Lattice Constants of High Cordierite Containing TiO₂ or ZnO Hiroyuki IKAWA, Kunio SATO*, Masaharu SUZUKI**, Kazuyori URABE and Shigekazu UDAGAWA, Yogyo-Kyokai-Shi, 94, 452-53 (1986) — The increase in lattice constants of high cordierite on doping with ZnO or TiO₂ were investigated. Gahnite was an impurity crystalline phase when cordierite glasses doped with ZnO were crystallized by heating at 1050°C for 48 h. Just a trace of gahnite was detected when 1.5 wt% of ZnO was added, but it increased with increased amount of the dopant. The unit cell volume of high cordierite increased on doping with ZnO up to 3 wt%, however, the volume did not increase on further doping. Rutile was one of the impurites when TiO₂ was added to the cordierite glass. In this case, however, solubility limit of TiO₂ in high cordierite was not so well defined as for the case of ZnO. That is, the unit cell volume of high cordierite increased continuously with increasing amount of TiO₂ up to 9 wt%, although the rate of increase in unit cell volume diminished with increased doping. Some differences in crystal chemistry between Zn²⁺ and Ti⁴⁺ ions were considered to be responsible for the observed difference.

1 fig., 1 table, 11 refs.

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