

## Selected Abstracts from *Yogyo-Kyokai-Shi*

As a service to readers and with the agreement of The Ceramic Society of Japan, selected English language Abstracts of the papers appearing in the *Journal of the Ceramic Society of Japan (Yogyo-Kyokai-Shi)* are reproduced here. The selection was made by Drs R. Stevens and P. Popper.

*Yogyo-Kyokai-Shi* **94** (1986)

**The Effect of  $\text{Al}_2\text{O}_3$  or  $\text{SiO}_2$  Addition on the Thermal Shock Resistance of  $\text{Y}_2\text{O}_3$ -Stabilized  $\text{ZrO}_2$**  Hiroshi WATANABE and Mitsuo CHIGASAKI, *Yogyo-Kyokai-Shi*, **94**, 255-60 (1986)—Thermal shock resistance of  $\text{ZrO}_2$ -(4-17)%  $\text{Y}_2\text{O}_3$  has been investigated. A relatively excellent thermal shock resistance is obtained in partially stabilized  $\text{ZrO}_2$ -7%  $\text{Y}_2\text{O}_3$ . The effect of  $\text{Al}_2\text{O}_3$  or  $\text{SiO}_2$  addition to  $\text{ZrO}_2$ -7%  $\text{Y}_2\text{O}_3$  has been also investigated to enhance its thermal shock resistance.  $\text{ZrO}_2$ -7%  $\text{Y}_2\text{O}_3$ -(0.5-4)%  $\text{Al}_2\text{O}_3$  indicated an improved thermal shock resistance. In  $\text{ZrO}_2$ -7%  $\text{Y}_2\text{O}_3$ , the monoclinic  $\text{ZrO}_2$  grains tended to coalesce into large clusters in the cubic  $\text{ZrO}_2$  matrix. As a result, large cracks are prone to initiate from the monoclinic phase during the martensitic transformation. On the other hand, the addition of  $\text{Al}_2\text{O}_3$  results in the isolated spherical grains of monoclinic  $\text{ZrO}_2$  phase, leading to an improved thermal shock resistance. 4 figs., 1 table, 19 refs. [H. W.]

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**Firing Shrinkage of Alumina Green Sheet** Shozo OHTOMO, Masatoshi KATO and Kimitake KOREKAWA, *Yogyo-Kyokai-Shi*, **94**, 261-66 (1986)—The alumina Green sheet (96% purity; Remainder 4% consists  $\text{SiO}_2$ ,  $\text{CaO}$  and  $\text{MgO}$ ) produced by doctor blade method is popularly used for the electronic ceramic parts. In this study,

the firing shrinkage behavior of alumina green sheet is discussed quantitatively by the mixing of certain ratio of two types of low soda aluminas selected from three commercial types of low soda aluminas. Higher green density is obtained by the mixing of two types. The green density of tape is predictable from particle size distribution. The linear shrinkage is determined by the dimensional distribution ratio which seems to be determined essentially by the anisotropy of alumina particles in the orientation effects by the tape processing equipment. 13 figs., 3 tables, 9 refs. [S. O.]

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**Milling Effects of Starting Hydrated Aluminum Sulfate on  $\eta \rightarrow \alpha$  Phase Transformation and Sinterability of Alumina** Keiji DAIMON and Etsuro KATO, *Yogyo-Kyokai-Shi*, 94, 273-80 (1986)—The milling effects of hydrated aluminum sulfate were studied. Dehydration of milled hydrated sulfate produced broken-egg shell-like particles composed of finer anhydrous sulfate particles through melting of the hydrate on heating. The anhydrous sulfate desulfated into aggregate grains of  $\eta$ - $\text{Al}_2\text{O}_3$  with irregular pore size distribution, and the transformation temperature from  $\eta$ - to  $\alpha$ -phase transformation was lowered about 100°C. It was confirmed that the enhancement of phase transformation was attributed to accelerated nucleation in  $\eta$ - $\text{Al}_2\text{O}_3$  grains, and that the sinterability of  $\alpha$ - $\text{Al}_2\text{O}_3$  was improved by the increase in the density of green compacts, which was caused by occurrence of many cracks in the skeletal grains of  $\alpha$ - $\text{Al}_2\text{O}_3$ . 11 figs, 1 table, 11 refs. [K. D.]

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**Measurement of Thermal Diffusivity of Boron Nitride and Sapphire by Laser Flash Method** Hiromichi OHTA and Yoshio WASEDA\*, *Yogyo-Kyokai-Shi*, 94, 295-99 (1986)—In order to measure the thermal diffusivity of ceramics at high temperatures, the correlation factor for radiative heat leak has been estimated with respect to normalized temperature decay rate after its maximum. The thermal diffusivity of boron nitride was obtained at temperature between 1073 and 1773 K with