

Glass-free low-temperature sintering and microwave dielectric properties of $\text{CaWO}_4\text{--Li}_2\text{WO}_4$ ceramics

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Abstract

The microwave dielectric properties of $(1-x)\text{CaWO}_4\text{--}x\text{Li}_2\text{WO}_4$ ceramics prepared by the solid state reaction were investigated as a function of Li_2WO_4 content ($0 \leq x \leq 0.14$). Due to the addition of Li_2WO_4 phase, the sintering temperature was reduced from 1150 °C to below the LTCC temperature limits. For the ceramics sintered at 900 °C and 950 °C, with the increase of x , the $Q \times f$ values first increased and then decreased. The density and permittivity presented the same behaviour as the $Q \times f$ values. At $x=0.1$, the $(1-x)\text{CaWO}_4\text{--}x\text{Li}_2\text{WO}_4$ ceramics sintered at 900 °C/2 h presented excellent microwave dielectric properties: $\epsilon_r=9.002$, $Q \times f=11.76 \times 10^4$ GHz and $\tau_f=-55$ ppm/°C. The ceramic was a chemically compatible with silver, which makes it a very promising candidate for LTCC application.

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Keywords: Microwave dielectric properties; LTCC; $\text{CaWO}_4\text{--Li}_2\text{WO}_4$; Low loss

1. Introduction

Intensive development of wireless communication systems has resulted in an increasing demand for microwave dielectric materials with low dielectric constant and low loss materials. In addition, it is desired to integrate the passive components to a function module by low temperature co-fired ceramic (LTCC) technology. Materials with low sintering temperature (<960 °C), high quality factor ($Q \times f$) value and near zero temperature coefficient of resonant frequency (τ_f) are necessary for the applications [1–3]. Currently, reducing the sintering temperature without deteriorating the microwave properties is an arduous problem in LTCC material research. Low melting glass addition, chemical processing, and small particle size of starting materials are the three main methods normally used to reduce the sintering temperature of these dielectric ceramics. However, fine powders prepared by wet chemical processing such as the sol–gel process may complicate the fabrication process and increase the cost. Addition of low melting glass either leads to poor microwave dielectric properties or significantly increases the possibility of chemical reaction with the metal electrode due to

the presence of complicated phases in the LTCC system [4,5]. In addition, glass addition can cause cracks during the soldering process due to the different thermal expansion coefficients between glasses and ceramic, especially when the glass addition amount is over 5 wt% [6,7].

Therefore, a glass-free LTCC material with low permittivity and appropriate microwave dielectric properties enables fabrication of devices for high speed applications [8,9]. Owing to their low permittivities (≤ 12) and high $Q \times f$ values ($> 50,000$ GHz), the scheelite ceramics, with the general formula MWO_4 ($\text{M}=\text{Ca}$, Sr , and Ba) were regarded as promising materials for microwave substrate applications [10]. Previously, it was reported that CaWO_4 sintered at 1100 °C for 3 h exhibited a low ϵ_r (≈ 10) and a high $Q \times f$ value ($\approx 75,000$ GHz) [11]. Meanwhile, Li_2WO_4 presents excellent microwave dielectric properties ($Q \times f \approx 62,000$ GHz) and low melting point (≈ 741 °C) [12], which might promote the sintering of CaWO_4 . In fact, Sung Joo Kim and Eung Soo Kim have investigated the effects of Li_2WO_4 on the microwave dielectric properties of $0.85\text{CaWO}_4\text{--}0.15\text{SmNbO}_4$ ceramics as a sintering additive, but not the effects as a second phase [13].

In this paper, we investigated the sintering behavior and microwave dielectric properties of $(1-x)\text{CaWO}_4\text{--}x\text{Li}_2\text{WO}_4$ to obtain a new glass-free LTCC material.

