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Transparent cubic-ZrO₂ ceramics for application as optical lenses

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Abstract

Optical transparent polycrystalline ZrO_2 ceramics were fabricated by solid-state sintering process using first vacuum sintering followed by hot isostatic pressing. In the visible wavelength range (400–800 nm), the in-line transmittance of 5.6-mm thick samples reaches 68% at exemplary wavelength 600 nm (corresponding to an in-line absorbance based on 10 of $A_{10} = 0.08 \, \text{cm}^{-1}$), which is approximately 90% of theoretical limit. The refractive indices of the ZrO_2 optoceramics at 630 nm (n_d) are varying between 2.10 and 2.20, depending on TiO_2 contents, the latter being used as sintering aid. The appearance of birefringence is strongly correlated to the addition of TiO_2 as sintering additive in the ceramic samples, whereas addition of TiO_2 and simultaneous increase in Y_2O_3 content resulted in a decrease of birefringence.

Keywords: ZrO2; Transparent; Optoceramics; Optical properties; Optical applications

1. Introduction

One of the main application fields of specialty glasses are lenses for objectives used in consumer, industrial as well as military optical systems. Due to the high compositional flexibility of the material glass a huge variety of lens materials can be realized at low costs in huge sizes tailored individually to a specific optical design. SCHOTT glass fulfils the optical design need in a variety of refractive indices, Abbe numbers, and partial dispersions as well as in high transmission.

However, increasing requirements from both sophisticated industrial as well as consumer mass markets have asked for optical transparent materials with extraordinary property combinations. For example the continuous trend to miniaturization of digital photographic devices like digital still cameras require optical materials with very high refractive indices up to 2.0 or higher whereas industrial devices like optical microscopes require materials with special dispersion characteristics. Here glass tends to its limits due to the fact, that compositions are needed that are prone to uncontrolled crystallization during cooling of the glass melt.

In contrast, transparent polycrystalline ceramics can address extraordinary property areas that glass cannot reach reliably.^{1,2} By controlled sintering of suitable powders optoceramics of

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high transmittance and high property homogeneity can be manufactured.^{3–5} This has been first demonstrated in the field of active YAG-based ceramic laser materials.⁶ Furthermore, Japanese companies have developed high refracting transparent ceramics made from cubic perovskite in qualities principally enabling the assembly as lenses into digital photographic devices.⁷

To our best knowledge, the first relevant reports concerning transparent c-ZrO₂ were published by Tsukuma (TOSOH CORP.), who studied the potential to make yttria-stabilized zirconia doped with TiO₂.^{8,9} TiO₂ is a standard agent for making non-transparent ceramics enabling very dense sintered bodies.¹⁰ Transparent samples were demonstrated showing transmittance up to approximately 65% measured on a sample of only 0.76 mm thickness. There is no information available on the measurement technique used (in-line, integral, real in-line). More recent papers from TOSOH CORP. stresses effect of sintering temperature on the transparency of samples made from tetragonal stabilized zirconia (t-ZrO₂).¹¹ In addition hypotheses for unusual optical properties that these samples exhibit are given based on phase transformation of t-ZrO₂ to c-ZrO₂.

Clasen produced a c-ZrO₂ optoceramic via electrophoretic deposition with an in-line transmittance (including reflection losses) of 53% at $600\,\mathrm{nm}$ at a sample thickness of around 1 mm. $^{12-14}$

According to Klimke and Krell 14 zirconia samples sized in the order of $30 \text{ mm} \times 50 \text{ mm}$ have been manufactured via pressing, but also gel-casting has been tested. The samples were

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manufactured in thicknesses between 2 and 6 mm, real-in-line-transmission (detection of transmitted light within 0.5° aperture angle; see Ref. [15]) values of a 4-mm thick sample are reported to be 75% of theoretical value (means approximately 57% or $A_{10} = 0.32 \, \mathrm{cm}^{-1}$; data at 640 nm).

In the present paper both benefits and drawbacks of using transparent ceramics in optical lens systems are summarized. On the example of cubic stabilized zirconia, an alternative high refracting material system beside perovskites SCHOTT has been working on, the technological challenge to reach suitable qualities with increased thickness ready for bulk measurement of refractive index dispersion will be discussed.

