

# Ceramic varistors based on ZnO–SnO<sub>2</sub>

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## Abstract

Model ceramic varistor formulations based on 98% ZnO–SnO<sub>2</sub> (plus 2% oxides of Bi, Co and Mn) were prepared by conventional powder processing routes; specimens were sintered at 1150–1275 °C. The product density increased with SnO<sub>2</sub> content, but decreased with increase in sintering temperature. The microstructures contained ZnO, Zn<sub>2</sub>SnO<sub>4</sub> spinel and very small bismuth rich phases; with increasing Sn content the spinel became the dominant phase. From I–V measurements, the non-linear coefficients were found to be in the range 24–32. For all compositions, the  $\alpha$  values decreased with increasing sintering temperature; the maximum  $\alpha$  value was obtained with samples containing 20 mol% SnO<sub>2</sub> sintered at 1200 °C. Breakdown fields were in the range 3000–11000 V/cm and increased with increasing tin content. Leakage currents were in the range 1–11  $\mu$ A. Tin substitution for Zn appears to cause a strong donor effect.

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**Keywords:** Grain boundary; Non-linear coefficient; Varistors; ZnO

## 1. Introduction

Varistors<sup>1</sup> are materials with unique non-linear, current–voltage characteristics that allow them to be employed in voltage surge protection applications. Today, the majority of varistors produced commercially are based on ZnO with minor additions<sup>2</sup> of typically Bi<sub>2</sub>O<sub>3</sub>, Sb<sub>2</sub>O<sub>3</sub>, CoO, MnO<sub>2</sub> and Cr<sub>2</sub>O<sub>3</sub>. The minor constituents segregate to the grain boundaries to help develop barriers to electrical conduction, and also form spinel phases Zn<sub>7</sub>Sb<sub>2</sub>O<sub>12</sub> at triple points and grain boundaries. The double, back-to-back Schottky barriers at the grain boundaries are the primary reason for the non-linear current–voltage (I–V) behaviour; the latter may conveniently be described in terms of a non-linear coefficient  $\alpha$  (which is defined by  $I = KV^\alpha$ , where  $K$  is a constant of proportionality). For a good quality ZnO based varistor, the non-linear coefficient is typically 40–50.

In recent years there has been growing interest in varistors of alternative formulations and these have included base compositions of tungsten oxide,<sup>3</sup> SrTiO<sub>3</sub>,<sup>4</sup> and TiO<sub>2</sub>.<sup>5–7</sup> Such varistors tend to exhibit compara-

tively small  $\alpha$  values (generally in the range 2–12) which limits their commercial exploitation, but SnO<sub>2</sub> based varistors<sup>8,9</sup> have emerged as one of the few novel systems where  $\alpha$  values in excess of 40 have been achieved. Additions of CoO to SnO<sub>2</sub> are reported to yield high-density products<sup>8</sup> but such varistors are not as easy to fabricate as the classical ZnO based products. Recently, Bernik and Daneu<sup>10</sup> prepared ZnO varistors (of commercial formulation) where the Sb<sub>2</sub>O<sub>3</sub> component (total 1 mol%) was systematically replaced by SnO<sub>2</sub>. They observed a significant increase in leakage current and increase in donor density with SnO<sub>2</sub> doping. In the present study we have examined microstructure development and the electrical characteristics of a model system based on ZnO (plus Bi<sub>2</sub>O<sub>3</sub>, Mn<sub>2</sub>O<sub>3</sub>, and Co<sub>2</sub>O<sub>3</sub>) where the ZnO was systematically replaced by SnO<sub>2</sub> (up to 20 mol%).

## 2. Experimental

Samples with nominal compositions (98– $x$ )ZnO– $x$ SnO<sub>2</sub> (plus 0.5% Bi<sub>2</sub>O<sub>3</sub>, 0.5% Mn<sub>2</sub>O<sub>3</sub> and 1.0% Co<sub>2</sub>O<sub>3</sub>; all in mol%) where  $x = 0, 5.0, 10.0, 15.0, 17.5$  and 20 mol%, were produced by the mixed oxide route using powders of 99.0–99.9% purity. For each composition, the powders were mixed in appropriate

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proportions and milled with propan-2-ol, using zirconia media for periods up to 24 h. After drying, the powders were pressed into discs (12 mm diameter, 2 mm thick) and fired at temperatures in the range 1150–1275 °C for 1 h. The heating and cooling rate was 4 °C/min.

