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Optical properties of multiferroic LuFeO₃ ceramics

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Abstract

We report on the optical behavior of pure orthorhombic perovskite LuFeO₃ ceramics in the energy range of 1.24–5 eV determined by spectroscopic ellipsometry and diffuse reflectance spectroscopy. The value of optical band gap for the LuFeO₃ was confirmed to be 2.76 ± 0.01 eV. Main mechanism of LuFeO₃ ceramics for the band gap possibly arises from the direct electronic transitions between the major channel Fe 3d to O 2p (t_{1g} (π) to t_{2g}) charge transfer excitation. © 2013 Elsevier Ltd and Techna Group S.r.l. All rights reserved.

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1. Intruduction

Multiferroic materials have attracted much attention for their abundant potential applications to modern optoelectronics and spintronics [1,2]. For the practical application, looking for the room temperature multiferroic materials is the urgent affairs. Materials showing occurrence of intrinsic multiferroism are rather rare. An interesting candidate for intrinsic multiferroism is LuFeO₃ [3–6]. The possible interaction in LuFeO₃ is different from that in original multiferroic compounds, and that will deeply affect the development of new spintronics and optoelectronics devices [2,7]. Furthermore, Zhang et al. [6] observed that a giant dielectric constant step with strong frequency dispersion in LuFeO₃ similar to that in LuFe₂O₄, Accompanied by its dielectric characteristic, this material possess particular properties, such as structural anisotropy and anti-symmetric spin coupling [8,9]. All of these advantages make LuFeO₃ a unique multiferroic material.

LuFeO₃, as a member of the rare-earth orthoferrites family, crystallize with the orthorhombic distorted perovskite structure in space group Pbnm. Both the oxygen environment of the Lu³⁺ cation and the octahedral environment of the Fe³⁺ ion are strongly distorted. The ions of transition metals in the compounds partially and even completely sometimes govern their optical,

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magnetic, and other properties [10]. Recently, there have been some reports on the optical properties and electronic structure of LuFeO₃. From the views of experimental, Wang et al. [11] investigated the optical spectra of hexagonal films of LuFeO₃, the optical band gap was assigned to 2.0 eV with crystal field energy splitting of Fe³⁺ and purely charge transfer theory. Pavlov et al. [12] studied optical absorption of hexagonal RFeO₃ (R=Ho, Er, Lu) films, the absorption bands were assigned to 2.27 and 2.97 eV. From the views of theoretical calculation, the band gap of LuFeO₃ has been studied by different groups [13], but nor have they reached consensus on it. However, the optical band gap (E_{o}) parameter of a given material is of particular importance for both scientific understandings and potential technological applications. And the previous studies of LuFeO3 mostly focused on the films, few researches have been conducted on bulk form. In order to clarify and elaborate the optical properties of LuFeO₃, the further investigation of optical spectroscopy is believed to be crucial. In this work, ultraviolet-visible-near-infrared (UV-vis-NIR), diffuse reflectance spectroscopy (DRS) and spectroscopic ellipsometry (SE) were employed to identify the optical parameters of LuFeO₃ ceramics.

2. Experimental

LuFeO₃ ceramics were prepared by a routine solid state reaction method. Lu₂O₃ (99.99%) and Fe₂O₃ (99%) were used as raw

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materials. Stoichiometric amounts of the raw materials were weighed and wet milled for 5 h using absolute alcohol and zirconia balls, dried in a non-vacuum evaporator at 353 K. The dried powders were compressed into pellets and then pre-sintered at 1073 K for 2 h using a muffle furnace. The sintered pellets were ground into powder and finally calcined for 2 h at 1473 K in air to form the ceramic samples. The procedure was repeated to increase sample homogeneity. The samples were double-side polished with a mechanical polishing process to smooth the surface. This process consists of three procedures: coarse grinding, fine grinding, and polishing. Then, the ceramics were rigorously cleaned in pure ethanol with an ultrasonic bath and rinsed several times by deionized water for spectral measurements. The phase purity of the sintered ceramics was determined by X-ray diffraction (XRD, Rigaku D/max-2200, with Cu K_{α} radiation). The surface micrograph of the sintered samples was characterized by a field emission scanning electron microscopy (FESEM, Philips XL30FEG). Raman scattering experiment of the ceramic was performed with a micro-Raman spectrometer (Jobin-Yvon LabRAM HR 800UV, using the 488 nm emission line of argon ion laser). The ellipsometric measurements were carried out in the photon energy range of 1.24-5 eV (245-1000 nm) with a spectral resolution of 1.5 nm by UV-vis-NIR SE (M-2000 by J.A. WOOLLAM CO. INC.). UV-vis-NIR DRS of the ceramics was recorded on a Varian Cary 500 UV-vis-NIR spectrometer equipped with the integration sphere in the wavelength region between 200 and 2000 nm.