2. Experimental

High purity cubic stabilized ZrO₂ powder (TZ-10YS, TOSOH CORP., Japan) was used as starting material. The powder was mixed with specially prepared sintering aids containing TiO₂ and binders and ball milled for 12 h in ethanol. Then, the alcohol solvent was removed by drying the milled slurry on a hot plate. The so-obtained powder was pressed with low pressure into required shapes in a metal mould and then cold isostatically pressed at 98 MPa.

Transparent ceramics made from cubic stabilized ZrO_2 (c- ZrO_2) were obtained after sintering under vacuum (1 × 10⁻³ Pa) at 1650 °C for 3 h followed by hot isostatic pressing at 1750 °C for 1 h at a pressure of 196 MPa. Because as-sintered samples were discolored into black due to the sintering in the reductive atmosphere, all the sample were heat-treated for oxidization in air at a temperature of around 1000 °C. The preparation procedure of ZrO_2 ceramic in this system is summarized in Fig. 1. The obtained optoceramic sample dimensions were 20 mm in diameter and 5–10 mm in thickness.

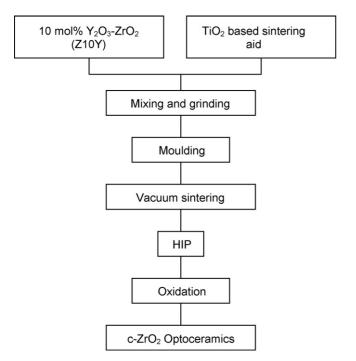


Fig. 1. Flow chart for preparation of c-ZrO₂ ceramics.

The refractive index $n(\lambda)$ was measured by the prism coupling method¹⁶ and V-Block measurement.¹⁷

The absorption spectra were measured in the wavelength range from 250 to 850 nm by means of a UV-vis-IR spectrophotometer using optically polished samples. The interaction of light with an optically transparent material is given by the addition of reflection, absorption, scattering and specular transmission (or in-line transmission after taking scattering into account). ¹⁸ The reflection losses are inherent to the material due to Snells law. An optimization of transmission is therefore only possible on scattering and absorption.

The total amount of light emerging from a material is termed as "total transmittance", while it's specularly transmitted portion is termed as in-line transmittance, after taking into account scattering as possible loss mechanism. Because inherent optical properties k are additive, and assuming that (1) small scattering and absorption $(k_{\rm abs} + k_{\rm scat})d \ll 1$, i.e. there is no diffusive light transport and (2) that light scattered once is not seen by the detector anymore, the absorption coefficient due to scatter can be calculated according to the following equation:

$$k_{\text{scatter}} = \frac{-\log(T_{\text{in-line}}/T_{\text{total}})}{d} = k_{\text{in-line}} - k_{\text{total}}$$
 (1)

At the band edge this analysis cannot be made anymore, as assumption (1) is not valid anymore.

The birefringence of the optoceramics can be seen by using crossed nichols in a polarizing reflected light optical microscope.

Other properties of the material were evaluated by XRD, TEM/EDX and Raman scattering.

3. Results and discussion

In order to measure refractive indices and dispersion data with high accuracy, samples are needed that have sufficient quality and size. To reach the quality a suitable sinter additive is necessary. Besides TiO₂, which is a well-known sinter aid for high performance technical c-ZrO₂ ceramics, ^{8,10} additional chemical species have been added and resulted in low pore contents in the final ceramic.

In Fig. 2 the in-line transmission of ZrO₂ optoceramics stabilized with 10 mol% Y_2O_3 (Z10Y) containing different amounts of TiO₂ (Z10Y–2 wt% TiO₂ and Z10Y–5 wt% TiO₂) is shown. Accordingly, the in-line transmission increased with increasing TiO₂ content, up to 5 wt% TiO₂ and reached a value of 51.6% at a wavelength of 600 nm for an 8.7-mm thick sample. It is obvious that the in-line transmission of the best sample of this study was drastically improved in comparison to those of the previously studies, $^{8,9,12-14}$ in which the in-line transmissions ranged from 1.2 to 40.3% after normalizing for the sample thickness of 8.7 mm. Nevertheless, in-line transmission of the 5 wt% TiO₂ sample at 600 nm was still not sufficient for optical applications.

