrodes from a colloidal solution of nanopowders using a quasi-continuous-wave UV laser (λ =355 nm) [18]. The laser-induced forward transfer of nanocrystalline TiO_2 was also developed using a pulse UV laser (λ =355 nm) [19]. A 355 nm laser beam was illuminated from the glass side to enhance the adhesion of nanoporous TiO_2 and the glass substrate, resulting in enhanced cell performance of DSSCs [20].

In our previous study, we demonstrated that KrF laser treatment on TiO₂ photoanodes can improve DSSC efficiency. The influence of the irradiation pulse number has been investigated. In this paper, we report the influence of laser power density on the properties of nanoporous TiO₂. Laser treatment induces the phase transformation of TiO₂ from anatase to rutile, increasing in severity as the

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power density rises. The performance of DSSCs with laser-treated TiO₂ photoanodes is also investigated. The improvement in cell performance mainly results from the improvement of the short-circuit current. An optimal power density is also identified.

2. Experimental details

Fluorine-doped tin oxide (FTO) coated glass slides (sheet resistance: $8\,\Omega/\Box$) were used as the substrates for the DSSCs. The TiO₂ pastes (E-solar P300, Everlight Chemical Industrial Co.) were first screen-printed onto the FTO glass, and then subsequently calcined at 510 °C for 15 min. This coating procedure was repeated three times for the TiO₂ layer thickness to reach 9 μ m. After calcination, the nanoporous TiO₂ layers were irradiated by a KrF excimer laser. The number of irradiated laser pulses was fixed at 1000. The laser power density was varied as

0, 40, 80, 120, 160, and 200 J/cm². The laser irradiation repetition rate was 10 Hz.

The laser-irradiated TiO_2 coated FTO substrates were then immersed in a mixed solution of acetonitrile (99.9%, J. T. Baker) and tertiary butyl alcohol (99.9%, J. T. Baker) containing 3×10^{-4} M of N719 (Solaronix) for 24 h. The dye-absorbed TiO_2 photoanodes were rinsed with ethanol and dried at room temperature. A 10-nm Pt was dc-sputtered onto an FTO glass substrate as the counter electrode, which was assembled with the dye-absorbed TiO_2 photoanode with spacers. Thereafter, the liquid electrolyte (E-solar EL 100, Everlight Chemical Industrial Co.) was injected into the assembled cells.

An AM1.5 solar simulator (WACOM, WXS-155S-L2) and a Keithley 2000 electrometer were used to measure the IV curves of the cells. The cells were tested with light illuminated through the FTO substrates. The illuminated cell area was 0.22 cm². The morphology of the laser-irradiated TiO₂ was examined by a scanning electron

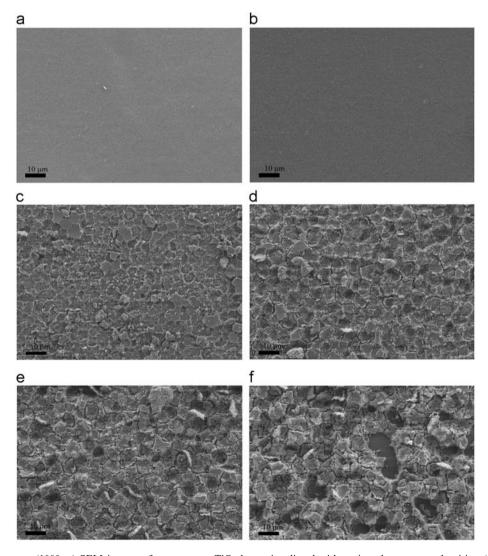


Fig. 1. Low magnifying rate (1000 ×) SEM images of nanoporous TiO₂ layers irradiated with various laser power densities: (a) no laser irradiation, (b) 40 mJ/cm², (c) 80 mJ/cm², (d) 120 mJ/cm², (e) 160 mJ/cm², (f) 200 mJ/cm².

microscope (SEM, Hitachi S-800). A UV–vis-NIR spectrophotometer (JASCO V-570) with an integration sphere was used to determine the absorption of the dye-anchored TiO_2 layers. A field emission scanning electron microscope (FE-SEM, LEO 1530) was used for TiO_2 surface morphology inspection. The TiO_2 crystal structures were characterized using an X-ray diffractometer (XRD, PANalytical X'Pert PRO) with Cu-K_{α} radiation (λ =0.15406 nm).

