Permittivity and the Form Factor of BaTiO₃ and Related Powders

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Abstract

Capacitance measurements of ferro- and paraelectric oxide powders in the frequency range of 1kHz-10 MHz have been performed in order to deduce a form factor which is related to the morphology of the particles and their arrangement in the powder. Using a capacitor measuring setup the frequency independent real part of the complex dielectric constant was measured in the range of 1 kHz-2 MHz as a function of the volume fraction. The interpretation follows a model derived for inhomogeneous dielectrics, which makes it possible to calculate a morphology dependent parameter provided that the permittivity of the solid is known. The results are verified for various BaTiO₃ powders and for aggregated and stepwise ground Ba_{1-x}Sr_xTiO₃ powders and are related to other particle shape parameters derived from SEM studies indicating the suitability of this method to describe the morphology of ferroelectric powders. For comparision the specific surface area BET data and the d_{50} values from grain size distribution measurements of the powders are also reported. The applicability of the method for checking the reproducibility of different lots of ferroelectric capacitor raw materials is shown for Z5U and X7R powders.

Zusammenfassung

Es wird über Kapazitätsmessungen an ferro- und paraelektrischen Oxidpulvern im Frequenzbereich 1 kHz bis 10 MHz und deren Auswertung zwecks Ermittlung eines Formparameters berichtet, der zur Morphologie der Partikel und deren Anordnung im Pulver in Beziehung steht. Mit Hilfe einer Kondensator—Meßanordnung wird für die Pulver im Bereich bis zu 2 MHz der frequenzunabhängige Realteil der komplexen Dielektrizitätskonstante in Abhängigkeit vom Volumenbruch ermittelt. Die Auswertung folgt den für inhomogene Dielektrika abgeleiteten Modellen, die bei Kenntnis der DK des

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betreffenden Feststoffes einen morphologieabhängigen Formparameter zu berechnen gestatten. Am Beispiel von verschiedenen BaTiO₃ Pulvern und von aggregierten und in mehreren Stufen fein gemahlenen $Ba_{1-x}Sr_xTiO_3$ Pulvern wird durch Vergleich mit REM-Studien erhaltenen Formanderen aus parametern gezeigt, daß diese Methode zur Beschreibung der Partikelmorphologie ferroelektrischer Pulver herangezogen werden kann. Zum Vergleich werden jeweils die BET-Werte für die spezifische Oberfläche und der d50 Wert aus der Korngrößenverteilung angegeben. Die Eignung der Methode zur Kontrolle der Chargenreproduzierbarkeit von ferroelektrischen Kondensatorpulvern wird am Beispiel von Z5U und X7R-Massen aufgezeigt. © 1998 Elsevier Science Limited. All rights reserved

1 Introduction

In a recently published paper, the permittivity measured in the powder state of a solid has been shown to be sensitively dependent on the shape of the particles. Using the models of inhomogeneous dielectrics as developed by Lorentz,² Lorenz,³ Rayleigh,⁴ Wiener,⁵ Bruggeman^{6,7} and other researchers, 8-10 such an interpretation relies on the premise that the dielectric constant of the bulk material is also valid for particles in the powder state. The latter should be sufficiently valid for solids with a linear dependence of polarization on the electric field strength. It has been shown that increasing bulkiness and roundness as defined by Medalia¹¹ indicating agglomeration or porosity of particles give rise to relatively high values of a new defined form factor β which was deduced from dielectric measurements. On the other hand, powders predominantly consisting of primary or less aggregated grains reveal β to be the more sensitive to particle shape the higher the permittivity of the material. The β values of powders, whose anisometry results from the crystal structure, e.g. wurtzite or alumina, have found to be consistent with the results of electronmicroscopic studies.

It is the aim of the present paper to check if the form factor β can be used for characterizing ferroelectric powders. Because of the non-linear polarization of these materials, the field dependence of the permittivity has to be taken into consideration. Furthermore, it seems questionable to use the dielectric constant of ferroelectric bulk ceramics for particles in the powder state. As a cardinal question the well known grain size dependence of the dielectric constant of $BaTiO_3^{12-14}$ has to be taken into account. Finally, the data obtained with the capacitor arrangement could become falsified due to piezoelectric effects when the volume fraction of the solid is varied by changing the pressure.