Another issue is homogeneity of optical properties. In a cubic transparent ceramic, the optical properties should be isotropic in all directions. Nevertheless, sinter additives and/or phase stabilizers may segregate at grain boundaries and cause fluctuations in optical properties. In Fig. 3 the birefringence observations of

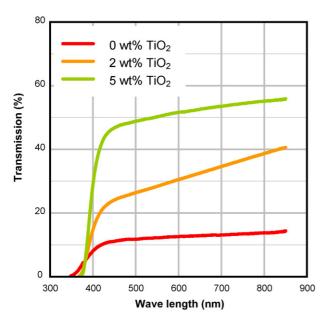


Fig. 2. In-line transmission of the samples of Z10Y-TiO₂ ceramics with different TiO₂ content. Transmission was measured on 6–9-mm thick specimen and normalized for the thickness of 8.3 mm.

the ceramic samples in the $ZrO_2-Y_2O_3-TiO_2$ system are presented. It can be observed that the appearance of birefringence is significantly correlated to the TiO_2 content in the ceramic samples. With lower TiO_2 birefringence decreases, however on cost on transmission. There is thus a competition of transmission versus homogeneity. This competition can be due to structural features and/or can be process related.

3.1. Structural features influencing transmission and homogeneity

In order to understand the structural origin of the birefringence of the $\rm TiO_2$ -containing c-ZrO₂, micro-Raman scattering of the $\rm ZrO_2$ ceramics has been carried out at several spots in the ceramic body. In Fig. 4 one exemplary spectrum out of 20 spectra carried out on different grains is shown and it is observed that the Raman spectra of the $\rm Z10Y-5$ wt% $\rm TiO_2$ ceramic is similar to the typical profile of $\rm TiO_2$ free c-ZrO₂ (Z10Y-0 wt% $\rm TiO_2$). Besides the Raman bands observed in $\rm Z10Y-0$ wt% $\rm TiO_2$, two supplementary bands are observed in $\rm Z10Y-5$ wt% $\rm TiO_2$. One appears at $\rm 700\,cm^{-1}$ which can be assigned to a $\rm Ti-O$ bond. Another relatively small band appears at $\rm 470\,cm^{-1}$.

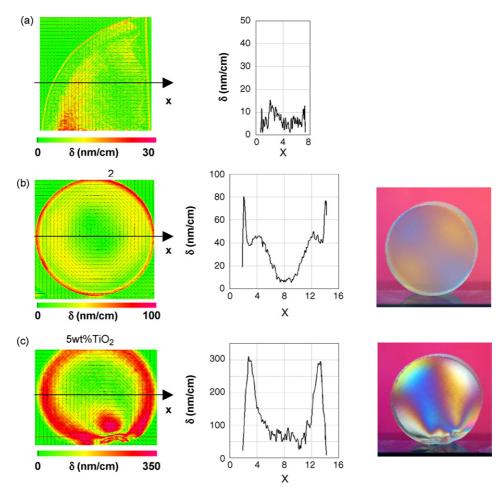
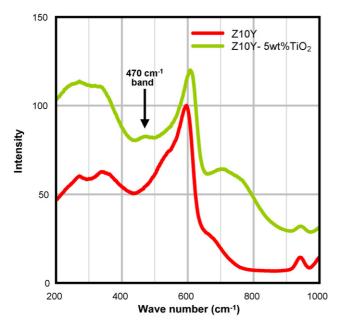


Fig. 3. Birefringence of the Z10Y-TiO₂ Ceramics. (a) Z10Y-0 wt% TiO₂, (b) Z10Y-2 wt% TiO₂ and (c) Z10Y-5 wt% TiO₂.





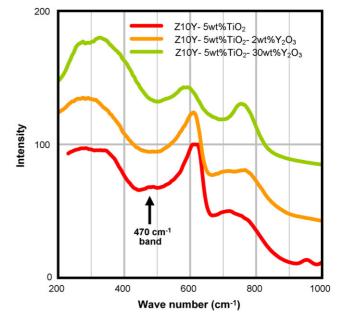


Fig. 6. Raman spectra of Z10Y-TiO₂ ceramics with increased Y₂O₃ content.

This observed band at 470 cm⁻¹ in Z10Y-5 wt% TiO₂ is considered to be associated with oxygen displacement in the fluorite-type related structure. Oxygen displacement is a characteristic feature observed in t-ZrO₂ ceramics, but it is well known that the t"-ZrO₂ phase which appears as metastable phase in the ZrO₂-Y₂O₃-system by rapid quenching from melt and following annealing, shows this 470 cm⁻¹ band, i.e. characteristic band for oxygen displacement.^{19,20} The unit cell of t"-ZrO₂ is cubic unlike for t-ZrO₂, however oxygen ions displace as well as oxygen ions of t-ZrO₂ displace along *c*-axis (Fig. 5).