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2. Experimental procedure

WO₃ (99%), CaCO₃ (99.9%) and Li₂CO₃ (99%) were used as starting materials. The powders were weighed according to the formula of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0 \leq x \leq 0.14$), and milled with ZrO₂ balls in distilled water for 12 h. The mixtures were dried and calcined at the temperature of 800 °C for 2 h. The calcined powders were reground for 12 h, then dried and mixed with 15 wt% PVA as binder and granulated. The granulated powders were uni-axially pressed into pellets with 12 mm in diameter and 5–6 mm in height under the pressure of 10 MPa. The pellets were sintered at the temperature ranging from 850 °C to 950 °C for 2 h with heating rate of 5 °C/min.

The crystalline phases of the sintered ceramics were analyzed by X-ray diffraction (XRD: DX-2700) using Cu K_α radiation. The micrographs were obtained by using scanning electron microscope SEM (SEM: INCA penta FETX3 oxford). The bulk density was measured by the Archimedes method. The microwave dielectric properties of the sintered samples were evaluated by the Hakki–Coleman method and Agilent N5230A (300–20 GHz) network analyzer in a resonant cavity. The temperature coefficient resonant frequency (τ_f) was also measured by the same method in a temperature ranging from 20 °C to 80 °C and calculated from the following equation: $\tau_f = (f_{80} - f_{20}) / [f_{20} \times (80 - 20)] \times 10^6$ (ppm/°C), where f_{80} and f_{20} represent the resonant frequency at 80 °C and 20 °C, respectively. The chemical compatibility with silver was investigated by co-firing the mixed powders with pure silver powders (20 wt % Ag) in ambient atmosphere at 900 °C for 2 h.

3. Results and discussion

Fig. 1 shows the XRD patterns of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0 \leq x \leq 0.14$) samples sintered at 900 °C for 2 h. All Li₂WO₄ added samples exhibited CaWO₄ (PDF #41-1431) main phase with trace amount of Li₂WO₄ (PDF #12-0760) phase. It was speculated that the Li₂WO₄ phase could coexist with CaWO₄ phase in the $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ system. CaWO₄ belonged to the tetragonal scheelite system (space group I4₁/a (No. 88)) [14], and the lattice parameters are $a=b=0.524$ nm, $c=1.138$ nm. Li₂WO₄ belonged to the rhombohedral phenacite structure (space group R $\bar{3}$ (No. 148)) [12], and the lattice parameters are $a=b=1.436$ nm, $c=0.960$ nm. And, it was observed that the intensities of diffraction peaks of the Li₂WO₄ phase increased gradually with the increase in proportion of Li₂WO₄ in the compounds.

The SEM micrographs of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0 \leq x \leq 0.14$) ceramics sintered at 900 °C for 2 h in air are shown in Fig. 2. The photos indicated that grain size became enlarged with increasing Li₂WO₄. However, the grains first gradually became denser with the increase of x from 0 to 0.1, and then abnormal grain growth and defects gradually increased with x of more than 0.12, which was mainly attributed to the excessive liquid phase of Li₂WO₄.

Fig. 3 shows the $Q \times f$ values of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0.06 \leq x \leq 0.14$) ceramics sintered at 850 °C, 900 °C and 950 °C for 2 h in air. With the increase of Li₂WO₄ content, the optimal sintering temperature gradually decreased. This was due to the low