2 Ferroelectric Powders and Characterization

The investigated powders are listed in Table 1. BaTiO₃ I and Ba_{1 x}Sr_xTiO₃ powders (x = 0.295 and x = 0.593) have been prepared following the mixed oxide route. Wet grinding of mixtures of BaCO₃ and SrCO₃ with highly dispersed rutile (8 m²/g) using agate spheres (10 and 20 mm \varnothing) in a cylinder mill was followed by calcination at 1150 or

1200°C, respectively. The coarse powders consisting predominantly of highly aggregated particles have been ground in order to study the dielectric properties as a function of the morphological state. Besides different commercial BaTiO₃ powders, Z5U and X7R materials have also been studied.

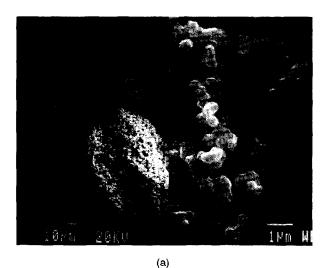
According to $d_s = 6/(\rho S)$ the BET specific surface area S gives an average value for the diameter d_S assuming spherical particles with ρ as the density. Grain size distribution was evaluated by light scattering measurements using the Microtrac apparatus. Particle shape was estimated from SEM studies providing the form parameters anisometry, bulkiness and roundness.11 Anisometry and bulkiness of particles were calculated from the equivalent ellipse having the same first and second degree moments about its central principal axes. For this purpose the particles were assumed to be planar and of uniform thickness and density. The anisometry was then defined as the ratio of the major axis to the minor axis of the equivalent ellipse and bulkiness as the ratio of the ellipse area to the particle area. The roundness is the square of the ratio of the particle perimeter to the perimeter of a circle having the same area as the particle and is given by (perimeter) $^2/(4\pi$ area). The three shape parameters are equal to 1 for spheres. The anisometry is high for elongated particles while

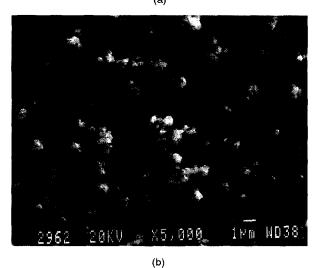
Table 1. Composition, preparation and morphological properties of the investigated powders

Compound	Preparation conditions	$\rho_{th} g/cm^3$	$\frac{S(BET)}{m^2/g}$	$d_S \mu m$	Grain size distribution			SEM information			β
					$d_{10}/\mu m$	$d_{50}/\mu m$	$d_{90}/\mu m$	Anisom.	Bulkin.	Roundn.	-
BaTiO ₃ I	1200°C 10 h	6.040	1.5	0.66	0.45	1.81	12.38	1.35	1.16	52.9	_
coarse											
ground	Milling 5 h		4.3	0.23	0.29	0.46	0.84	1.67	1.06	1.72	0.385
BaTiO ₃ II	Commercial Degussa		8.0	0.12	0.23	0.34	0.54	1.47	1.05	1.93	0.45
BaTiO ₃ III	Commercial Nippon Chem. BT3L		2.3	0.43	0.28	0.41	0.63	1.62	1.05	1.71	0.37
BaTiO ₃ IV	Commercial Nippon Chem. BL1		1.5	0.66	0-31	0.61	1.69	1.44	1.03	1.43	0.31
BaTiO ₃ V	Commercial Ferro 6A		2.2	0.45	0.31	0.58	1.49	1.47	1.04	1.51	0.32
$Ba_{1-x}Sr_xTiO_3$ $x = 0,295$ coarse	1150°C 10 h	5.783	0.9	1.15	1.46	5.92	42.98	1.40	1.21	28.0	-
ground (a)	Milling 5 h		1.1	0.94	0.56	3.73	17.30	1.44	1.06	2.64	0.48
ground (b)	Milling 27 h		1.2	0.86	0.32	1.74	3.50	1.52	1.04	1.83	0.31
$Ba_{1-x}Sr_xTiO_3$ $x = 0.563$	1150°C 10 h	5.541	0.9	1.20	1.19	4.03	12.59	1.54	1.18	67.6	_
coarse											
ground (a)	Milling 5 h		1.2	0.90	0.32	2.77	6.87	1.61	1.09	6.67	0.55
ground (b)	Milling 27 h		1.2	0.90	0.32	1.85	3.72	1.60	1.06	1.88	0.39
Z5U ε ₈₇₀₀ lot 1	Commercial	6.0	3.1	0.32	0.32	0.49	0.82	1.42	1.03	1.68	0.255
Z5U ε ₈₇₀₀ lot 2	Commercial		2.9	0.34	0-31	0.49	0.82	1.42	1.02	1.37	0.195
X7R ε_{3000} lot 1	Commercial	6.0	5.0	0.20	0.30	0.46	0.76	-	-	-	0.225
X7R ε ₃₀₀₀ lot 2	Commercial		4.2	0.24	0.31	0.49	0.79	1.33	1.02	1.40	0.205

