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# Spark plasma sintering of graphene reinforced zirconium diboride ultra-high temperature ceramic composites

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#### **Abstract**

Spark plasma sintering (SPS) of monolithic ZrB<sub>2</sub> ultra-high temperature ceramic and 2-6 vol% graphene nanoplates (GNPs) reinforced ZrB<sub>2</sub> matrix composites is reported. The SPS at 1900 °C with a uni-axial pressure of 70 MPa and soaking time of 15 min resulted in near-full densification in ZrB<sub>2</sub>-GNP composites. Systematic investigations on the effect of GNP reinforcement on densification behavior, microstructure, and mechanical properties (microhardness, biaxial flexural strength, and indentation fracture toughness) of the composites are presented. Densification mechanisms, initiated by interfacial reactions, are also proposed based on detailed thermodynamic analysis of possible reactions at the sintering temperature and the analysis of in-process punch displacement profiles. The results show that GNPs can be retained in the ZrB2 matrix composites even with high SPS temperature of 1900 °C and cause toughening of the composites through a range of toughening mechanisms, including GNP pull-out, crack deflection, and crack bridging.

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## 1. Introduction

Ceramics with melting point greater than 3200 °C are generally classified as ultra-high temperature ceramics (UHTC). The most important application of UHTC is in the leading edges and nose caps of space vehicles re-entering the earth's atmosphere. These materials also find applications in high temperature crucibles and parts of electrical heaters and igniters [1-4]. Among the several UHTC systems (TaC, HfC, NbC, ZrC, HfB2, ZrB2, and TiB<sub>2</sub>), ZrB<sub>2</sub> based ceramics are the most promising for thermal protection systems due to their good combination of elevated temperature mechanical properties and oxidation resistance. However, there is a critical need to further improve their oxidation resistance, thermal conductivity, and fracture toughness to realize the full potential of ZrB<sub>2</sub>based ceramics as next generation materials for sharp

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leading edge re-entry space vehicles [5-8]. In general, reinforcements are added to ZrB2 ceramic matrix to improve sinterability and mechanical, thermal, and oxidation properties of the composites. Several investigations have reported improved oxidation resistance and/or fracture toughness of SiC reinforced ZrB<sub>2</sub> ceramic composites [9–14]. Recently, carbon nanotubes (CNTs) have also been used as nano-scale filler in the ZrB<sub>2</sub> ceramics [14–15]. The CNT reinforcement resulted in improved densification particularly in the early (particle rearrangement) and final stages (diffusion) of densification. The CNT reinforcement also resulted in improvement in room temperature fracture toughness [14]. While the effectiveness of CNT reinforcement in improving densification and fracture toughness was demonstrated, exact mechanisms of densification and toughening are still not fully understood. Furthermore, uniform dispersion of the CNTs in the UHTC composites is often difficult [16].

Recently, graphene nanoplates (GNPs) are attracting significant attention as potential nano-scale reinforcement

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in advanced composites. Graphene is a single atom thick sheet of sp<sup>2</sup> hybridized carbon atoms arranged in a honeycomb structure. It is the building block for other sp<sup>2</sup> hybridized carbon forms such as single and multiwalled CNTs. It also exhibits unique mechanical, electrical, and thermal properties [17–22]. These properties have also been extended in bi- and few-layer graphene [23-25]. Unlike CNTs, a fairly uniform distribution of GNPs is possible in the ceramic matrices using traditional processing approaches [26]. A number of studies have reported on reinforcement of GNPs to improve the mechanical properties of polymer, metals/alloys, and ceramic matrix composites [27-33]. Most recently, Walker et al. [31] used GNPs as nano-reinforcement in silicon nitride ceramic matrix composites. The GNP reinforcement significantly improved the fracture toughness of Si<sub>3</sub>N<sub>4</sub> through a range of toughening mechanisms including wrapping of GNPs to resist sheet pull-out, crack bridging, and crack deflection. A three-dimensional crack deflection mechanism, induced by GNPs, for observed improved toughness of the ceramic matrix was also proposed [31]. As with CNT reinforcement, the mechanisms of densification in presence of GNPs are not well investigated. This is particularly important in case of high temperature sintering of highmelting point ceramics because of potential structural damage to the GNPs and consequent degradation of properties.