3. Results and discussion

Fig. 1 shows the $1000 \times SEM$ images of the nanoporous TiO_2 layers irradiated with 1000 laser pulses of various laser power densities. The as-calcined nanoporous TiO_2 surface is pretty smooth under this magnifying rate, as shown in Fig. 1(a). For the TiO_2 irradiated with the laser power density of 40 mJ/cm², there is some fine structure alternation (shown in Fig. 1(b)) that can be observed more clearly with a $10000 \times magnifying$ rate (Fig. 2(b)). As the power reaches 80 mJ/cm^2 , the TiO_2 surface forms textures

and some minor ablation occurs (Fig. 1(c)) Apparent ablation can be clearly identified when the power density is greater than 120 mJ/cm², increasing in severity as the power density increases (Fig. 1(d)-(f)). A large area of flat surface underneath is revealed as power density reaches 200 mJ/cm², as shown in Fig. 1(f). Fig. 2 shows SEM images at a higher magnifying rate $(10000 \times)$. The as-calcined nanoporous TiO₂ layer is shown in Fig. 2(a). Nanopores can barely be observed. As shown in Fig. 2(b), some cracking can already be observed on the TiO₂ surface after laser irradiation with 1000 shots of 40 mJ/cm² pulse; some of the nanopores have also expanded. As the laser power density reaches 80 mJ/cm², some remelting and solidification can be identified on the TiO2 surface; the TiO₂ surface forms a chip-like structure, as shown in Fig. 2(c). In Fig. 2(d)–(f), it can be clearly observed that the chip size becomes larger as the laser power density increases. Some of the nanopores continue expanding, while some are closed owing to the surface remelting and solidification processes.

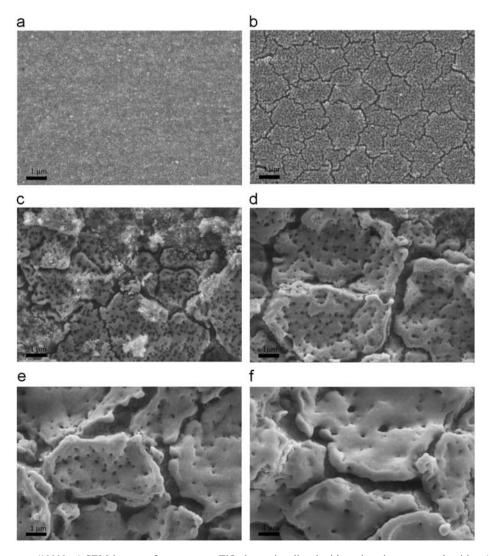


Fig. 2. High magnifying rate ($10000 \times$) SEM images of nanoporous TiO₂ layers irradiated with various laser power densities: (a) no laser irradiation, (b) 40 mJ/cm^2 , (c) 80 mJ/cm^2 , (d) 120 mJ/cm^2 , (e) 160 mJ/cm^2 and (f) 200 mJ/cm^2 .

Fig. 3 shows the XRD patterns for the TiO₂ irradiated with various laser power densities. The anatase content f_A is calculated by $f_A = (1 + 1.26 I_R/I_A)^{-1}$, where I_R is the (110) rutile phase XRD peak, and I_A is the (101) anatase XRD peak [21,22]. The anatase contents are listed on the right of the XRD patterns. The anatase content significantly decreases as the laser irradiation power density increases. The anatase content is only 46% for TiO₂ irradiated with the 200 mJ/cm² power density, in comparison to 89% for TiO₂ without any laser treatments. The laser power density is the dominant factor for the TiO₂ phase transformation. Phase transformation of amorphous TiO₂ thin films induced by excimer laser irradiation has been studied by Overschelde et al. [23,24]. According to their studies, TiO2 films undergo amorphous-to-anatase and amorphous-to-rutile transformations for low and high fluence KrF excimer laser irradiations, respectively. The anatase phase is more stable in smaller particles, whereas the rutile phase is more stable in larger TiO₂ particles. In our case, the used TiO₂ pastes contain TiO₂ nanoparticles. The calcination process only slightly connects the nanoparticles. After laser irradiation, the anatase TiO2 nanoparticles undergo the remelting and solidification process to merge into a larger piece, in which rutile is the more stable phase. This leads to the anatase-to-rutile phase transformation. The electron transport is slower yet light scattering is more efficient in the rutile phase [25–27]. The increased amount of rutile phase content in the TiO₂ layer is favorable optically, but unfavorable electrically.

Fig. 4 shows the absorption spectra for the dye-adsorbed ${\rm TiO_2}$ layers treated with various laser power densities. The absorption A is defined as A=1-T-R, where T and R represent transmission and reflection measured by integration spheres. The absorption increases first and then decreases as the laser power density increases. For the cases of $160~{\rm mJ/cm^2}$ and $200~{\rm mJ/cm^2}$, absorption is lower than that without any laser treatments, caused by the

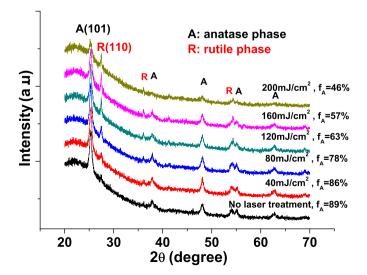


Fig. 3. XRD patterns for nanoporous TiO₂ layers irradiated with various laser power densities.