bulkiness and roundness are high for rugged ones. The roundness is also sensitive to elongation.

SEM photographs indicate that the powders prepared via the mixed oxide route are highly aggregated and rugged. Figure 1(a) shows the coarse grain structure of BaTiO₃ I, Fig. 1(b) one of the ground powders consisting predominantly of





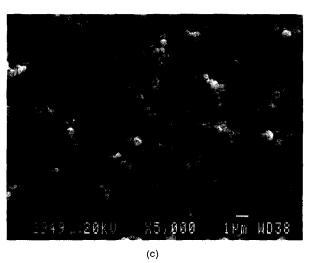


Fig. 1. SEM photographs of coarse [Fig. 1(a)] and ground [Fig. 1(b)] BaTiO₃ I powders and of commercial BaTiO₃ IV powder [Fig. 1(c)].

primary particles and Fig. 1(c) a commercial powder BaTiO₃ IV for comparison.

3 Dielectric Measurements

The capacitance $C = \varepsilon_0 \varepsilon_r A/L$ of the powders was measured in air at room temperature as a function of frequency using a capacitor arrangement with a guard ring at the displaceable electrode, as described elsewhere. 1 Analysis of the complex dielectric constant $\varepsilon^* = \varepsilon' + i\varepsilon''$ makes it possible to check the frequency independent range of values of the real part $\varepsilon' = \varepsilon_r$. Figures 2(a) and (b) show the data for BaTiO₃ and Ba_{1-x}Sr_xTiO₃ powders at a given volume fraction v2. In order to obtain data at several volume fractions in the frequency independent range, different degrees of compression caused by the movable electrode have to be applied. In the case of ferroelectric powders, it is necessary after compression to reduce the pressure to zero, in order to avoid piezoelectric contributions to the measuring signal. The different effect of densification pressure increasing the volume fraction for ferroelectric Ba_{0.705}Sr_{0.295}TiO₃ and paraelelectric Ba_{0.437}Sr_{0.563}TiO₃ is demonstrated by the data of Fig. 3.

4 The Frequency Independent Real Part of the Permittivity of the Powders

The analysis follows the model which has been proposed in the preceding paper.¹ Starting from the equations of Bruggeman⁶ with ε'_{sph} for spheres with permittivity ε_2 and volume fraction v_2 randomly embedded in a continuous medium with ε_1 (e.g. air) given by:

$$1 - v_2 = \frac{\varepsilon_2 - \varepsilon'_{\text{sph}}}{\varepsilon_2 - \varepsilon_1} 3 \sqrt{\frac{\varepsilon_1}{\varepsilon'_{\text{sph}}}}$$
 (1)

and with ε'_{lam} for lamella-like particles also randomly distributed in a continuous medium given by:

$$\varepsilon_{\text{lam}}' = \varepsilon_2 \frac{3\varepsilon_2 + 2v_2(\varepsilon_2 - \varepsilon_1)}{3\varepsilon_2 - v_2(\varepsilon_2 - \varepsilon_1)} \tag{2}$$