In this paper, spark plasma sintering (SPS) of GNP reinforced ZrB<sub>2</sub>-based UHTCs is reported. Detailed investigations on the effect of GNP reinforcement on densification behavior, microstructure, and mechanical properties (micro-hardness, bi-axial flexural strength, and indentation fracture toughness) of the composites are presented. The SPS is a relatively novel process where powders are sintered under simultaneous influence of pulsed direct current and uni-axial pressure. While there is no general agreement on the nature of active sintering mechanisms during SPS, it is widely accepted that Joule heating at the particle contacts causes localized heating and solid state diffusion. Due to unique mechanisms of sintering, SPS often allows sintering at relatively lower temperature and in shorter sintering time compared to conventional hot pressing. In this investigation, in-process punch displacement profiles and thermodynamic analysis were used to seek better understanding of the densification mechanisms in presence of GNPs at sintering temperature of 1900 °C. Detailed results of Raman spectroscopy and SEM microscopy are also presented with critical analysis of structural changes in GNPs during SPS and potential toughening mechanisms in the ZrB<sub>2</sub>-GNP composites.

## 2. Experimental procedure

Commercially available  $ZrB_2$  powder (1–2  $\mu$ m diameter; Alfa Aesar, USA) and graphene nanoplates (6–8 nm thick, 5  $\mu$ m diameter; XG Sciences, USA) were used as starting materials for the processing of monolithic  $ZrB_2$ 

and ZrB<sub>2</sub>-GNP composites. ZrB<sub>2</sub>-GNP composite powder mixtures, with 2, 4, and 6 vol% GNPs were prepared using colloidal processing [34]. Pre-measured quantity of GNPs was stirred in 100 ml acetone for 1.5 h using a high speed magnetic stirrer. ZrB<sub>2</sub> powder was then added into the mixture and stirred for additional 1.5 h. The slurry was then heated at 70 °C for 3 h using a hot plate followed by 24 h drying in a fume hood. Fully dried composite powder mixture was then milled in a high energy ball mill (Fritsch, USA) with a speed of 500 revolutions per minute and ballto-powder weight ratio of 5:2 for 5 min. For preparing monolithic ZrB<sub>2</sub>, as-received powder was ball milled with the milling parameters same as those for composite powder mixtures. Tungsten carbide (WC) balls and jar were used for ball milling. It has been reported that ball milling results in exfoliation of graphene into fewer or even single layer GNPs in the composite powder mixture [35]. The ball milled ZrB<sub>2</sub>, ZrB<sub>2</sub>-GNP composite powders were sintered at 1900 °C with a uniaxial pressure of 70 MPa and soaking time of 15 min under argon atmosphere using a commercial SPS system (Thermal Technology LLC, USA). Initial heating rate of 100 °C/min was used to attain sintering temperature of 1900 °C. Disc-shaped samples of 20 mm diameter and 2 mm thickness were sintered using graphite dies and punches. To investigate densification behavior of monolithic ZrB<sub>2</sub> and ZrB<sub>2</sub>-GNP composites, punch displacement was continuously monitored during the SPS cycle. Archimedes' principle was used to measure bulk densities of the sintered samples. The theoretical densities of the composites were calculated using the rule of mixtures. For phase analysis of the samples, a X-ray diffractometer (Philips Norelco, USA) operating with Cu  $K_{\alpha}$  ( $\lambda = 1.54178 \text{ Å}$ ) radiation was used. Raman spectroscopy was carried out to characterize GNPs on the fracture surfaces of sintered composites using 532 nm laser excitation, 0.8 mW laser power, and 20 µm spot size (WITec Instruments Corp., Germany). A Vickers hardness tester (Clark Instrument, USA), operated with a normal force of 9.8 N and holding time of 15 s, was used to determine indentation hardness of the samples. The indentation fracture toughness of the samples was calculated by measuring lengths of diagonal cracks emanated at the indentation corners. The fracture toughness values are based on three samples with five indents per sample. The indentation fracture toughness,  $K_{IC}$ , is given by

$$K_{\rm IC} = 0.016 \left(\frac{E}{H}\right)^{1/2} \frac{P}{c^{3/2}},$$
 (1)

where E is the Young's modulus of the composites (Young's moduli of  $ZrB_2$  and GNP are 500 and 1000 GPa, respectively), H is the Vickers hardness (GPa), P is the applied load (N), and c is the diagonal crack length (m) [36]. The flexure strength for the samples, 20 mm in diameter and 2 mm in thickness, was determined using ASTM C1499-05 ring-on-ring test method (Instron, USA). The support and loading ring diameters were 15 mm and

Table 1 Relative density and mechanical properties.