3. Results and discussion

X-ray diffraction (XRD) patterns recorded at room temperature of the LuFeO₃ ceramics sintered at 1473 K for 2 h is shown in Fig. 1. All reflections could be adequately indexed according to LuFeO₃ (ICDD-PDF # 74-1483) with orthorhombic perovskite structure. It indicates that the samples are pure perovskite structure and there are no additional or intermediate phases, which also can be confirmed by the following Raman spectroscopy. Lattice constants a, b, c and volume of unit cell for the LuFeO₃ ceramics by Bragg's law are 5.216 Å, 5.552 Å, 7.584 Å (magnitude of error, \pm 0.001 Å) and 219.63 Å³, respectively. In comparison with the standard data for LuFeO₃ powder crystals obtained from JCPDS # 74-1483, the lattice constants obtained

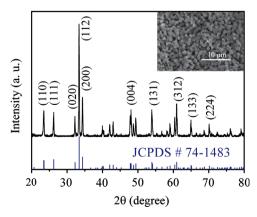


Fig. 1. X-ray diffraction pattern for LuFeO₃ ceramics sintered at 1473 K in air for 2 h. The inset is SEM micrograph of the surface of LuFeO₃ ceramics.

here are slightly larger. There are two possible reasons for this change in lattice constants. One is grain clamping effect in process of formation of ceramics and the stress maintain vacancy defects found in grain boundary, the other is intrinsic vacancy defects caused by slight non-stoichiometry. The inset of Fig. 1 illustrates the SEM image of the sample sintered at 1473 K. It can be seen that there are many square-shaped grains stacking onto each other with the average grain size about 3 µm.

Fig. 2 shows the Raman spectra of polycrystalline LuFeO₃ ceramics, measured at room temperature using 488 nm excitation lines. Raman spectroscopy is also the powerful method to analyze the structure and the phase purity of the samples. In order to obtain the exact peaks, which represent the presence of Raman active modes, the measured spectra were fitted, and the fitted curves were decomposed into individual Lorentzian components, the peaks positions can be deduced, which are summarized in Table 1. There are four LuFeO₃ units within the unit cell, and the 12 oxygen atoms are distributed among sites of two different symmetries within the unit cell, as shown in inset of Fig. 2. Since there are 20 atoms in the primitive cell, the phonon dispersion relation has three acoustical and 57 optical phonon branches. The site symmetries of all the ions are as follows:

$$Lu^{3+}(Cs); Fe^{3+}(C_i); O_I^{2-}(Cs); O_{II}^{2-}(C_1).$$

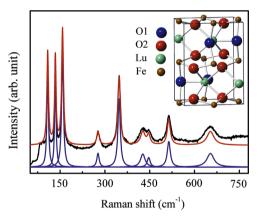


Fig. 2. Measured Raman spectra (black line), together with their fitted spectra (red line) and the decomposed active modes (violet line), for LuFeO₃. The unit cell of LuFeO₃ crystals is shown as inset. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

Table 1 Comparison of Raman peaks between experimental results and literature.

	Raman peak								
Our work $(\pm 2 \text{ cm}^{-1})$	108	134	158	277	347	427	446	514	653
literature ^a	110	136	158	278	350	427	450	516	654
Phonon mode	A_g	A_g/B_{1g}	A_g	A_g/B_{1g}	A_g/B_{1g}	A_g/B_{1g}	A_g	A_g/B_{1g}	A_g

^aRef. [14].