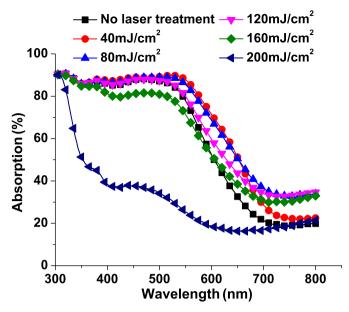


Fig. 4. Absorption spectra of dye-anchored TiO₂ layers irradiated with various laser power densities.

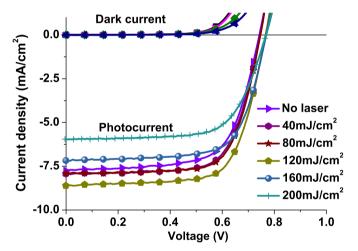


Fig. 5. IV characteristics of DSSCs with TiO₂ photoanodes irradiated with various laser power densities.

significant ablation on the TiO_2 films. The laser treatment creates surface textures to enhance light trapping and absorption in the nanoporous TiO_2 layers. The increased amount of rutile phase after laser irradiation may also enhance light scattering, increasing the light absorption. In the meantime, the surface remelting and solidification process as well as the ablation of the TiO_2 layers lead to the decrease of total surface area for dye-anchoring. This will in turn decrease the total absorption.

Fig. 5 shows the IV characteristics of the assembled DSSCs. The cell performance parameters extracted from Fig. 5 are listed in Table 1. Cell efficiency, strongly correlated with the photocurrent level, increases first and then decreases as the power density of the photoanode laser treatment increases. This trend is consistent with the results of the absorption spectra of the dye-anchored TiO₂, shown in

Table 1
The DSSC performance parameters extracted from Fig. 5.

Laser power (mJ/cm ²)	Film thickness (μm)	Voc (V)	Jsc (mA/cm ²)	Fill factor (%)	Efficieny (%)
X	8.3	0.742	7.76	66.6	3.84
40	8.2	0.746	7.92	69.7	4.12
80	7.6	0.755	7.93	69.8	4.18
120	7	0.766	8.61	69.7	4.6
160	5.8	0.769	7.17	69.6	3.84
200	4.8	0.769	5.96	68.0	3.1

Fig. 4. The TiO₂ surface textures, created by the laser treatments, enhance light scattering/trapping, improving the photocurrent levels and cell efficiency. The increased amount of rutile TiO₂ after laser irradiation may also enhance light scattering, improving the photocurrent levels and cell efficiency. However, when the laser power density is beyond 120 mJ/cm², the strong ablation of the TiO₂ film reduces the total TiO₂ volume, leading to the decreasing amount of adsorbed dyes. This reduces the amount of light being absorbed, thereby lowering the photocurrent level and cell efficiency for the DSSCs with photoanodes treated with higher laser power densities. Because the short-circuit current is associated with the photocurrent, the short circuit current increases first and then decreases as the laser irradiation power density increases. The increase of the short-circuit current at the lower laser power density results from the enhancement of light scattering/trapping which improves the light absorption efficiency. On the other hand, the decrease of the short-circuit current at the higher laser power density owes to the reduction of the TiO2 thickness, which causes a decreasing amount of dyes being adsorbed, leading to the reduction of light absorption. The fill factor varies slightly with the laser treatments on the photoanodes. The opencircuit voltage V_{oc} reveals a monotonic increase with the laser power density applied on the TiO₂ photoanodes. The surface remelting and solidification process may decrease the density of the surface recombination centers, resulting in the increase in $V_{\rm oc}$ [25–27].

4. Conclusions

We investigated the performance of dye-sensitized solar cells (DSSCs) with TiO₂ photoanodes irradiated by KrF excimer laser beams of various power densities. Laser treatments create textures on the nanoporous TiO₂ surfaces. TiO₂ proceeds with an anatase-to-rutile phase transition after laser irradiation. The amount of transformed rutile TiO₂ increases with the irradiation laser power density on the TiO₂. Cell efficiency, strongly correlated with the photocurrent level and the absorption spectra of the dye-anchored TiO₂, increases first and then decreases as the laser power density increases. The opencircuit voltage increases slightly with the applied laser irradiation power density. The enhanced cell efficiency is attributed to improved light trapping, thereby increasing

the photocurrent (or short-circuit current) level and cell efficiency. At high laser irradiation power density, the drop in cell efficiency is attributed to the ablation of the ${\rm TiO_2}$ layer that causes the reduced amount of absorbed dyes. The surface remelting and solidification process also reduces the density of the surface recombination centers, resulting in the increase of the open-circuit voltage.

Acknowledgments

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