Sample	Relative density (%)	Hardness (GPa)	Fracture toughness (MPa m <sup>1/2</sup> )	Flexural strength (MPa)
$ZrB_2$ $ZrB_2+2\%$ GNPs $ZrB_2+4\%$ GNPs $ZrB_2+6\%$ GNPs	84.8 84.5 96.5 96.9	$16.64 \pm 0.90$ $13.53 \pm 0.25$ $15.90 \pm 0.84$ $14.00 + 0.60$	$1.51 \pm 0.02$ $2.10 \pm 0.43$ $2.15 \pm 0.24$ $2.77 + 0.06$	$ 162 \pm 31  204 \pm 34  219 \pm 23  316 + 85 $

5 mm, respectively, with a displacement controlled loading rate of 0.5 mm/min. The flexure strength,  $\sigma_{RoR}$ , is given by

$$\sigma_{\text{RoR}} = \frac{3P}{2\pi t^2} \left( \frac{(1-v)(a^2 - r^2)}{2R^2} + (1+v)\ln\frac{a}{r} \right),\tag{2}$$

where P is the applied load (N), v is the Poisson's ratio of the composites (v of  $ZrB_2$  and GNP are 0.15 and 0.165, respectively), a is the radius of the support ring, r is the radius of the load ring (m), and R and t are the sample radius and thickness (m), respectively [37]. Three flexure tests were conducted for each sample condition and average values along with positive and negative error bars are reported. The fracture surfaces of the samples were analyzed using SEM to investigate microstructure development and fracture behavior. The relative density and mechanical properties of sintered monolithic  $ZrB_2$  and  $ZrB_2$ –GNP composites are listed in Table 1.

#### 3. Results and discussion

## 3.1. Densification behavior and microstructure development

Fig. 1 presents the variation of relative density of ZrB<sub>2</sub>-GNP composite samples with GNP reinforcement content (0-6 vol%). The average relative density of monolithic ZrB<sub>2</sub> (0 vol% GNPs) and ZrB<sub>2</sub>-GNP (2 vol% GNPs) composites was about 85%. For the similar SPS processing parameters (sintering temperature of 1900 °C, uniaxial pressure of 70 MPa, and soaking time of 15 min), the relative density of ZrB2-GNP composites increased with GNP reinforcement content. The ZrB<sub>2</sub>-GNP composites with 6 vol% GNPs exhibited highest relative density of about 97%. The SEM micrographs from the fracture surface of the ZrB<sub>2</sub> and ZrB<sub>2</sub>–GNP composites are shown in Fig. 2. As seen in the figure, the monolithic ZrB<sub>2</sub> samples exhibited uniformly distributed porosity (relative density of about 85%) with average grain size of about 2.1 μm, which is slightly greater than starting ZrB<sub>2</sub> particle size (1-2 µm). The ZrB<sub>2</sub>-GNP composite reinforced with 2 vol% GNPs had comparable density of about 85% and average grain size of about 2.3 µm. The ZrB<sub>2</sub>-GNP composites reinforced with 4 and 6 vol% GNPs exhibited dense microstructure with relatively flatter features. Some distributed closed porosity and grain pull-out depressions can be seen on the fracture surfaces of these composite samples. The average grain size for these dense ZrB<sub>2</sub>-GNP composites (4–6 vol% GNPs) was about 4.1–4.7 μm.

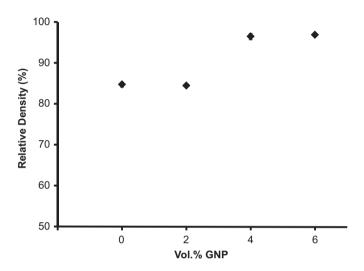


Fig. 1. Variation of relative density of  ${\rm ZrB_2\text{--}GNP}$  composites with reinforcement content.

Furthermore, the SEM micrographs clearly show distributed GNPs pulled out at the grain boundaries. Clearly, the reinforcement of GNPs above 2 vol% in ZrB<sub>2</sub> UHTC resulted in significant improvement in densification with minimal grain growth. Similar improvements in the densification of ZrB<sub>2</sub> UHTC were also observed with reinforcement of carbon nanotubes (CNTs) and carbon [14].

To get better insight into densification behavior of the monolithic ZrB2 and ZrB2-GNP composites, the data on punch displacement was recorded during SPS. The punch displacement profiles during initial heating and soaking stages of the SPS cycle for ZrB<sub>2</sub> and ZrB<sub>2</sub>–GNP composites are shown in Fig. 3. The temperature and pressure profiles during sintering cycle are also plotted in the figure. Note that temperature and pressure were simultaneously increased during sintering. The desired pressure of 70 MPa was reached in first 8 min of the sintering cycle. The sintering temperature of 1900 °C was reached in first 17 min of the cycle. With the soaking time of 15 min, the total sintering time (heating and soaking stages) was about 32 min. The punch displacement profiles clearly indicate three distinct trends: initial increasing trend (compaction) corresponding to increasing pressure (stage I), intermediate decreasing trend corresponding to thermal expansion (stage II), and final increasing trend corresponding to final densification stages (stage III). It can be seen from the figure that initial compression in monolithic ZrB<sub>2</sub> was about 0.25 mm. This compression was significantly higher for ZrB2-GNP composites reaching about 0.95 mm