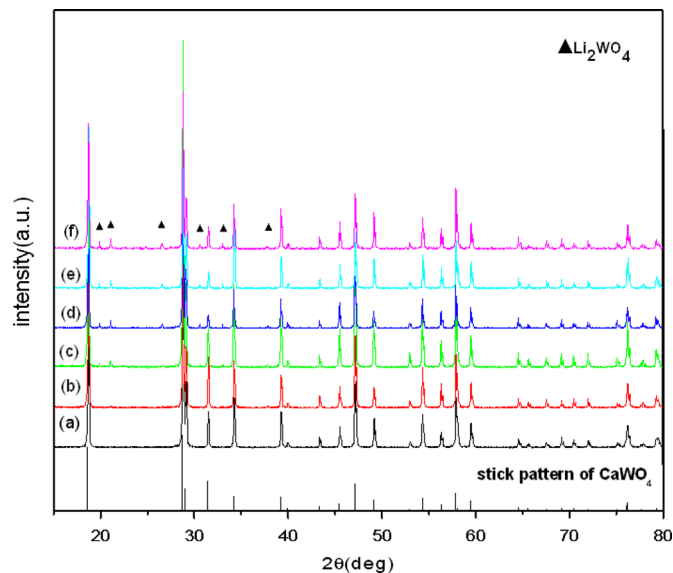


Fig. 1. XRD patterns of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ceramics (a) $x=0$, (b) $x=0.06$, (c) $x=0.08$, (d) $x=0.10$, (e) $x=0.12$, and (f) $x=0.14$ sintered at 900 °C for 2 h in air.

melting point of Li₂WO₄. When the sintering temperature was at 850 °C, the $Q \times f$ values of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0.06 \leq x \leq 0.14$) ceramics clearly increased but not yet saturated. However, when the ceramics sintered at 900 °C and 950 °C, with the increase of x , the $Q \times f$ value increased to a maximum and then decreased. This might be related to the density of the samples. As shown in Figs. 2 and 3, we could see that the $Q \times f$ values had roughly the same trend as the density. This could be related to the abnormal grain growth, which results in more pores. It is known that the microwave dielectric loss includes the intrinsic and the extrinsic losses [2]. The intrinsic losses are dependent on the crystal structure and mainly caused by lattice vibration modes [15], while the extrinsic losses are associated with many factors, such as impurities, microstructural defects, densification or porosity, grain size, second phase, presence of liquid-phase or grain boundaries [16]. Therefore, our results might become apparent because extrinsic factors play a major role, especially the porosity.

The changes in relative density and permittivity of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0.06 \leq x \leq 0.14$) ceramics sintered at 850 °C, 900 °C and 950 °C for 2 h in air are shown in Figs. 4 and 5. As expected, permittivity closely followed density in all cases. It could be seen that the relative density of the samples clearly increased with the increase of Li₂WO₄ content, when the sintering temperature was at 850 °C. However, when the samples were sintered at 900 °C and 950 °C, with the increase of Li₂WO₄ content, the relative density increased to a maximum and then decreased. The ceramic sample had a relative density above 97%. As shown in Fig. 2, the SEM photos show that the increase in porosity with x of more than 0.12 lead to deteriorated density. Also, the calculated density of the sample was obtained by the following equations [17]:

