

Journal of the European Ceramic Society 21 (2001) 1681–1684

www.elsevier.com/locate/jeurceramsoc

# Effects of acceptor and donor additives on the properties of MgTiO<sub>3</sub> ceramics sintered under reducing atmosphere

C. Vigreux, B. Deneuve, J. El Fallah, J.M. Haussonne\*

Laboratoire Universitaire des Sciences Appliquées de Cherbourg, Université de Caen Basse Normandie, LUSAC, Site Universitaire de Cherbourg, BP 78, 50130 Octeville, France

Received 4 September 2000; accepted 5 October 2000

#### Abstract

Sintered in reducing atmospheres, MgTiO<sub>3</sub> ceramics are known to own ionised oxygen vacancies and conduction electrons which give rise to very poor insulation resistance. Prompted by the example of BaTiO<sub>3</sub> ceramics where acceptor doping agents can trap the conduction electrons and donor doping agents compensate the acceptors in place of oxygen vacancies (Hagemann, H. and Hennings, D., Reversible weight change of acceptor-doped BaTiO<sub>3</sub>. J. Am. Ceram. Soc., 1981, 64, 590-594), we have investigated on the effects of Mn acceptor and W donor doping agents on the properties of MgTiO<sub>3</sub> ceramics sintered under a reducing atmosphere of moist  $N_2-1\%H_2$ . Resistivities higher than  $10^5$  M $\Omega$ .cm have effectively been obtained in these doped ceramics and dielectric constant of pure MgTiO<sub>3</sub> was proved not to be modified by the presence of doping agents. In order to study the possibility of employing these materials in the production of type-I ceramic multilayer capacitors with base-metal electrodes, co-sintering tests with nickel were investigated. © 2001 Elsevier Science Ltd. All rights reserved.

Keywords: Base metal electrodes multilayer capacitors; Capacitors; Dielectric properties

### 1. Introduction

Magnesium titanate based dielectric materials are classically used for producing type-I ceramic multilayer capacitors. Being sintered at a temperature higher than 1350°C, precious metal electrodes are to be used. In order to reduce the ceramic multilayer capacitors cost, one way is to replace the expensive noble metals by cheaper ones as nickel<sup>2</sup> or copper. This solution implies to sinter the capacitors in atmospheres such as  $N_2/H_2$  or  $CO/CO_2$ . But in such reducing atmospheres MgTiO3 as BaTiO3 based dielectric ceramics are known to get large numbers of ionised oxygen vacancies and conduction electrons, giving rise to very poor insulation resistance. BaTiO<sub>3</sub> ceramics containing Mn acceptors and Mo, W or Nb donors were proved to show a high insulation resistance when fired in reducing atmosphere. 1 It has been shown that the conduction electrons can be trapped by incorporation of Mn<sup>2+</sup> acceptor ions in the perovskite lattice and that the acceptors can be compensated in place of oxygen vacancies by donor ions like W6+. Then it is assumed

E-mail address: jmhaussonne.lusac@chbg.unicaen.fr (C. Vigreux).

that donors and acceptors form complexes which mobility in the electric field is extremely low:<sup>3</sup>

Prompted by these results, we have investigated on the effects of Mn acceptor and W donor doping agents on the properties of ilmenite MgTiO<sub>3</sub> ceramics sintered under a reducing atmosphere of moist N<sub>2</sub>-1%H<sub>2</sub>.

The experimental procedure followed is first described. The efficiency of the donor and acceptor doping agents on the electric properties of MgTiO<sub>3</sub> sintered under a reducing atmosphere was demonstrated and the possibility of co-sintering such ceramics with nickel investigated.

#### 2. Experimental procedure

The studied ceramic materials have the following general composition:

$$Mg(Ti_{1-x}Mn_{x/2}W_{x/2})O_{3-z}$$

with  $0 \le x \le 0.02$  and z close to zero.

0955-2219/01/\$ - see front matter © 2001 Elsevier Science Ltd. All rights reserved.