Specimen densities were determined from weight and dimension measurements. The phases present were identified by X-ray diffraction (XRD) techniques (Philips X'Pert system) using Cu  $K_\alpha$  radiation. For microstructure analysis, sintered products were ground and polished down to 1  $\mu\text{m}$  diamond paste. Selected samples were etched either thermally (by heating rapidly to 200 °C below the sintering temperature and holding for 10 min) or by the use of dilute acid (5% hydrochloric acid for 5–10 s). Morphologies were examined optically and by scanning electron microscopy (SEM) using Philips 505 SEM equipped with energy dispersive spectrometer. Grain sizes were determined by the linear intercept method. Second phases were examined by transmission electron microscopy (TEM using a Philips EM430). Samples were mechanically thinned and ion beam thinned using an ION TEC 791 ion beam thinner.

For electrical measurements the specimen thickness was reduced to 0.5–1.0 mm and silver electrodes applied to both sides of the discs. Each specimen was fired at 550 °C for 60 min, using 4 °C/min heating and cooling rates. DC-measurements were undertaken using a Brandenburg power supply (475A) in conjunction with digital multimeters (Fluka 8012A). The non-linear coefficient  $\alpha$  was determined across the current density range 0.1–1.0  $\text{mAcm}^2$  via the relationship

$$\alpha = (\log J_2 - \log J_1) / (\log E_2 - \log E_1) \quad (1)$$

where  $J_2$  and  $J_1$  are current densities at applied electric fields  $E_2$  and  $E_1$ . The breakdown fields  $E_b$  were determined at current densities of 0.5  $\text{mA/cm}^2$ . The leakage current  $I_L$  was determined at 0.75  $E_b$ .

### 3. Results and discussion

Good quality, high-density products were obtained for sintering temperatures of 1200 °C or higher. Fig. 1 summarises densification behaviour of the specimens as a function of sintering temperature. With increasing substitution of ZnO by  $\text{SnO}_2$ , the density increases as a greater proportion of the heavier element is incorporated into the samples. However, independently of composition, the density tends to decrease with increasing sintering temperature, probably due to evaporation of ZnO at the higher temperatures.

Typical examples of the microstructures for samples sintered at 1250 °C are presented in Fig. 2. At low levels of  $\text{SnO}_2$  substitution (up to 5%) the microstructure is still dominated by large ZnO grains (about 10  $\mu\text{m}$  in size) exhibiting inversion boundaries with spinel phases

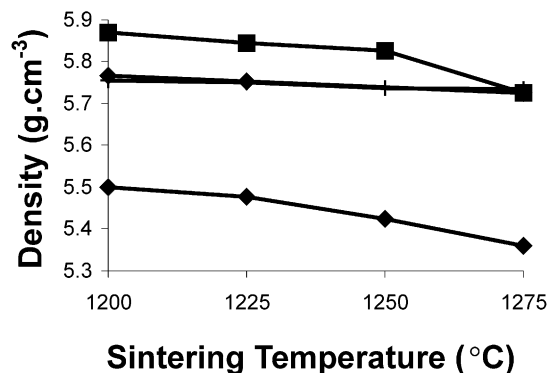


Fig. 1. Specimen density as a function of sintering temperature for samples prepared with different amounts of  $\text{SnO}_2$  (◆ 5%, ◑ 15%, ▲ 17.5%, ■ 20%).

at triple points and along ZnO grain boundaries. Although the sintering temperature had little effect on grain size, there were significant changes in microstructure and grain size with specimen composition. As  $\text{SnO}_2$  replaced the ZnO there was a rapid increase in the amount of the second phase and a distinct decrease in the size of the ZnO grains. Simple image analysis of the microstructures suggested that the second phase constituted approximately 35% of the samples when the  $\text{SnO}_2$  content was 5% (Fig. 2b), but the “second phase” fraction increased to almost 60% when the  $\text{SnO}_2$  content in the starting mixture was 20% (Fig. 2d). Clearly the tin rich second phase effectively pins the grain boundaries, limiting the growth of ZnO grains. It is also evident that whilst there are many small second phase grains, the second phase was able to grow rapidly to individual and aggregated grains of 20  $\mu\text{m}$  or larger.