$$\rho = \frac{w_1 + w_2}{w_1/\rho_1 + w_2/\rho_2} \quad (1)$$

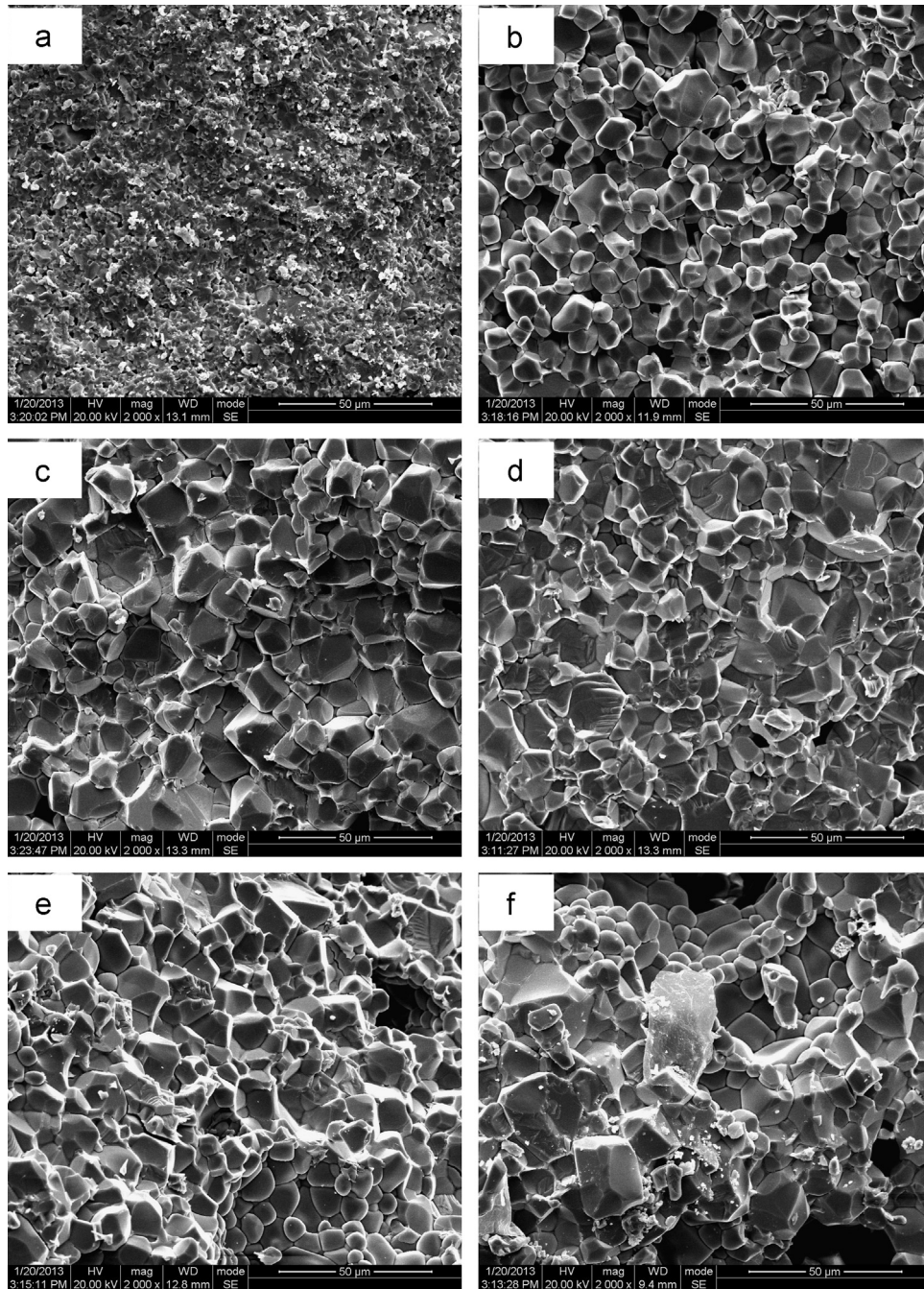


Fig. 2. SEM micrographs of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ceramics. (a) $x=0$, (b) $x=0.06$, (c) $x=0.08$, (d) $x=0.10$, (e) $x=0.12$, and (f) $x=0.14$ sintered at 900°C for 2 h in air.

Where ρ_1 and ρ_2 are the theoretical density of CaWO_4 and Li_2WO_4 , w_1 and w_2 are the weight fraction of CaWO_4 and Li_2WO_4 . Therefore, with the increase of Li_2WO_4 content, the density of the compound decreased, as Li_2WO_4 was less dense than CaWO_4 . The similar behavior of permittivity and density indicated that the permittivity was significantly affected by the density.

Fig. 6 shows the τ_f values of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0.06 \leq x \leq 0.14$) ceramics sintered at 900°C for 2 h in air. It could be seen that the τ_f values were all negative. This could be overcome by adding a positive τ_f material. As the Li_2WO_4

content increases, the absolute value of τ_f increased. This result was different from the report [13] that τ_f increased to the positive value when Li_2WO_4 acting as a sintering additive only. In a composite ceramic system, τ_f of the samples can be calculated by the following equations [18]:

$$\tau_f = V_1\tau_{f1} + V_2\tau_{f2} \quad (2)$$

Where V_1 and V_2 are the volume fraction of CaWO_4 and Li_2WO_4 , τ_{f1} and τ_{f2} are the τ_f values of CaWO_4 and Li_2WO_4 phase, respectively. CaWO_4 had a higher τ_f of $-53 \text{ ppm}/^\circ\text{C}$ [19], while Li_2WO_4 had a lower τ_f of $-146 \text{ ppm}/^\circ\text{C}$ [12].