 ${\rm Fe}^{3+}$ ions possessing the inversion symmetry generate no Raman active mode. First order Raman scattering is expected from 24 active modes with the representations ${\rm A}_{1g}$, ${\rm B}_{1g}$, ${\rm B}_{2g}$ and ${\rm B}_{3g}$ [14]. Only nine modes could be detected here. The other modes (${\rm B}_{2g}$ and ${\rm B}_{3g}$) probably exhibit either too weak intensities or have frequencies outside the measured spectral range. Furthermore, it should be noted that also for other perovskite oxides belonging to the Pnma space group, e.g. LaMnO₃, BiCrO₃ [15–17], a smaller number of modes than theoretically predicted were observed experimentally. As shown in Table 1, the peak positions are in good agreement with the previous measurements in a single crystal of LuFeO₃ [14], further indicating that the LuFeO₃ sample with pure orthorhombic pervoskite phase.

As a sensitive and nondestructive optical method, SE based on the reflectance configuration is easily acceptable for the determination of optical constants [18]. In order to extract the optical function from the ellipsometric spectra, a dielectric function model should be applied. The optical properties of the LuFeO₃ ceramics are analyzed via fitting the reflectance spectra with reasonable dielectric function model. The complex reflectance ratio, ρ , of ceramics is a function of ellipsometry parameters of Ψ and Δ . The fundamental equation of ellipsometric is described as: ρ =tan Ψ exp(i Δ). The dielectric function (ε) of ceramic bulk is a common representation via the two-phase (air/ceramic) model [19,20],

$$\varepsilon = \varepsilon_1 + i\varepsilon_2 = \sin^2\phi + \sin^2\phi \tan^2\phi \frac{(1-\rho)^2}{(1+\rho)^2},$$

where φ is the angle of incidence, ρ is the complex reflectance ratio. The diectric functions are related to the refractive index, n, and extinction coefficient, k, by the Fresnel formula: $\varepsilon_1 = n^2 - k^2$, and $\varepsilon_2 = 2nk$.

Fig. 3 shows the experimental ellipsometric data (Ψ and Δ) for the LuFeO₃ ceramic from near-infrard to ultraviolet photon energy region recorded at the incident angle of 70° . Inset of Fig. 3 shows the calculations of refractive index (n) and extinction coefficient (k). Bahng et al. [21], Shen et al. [22], and Chu et al. [23] proposed that the values of refractive index peaks generally correspond to the band gap energies of the dielectrics and semiconductors. This can be understood from the Kramers–Kronig relation,

$$n(\nu) = 1 + \frac{c}{\pi} \int_0^\infty \frac{d\alpha(\nu')}{\nu'} \log(\frac{\nu' + \nu}{\nu' - \nu}) d\nu'$$

where c is the light velocity, ν' and ν are frequencies of light, and $\alpha(\nu')$ is the absorption coefficient at frequency ν' . In the energy region near the fundamental absorption region, the value of $d\alpha(\nu')/d\nu'$ goes up and the refractive index increases as the photon energy enhances. However, when the photon energy reaches the gap energy, i.e., interband critical point, the absorption curve changes its slope and becomes flatter, and then the value of $d\alpha(\nu')/d\nu'$ decreases above the gap energy. Thus, peaks appear in the refractive index at the gap energy. The refractive index peak site is 2.78 eV. This value is in good agreement with the values determined by Tauc's power law, and it is also similar to the other perovskite oxides, e.g. BiFeO₃ (Ref. 2).

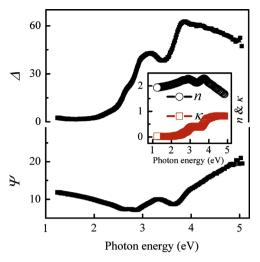


Fig. 3. Experimental ellipsometric data (Ψ and Δ) for the LuFeO₃ ceramic from near-infrard to ultraviolet photon energy region recorded at the incident angle of 70° . The inset shows the refractive index, n, and extinction coefficient, κ .

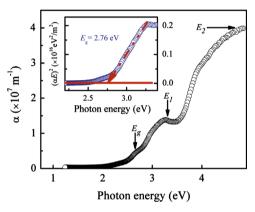


Fig. 4. Optical absorption coefficient α of LuFeO₃ ceramic. The inset shows E_g determination by Tauc's power law.