As the t"-ZrO₂ phase appeared via conventional solid-state reaction in the TiO_2 -containing system, it may be relatively stable. This is in contrast to the t"-ZrO₂ phase in the ZrO_2 -Y₂O₃-system without TiO_2 . Thus, TiO_2 might act as tetragonal phase, i.e. stabilizing oxygen displacement as predicted in the ZrO_2 - TiO_2 phase diagram.⁶ This displacement of oxygen breaks the cubic symmetry and might cause the birefringence as shown in Fig. 3.

In order to evaluate the influence of yttria content on birefringence of TiO_2 -containing ZrO_2 ceramics, ZrO_2 ceramics with different amount of Y_2O_3 were prepared.

Raman spectra of the samples with higher Y_2O_3 content are shown in Fig. 6. The $470\,\mathrm{cm}^{-1}$, i.e. characteristic for the displacement of oxygen ions, disappeared by additional doping of Y_2O_3 regardless of the coexistence of TiO₂.

Probably due to the elimination of oxygen displacement, the birefringence of the TiO₂-containing ZrO₂ ceramics was drastically reduced as shown in Fig. 7.

The sample with additional $30 \, \text{wt}\% \, Y_2O_3$ (Z10Y–5 wt% TiO₂–30 wt% Y_2O_3 , no post-annealing) showed relatively high birefringence ($\sim 30 \, \text{nm/cm}$) at the rim edge due to the observed residual stress. However, the birefringence of the central region of the sample was 5 nm/cm or below which was comparable with the specification of typical optical glasses.

Fig. 8 shows the in-line and total transmission of the same sample Z10Y-5 wt% TiO_2 -30 wt% Y_2O_3 shown in Fig. 7. Although slight absorption at 500 to 550 nm probably due to

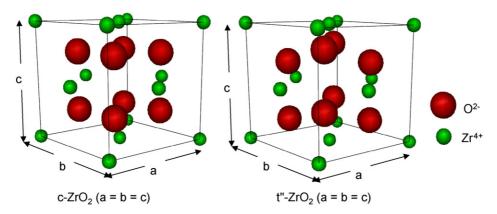


Fig. 5. Schematic structure of c-ZrO₂ and t"-ZrO₂.

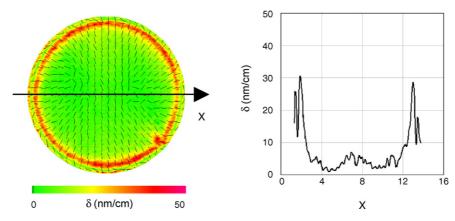


Fig. 7. 2D mapping and 1D profile of the birefringence, cross-Nichols image of Z10Y-5 wt% TiO₂-30 wt% Y₂O₃ ceramics.

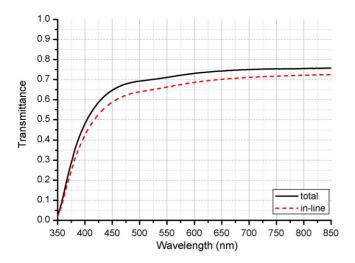


Fig. 8. In-line and total transmission of Z10Y–5 wt% TiO_2 –30 wt% Y_2O_3 ceramics. Transmission was measured on a 5.76-mm thick specimen.

impurity from the Y_2O_3 raw powder material, the in-line transmission was improved from ${\sim}60\%$ at 600 nm to ${\sim}69\%$ at a sample thickness of 5.8 mm by elimination of birefringence and optical scattering through an increased amount of Y_2O_3 inside the ZrO_2 ceramic.

In Fig. 9 a photograph of Z10Y–5 wt% TiO_2 –30 wt% Y_2O_3 is presented. The sample shows slight yellowish discoloration



Fig. 9. Picture of Z10Y–5 wt% TiO₂–30 wt% Y_2O_3 ceramics (the sample is \sim 1 cm spaced from the background).

caused by the TiO_2 addition. Considering the requirements for optical materials for lenses these discoloration needs to be removed, consequently, it is preferable to avoid addition of TiO_2 inside ZrO_2 optoceramics.