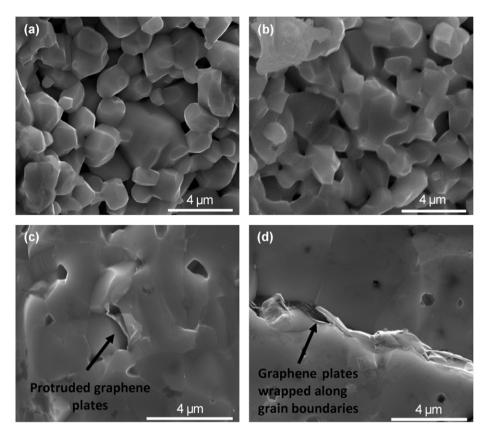


Fig. 2. SEM micrographs from the fracture surfaces of (a) monolithic  $ZrB_2$ , (b)  $ZrB_2+2$  vol% GNPs, (c)  $ZrB_2+4$  vol% GNPs, and (d)  $ZrB_2+6$  vol% GNPs composites.

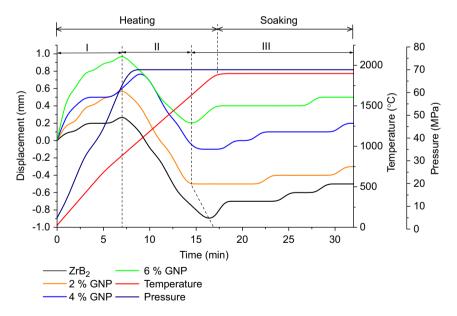


Fig. 3. Ram displacement, temperature, and pressure profiles during heating and soaking stages of SPS cycles for monolithic  $ZrB_2$  and  $ZrB_2$ —GNP composites.

for composites reinforced with 6 vol% GNPs. While GNPs seem to promote early stages of densification dominated by particle re-arrangement, the GNPs also shift the stage II-III transition point to left (earlier times) indicating enhanced densification in later stages (indicated by dotted line). For the monolithic ZrB<sub>2</sub>, the stage III densification begins at the start

of soaking cycle at 1900 °C (t=17 min). This stage of densification begins in heating cycle at the temperature of about 1700 °C (t~13 min) for ZrB<sub>2</sub>–GNP composites.

While the exact nature of mechanisms of densification in presence of GNPs is not fully understood, it seems that GNPs promote early stage ZrB<sub>2</sub> particle rearrangement

due to its self lubricating effects. In the later stages of sintering of ZrB<sub>2</sub>–GNP composites, GNPs act more as sintering aid. Carbon has long been used as a sintering aid for the densification of ceramics, and it is often proposed that carbon removes the native oxides on the surfaces of the non-oxide ceramics by some interfacial reactions and promote the densification. Zhu et al. reported nearfull densification of carbon-coated ZrB<sub>2</sub> ceramic using pressureless sintering and proposed possible interfacial reactions based on thermodynamic analysis. For ZrB<sub>2</sub> UHTC, the native oxides on the surfaces are ZrO<sub>2</sub> and B<sub>2</sub>O<sub>3</sub>. During high temperature sintering, GNPs can react with these native oxides according to carbothermal reduction reactions [38]:

$$ZrO_2 + B_2O_3 + 5C \rightarrow ZrB_2 + 5CO$$
 (3)

$$ZrO_2 + 3C \rightarrow ZrC + 2CO$$
 (4)

The free energies of these reactions are plotted over a range of temperatures from 0 to 3000 °C in the standard state (pressure  $\sim 1$  atm) in Fig. 4. From these thermodynamic calculations, it can be seen that the reactions (3) and (4) are feasible at temperatures above 1500 and 1656 °C, respectively, in the standard state as long as the activation barrier is surmounted by the available thermal energy in the medium. With the reduced pressure during SPS, these reactions are expected to be feasible even at lower temperatures. According to these reactions, the surface oxides get converted back to ZrB2 or form interfacial product (ZrC). The removal of oxide impurities in presence of graphene nanoplates possibly promotes densification in ZrB2-GNP composites in later stages of sintering as indicated by punch displacement profiles. As the fracture surfaces showed intact GNPs at the grain boundaries, the surface reactions leading to enhanced densification seem to occur at ZrB<sub>2</sub>/GNP interface.