Fig. 4 shows the absorption coefficient ($\alpha = 4\pi\kappa/\lambda$) as a function of photon energy $(h\nu)$. It can readily identified two features at about 3.3 eV (E_1) and 4.7 eV (E_2) . We demonstrate that the absorption coefficient in the spectral region around E_1 and E_2 is extremely large, with values of α in excess of $10^7 \,\mathrm{m}^{-1}$. This should compare to the values of $10^2 - 10^6 \,\mathrm{m}^{-1}$ in the spectral region immediately above the optical gap [11]. The band edge absorption is completely negligible on the scale of the data because of the band edge absorption is weak and the density of states at the band edge is comparatively small [24–26]. Thus, the optical gap of LuFeO₃ should be located around E_g region (2–3 eV). The optical gap, E_g , values of the LuFeO₃ ceramics can be evaluated from Tauc's power law, $\alpha E \propto (E - E_g)^{\eta}$, where E = hv the photon energy, and η is a constant with value 1/2 for a direct band gap type and 2 for an indirect one. Inset of Fig. 4 shows a plot of $(\alpha E)^2$ $(\eta = 1/2)$ versus photon energy. The extrapolated linear portions show that the optical gap for direct transition is 2.76 ± 0.01 eV.

To better explain electronic structure of the orthorhombic LuFeO₃, the DRS measurement was performed. Fig. 5 shows the DRS absorbance spectra of the LuFeO₃ ceramics. It can be easily found two evident peaks at 1.22 and 1.73 eV, and a steep

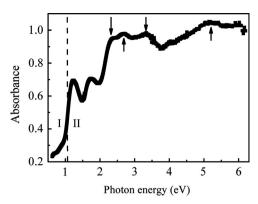


Fig. 5. Optical absorption spectra of the $LuFeO_3$ ceramic using diffuse reflectance spectroscopy.

slope around 1 eV, not similar to the reported optical properties of hexagonal LuFeO₃ films. This discrepancy is due to the form of material and characterization methods. The origin of the peaks detected in the sample could be manifold. We analyze the possible origin of the observed absorption bands here on the basis of crystal filed theory and charge transfer transition. The spectra can be roughly divided by the dashed line into two parts. The first part (I) is called as Drude region, and the other one (II) is the charge transfer (CT) region [27]. Valence shell structures of Lu, Fe, and O are 5s5p5d4d6s, 3s3p4s4p3d, and 2s2p states, respectively [13]. $LuFeO_3$ has a simple magnetic structure because of the complete fshell of lutetium. One Fe³⁺ contains five 3d electrons, which reside in a high spin state and occupy the only spin sextet and orbital singlet ⁶S lying below the other terms [10]. There are 18 oxygen O 2p atomic orbitals in the octahedral FeO₆. The results of the evident peak position are in good agreement with the literatures [10]. 1.22 and 1.73 eV are the transitions from ${}^6A_{1g}$ to ${}^4T_{1g}$ and ⁴T_{2g} split line, respectively, which can be summarized to crystal field d-d transitions [2,10]. There are also four broad peaks at 2.34, 2.73, 3.35 and 5.14 eV marked by the arrows in Fig. 5. The local charge transfer energy to move an electron from the O^{2-} ion to the Fe³⁺ ion is essentially given by the difference in the ionization potential of O and the electron affinity of Fe in the actual solid [28]. Thus, the electron excitation around 5.14 eV is probably strong hybridized major channel O 2p and Fe 3d to Lu 5d state excitation. The low-energy intense band around 3.35 eV can be assigned to the strong dipole-allowed on-center $t_{2\mu}$ (π) to t_{2g} $p\!-\!d$ CT transition as in literature [29]. Possible mechanisms for the inflection points at 2.34 is Fe³⁺ d-d crystal field transition, and points at 2.73 eV is the optical band gap, possibly the major channel Fe 3d to O 2p $(t_{1g}(\pi))$ to t_{2g} charge transfer excitation. This result of optical band gap of DRS measurement was further explained with the previous SE investigation. Furthermore, defect and slight non-stoichiometry in the sample might ascribe to the broad peak in the spectral, and this requires a more thorough investigation, especially by high resolution electron microscopy.

4. Conclusions

In conclusion, XRD and Raman scattering results show that LuFeO₃ ceramics sintered at 1473 K for 2 h possesses

orthorhombic perovskite structure. The optical band gap for LuFeO₃ is 2.76 ± 0.01 eV by Tauc's power law. The value of refractive index peak is 2.78 eV generally correspond to the band gap energies. The detailed electronic transitions of multiferroic LuFeO₃ have been further investigated using UV–vis diffuse reflectance spectroscopy, and the band gap obtained by DRS is 2.73 eV.

Acknowledgments

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