The microstructures for these ZnO– $\text{SnO}_2$  ceramics are in marked contrast to those of (1) the  $\text{SnO}_2$  based ceramics,<sup>8,9</sup> where there is predominantly a monophase microstructure of relatively uniform size, and (2) the classical ZnO based systems, where the primary grains still dominate the Zn–Sb spinel second phases.<sup>11</sup> In the related study of Bernik and Daneu<sup>10</sup> the initial 1%  $\text{Sb}_2\text{O}_3$  was systematically replaced by  $\text{SnO}_2$ . They found very little change in grain size (all ZnO grains around 11.5  $\mu\text{m}$ ) and the second phase was small in size and volume, being primarily confined to triple points and grain boundaries. However, as the Sb/Sn ratio changed in the starting mixtures, there was a clear change in the nature and composition of the spinel phase ( $\text{Zn}_7\text{Sb}_2\text{O}_{12}$  and/or  $\text{Zn}_2\text{SnO}_4$ ) and occasionally traces of pyrochlore were detected. As the ‘model’ composition used in this study did not include  $\text{Sb}_2\text{O}_3$  the resulting phase development was much simpler than that reported by Bernik and Daneu.<sup>10</sup> EDS analysis of the second phases in Fig. 2 confirmed the presence of only one type of spinel,  $\text{Zn}_2\text{SnO}_4$ , and these contained small amounts of cobalt and manganese. Indeed XRD spectra for all the Sn-bearing samples revealed only two phases, ZnO and

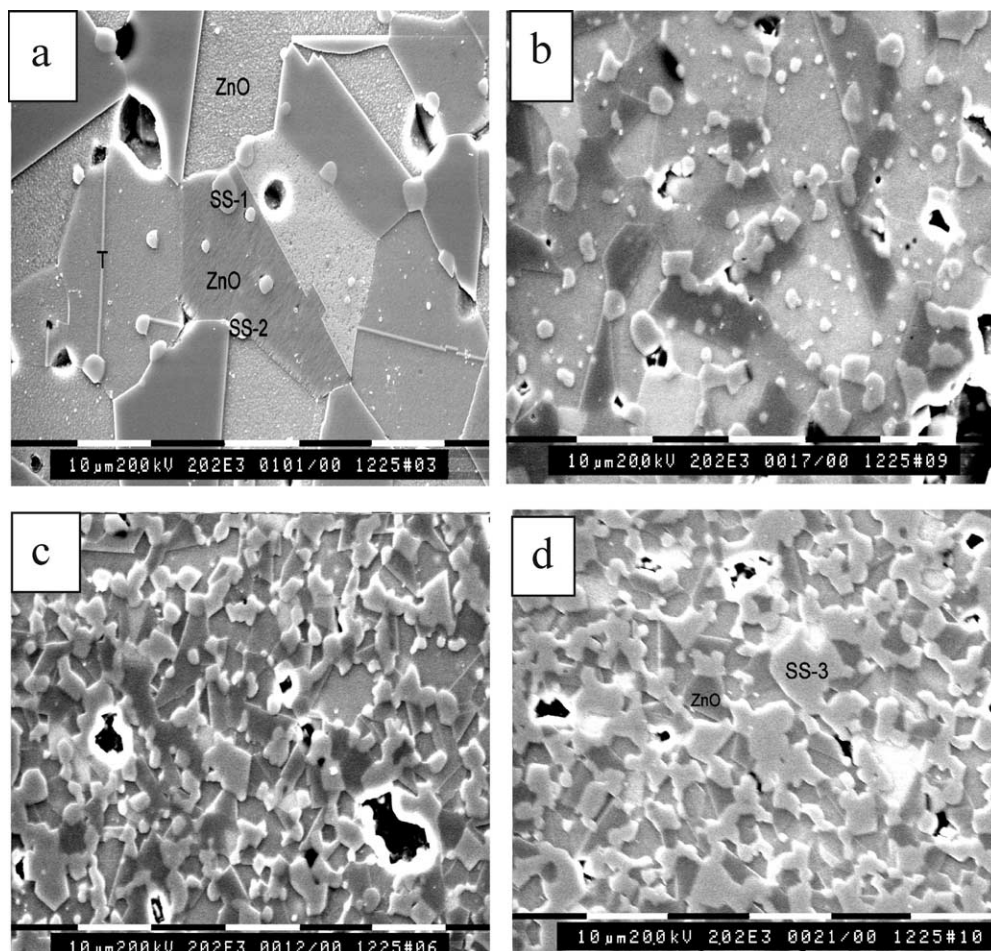


Fig. 2. SEM micrographs of specimens prepared with different amounts of  $\text{SnO}_2$ : (a) 1%, (b) 5%, (c) 15%, (d) 20%. SS denotes second phase grains.