The form parameter u as defined by Wiener⁵ according to

$$\frac{1}{\varepsilon' + u} = \frac{v_1}{\varepsilon_1 + u} + \frac{v_2}{\varepsilon_2 + u}$$
and
$$\frac{\varepsilon'}{\varepsilon' + u} = \frac{v_1 \varepsilon_1}{\varepsilon_1 + u} + \frac{v_2 \varepsilon_2}{\varepsilon_2 + u}$$
with $0 < u < \infty$ (3)

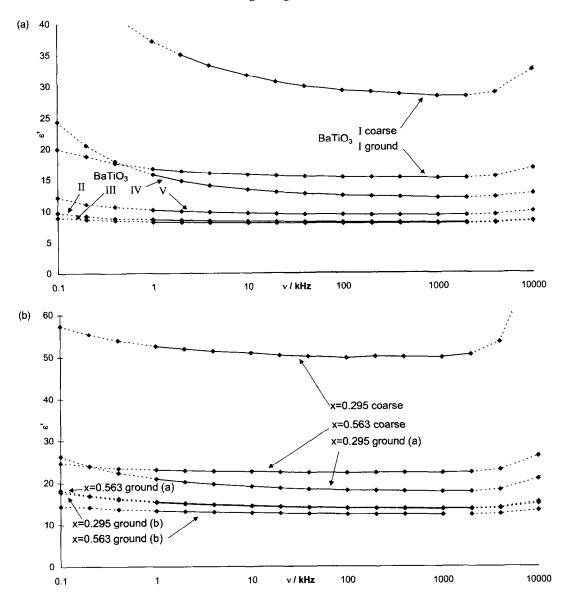


Fig. 2. Measured powder dielectric constant ε' for BaTiO₃ [Fig. 2(a)] powders (I) coarse ($v_2 = 0.196$), ground ($v_2 = 0.421$), and the commercial powders (II) ($v_2 = 0.280$), (III) ($v_2 = 0.312$), (IV) ($v_2 = 0.409$) and (V) ($v_2 = 0.359$) and for Ba_{1-x}Sr_xTiO₃ [Fig. 2(b)] powders with x = 0.295 coarse ($v_2 = 0.266$), ground (a) ($v_2 = 0.356$), ground (b) ($v_2 = 0.414$) and x = 0.563 coarse ($v_2 = 0.282$), ground (a) ($v_2 = 0.371$) and ground (b) ($v_2 = 0.381$).

together with the fundamental relations for parallel and series interconnection in the Effective Medium Theory (EMT)

$$\varepsilon_p = \mathbf{v}_1 \varepsilon_1 + \mathbf{v}_2 \varepsilon_2 \tag{4a}$$

and

$$\varepsilon_s = \frac{1}{\frac{v_1}{\varepsilon_1} + \frac{v_2}{\varepsilon_2}} \tag{4b}$$

lead to the equation

$$\varepsilon' = \frac{\varepsilon_s (\varepsilon_p u + \varepsilon_1 \varepsilon_2)}{\varepsilon_1 \varepsilon_2 + u \varepsilon_s} \tag{5}$$

The combination of (5) with (1) or (2) expresses $\varepsilon'_{\rm sph}$ and $\varepsilon'_{\rm lam}$ as a function of the parameter $u_{\rm sph}$ or

 u_{lam} , respectively. For modelling a real powder consisting of polyhedral particles of varying shape, geometric averaging of u_{sph} and u_{lam} can be justified; the mixing parameter β for continuous adaption has accordingly been defined,

$$u_{\text{pol}} = 2\sqrt{u_{\text{lam}}^{\beta} u_{\text{sph}}^{2-\beta}} \text{ with } 0 \le \beta \le 2$$
 (6)

thus leading to

$$\varepsilon_{\text{pol}}' = \frac{\varepsilon_S (\varepsilon_p u_{\text{pol}} + \varepsilon_1 \varepsilon_2)}{\varepsilon_1 \varepsilon_2 + u_{\text{pol}} \varepsilon_S} \tag{7}$$