PII: S0955-2219(01)00092-9

<sup>\*</sup> Corresponding author. Tel.: 0033-02-3301-4214; fax: 0033-02-

MgO, TiO<sub>2</sub> and dopant oxides MnO<sub>2</sub> and WO<sub>3</sub> were weighed and milled together for 1 h in alcohol, using 2 mm in diameter zircon balls. Three compositions were prepared corresponding to x=0.005; 0.01 and 0.02 and labelled  $M_1$ ,  $M_2$  and  $M_3$ , respectively. Calcination temperatures were estimated thanks to dilatometric measurements, which were performed in air on a TMA92 SETARAM dilatometer, and using a heating rate of 300°C h<sup>-1</sup> and a cooling rate of 900°C h<sup>-1</sup>. The dilatometric curves are shown Fig. 1. The phase formation begins around 800°C and finishes between 1100 and 1200°C, temperatures for which densification takes place. Three series of  $M_i$ , i=1, 2 or 3, so-called  $M_i$ (900),

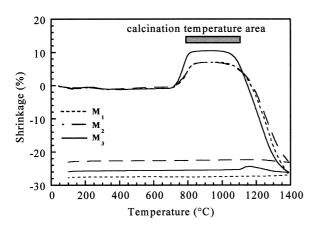


Fig. 1. Dilatometric curves before calcination for samples  $M_1$ ,  $M_2$  and  $M_3$ .

 $M_i(1000)$  and  $M_i(1200)$  were finally prepared, after one-hour calcination at 900, 1000 and 1200°C, respectively.

The calcined materials were systematically analysed by X-ray diffraction. The diffraction patterns were collected using a SIEMENS 5005 diffractometer with  $CuK_{\alpha}$  ( $\lambda = 1.5405$  Å), in the range 15–75°C (2 $\theta$ ). All of them show a single phase with ilmenite structure (space group R(-3); structure shown Fig. 2). As an example, Fig. 3 shows the diffraction pattern obtained for  $M_2(1000)$ . After calcination, new dilatometric measurements were performed on the different samples in order to estimate the firing temperature. The shrinkage curves recorded

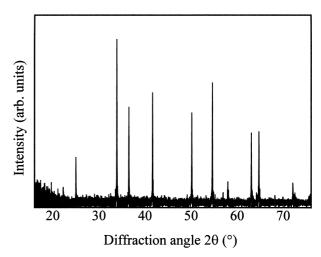


Fig. 3. X-ray diffraction pattern of  $M_2(1000)$ . The diffraction peaks correspond to a single phase with ilmenite structure.

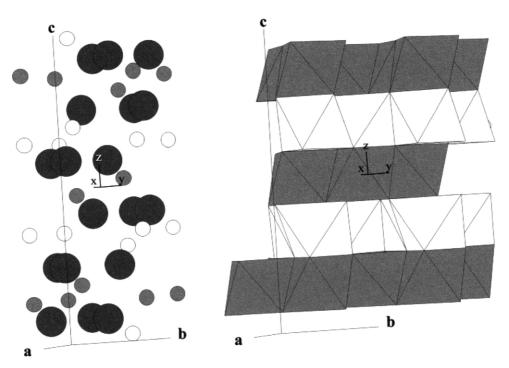


Fig. 2. Ilmenite structure. On the left are represented the different atoms: (i) the oxygen atoms in dark grey; (ii) the titanium atoms in clear grey and (iii) the magnesium atoms in white. The octahedral coordination polyhedrons of titanium (in clear grey) and magnesium (in white) are represented on the right of the figure.

on the samples  $M_1(900)$ ,  $M_1(1000)$  and  $M_1(1200)$  are reported in Fig. 4. Shrinkage value decreases when the calcination temperature increases, while the temperatures

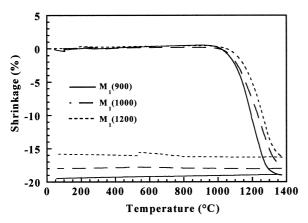


Fig. 4. Dilatometric curves of  $M_1$  after calcination at 900, 1000 or  $1200^{\circ}\mathrm{C}$ .