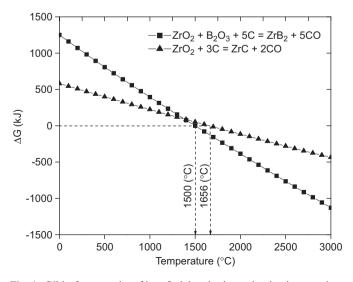


Fig. 4. Gibbs free energies of interfacial carbothermal reduction reactions as a function of temperature.

## 3.2. X-ray diffraction and Raman spectroscopy analysis

X-ray diffraction (XRD) patterns from sintered monolithic  $ZrB_2$  and  $ZrB_2$ –GNP composites are presented in Fig. 5. While all the samples showed characteristic peaks of  $ZrB_2$  in the XRD patterns, the characteristic (002) peak of graphene at  $2\theta$ =26.6° was not observed in the patterns from composites. Furthermore, no additional peaks corresponding to possible interfacial reaction product ZrC were observed. This is not surprising as the XRD analysis often does not show additional peaks when the reinforcement content is very low and the interfacial reactions are limited to near-interface regions. Furthermore, some of the interfacial products, possibly ZrC here, exhibit very high solubility in the matrix  $ZrB_2$  phase and may not show up in XRD patterns.

Raman spectroscopy was also used to characterize ZrB<sub>2</sub>-GNP composites. Raman acquisitions reveal the state of GNPs during the SPS of the composites. A typical graphitic carbon exhibit characteristic Raman peaks at D (1350 cm<sup>-1</sup>) and G (1580 cm<sup>-1</sup>) representing in-plane stretching and breathing modes, respectively, while the peaks at D' (1620 cm<sup>-1</sup>) and G' (2700 cm<sup>-1</sup>) are attributed to their respective higher order modes [39–41]. The modes corresponding to D and D' peaks are forbidden in the perfect sp<sup>2</sup> hybridized carbon due to symmetry. However, the presence of defects creates structural disorder that allows the breathing mode [39–41]. Hence, the D and D'peaks are observed only in graphitic carbon with disorder in its crystal structure. In the present study, we analyzed the characteristic Raman peaks of GNPs at the fractured surfaces of the composites. Fig. 6(a) shows the Raman spectra of ZrB<sub>2</sub>-GNP composites (2, 4 and 6 vol% GNPs) along with sintered GNPs and ZrB2. It is obvious from the Raman spectrum of ZrB2 that it has no detectable Ramanactive vibrational modes in the frequency range of 1000-3000 cm<sup>-1</sup>. Hence, the Raman spectra acquired from the composites exhibit the characteristic peaks of GNPs. Further, it is apparent that the D and G peaks in ZrB2-GNP

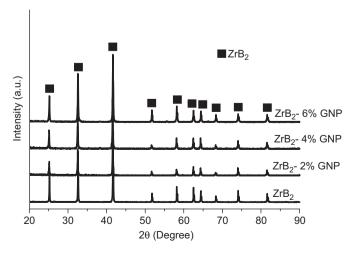
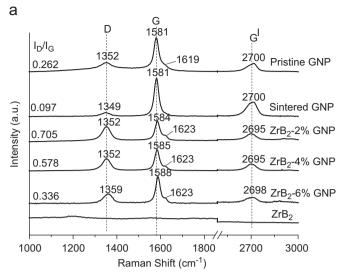


Fig. 5. XRD patterns from sintered ZrB<sub>2</sub> and ZrB<sub>2</sub>-GNP composites.



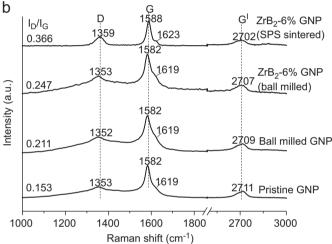


Fig. 6. Raman spectra of (a) ZrB<sub>2</sub>, ZrB<sub>2</sub>–GNP composites, pristine GNPs, and sintered GNPs; and (b) pristine GNPs, ball milled GNPs, ball milled ZrB<sub>2</sub>–6% GNPs powder, and sintered ZrB<sub>2</sub>–6% GNPs composite.

composites have shifted to higher energy, while the *G'* band has shifted to lower energy. The position of *G* and *G'* peaks are influenced by the following factors: (i) density of defects in the GNPs incurred during SPS processing [41], (ii) the thermal residual stress [42–44] that evolves during the cooling step (i.e., thermal contraction of ZrB<sub>2</sub> matrix surrounding GNPs), and (iii) reduction in number of graphene layers (nGLs) [45–46].