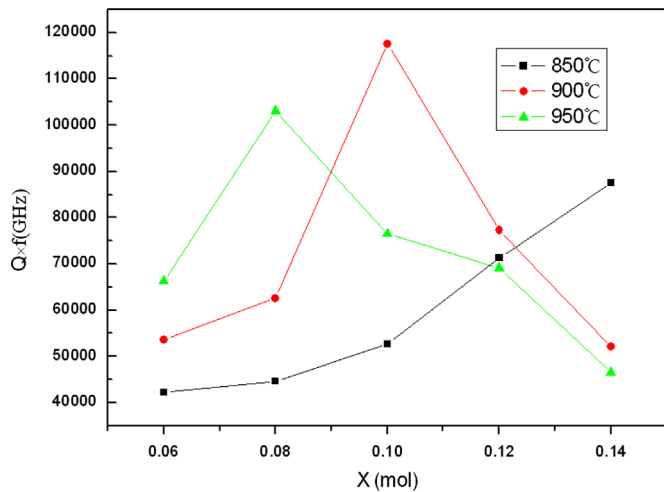


Fig. 3. The $Q \times f$ of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($x=0.06$, $x=0.08$, $x=0.10$, $x=0.12$, $x=0.14$) ceramics sintered at 850 °C, 900 °C and 950 °C for 2 h in air.

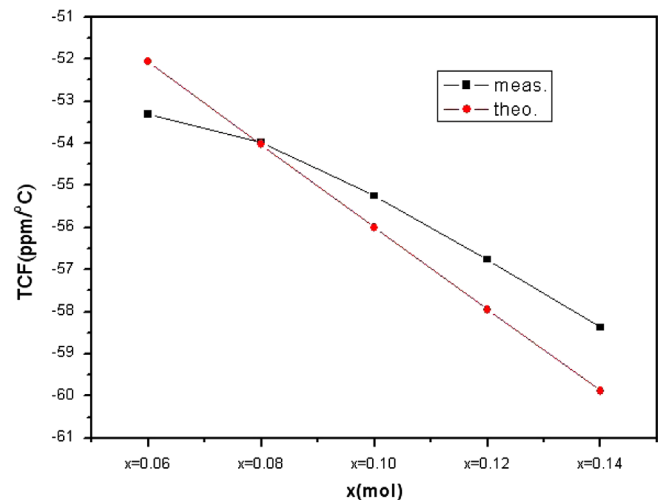


Fig. 6. The τ_f of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0.06 \leq x \leq 0.14$) ceramics sintered at 900 °C for 2 h in air.

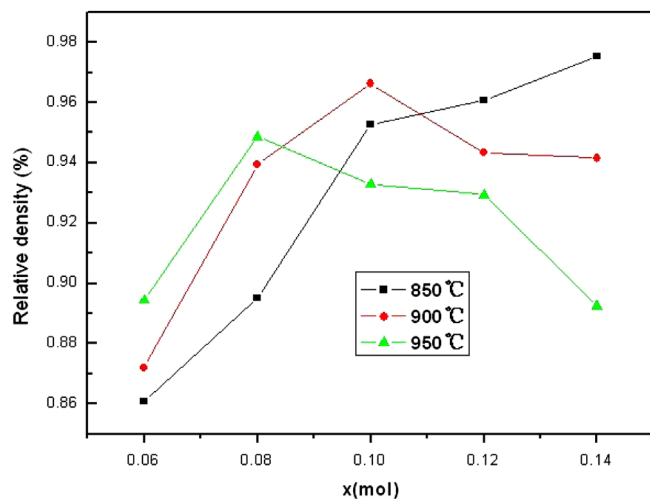


Fig. 4. The relative density of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0.06 \leq x \leq 0.14$) ceramics sintered at 850 °C, 900 °C and 950 °C for 2 h.