Knowing ε_1 of air to be approximately 1.00, the measuring data ε'_{pol} are only determined for pairs of ε_2 and β , i.e. it is necessary to fix ε_2 in order to

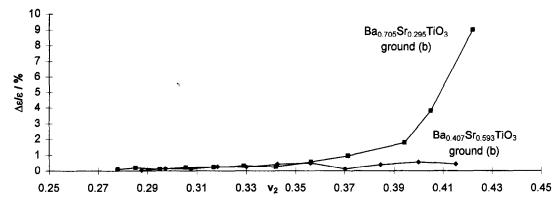


Fig. 3. Effect of densification pressure on the relative enhancement of the dielectric constant ε' of Ba,SrTiO₃ powders.

determine the form factor β or inversely. However, theoretical reasons suggest limits for the validity of EMT.¹⁵ Particles of inhomogeneous dielectrics must have sizes, which lie between certain limits if the EMT's are to be applicable.

4.1 Dielectric constant of BaTiO₃ in the powder state and the form factor β

At 2MHz the dielectric constant of ferroelectric BaTiO₃ results from electronic polarization and phonon lattice dynamics (ionic polarization) and domain wall motion. In bulk BaTiO₃ ceramics, the grain size has a strong effect on the permittivity at room temperature for grain sizes below about $10 \,\mu\text{m}$. The permittivity rises as the grain size decreases down to about $0.7 \mu m$, which has been explained by the increasing contributions of domain wall motion. The domain thickness becomes diminished and subsequently the domain wall density is growing per unit of volume.¹² Below the maximum shown in Fig. 4, the permittivity sharply drops and at the same time the tetragonality and remanent polarization are reduced. 12-14,17 Below about 0.8 µm single domain particles prevail.¹² Obviously, the missing contributions from domain wall motion act as one of the reasons for the decrease in the permittivity in this lower range of the grain size. Other arguments are based on

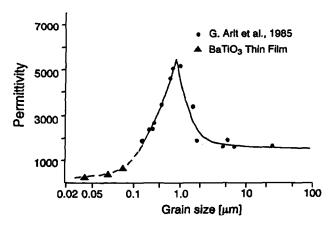


Fig. 4. Grain size dependence of the relative dielectric permittivity of BaTiO₃ ceramics according to Waser. 14

thermodynamics, ¹³ space charge layers, ¹⁸ and the transition to a superparaelectric state. ¹⁴

The BaTiO₃ powders of Table 1 have grain sizes between 0.3 and $0.6\,\mu\mathrm{m}$; therefore domain wall contributions are suggested to be negligible. Assuming that in this range of small grain sizes the permittivity of the powder is predominantly determined by electronic and ionic polarization as in the bulk ceramic state, the values of the dielectric constant have been taken for this band of grain sizes from the falling branch of the curve of Fig. 4 in order to calculate the form factor β .

Figure 5 shows data for ε' as a function of the volume fraction v_2 for the ground powder BaTiO₃I together with the fitting curves obtained by varying the permittivity ε_2 in relation with the form factor β . Extrapolation to $v_2 = 1$ provides $\varepsilon' = \varepsilon_2$. When ε_2 is increased in the range of 2000–4000 according to the grain size 0.3–0.6 μ m, β is found to have values between 0.38 and 0.41. Of course, β is more sensitive to shape than to size. Furthermore, it is clearly indicated that at high particle dielectric constants, β is relatively less dependent on ε_2 . This means that the grain size dependence of the dielectric constant has only a secondary effect on the derived form factor β .