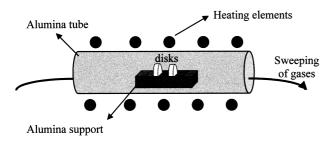


Fig. 5. Positioning of the disks in the tubular furnace.

corresponding to the beginning of the shrinkage are very close to each other, i.e.  $\approx 1050^{\circ}\text{C}$ . For the three samples, the rate of shrinkage slows down beyond approximately  $1200^{\circ}\text{C}$  and sintering process seems finished at  $1350^{\circ}\text{C}$ . The same behaviour was observed for  $M_2$  and  $M_3$ , so that the firing temperature was chosen at  $1350^{\circ}\text{C}$ .

All the calcined mixtures were pressed at 2000 kg cm<sup>-2</sup> to obtain disks 15 mm in diameter and around 2 mm in thickness. The disks were then fired in a reducing atmosphere of moist N<sub>2</sub>-1%H<sub>2</sub>, with heating and cooling rates of 150°C h<sup>-1</sup> and different soak-times ranging from 1 to 5 h at 1350°C. The position of the disks in the tubular furnace is described in Fig. 5. X-ray diffraction analysis of the fired disks confirmed all the ceramics to be single phased with ilmenite structure.

The microstructures of the sintered samples were observed with a scanning electron microscope HITACHI S-2460-N. The dielectric constant  $\varepsilon_{\rm r}$  and the dielectric losses  $\tan\delta$  of the disks were measured at 1 kHz and the electric properties were recorded by applying a 1 kV DC bias at room temperature. The density of the samples was estimated using a helium Micromeritics Accu-Pyc1330 pycnometer.

#### 3. Results

The first evidence is the efficiency of the doping agents Mn<sup>4+</sup> and W<sup>6+</sup> on the insulation resistance of MgTiO<sub>3</sub> sintered in a reducing atmosphere. Indeed, high insulation

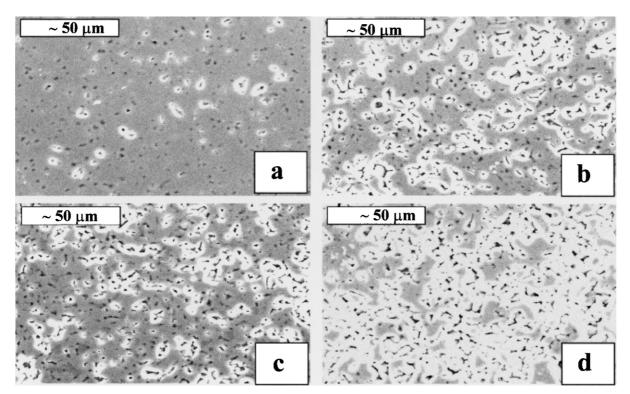


Fig. 6. Microstructures observed for (a) MgTiO<sub>3</sub>, (b)  $M_1(1000)$ , (c)  $M_2(1000)$  and (d)  $M_3(1000)$ .

Table 1 Samples characteristics after 1 h sintering at 1350°C under a reducing atmosphere of moist  $N_{2}$ -1% $H_{2}$ 

	Shrinkage (%)	Percentage of theoretical density	Electric resistivity (MΩ cm)	Tanδ (%)	Dielectric constant $\varepsilon_r$
MgTiO <sub>3</sub>	18.5	96.7	Conductor	387	4.87e + 05
$M_1(1000)$	18.3	93.3	1.85e + 06	0.3	46.9
$M_2(1000)$	17.8	93.5	1.52e + 05	3.4	32.9
$M_3(1000)$	15.3	94.7	3.44e + 05	0.7	44.8

resistances (higher than  $10^3~\mathrm{M}\Omega$  cm) were obtained with the different doped ceramics quoted before, whatever the x value (0.005; 0.01 or 0.02) and the calcination temperature (900, 1000 or 1200°C) are. On the other hand, pure MgTiO<sub>3</sub> was confirmed to be highly conductive.