A routinely used measure of density of defects in graphitic carbon materials is the ratio of intensities of D to G peaks  $(I_D/I_G)$  [39,41].  $I_D/I_G$  values for GNPs in ZrB<sub>2</sub>–GNP composites (2, 4 and 6 vol% GNPs), sintered GNPs, and pristine GNPs are displayed in Fig. 6(a) (in fact, these values are close to average values of  $I_D/I_G$ ). It is evident from the values of  $I_D/I_G$  that the average density of defects has increased by more than 2 times in ZrB<sub>2</sub>–GNP composites reinforced with 2 and 4 vol% GNPs. The density of defects is comparatively higher in 2 vol% than 4 and 6 vol% GNP reinforced composites and it scales inversely

with the concentration of GNPs. In general, high temperature (> 1600 °C) sintering of nanocrystalline graphite increases the average crystallite size [47]. This situation is directly observed in sintering of pure GNPs. The  $I_D/I_G$ value of sintered GNPs is close to zero (Fig. 6a) which is reminiscent of crystalline graphite. This point is further corroborated by G peak position being at  $1581 \,\mathrm{cm}^{-1}$ (Fig. 6a), which is in close agreement with literature [41]. Hence, it is inferred that  $I_D/I_G$  has increased for  $ZrB_2$ composites because of interfacial reactions between GNPs and ZrB2 matrix at high SPS processing temperature (1900 °C).  $I_D/I_G$  is higher at low concentrations of GNPs possibly due to two reasons: (i) less GNP aggregation and therefore, more carbon surface interaction with ZrB<sub>2</sub> per GNP; and (ii) higher multiplication of GNPs due to exfoliation at lower GNP concentration. The exfoliation of GNPs provides more GLs in contact with ZrB<sub>2</sub> matrix in 2 and 4 vol% GNP reinforced composites and it will result in a higher  $I_D/I_G$  [48]. This point is supported by the positions of G and G' peak.

In addition, the G peak frequency in ZrB<sub>2</sub>-GNP (2, 4 and 6 vol%) for corresponding  $I_D/I_G$  values is expected to decrease simultaneously with increasing concentration of GNPs [43]. According to Ferrari et al., when the  $I_D/I_G$ value increases from 0 to 2 due to increasing concentration of defects in nanocrystalline graphite, a high energy shift of the G peak from 1580 to  $1600 \text{ cm}^{-1}$  is observed concomitantly [41]. On the contrary, we observe the G peak frequencies at 1584 and 1585 cm<sup>-1</sup> for 2 and 4 vol% GNPs reinforced composites, respectively (Fig. 6a). This shift is lower than the expected G peak shift by 6 and 5 cm<sup>-1</sup>, respectively (estimated from Ref. 43 using corresponding  $I_D/I_G$  values of 2 and 4 vol% GNPs). This discrepancy is attributed to reduction in nGLs in GNPs of 2 and 4 vol\% reinforced composites that occurred due to exfoliation [48,49]. On the other hand, the G peak of 6 vol% GNP reinforced composite (at 1588 cm<sup>-1</sup>) has shifted to higher energy than expected G peak (1583 cm $^{-1}$ ) that is suggestive of the residual compressive stresses acting on GNPs incurred during thermal contraction of ZrB<sub>2</sub> matrix [43–45]. The thermal stresses are more prominent in 6 vol% than 2 vol% GNP reinforced composite due to higher densification of the ceramic matrix. Concurrently, we observe a higher frequency shift in the D peak in 6 vol% GNP reinforced composite due to residual compressive stresses.

In order to investigate the process of exfoliation we focused on the G' band. It is evident from Fig. 6(a), that the G' band of the composites has shifted to lower energy. This lower shift in G' band implies that the nGLs has decreased [48,49]. Alternatively, the shift in the G' peak is also possible due to interfacial reactions incurred during the high temperature processing of composites. As discussed from the thermodynamic analysis, the possible interfacial reaction could be formation of ZrC or solid solution of ZrC in ZrB<sub>2</sub>. Another possibility is the doping of GNPs with B atoms. Both possible interfacial interactions are expected to

result in high energy shift in G' band [50]. On the other hand, we observe a low energy shift in the G' band that suggests reduction in nGLs has the dominant influence on the G' band frequency. To further investigate the origin behind exfoliation of graphene, we acquired Raman spectra in ball-milled GNPs with and without  $ZrB_2$  powder as shown in Fig. 6(b). The lower energy shift in the G' band indicates a decrease in nGLs during ball-milling. In addition, the trend in  $I_D/I_G$  as well as G peak frequency suggests that the ball-milling has induced minimal damage. Additional shift in the G' band of sintered composite suggests further decrease in nGLs during SPS processing of the ball-milled mixture of  $ZrB_2$  and GNPs.