Figure 6 shows the experimental data for the powder permittivity ε' as a function of v_2 for the five different BaTiO₃ samples together with the curves which result from model calculation using the grain size dependent ε_2 values according to the correlation of Fig. 4. The d_{50} values of Table 1 have been applied for estimating ε_2 . In accordance with electronmicroscopic investigations, there is only a small variation in the shape of the commercially available BaTiO₃ samples and therefore the form factors β agree within small limits. The ground state of the BaTiO₃ I closely relates to the other highly dispersed powders which are commercially available. On the other hand, presumably because of the high degree of roundness (Table 1), the coarse state of powder I consisting of relatively large rugged particles [Fig. 1(a)] formed as a result

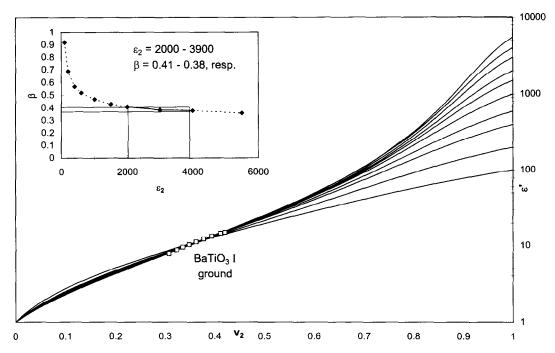


Fig. 5. Powder dielectric constant ε' of BaTiO₃ (I) measured at 2 MHz and 1 V/cm in dependence on the volume fraction v_2 . The dependence of β upon variation of ε_2 is shown in the insert.

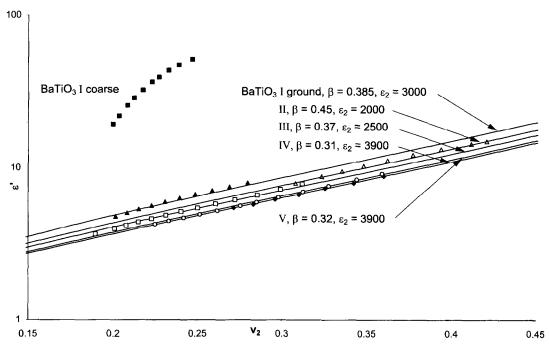


Fig. 6. Permittivity ε' in dependence on the volume fraction v_2 for BaTiO₃ I (coarse, \blacksquare) and (ground, \triangle), II (\triangle), III (\square), IV (\spadesuit) and V (\bigcirc) powders according to Table 1. Solid lines are calculated ε'_{pol} according to (7).

of mixed oxide preparation could not be included in the modelling. Indeed, the particles are extraordinary large and therefore out of the limits of validity of the theoretical prerequisites.

4.2 Dielectric constant of $Ba_{1-x}Sr_xTiO_3$ powders with x = 0.295 and 0.563

The formation of single phase solid solutions for $Ba_{1-x}Sr_xTiO_3$ is quite well documented¹⁹ resulting in a shift of the ferroelectric Curie temperature T_C to lower temperatures.²⁰ Figure 7 shows

the temperature dependence of ε_2 for the two ceramic samples Ba_{0.705}Sr_{0.295}TiO₃ and Ba_{0.437}Sr_{0.563}TiO₃. Curie Weiss Law analysis provides the values $T_C = 36^{\circ}\text{C}$ ($\theta = 37^{\circ}\text{C}$) and $T_C < -55^{\circ}\text{C}$ ($\theta = -48^{\circ}\text{C}$), respectively. They are in satisfactory accordance with data from the literature. However, the grain size dependence of ε_2 in the ferroelectric state of this solid solution series seems to be unknown up to now.