The second evidence is that the presence of doping agents in MgTiO<sub>3</sub> gives rise to shrinkages and densities values smaller than the one observed with pure manganese titanate, related to a more important porosity. These parameters are still to be improved. As an illustration, microstructures of MgTiO<sub>3</sub>,  $M_1(1000)$ ,  $M_2(1000)$  and  $M_3(1000)$  are compared in Fig. 6.

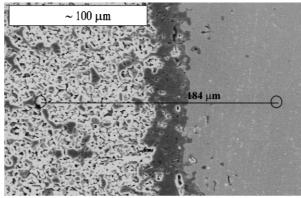
Then, it was shown that doping agents did not modify the dielectric constant of pure MgTiO<sub>3</sub>, so that doped ceramics can be used as type-I capacitors. Low dielectric losses could finally be obtained, typically lower than 1%.

The best results after 1 h sintering in a reducing atmosphere at  $1350^{\circ}$ C were obtained for the samples calcined at  $1000^{\circ}$ C during 1 h. They are summarised Table 1. The percentage of theoretical density was obtained making the ratio of the measured density on the theoretical density of MgTiO<sub>3</sub> (d=3,894 g cm<sup>-3</sup>). We can see in Table 1 that remarkable high resistivities could be obtained for these doped samples, since higher than  $10^{5}$  M $\Omega$  cm.

These materials are to be co-sintered with base-metal electrodes in multilayer capacitors. So, co-sintering tests of the ceramics with nickel were then carried out. Disks made up of a layer of nickel intercalated between two layers of ceramics were elaborated and then sintered at  $1350^{\circ}$ C during one hour in a reducing atmosphere. Fig. 7 represents the interface between nickel (on the right) and the  $M_2(1000)$  ceramic (on the left) after sintering. One can see that nickel diffused in the ceramic into an approximately  $25 \, \mu m$  zone, which makes impossible the realisation of multilayer capacitors with such ceramics and metal.

#### 4. Conclusion

Like in BaTiO<sub>3</sub>, <sup>1</sup> the presence of Mn acceptor and W donor doping agents in MgTiO<sub>3</sub> allowed to obtain ceramics with high resistivities (higher than  $10^5$  M $\Omega$  cm) after sintering under a reducing atmosphere of moist N<sub>2</sub>-1%H<sub>2</sub>. Furthermore, it was shown that introducing these doping agents in the titanate magnesium ilmenite



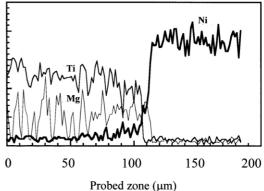


Fig. 7. Photograph and X-ray analysis of the transverse section of the  $M_2(1000)$  ceramic co-sintered with nickel.

structure did not modify the dielectric constant. But it was proved that these doped ceramics could not be employed in the production of type-I ceramic multilayer capacitors with nickel electrodes, since this metal diffuses in the ceramic during sintering.

Within the framework of our collaboration with the TEKELEC-TEMEX society, we are actually studying the possibility of realising such type-I ceramic multilayer capacitors with copper electrodes. Using such metal requires to work at lower temperature (<1000°C), so that the role of different compounds as sintering aids is being investigated.

## Acknowledgements

The authors are indebted to the TEKELEC-TEMEX society for collaborating with and allowing them to publish in this review.

### References

- Hagemann, H.-J. and Hennings, D., Reversible weight change of acceptor-doped BaTiO<sub>3</sub>. J. Am. Ceram. Soc., 1981, 64, 590-594.
- Hennings, D., Dielectric ceramic materials for multilayer capacitors with Ni electrodes. In: *Proceedings K1:IL06 CIMTEC*, ed. P. Vincenzini, Florence, 1998, pp. 88-91.
- Albertsen, K., Hennings, D. and Steigelmann, O., Donor-acceptor charge complex formation in barium titanate ceramics: role of firing atmosphere. *J. Electroceram.*, 1998, 2, 193–198.