In Fig. 10 the scattering coefficient calculated from $k_{\text{in-line}} - k_{\text{total}}$ of the prepared samples is given. Keeping the amount of Y_2O_3 constant a decrease in scattering is observed

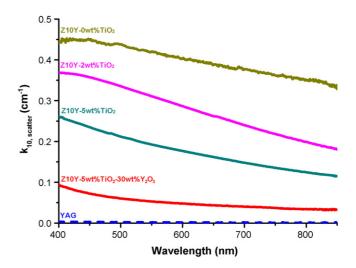


Fig. 10. Scattering coefficient calculated from $k_{\text{in-line}} - k_{\text{total}}$ of the prepared samples.

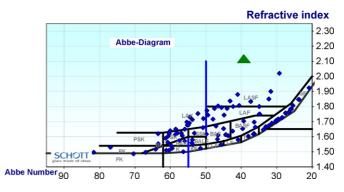


Fig. 11. Position of sample Z10Y–5 wt% TiO_2 –30 wt% Y_2O_3 in Abbe diagram (triangle).

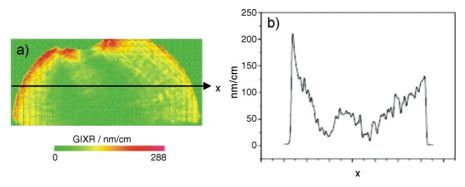


Fig. 12. 2D mapping of birefringence (a) and profile of birefrengence along x-axis (b) of Z78.

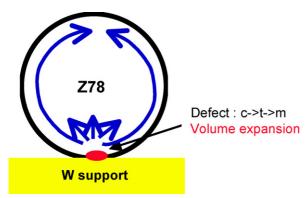


Fig. 13. Possible-induced compression stress by volume expansion at a defect.

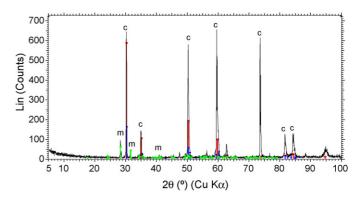


Fig. 14. XRD pattern from the defect on the round surface of Z10Y-5 wt% TiO_2 ceramics which was formed by a at the contact with W support during vacuum sintering. m- ZrO_2 (indicated by m) was detected as well as c- ZrO_2 (marked by c). (For interpretation of the references to color in citation of this figure, the reader is referred to the web version of the article.)

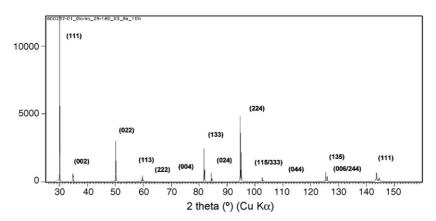


Fig. 15. XRD pattern of Z10Y-5 wt% TiO₂ ceramics (face not in contact with support).

Table 1 Mol fraction, refractive index at 633 nm, scattering as compared to Konoshima YAG and birefringence of prepared ZrO₂ optoceramics refractive indices have been measured with an accuracy of ± 0.0001 ($(_{\rm d} = (n_{\rm 588\,nm} - 1)/(n_{\rm 486\,nm} - n_{\rm 656\,nm})$

Sample	Mol fraction (mol%)			n at	$n_{\rm d}$ at	n at	Dispersion	Scatter (times	Birefringence level in
	ZrO_2	Y ₂ O ₃	TiO ₂	- 633 nm	588 nm	780 nm	$(v_{\rm d})$	YAG)	centre (nm/cm)
Z10Y–0 wt% TiO ₂	90.0	10.0	0.0	2.161	N.A.	N.A.	N.A.	~260×	5–10
Z10Y–2 wt% TiO ₂	87.0	9.7	3.3	2.177	N.A.	N.A.	N.A.	~185×	10-50
Z10Y-5 wt% TiO ₂	82.7	9.2	8.1	2.196	N.A.	N.A.	N.A.	~115×	50-100
Z10Y-5 wt% TiO ₂ -2 wt% Y ₂ O ₃ Z10Y-5 wt% TiO ₂ -30 wt% Y ₂ O ₃	81.6 64.2	10.2 26.6	8.2 9.2	N.A. 2.1090	N.A. 2.1176	N.A. 2.0940	N.A. 39 ± 1	N.A. $\sim 30 \times$	N.A. ∼5