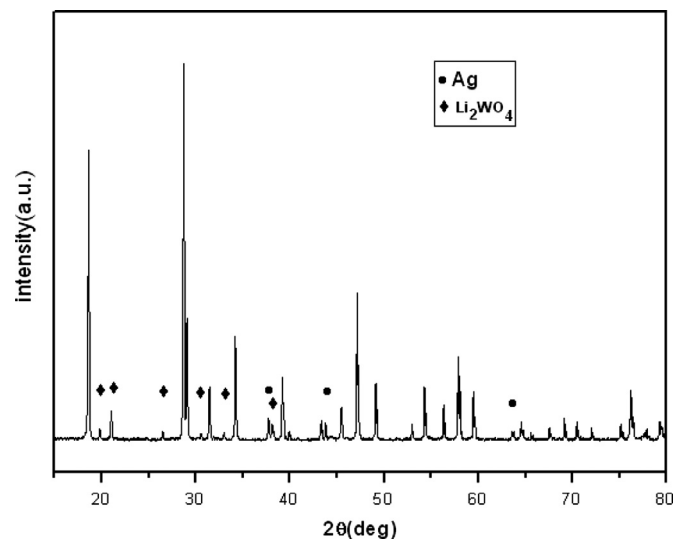


Fig. 7. Powder XRD pattern of $0.9\text{CaWO}_4-0.1\text{Li}_2\text{WO}_4$ mixed with 20 wt% Ag powders sintered at 900 °C for 2 h in air.

The theoretical τ_f value of the sample decreased with the increasing Li_2WO_4 content. Similarly, the measured τ_f value of the sample also decreased with increasing Li_2WO_4 content.

Fig. 7 shows the XRD pattern of the $0.9\text{CaWO}_4-0.1\text{Li}_2\text{WO}_4$ mixed with 20 wt% Ag powders sintered at 900 °C/2 h in air. All the major peaks of Ag were observed in the XRD pattern and no phase of silver compounds presented. It seems to show that no chemical reaction has taken place between the $0.9\text{CaWO}_4-0.1\text{Li}_2\text{WO}_4$ ceramic and Ag.

4. Conclusion

CaWO_4 ceramics modified with Li_2WO_4 were investigated as a promising microwave dielectric material for LTCC applications. All the samples exhibited CaWO_4 main phase with trace amount of Li_2WO_4 phase. The Li_2WO_4 reduced the sintering temperature and the microwave dielectric loss. All samples exhibited negative τ_f values. High performance of microwave dielectric properties were

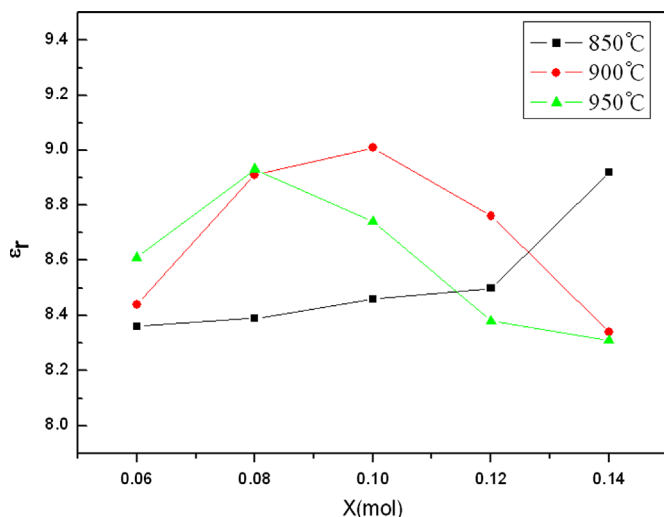


Fig. 5. The permittivity of $(1-x)\text{CaWO}_4-x\text{Li}_2\text{WO}_4$ ($0.06 \leq x \leq 0.14$) ceramics sintered at 850 °C, 900 °C and 950 °C for 2 h.

obtained in the $0.9\text{CaWO}_4-0.1\text{Li}_2\text{WO}_4$ ceramics sintered at 900°C with a microwave permittivity of 9.002, $Q \times f = 11.76 \times 10^4$ GHz and $\tau_f = -55$ ppm/ $^\circ\text{C}$. The ceramic was a chemically compatible with silver, which makes it a very promising candidate for LTCC application.

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