$\text{Zn}_2\text{SnO}_4$ ; as the Sn content increased the intensity of the spinel peaks grew at the expense of the ZnO peaks. Limited TEM analysis confirmed the presence of  $\text{Bi}_2\text{O}_3$ -rich phases along the grain boundaries. These tended to be in the shape of laths and predominantly of bismuth oxide in composition, with small amounts of cobalt and manganese. In the presence of the very strong peaks for ZnO and  $\text{Zn}_2\text{SnO}_4$  it was not possible to detect the bismuth rich phases by XRD.

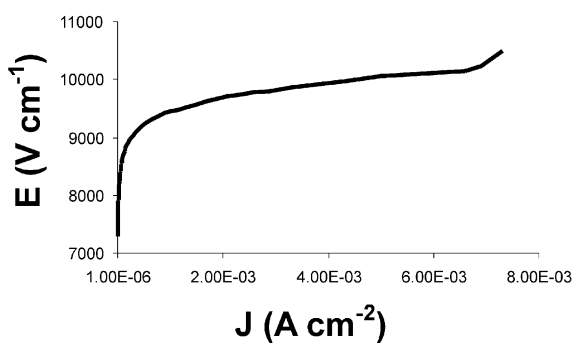


Fig. 3. Electric field current density ( $E$ - $J$ ) profile for sample prepared with 20%  $\text{SnO}_2$  sintered at 1275 °C.

Due to the high resistivity of the specimens, it was necessary to reduce specimen thickness below 1 mm in order to perform I-V measurements. A typical I-V profile is shown in Fig. 3. In some samples there was a suggestion that the I-V profile included part of the pre-breakdown region (as in Fig. 3), but most terminated in the non-linear region. Using Eq. (1) the non-linear coefficients ( $\alpha$ ) were determined; results are presented in Fig. 4. There is a clear dependence of  $\alpha$  values on sample composition and sintering temperature. With increasing  $\text{SnO}_2$  content, particularly above 15%, there is a small increase in  $\alpha$  value; this in part reflects the improvement in densification (Fig. 1). If this trend continues then significantly higher  $\alpha$  values would be expected in the  $\text{SnO}_2$  based varistors. Indeed for  $\text{SnO}_2$  varistors containing small amounts of  $\text{CoO}$ ,  $\text{Cr}_2\text{O}_3$ ,  $\text{Ta}_2\text{O}_5$ , non-linear coefficients in excess of 40 have been reported.<sup>8,9</sup> For the ZnO- $\text{SnO}_2$  samples in the present study, the non-linear coefficients decrease, almost linearly with increasing sintering temperature (reflecting the decrease in density). At 1200 °C the maximum  $\alpha$  values of 28–32 were achieved, with the highest value being that for samples prepared with 20%  $\text{SnO}_2$ . In the  $\text{SnO}_2$ -

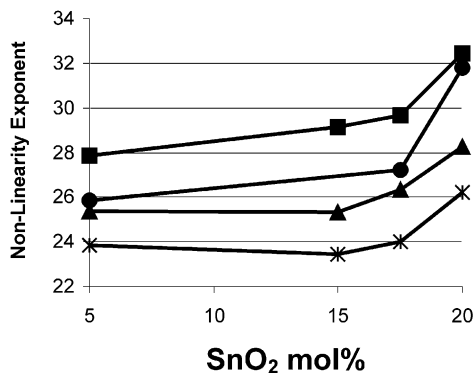


Fig. 4. Non-linear exponent as a function of SnO<sub>2</sub> content (and sintering temperature: ■ 1200 °C, ● 1225 °C, ▲ 1250 °C, \* 1275 °C).