while incorporating TiO₂ into the structure. This corresponds to the improvement of microstructure of the samples and confirmed that TiO₂ is a good sintering aid except for introducing inhomgeneity fluctuations which cause the above discussed birefringence. In the wavelength range between 500 and 800 nm the scatter of the best ZrO₂ optoceramic (Z10Y–5 wt% TiO_2 -30 wt% Y_2O_3) is around 30× higher than that of ceramic YAG from Konoshima Chemical Co. Ltd. (Japan) or that prepared by Ikesue.²¹ The former was qualified for having optical quality. Therefore, a further reduction of scattering by a factor of at least 10× is needed to be comparable to ceramic YAG (Fig. 10) will be needed to use these optoceramics as lenses and thus needing substantial development efforts in terms of process conditions and additives. Principally, there are still some fine macroscopic "flitters" or patterns observed in all samples by eye that seem to reflect light, i.e. decrease in transmittance and scatter sites.

The refractive index of the prepared ZrO_2 optoceramics varied between 2.15 and 2.20. By incorporation TiO_2 into the ceramics, the refractive index increased (Table 1) which is a well known effect in silicate glasses.²² Increasing the amount of Y_2O_3 while keeping the amount of TiO_2 constant resulted in a decrease in refractive index from 2.20 to 2.11,

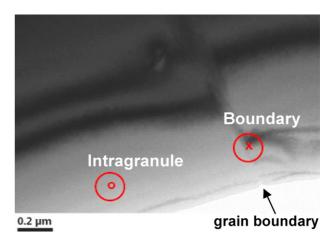


Fig. 16. TEM micrograph of Z10Y-5 wt% TiO₂ ceramics.

which has also been observed in Y_2O_3 -stabilized ZrO_2 single crystals.²³

Calculation of reliable dispersion data (Pg,F) requires extremely correct refractive index data (5th digit). Nevertheless, a tentative value of $\sim\!39\pm1$ for the best ZrO $_2$ optoceramic sample (Z10Y–5 wt% TiO $_2$ –30 wt% Y $_2$ O $_3$) has been calculated

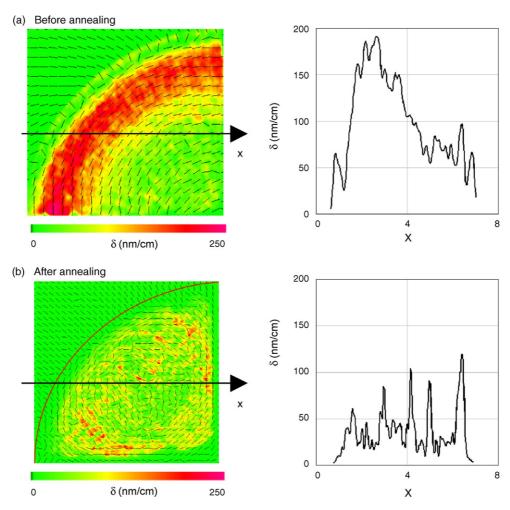


Fig. 17. Birefrengence of Z10Y-5 wt% TiO2 ceramics before (a) and after (b) post-annealing at 1600 °C for 3 h.

(Fig. 11). Further measurements are currently under way in order to get more accurate refractive index data.

3.2. Process related steps influencing transmission and homogeneity

During sintering in the vacuum furnace a whitish ring formed on the outside of the optoceramic samples doped with TiO_2 . A high intensity of stress birefringence can be observed at the outer rim of the ceramic sample (Fig. 12).

This inhomogeneous distribution of the birefringence starts from a defect at the outer RIM where monoclinic ZrO_2 (m- ZrO_2) could be detected. Thus, Y_2O_3 was extracted from c- ZrO_2 by reaction with the W-containing support plate during sintering and during cooling transformation to tetragonal ZrO_2 (t- ZrO_2) and then to m- ZrO_2 occurred.

Volume expansion coming along these phase transformations (c->t: $2.4 \, \text{vol}\%$, t->m: $7.4 \, \text{vol}\%$) will induce compression stress around the defect and the outer region of the Z10Y–5 wt% TiO₂ ceramic as shown in Fig. 13. In Fig. 14 the XRD pattern from the defect at the outer rim surface of Z10Y–5 wt% TiO₂ which is formed by reaction with the W support during vacuum sintering is shown. Monoclinic ZrO₂ (indicated by green marker) was detected together with c-ZrO₂ (marked by red).