## 3.3. Mechanical properties

The variation of indentation hardness with GNP reinforcement content in the ZrB2-GNP composites is presented in Fig. 7. Among all the compositions investigated in this study, the monolithic  $ZrB_2$ , with  $\sim 85\%$  relative density, exhibited highest average hardness of about 16.6 GPa. Even though ZrB<sub>2</sub>-GNP composites exhibited higher relative density, no improvement in the hardness was observed with GNP reinforcement in the composites. In fact, the hardness of composites was relatively lower compared to that of monolithic ZrB2. The average hardness of the composites (2-6 vol% GNPs) was in the range of 15.5-15.9 GPa without any specific trend with increasing GNP reinforcement content. The lower hardness of the composites could be due to competing effects of densification, GNP induced strengthening, grain growth, and interfacial reactions on the hardening of ceramics. On the other hand, all the composites showed improved biaxial flexural strength compared to the monolithic ZrB<sub>2</sub> (Fig. 8). The monolithic ZrB<sub>2</sub> had the average flexural strength of about 162 MPa. The ZrB2-GNP composites reinforced with 2 and 4 vol% GNPs had the flexural strength of about 204 and 219 MPa, respectively. The composites reinforced with 6 vol% GNPs exhibited highest flexural strength (316 MPa), which is about two times that of

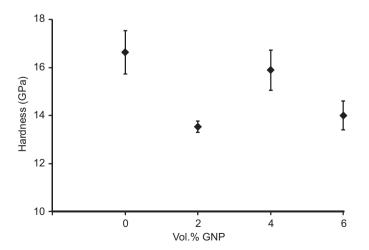


Fig. 7. Hardness of ZrB<sub>2</sub>–GNP composites as a function of GNP reinforcement content.

monolithic ZrB<sub>2</sub>. Clearly, the biaxial flexural strength, which better represents the strength-related property of the bulk samples than the surface microhardness, shows much better correlation with the GNP reinforcement content or the relative density of the ZrB2-GNP composites. Fig. 9 shows the variation of indentation fracture toughness of the ZrB2-GNP composites with GNP reinforcement content. The fracture toughness of monolithic  $ZrB_2$  samples (1.5 MPa m<sup>1/2</sup>) is also indicated in the figure. The fracture toughness values reported here for monolithic ZrB<sub>2</sub> samples are relatively lower than those reported for nearly fully dense ZrB<sub>2</sub> samples (3–4 MPa m<sup>1/2</sup>). In general, the fracture toughness of the composites was higher than that of monolithic ZrB<sub>2</sub> and increased with increasing GNP content. Among all the composite compositions investigated, the ZrB<sub>2</sub>-GNP composites reinforced with 6 vol% GNPs exhibited highest average fracture toughness of 2.8 MPa m<sup>1/2</sup>, an increase of about 83% over the monolithic ZrB<sub>2</sub>. While toughening effect of GNPs is clearly observed, the measurement of crack lengths using SEM seems to have resulted in underestimation of fracture toughness. Note that there are controversies surrounding the use of indentation methods for characterizing fracture toughness of ceramics reinforced with nano-scale fillers [51–53]. The indentation fracture toughness values obtained by indentation methods may not be representative of the bulk samples due to complexity of stress conditions associated with indentation process. For better characterization of the fracture toughness of the GNP reinforced ceramic samples, single edge V-notch bending (SEVNB) and rising R-curve methods will be used [53]. SEM micrographs showing morphologies of cracks emanated from the indentation corners for ZrB<sub>2</sub> and ZrB<sub>2</sub>-GNP composites are presented in Fig. 10. The crack front was relatively straight for monolithic ZrB<sub>2</sub> samples without significant toughening mechanisms such as crack deflection or bending (Fig. 10(a)). In these samples, the cracks propagate along the grain boundaries and meet

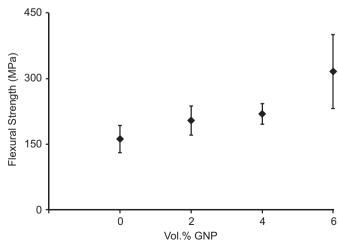


Fig. 8. Flexural strength of ZrB<sub>2</sub>–GNP composites as a function of GNP reinforcement content.

open porosity of the samples. The SEM micrograph from the fracture surface indicated clear intergranular fracture in these porous (relative density of about 85%) ZrB<sub>2</sub> samples (Fig. 2(a)). For the ZrB<sub>2</sub>–GNP composites (2 vol% GNPs), the morphology of the indentation cracks is similar to that for ZrB<sub>2</sub> samples except that the cracks encounter agglomerates of GNPs. The GNP agglomerates at the grain boundaries seem to deflect the cracks along the grain boundaries leading to intergranular