Neglecting this aspect and relying on the conclusions deduced for BaTiO₃ powders, the room

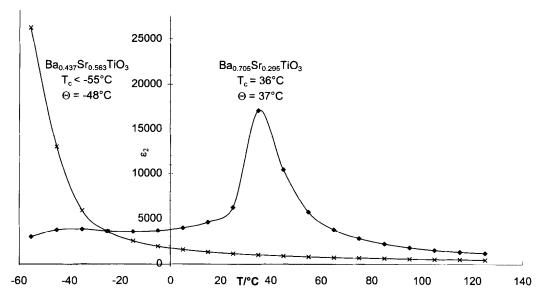


Fig. 7. Temperature dependence of the dielectric constant of Ba_{0.705}Sr_{0.295}TiO₃ and Ba_{0.437}Sr_{0.563}TiO₃ ceramics.

temperature dielectric constant $\varepsilon_2 = 5560$ of the ferroelectric ceramic sample Ba_{0.705}Sr_{0.295}TiO₃ was applied as a trial value for calculating the form factor β of the powder. As an example, for the ground state (b) mentioned in Table 1, $\beta = 0.31$ is found. Such a proceeding appears justified, because if ε_2 were to change in the range 5560 ± 3000 , β would only vary between 0.29 and 0.34.

The results shown in Fig. 8 clearly indicate the effect of grinding. Again the coarse powder from the mixed oxide preparation route cannot be interpreted. On the other hand, continuous grinding leading to the powder states (a) with $d_{50} = 3.7$ and (b) with $d_{50} = 1.7 \,\mu\text{m}$ is characterized by

decreasing β values 0.48 and 0.31, respectively. Of course, as a result of grinding, the spherical contribution to the polyhedral powder state is expected to increase. Similar results were obtained for the Ba_{0.437}Sr_{0.563}TiO₃ powders. Because of the paraelectric state of Ba_{0.437}Sr_{0.563}TiO₃, the interpretation is less questionable. Taking $\varepsilon_2 = 1170$ from the diagram of Fig. 7, the β values 0.55 and 0.39 were found corresponding to 2.8 μ m and 1.9 μ m for the two ground powders (a) and (b).

It has to be noticed, in spite of the essential difference in the permittivity of ferroelectric $Ba_{0.705}Sr_{0.295}TiO_3$ and paraelectric $Ba_{0.437}Sr_{0.563}$ TiO₃ that the ground powder states (b) showing

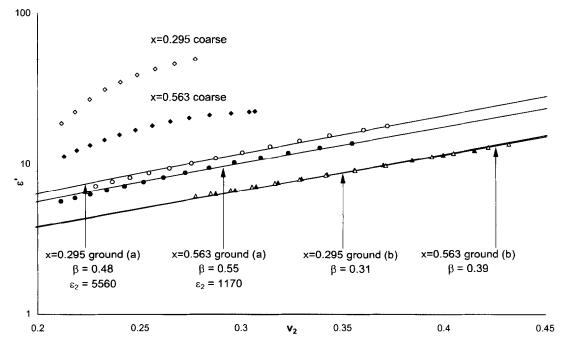


Fig. 8. Permittivity ε' in dependence on the volume fraction v_2 for powders $Ba_{1-x}Sr_xTiO_3$ in the coarse state and after grinding according to Table 1 with x=0.295 (coarse, \diamondsuit), (a,\bigcirc) and (b,\triangle) and (a,\bigcirc) and (a,\bigcirc)

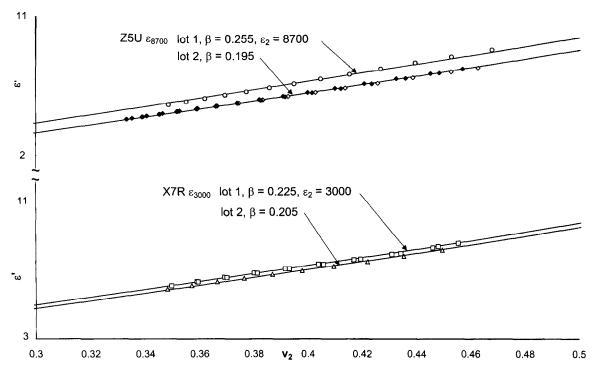


Fig. 9. Permittivity ε' in dependence on the volume fraction v_2 for different lots of Z5U $(\bigcirc, \diamondsuit, \spadesuit)$ and X7R (\square, \triangle) capacitor mass powders according to Table 1. Solid lines are calculated ε'_{pol} according to (7).

comparable average grain sizes (Table 1) tend to provide less different values $\beta = 0.31$ or 0.39, respectively.