XRD analysis revealed further that only the cubic ZrO₂ phase developed inside the sample after sintering of the Y-stabilized c-ZrO₂-containing 5 wt% TiO₂ (Z10Y–5 wt% TiO₂, Fig. 15). Microscopic fluctuation of Y₂O₃ and/or TiO₂ within the ceramic was also checked by energy dispersive X-ray spectrometer equipped on a TEM (Fig. 16). Although slight accumulation of TiO₂ at grain boundaries by around 1 mol% was detected, no significant deviation of composition generating tetragonal and/or monoclinic ZrO₂ and/or other TiO₂ and/or Y₂O₃ related compounds, which might cause birefringence, was detected. Nevertheless, besides the structural change, this local concentration of TiO₂ at grain boundaries could cause strong local fluctuation of refractive index and/or local strains enough to generate the observed scatter.

In conclusion, the actual observed birefringence in the $c\text{-}ZrO_2$ ceramics-containing TiO_2 is a superimposition of birefringence come from t"- ZrO_2 and also from mechanical/residual stress as described above.

In order to release the residual strain, post-annealing was performed at 1600 °C for 3 h in air using low heating and cooling rates. After post-annealing the high intensity of birefringence at the outer rim of the samples has been drastically reduced. It was confirmed that the birefringence due to residual stress, which was typically appearing at the rim edge of the specimen and whose intensity was higher than 150 nm/cm in the most cases, was clearly eliminated by the annealing as shown in 2D and 1D mappings of Fig. 17. The birefringence in the sample centre is still remaining. So the displacement of oxygen has not been reduced by the post-annealing. Unfortunately, the transmission was degradated by the post-annealing which could be due to bubbles, i.e. evaporation of Ar dissolved in the crystal lattice during HIPing.

4. Conclusions

Remarkable progress has been made in the field of transparent ceramics in the past years towards real optical quality. YAG ceramics are already available in qualities close to that of optical glass enabling application as laser host. Nevertheless, sophisticated optical imaging in the UV–vis requires even better "glass like" quality in terms of scattering and optical homogeneity. High refracting tunable materials with $n_{\rm d} > 2.0$ (zirconia, perovskites) are interesting for consumer optics. However, the successful and sustainable introduction of a new optical material into the optical industry requires identification and assessment of other/additional uniqueness of optoceramics in terms of their optical properties.

Transparent ceramics have been obtained in the $ZrO_2-Y_2O_3$ -system using TiO_2 as sintering agent. Optical properties have been evaluated and a variation of refractive index between 2.10 and 2.20 has been observed, depending upon addition of TiO_2 as sintering aid.

It has been noted that the addition of TiO₂ results in the appearance of a yellowish discoloration and of birefringence in the optoceramic samples. However, a decreased TiO₂ content gives negative impact on the transmission of the sintered body.

Oxygen displacement in c-ZrO₂ crystal structure induced by addition of TiO₂ and residual stress induced during sintering and post-oxidation process was suggested as the origins of the birefringence. Post-annealing was effective to relax the residual stress at the outer rim edge of the samples and thus reduced the birefringence induced by this stress. But through this process no reduction of birefringence in the sample centres could be obtained. This centre birefringence has been determined to be caused by oxygen displacement in c-ZrO₂ lattice. Nevertheless, a post-annealing treatment is not a practical measure to remove the birefringence, because at the same time transmission is degradated.

Increasing the amount of Y_2O_3 without changing the TiO_2 contents resulted in a decrease of birefringence in the sample centre, thus eliminating the oxygen displacement in the ZrO_2 ceramics-containing TiO_2 .

In order to improve the optical quality of the ZrO2 transparent ceramics needs substantial development efforts in terms of (a) avoidance of contamination of the sample through the support material of the sample in vacuum sintering furnace, i.e. change the support from W to ZrO₂ and thus avoiding possible chemical reaction of W with sample; (b) change heating elements in HIP (avoid effect of pollution effects); (c) improvement of the sintering, and post-sintering heat treatment processes which are related to the induction of residual stresses in the ceramic and use of different; (d) usage of different additives instead of TiO₂.

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