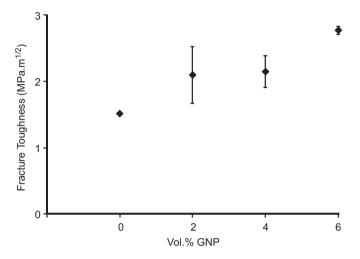


Fig. 9. Fracture toughness of ZrB<sub>2</sub>-GNP composites as a function of GNP reinforcement content.

fracture (Figs. 10(b) and 2(b)). In ZrB<sub>2</sub>-GNP composites with higher amount of GNP reinforcement (4 and 6 vol%), the indentation cracks propagated both along the grain boundaries and through the grains with crack deflection at the graphene agglomerates at the grain boundaries (Fig. 10(c,d)). The fracture surface also shows flatter surface features with predominantly transgranular fracture through relatively coarse grains (Fig. 2(c,d)). The figure clearly shows the pulled out graphene plates at the grain boundary (Fig. 2(c)). The graphene plates wrapped along the grain boundaries over multiple grains can also be seen in the composite sample reinforced with 6 vol\% GNPs (Fig. 2(d)). The difference in focus between lower left and upper right regions of this micrograph (Fig. 2(d)) indicate crack deflection at the graphene plates at the grain boundaries. The graphene plates also seem to bridge the indentation crack propagating perpendicular to the plates as seen in Fig. 11. While the ZrB<sub>2</sub>–GNP composite samples exhibited relatively higher relative density and grain size than monolithic ZrB<sub>2</sub> samples, the fracture surfaces clearly indicate toughening mechanisms such as crack deflection and crack bridging with the reinforcement of GNPs in the composites. The effect of grain size on toughening in ceramics is often very weak. As the ZrB2-GNP composites reinforced with 4 and 6 vol% GNPs exhibited fairly similar relative density (about 97%), higher indentation fracture toughness of composite reinforced with 6 vol% GNPs is indicative of increased contribution of GNPs in toughening the ceramic.

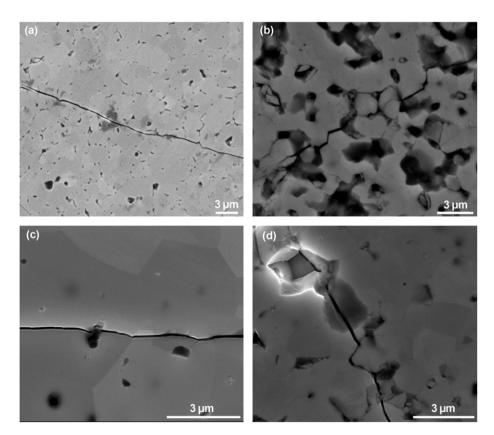


Fig. 10. Morphologies of indentation cracks for (a) monolithic ZrB<sub>2</sub>, (b) ZrB<sub>2</sub>-2% GNPs, (c) ZrB<sub>2</sub>-4% GNPs, and (d) ZrB<sub>2</sub>-6% GNPs composites.

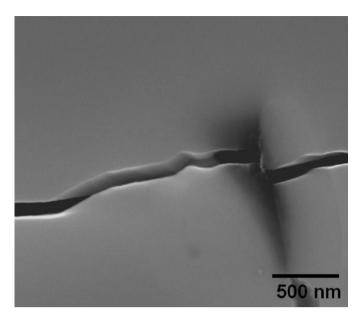


Fig. 11. A high magnification SEM micrograph showing bridging and deflection of an indentation crack front by graphene nanoplates.

#### 4. Conclusions

Graphene nanoplate (GNP) reinforced ZrB<sub>2</sub> ultra-high temperature ceramic composites were successfully sintered using spark plasma sintering. In-process punch displacement profiles and thermodynamic analysis indicated that the reinforcement of GNPs favors the densification of composites, seemingly through interfacial reactions. Raman spectroscopy analysis indicated that the GNPs, with some extent of exfoliation or interfacial reactions, can be retained in the composites sintered at high temperature of 1900 °C. The analysis of the ratio of D and G band intensities  $(I_D/I_G)$ indicated that the average defect density in graphitic carbon has increased by more than 2 times for ZrB<sub>2</sub>-GNP composites and it scales inversely with the concentration of GNPs. The  $I_D/I_G$  has increased for ZrB<sub>2</sub> composites because of interfacial reactions between GNPs and ZrB2 matrix, leading to removal of oxide impurities or formation of ZrC, at high SPS processing temperature (1900 °C). Toughening effects of GNPs such as crack deflection and crack bridging were directly observed from high magnification images of the fracture surfaces and indentations crack morphologies.

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