4.3 Ferroelectric capacitor raw materials

Figure 9 shows the experimental results for two lots of commercial Z5U and X7R powder materials leading in the ceramic state at room temperature to the permittivity of $\varepsilon_2 = 8700$ and $\varepsilon_2 = 3000$, respectively. Fitting of the experimental data of ε' is possible with these values, although because of the complex composition of the raw materials the application of the ceramic permittivity to powder particles is to be seen as questionable. On the other

hand, the resolution to detect small differences between lots is found to be relatively high. Therefore β is more an empirical factor suitable for controlling the constancy of the morphological state of charges. The two lots of the X7R raw material indicate the reproducibility that has been achieved. The measurement of the Z5U powder lot 2 has been carried out three times in order to check the reproducibility. The difference between lot 1 and lot 2 results in part from the type of the vitreous sintering aid in the powder. It is interesting to note that this small difference may be detected by the difference of the β value of 0·195 (lot 2) or 0·255 (lot 1), respectively.

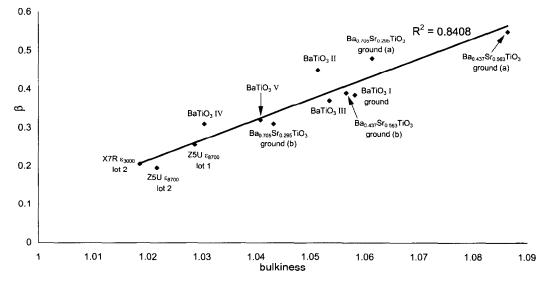


Fig. 10. Correlation of the form factor β to the particle bulkiness. Pearson's correlation coefficient R is given.

5 Comparision of the Form Factor β with Anisometry, Bulkiness and Roundness

The morphological significance of the form factor β is demonstrated in Fig. 10 showing the dependence of β on the bulkiness of the investigated powders. In accordance with previous studies on dielectric powders, 1 a good correlation between the bulkiness of the grains and the observed β value is found also for ferroelectric particles. However, a high bulkiness of the grains, together with the high roundness listed in Table 1 seems to lead to a dielectric behaviour of the powders that cannot be interpreted in the frame of dielectric models based on the assumptions of the EMT's. Therefore the form factors β for the coarse BaTiO₃ I and the coarse $Ba_{0.437}Sr_{0.563}TiO_3$ and $Ba_{0.705}Sr_{0.295}TiO_3$ powders have not been used for calculation of linear regression. Moreover, β shows a weak increasing correlation with anisometry which may be due to a dominating influence of bulkiness. Consequently for ferroelectric particles with a low bulkiness the β value seems to be a very sensitive measure of particle shape.

6 Summary and Conclusions

Ferroelectric powders can be described in terms of a particle shape dependent form factor β deduced from dielectric measurements using models of the EMT which have been developed for inhomogeneous dielectrics. The permittivity of the solid particles ε_2 and of the continuous medium ε_1 have to be known, while the volume fraction v₂ and the dielectric constant ε' of the mixture are measured. At high values of the permittivity of the ferroelectric particles, the form factor β is found to be independent of ε_2 . On the other hand, β becomes the more sensitive to shape as the permittivity rises. Below about $0.7 \,\mu\text{m}$, i.e. in the range of single domain grains, the dielectric constants ε_2 of the ferroelectric ceramics and their dependence on temperature and grain size are assumed to be also valid for powders. A comparision of the form factor β with the shape parameters (anisometry, bulkiness and roundness) shows a dominating influence of bulkiness. β increases with increasing anisometry, bulkiness and roundness. However, the method seems to be unsuitable for powders, whose grains sizes lie outside the theoretical validity of EMT's. The possibility of using these methods to monitor the morphologies of X7R and Z5U

powders is elucidated. They show low roundness and good approximation to spheres embedded in air. Relatively small differences in the morphological state of ferroelectric capacitor raw materials have been successfully detected.

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