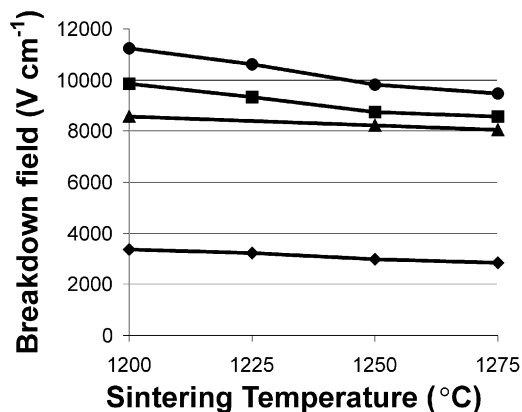


Fig. 5. Breakdown field as a function of sintering temperature for samples prepared with different amounts of SnO<sub>2</sub> (◆ 5%, ▲ 15%, ■ 17.5%, ● 20%).

doped ZnO varistors, Bernik and Daneu,<sup>10</sup> found that the low levels of tin oxide (up to 1 mol%) had comparatively little effect on microstructure, except for changes in spinel phase composition, in the traditional ZnO based formulation including Sb<sub>2</sub>O<sub>3</sub>; they were able to achieve  $\alpha$  values of 37–46. It would be interesting to explore the effect of Sb<sub>2</sub>O<sub>3</sub> and five valent oxide additions on the electrical properties of these ZnO–SnO<sub>2</sub> based ceramics.

Electrical breakdown fields are summarised in Fig. 5.  $E_b$  depends critically on composition but is less sensitive to sintering temperature. The breakdown field  $E_b$  is almost independent of sintering temperature, but increases almost linearly with SnO<sub>2</sub> content from 3800 to approximately 11 000 V/cm (in samples prepared with 20% SnO<sub>2</sub>). At the same time it was found that the leakage current  $I_L$  increases slightly with increasing sintering temperature, but decreases as the SnO<sub>2</sub> content increases. For example in samples sintered at 1250 °C,  $I_L$  decreases from 9  $\mu$ A at 15% SnO<sub>2</sub> to 4  $\mu$ A at 20% SnO<sub>2</sub>. These leakage currents are lower than those reported by Bernik and Daneu for SnO<sub>2</sub> doped ZnO varistors,<sup>10</sup> and the  $E_b$  values achieved here for the mixed ZnO–SnO<sub>2</sub> varistors are at least comparable with

those reported by Santos et al. for SnO<sub>2</sub> varistors.<sup>8,9</sup> Thus whilst the mixed ZnO–SnO<sub>2</sub> varistors exhibit slightly lower  $\alpha$  values than some competing formulations, they do have attractive breakdown fields and leakage currents and can be fabricated relatively easily.

On the basis of C–V measurements, Bernik and Daneu<sup>10</sup> deduced that substitution of Sb<sub>2</sub>O<sub>3</sub> by SnO<sub>2</sub> gives rise to a strong donor effect causing a slight increase in leakage current. Preliminary CV investigations of SBSN samples by Lee<sup>12</sup> indicate a similar mechanism when SnO<sub>2</sub> replaces ZnO.

#### 4. Conclusions

The densities of ZnO–SnO<sub>2</sub> based varistors increased with SnO<sub>2</sub> content, but decreased with increase in sintering temperature. The microstructures of all samples comprised ZnO primary grains, Zn<sub>2</sub>SnO<sub>4</sub> spinel and very small bismuth rich phases at the grain boundaries. As SnO<sub>2</sub> replaced ZnO, the size of the ZnO grains decreased and the spinel became the dominant phase. From I–V measurements, the non-linear coefficients were found to be in the range 24–32. The  $\alpha$  values were almost independent of SnO<sub>2</sub> content up to 15 mol%, but increased at higher levels of tin replacement. For all compositions the  $\alpha$  values decreased with increasing sintering temperature; the maximum  $\alpha$  value was achieved for samples prepared with 20 mol% SnO<sub>2</sub> sintered at 1200 °C. Breakdown fields were in the range 3000–11 000 V/cm and increased with increasing tin content. Leakage currents were in the range 1–11  $\mu$ A and tended to decrease with increasing SnO<sub>2</sub> content. Related studies suggest that tin substitution for Zn has a strong donor effect.

Thus replacement of Zn by Sn in ceramic varistors of model formulation has a dramatic effect on the microstructure with the development of extensive regions of Zn<sub>2</sub>SnO<sub>4</sub> spinel. Whilst the non-linear coefficients are not quite as high as those in the best ZnO varistors, the  $E_b$  values are uniformly high and leakage currents low, making them potentially attractive for further investigation and evaluation.

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